Rates and Equilibria for a Photoisomerizable Antagonist at the Acetylcholine Receptor of *Electrophorus* Electroplaques

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**ABSTRACT** Voltage-jump and light-flash experiments have been performed on isolated *Electrophorus* electroplaques exposed simultaneously to nicotinic agonists and to the photoisomerizable compound 2,2'-bis-[a-(trimethylammonium)methyl]-azobenzene (2BQ). Dose-response curves are shifted to the right in a nearly parallel fashion by 2BQ, which suggests competitive antagonism; dose-ratio analyses show apparent dissociation constants of 0.3 and 1 μM for the cis and trans isomers, respectively. Flash-induced trans → cis concentration jumps produce the expected decrease in agonist-induced conductance; the time constant is several tens of milliseconds. From the concentration dependence of these rates, we conclude that the association and dissociation rate constants for the cis-2BQ-receptor binding are \(-1 \times 10^8 \text{M}^{-1} \text{s}^{-1}\) and \(60 \text{s}^{-1}\) at 20°C; the Q10 is 3. Flash-induced cis → trans photoisomerizations produce molecular rearrangements of the ligand-receptor complex, but the resulting relaxations probably reflect the kinetics of buffered diffusion rather than of the interaction between trans-2BQ and the receptor. Antagonists seem to bind about an order of magnitude more slowly than agonists at nicotinic receptors.

**INTRODUCTION**

According to present concepts of competitive antagonism at nicotinic receptors, d-tubocurarine and related drugs interact directly with the agonist site; this occupation prevents the binding of agonists and the resulting activation of the channel. Most electrophysiological studies suggest that the antagonist-receptor interaction is characterized by a single dissociation constant of \(\sim 10^{-7} \text{M}\), although more recent binding studies reveal two separate binding reactions whose association constants flank this value (Neubig and Cohen, 1979; Sine and Taylor, 1979).

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This paper is concerned with the kinetics of the antagonist-receptor interaction.

Hill (1909) first studied the response to a sudden change of tubocurarine concentration in the solution bathing a frog muscle. Although he hoped to determine the molecular rates of the antagonist-receptor interaction, the measurements were limited by the drug’s diffusion into the muscle rather than by the molecular rates of interest. del Castillo and Katz (1957) measured the responses to brief iontophoretic pulses of both antagonist and agonist from twin pipettes onto the same synapse. In their experiments, tubocurarine’s inhibitory effect decayed over several seconds, much more slowly than expected from the diffusion of tubocurarine in solution. It was suggested that this slow recovery represented the molecular rate of dissociation of the tubocurarine-receptor complex. Less direct experiments have since suggested much faster rates (Blackman et al., 1975; Sheridan and Lester, 1977; Colquhoun et al., 1979), and it has been shown that the iontophoresis experiment was distorted by the phenomenon of buffered diffusion within the synaptic cleft: tubocurarine’s macroscopic rate of action is at least an order of magnitude slower than its molecular binding rates because its rate of diffusion is slowed by high-affinity binding to its receptors (Armstrong and Lester, 1979).

Other relaxation techniques have been of limited use for kinetic data on the antagonist-receptor interaction. The binding is not thought to be voltage-sensitive, so that voltage jumps couple only indirectly to this interaction and are instead dominated by the voltage-dependent processes of (a) channel gating and (b) open channel blockade by the antagonist (Manalis, 1977; Colquhoun et al., 1979). Single channel measurements would be complicated by (a) multiple closed states of the receptor and (b) channel gating produced by the antagonists themselves (Trautmann, 1982; Morris et al., 1983); this activation might not involve the same binding mechanisms as the competitive antagonism.

The light-flash technique (Lester and Chang, 1977; Lester and Nerbonne, 1982) does enable the experimenter to jump the concentration of ligand within the synaptic cleft. Although the prototypical competitive antagonist d-tubocurarine is not photosensitive, 2,2'-bis-[α-(trimethylammonium)methyl]azobenzene (2BQ) (Fig. 1) is photoisomerizable and displays competitive antagonism. The apparent dissociation constant is similar to that for d-tubocurarine (Lester et al., 1980). The present study was therefore undertaken to measure directly the rate constants for the antagonist-receptor interaction. The data have already been presented briefly in abstract form (Krouse et al., 1982, 1984) and as part of a Ph.D. dissertation (Krouse, 1984).

METHODS

Reagents
trans-2BQ was synthesized as described (Wassermann and Erlanger, 1981). The pure cis isomer was obtained by exposing a nearly saturated solution (2 mM) to ultraviolet (UV) light, followed by high-performance liquid chromatography (Nerbonne et al., 1983). Tetrodotoxin and the agonists (acetylcholine, carbachol, and suberyldicholine) were purchased from Sigma Chemical Co., St. Louis, MO.
Electrophysiology

In a typical experiment, 5–11 applications of agonist and numerous washes were completed on a single isolated electroplaque in an apparatus similar to that described by Sheridan and Lester (1977). Two preliminary agonist applications were performed to check the reproducibility of the agonist-induced currents and to provide a measure of the time of diffusion of agonist molecules from pool A (the solution bathing the innervated face) to receptors. During subsequent agonist applications, voltage-jump episodes were recorded after the estimated diffusion time had elapsed. After each agonist application, the solution in pool A was exchanged for Ringer three times over a period of 5 min. At the end of the wash cycles, agonist-induced currents had disappeared. For measurements of agonist-induced currents in the presence of 2BQ, pool A was preincubated with 2BQ alone for 5–10 min. After trials in agonist and 2BQ, pool A was washed several times with Ringer over 15 min. All experiments in the presence of 2BQ were preceded and followed by trials with agonist alone. Agonist-induced currents were measured by subtraction of the passive membrane responses; as detailed in previous publications, these passive responses could be isolated either in trials without agonist or, in the presence of agonist, by evoking voltage jumps in a region of membrane potential where the agonist-induced conductance is negligible (Sheridan and Lester, 1977).

Rate constants were measured for the relaxations in response to both voltage jumps and light flashes. Voltage-jump relaxations could usually be described with a single exponential, using a least-squares fit to the semilogarithmic plot of the approach to steady state (Figs. 9 and 10). Some voltage-jump relaxations, and most light-flash relaxations, required two exponentials; these were fit by the method of least squares.

Figure 1. Structure of 2BQ and optical absorption spectra of the cis and trans isomers. The absorption maximum of the trans isomer is at 327 nm ($\epsilon = 1.9 \times 10^4$ l/mol·cm$^{-1}$); an isosbestic point is observed at 287 nm ($\epsilon = 0.54 \times 10^4$ l/mol·cm$^{-1}$).
Photochemistry

The thermal cis-trans isomerization of 2BQ in solution is negligibly slow at or below room temperature. The two isomers interconvert in the presence of light; the mole fraction of each isomer in a solution depends on the wavelength of irradiation. This mole fraction is determined from the optical spectra of the solutions by linear interpolation between the $A_{281}/A_{281}$ ratios for the pure isomers (cf. Fig. 1). These concepts are used to measure the potency of a given flash for trans $\rightarrow$ cis and cis $\rightarrow$ trans photoisomerization (Nass et al., 1978; Lester and Nerbonne, 1982; Sheridan and Lester, 1982). Table I presents the photoisomerization potencies for the filters and flashes used with 2BQ in the experiments.

The optical arrangement was similar to that previously described (Nass et al., 1978). There were two improvements: (a) the xenon short-arc flash tube was placed at one focus of a polished ellipsoidal mirror rather than in a standard lamp housing; and (b) the energy storage bank had a capacitance of 2,100 J and was charged to 450 V. The flash intensity rose to a peak in <100 μs and fell with a time constant of 500 μs. This arrangement yielded a fivefold increase in the UV light intensity at the preparation, as compared with the previous apparatus (Nass et al., 1978).

Light from the flash-lamp was filtered to remove λ < 290 nm (WG 295, Schott Optical Glass, Inc., Duryea, PA) and focused on the preparation with a quartz lens (focal length, 75 mm; numerical aperture, 0.16) (Nargeot et al., 1982). Filters were placed in the light path between the ellipsoidal mirror and the focusing lens.

A few experiments required briefer flashes; these were delivered at 440 nm with the flash-lamp–pumped dye laser described by Sheridan and Lester (1982).

### Table I

<table>
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<th>Filter</th>
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<th>Isomerization potency $k_{\text{trans} \rightarrow \text{cis}}$</th>
<th>Isomerization potency $k_{\text{cis} \rightarrow \text{trans}}$</th>
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<tr>
<td>OG550 (orange)</td>
<td>&gt;550</td>
<td>&lt;0.01</td>
<td>&lt;0.01</td>
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**RESULTS**

**Equilibrium Measurements**

In Electrophorus electroplaques, steady state agonist-induced conductances are conveniently measured during hyperpolarizing voltage jumps (Sheridan and...
Lester, 1975, 1977). Our experiments employed the three agonists suberyldicholine (Fig. 2), carbachol, and acetylcholine. Results with agonist alone generally agreed with previous studies showing that steady state conductances increase with increasing agonist concentration and with more negative membrane potential (Adams, 1975; Neher and Sakmann, 1975; Sheridan and Lester, 1975, 1977). 2BQ induces no detectable conductance by itself, but it does block agonist-induced conductances.

Dose-ratio studies were performed in order to characterize the 2BQ-receptor interaction. In such experiments, one measures steady state conductances at several agonist concentrations in the presence of varying concentrations of 2BQ, and from these data abbreviated log-dose response curves are constructed (Fig. 3). Concentrations were selected to give agonist-induced conductances between

30 and 60% of the maximal response. The measure of interest is the ratio of agonist concentrations giving equal steady state conductance levels in the presence and absence of 2BQ. With 2BQ at concentrations of ≤4.0 μM, dose-response curves were shifted to higher agonist concentrations with no change in slope, and 2BQ did not decrease the maximum agonist-induced current at high agonist concentrations (Krouse, 1984). Under these conditions, 2BQ therefore behaves as a competitive antagonist. At concentrations of >4 μM, 2BQ did exert noncompetitive blockade; this action is discussed more fully below.

From the dependence of the dose ratio on 2BQ concentration, one obtains an apparent dissociation constant $K_i$ of 0.7 μM for the trans-photostationary state (Fig. 4). As found previously for d-tubocurarine (Lester et al., 1975), this value was the same for all voltages in the range tested (−60 to −150 mV), which implies
that the apparent affinity of 2BQ for the receptor is voltage-independent. It should be noted that the assumptions of the dose-ratio method may be inappropriate for nicotinic acetylcholine receptors. The Discussion contains a more detailed evaluation of the shape of the dose-ratio plot.

The most interesting result of the dose-ratio studies was of course the variation in inhibition with the isomeric composition of the 2BQ solution. One would expect the apparent affinity constant $1/K_i$ to depend linearly on the mole fraction

\[ \text{Dose Ratio} = \frac{1}{K_i} \times \text{isomeric composition} \]
of each isomer; therefore, it is more revealing to plot this parameter than $K_i$ (Fig. 5). The pure cis isomer has an equilibrium binding constant of 0.33 µM for the receptor; the pure trans isomer binds one-third as strongly.

**Kinetic Measurements: Voltage-Jump Relaxations**

Although the voltage-jump experiments described above were designed primarily to give information about equilibrium binding properties, they also gave preliminary kinetic information. In the experiment of Fig. 2, for instance, the presence of 2BQ caused a reduction of agonist-induced conductance, but there was little or no change in the rate constant of the voltage-jump relaxation. This suggests that the blockade re-equilibrates more slowly than the time scale of the voltage-jump relaxation. We therefore expected that more direct experiments, based on flash-induced concentration jumps of 2BQ, would reveal a distinct component caused by the 2BQ-receptor interaction on a time scale slower than the agonist-receptor interaction. This expectation was confirmed.

**Concentration-Jump Relaxations**

**UV flashes** In solutions containing mostly trans-2BQ, UV light increases the mole fraction of cis in solution. This "antagonist concentration jump" leads to a net decrease in the conductance and provides the clearest data on the kinetics of the interaction between cis-2BQ and the receptor.

The typical UV-flash relaxation (Fig. 6) consists of two components. The faster component, termed $I_{v_f}$, is either a delay (period of zero slope) or a small increase in agonist-induced current; it is best seen under different conditions and will therefore be described later. The slower component, $I_{v_a}$, is the decrease in agonist-induced current expected to accompany an increase in the antagonist concentration. The point of greatest interest is that this decrease occurred on a
time scale roughly one order of magnitude slower than the voltage-jump relaxations thought to reflect primarily the agonist-receptor interaction.

Our experiments were guided by the idea that component $2_{i-f}$ is dominated by the molecular kinetics of the interaction between cis-2BQ and the nicotinic receptor. Several detailed models are presented in the Discussion; for present purposes, we employ the simple equation

$$cis-2BQ + R \xrightarrow{k_+} cis-2BQ \cdot R.$$ (1)

**Figure 6.** A UV-flash relaxation during a voltage-clamp trial. At the start of the trial, 2BQ is present in the trans-photostationary state (2 $\mu$M); the agonist is carbachol (500 $\mu$M). The trial consists of three episodes at intervals of 3 s. Panel A shows agonist-induced currents; passive and capacitive currents have been subtracted. 50 ms after the start of each episode, the voltage was jumped from +60 to -150 mV. As in the experiment of Fig. 2, the agonist-induced current approached a new steady state value along an approximately exponential time course. The rate constants for these voltage-jump relaxations are 150–200 s$^{-1}$. Because the time scale is five times longer than that of Fig. 2, the traces also show a slow decrease in agonist-induced current, presumably caused by open channel block by 2BQ (see text). The flash was delivered at the arrow in the second episode, jumping the cis-2BQ concentration from 0.4 to 0.98 $\mu$M. After an initial delay of a few milliseconds (see text), the conductance decreased with a rate constant of $\approx$ 25 s$^{-1}$. Panel B presents the difference between traces 2 and 3 at higher gain to show the UV-flash relaxation. Temperature, 6°C. Experiment 1122T24.

For Scheme 1, the relaxation rate constant for component $2_{i-f}$ is expected to equal the sum $k_- + fk_+[cis-2BQ]$. The dissociation rate constant $k_-$ would dominate at low 2BQ concentrations; at higher concentrations, the relaxation rate constant would be determined by the forward binding rate $fk_+[cis-2BQ]$. The factor $f$, which is between 0 and 1, decreases with agonist concentration and reflects the more rapidly equilibrating occupancy of binding sites by the agonist;
this occupancy effectively reduces the probability of forming the cis-2BQ-receptor complex. The most important evidence for the view that Scheme 1 dominates the kinetics is the behavior of component \(2_{f,r}\) as a function of drug concentration (Fig. 7). As expected, (a) \(1/r\) for component \(2_{f,r}\) increases with 2BQ concentration, and (b) this increase is less steep with increased agonist concentration.

![Relaxation rate constants vs. [cis-2BQ]](image)

**Figure 7.** Relaxation rate constants vs. [cis-2BQ]. Data are from experiments like that of Fig. 6, at various concentrations of 2BQ and carbachol. The lines are least-squares fits to the data. (A) 6°C. Data at 150 (□) and 250 (■) μM carbachol are from the same cell. △, 400 μM carbachol. (B) 20°C. ○, 150 μM carbachol; ●, 400 μM.

The data of Fig. 7 thus allow preliminary estimates of the rate constants for association and dissociation of cis-2BQ. From the zero-concentration intercepts, one obtains dissociation rate constants \(k_\text{-}\) of 10.7 ± 0.8 s\(^{-1}\) (mean ± SD, 6 cells) at 6°C and 60.4 ± 13.9 s\(^{-1}\) (17 cells) at 20°C. These intercepts did not change with voltage over the range −90 to −150 mV. From the slope of the steepest line at each temperature, one can conclude that the forward binding rates are at...
least $2 \times 10^7 \text{M}^{-1} \text{s}^{-1}$ at 6°C and at least $7 \times 10^7 \text{M}^{-1} \text{s}^{-1}$ at 20°C. The Discussion contains a more complete treatment of the factor $f$, allowing for better estimates of the forward binding rate constants.

**ALTERNATIVE EXPLANATIONS** Because our strategy relies so heavily on the interpretation of component $2_{c\cdot f}$, it is well to emphasize how the observations rule out alternative explanations. For instance, is component $2_{c\cdot f}$ dominated by buffered diffusion? The UV flash increases the cis-2BQ concentration both within the synaptic cleft and in the external solution. However, the synaptic cleft has such a high receptor concentration that equilibrium inhibition might be established only after cis-2BQ molecules diffuse into the cleft from the external solution. This diffusion would be slowed by repeated antagonist binding (Armstrong and Lester, 1979). Two observations argue against buffered diffusion in our experiments. (a) Buffering would be reduced by decreasing the number of binding sites for 2BQ (Armstrong and Lester, 1979), for instance by increasing the agonist concentration. However, the relaxation rate constants decreased with increasing agonist concentration (Fig. 7), contrary to the expectations of the buffering model. (b) Rather low temperature dependences ($Q_{10} \sim 1.3$) are indicative of buffered diffusion (Armstrong and Lester, 1979), but our most reliable kinetic data—the zero-concentration intercept of plots like those of Fig. 7—show a $Q_{10}$ of \sim3 over the range 6–30°C. A $Q_{10}$ of 3 is within the range expected for ligand-protein interactions. Thus, buffered diffusion does not seem to dominate component $2_{c\cdot f}$.

Are the relaxations dominated by the interaction of trans-2BQ and the receptor? In most of our experiments (e.g., that of Fig. 7), the cis-2BQ concentration after a single flash was varied by bathing the cell in varying initial concentrations of TPSS-2BQ. Other experiments were simply conducted at a fixed TPSS-2BQ concentration; relaxations were measured as a series of UV flashes converted the solution to the CPSS (Krouse, 1984). The rate constants increased during the series, contrary to the expectation for a model that is similar to Scheme 1 but in which kinetics are dominated by trans-2BQ.

**Molecular Rearrangements**

In addition to concentration-jump experiments, the light-flash technique makes possible a unique class of experiments in which ligand molecules already bound to receptors are manipulated photochemically to produce a molecular rearrangement of the ligand-receptor complex (Nass et al., 1978; Lester and Nerbonne, 1982). Because of this direct action, such perturbations generally produce more rapid responses than do concentration jumps. The response to a molecular rearrangement does not depend on the direction or amplitude of the bulk concentration changes, although later events may reflect these changes.

We have no direct information about the photoisomerization efficiency of bound 2BQ as opposed to 2BQ in solution. Spectral measurements with a covalently bound azobenzene agonist suggest that such differences are minor (Sheridan and Lester, 1982); however, in the absence of relevant data for 2BQ, we shall describe the amplitudes of these relaxations in a qualitative fashion.

**EXPERIMENTS AT THE CIS-PHOTOSTATIONARY STATE** With predominantly
FIGURE 8. Visible-flash relaxations at 6°C and -150 mV in the simultaneous presence of suberyldicholine (3 μM) and 2BQ (1 μM). At the beginning of the trial, the 2BQ was in the cis-photostationary state. A visible flash (during the second episode) jumped the cis-2BQ concentration from 0.85 to 0.23 μM. The resulting conductance increase is dominated by a rapid relaxation called component 1. Semilog plots are shown at the top; the relaxation rate constants are nearly equal for the voltage-jump and light-flash relaxations. Experiment 682IT80.

cis-2BQ solutions and unfiltered flashes, the molecular rearrangements were expected to arise predominantly from cis → trans photoisomerizations of bound cis-2BQ molecules, because (a) this isomer binds more strongly to receptors, (b) it was present in higher concentrations, and (c) cis → trans flux predominates in unfiltered flashes (see Table I). We therefore expected that these experiments would reveal the kinetics of the trans-2BQ-receptor interaction.

Such flashes did produce increases in the conductance as the newly created trans-2BQ molecules left the receptors. When acetylcholine and suberyldicholine were the agonists, the increase occurred with a rate constant that was indistin-

FIGURE 9. Laser-flash relaxations at 20°C in the simultaneous presence of carbachol (200 μM) and 2BQ (1 μM). The protocol is the same as in Fig. 8. The laser flash jumps the cis-2BQ concentration from 0.85 to 0.20 μM. Note that the voltage-jump relaxations are faster than the light-flash relaxation. See text. Experiment 617T70.
guishable (±10%) from that of the voltage jump in the next episode (Fig. 8). At temperatures of 20°C or higher with carbachol as the agonist, the agonist-receptor interaction was characterized by rate constants greater than ~2 ms⁻¹. Under these conditions, the flash-lamp discharge could no longer be considered instantaneous on the time scale of the relaxations. Therefore, the light-flash relaxations were evoked with a pulsed laser (Fig. 9). Under these conditions, the light-flash relaxation was clearly slower than the voltage-jump relaxation, which suggests that the dissociation of trans-2BQ had become rate-limiting.

![Diagram of light-flash relaxations](image)

**Figure 10.** Agonist-induced currents from episodes in the simultaneous presence of carbachol (1 mM) and 2BQ (2 μM). At the start of each episode, the 2BQ was in the trans-photostationary state ([cis-2BQ] = 0.4 μM); the relaxations were superimposed on a steady agonist-induced current of 5.1 mA/cm². The flash was filtered differently for each episode. See text and Table I. Voltage, -150 mV; temperature, 6°C. Experiment 1122T64.

### EXPERIMENTS AT THE TRANS-PHOTOSTATIONARY STATE

In cells exposed simultaneously to TPSS-2BQ and to agonist, the kinetics depend strongly on how the flash is filtered (Fig. 10). As the cis → trans photoisomerization potency increases (cf. Table I), the rapid conductance increase, component 1, is larger. This result is expected because (a) roughly half of the inhibition is produced by the 20% of cis-2BQ in TPSS-2BQ, and (b) a flash converts the bound cis-2BQ to trans-2BQ. Fig. 11 explains how the subsequent slower conductance decrease,

<table>
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<th>[cis-2BQ] after flash</th>
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<tr>
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<td>b</td>
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<tr>
<td>OG550 (orange)</td>
<td>0.40</td>
<td>—</td>
<td>d</td>
</tr>
<tr>
<td>None (visible)</td>
<td>0.40</td>
<td>6.1</td>
<td>e</td>
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component 2$_{t-f}$ is also expected as equilibrium is re-established. An unexpected finding in the experiment of Fig. 10 is that component 2$_{t-f}$ also becomes slower as the cis $\rightarrow$ trans photoisomerization potency increases. Although traces $a$ and $b$ have nearly the same initial and final state, the decrease in trace $b$ occurs with a rate constant roughly half as large as that of trace $a$. With unfiltered flashes (trace $e$), the relaxation is slowed even further, to a rate constant less than one-third that of the UV flash. We do not have a full explanation for this slowing; possibilities are presented in the Discussion.

**DISCUSSION**

This paper completes our investigation of acetylcholine receptors in *Electrophorus* electroplaques with photoisomerizable azobenzene derivatives. Previous studies dealt with agonists and with open channel blockers (Lester and Nerbonne, 1982;
Sheridan and Lester, 1982; Nerbonne et al., 1983); the present data describe inhibition by a competitive antagonist. Like other members of the series, the 2BQ molecule has the important property that the cis and trans isomers differ in pharmacological potency. We have exploited this property to study the kinetics of the interaction between 2BQ and the receptor; these kinetics have been interpreted in terms of the molecular rate constants for the association and dissociation events.

The experiments are somewhat limited by the fact that the two isomers differ by only a factor of 3 in their apparent affinities for the receptor; therefore, a concentration jump from zero antagonist activity could not be produced. Also, it is not possible to obtain reliable data at either very large or very small concentrations of the agonist or antagonist, because these conditions involve only small changes of the conductance. Nonetheless, it is possible to give a reasonably consistent kinetic and equilibrium description of the interaction between cis-2BQ and the receptor.

2BQ as a Competitive Antagonist

The dose-response data, i.e., the parallel shift in the dose-response curves, are consistent with competitive antagonism as the major action of 2BQ under the conditions of our studies (concentrations of \( \leq 4 \) \( \mu \)M and voltages between -60 and -150 mV). As usual with such studies, one needs to exclude the possibility that 2BQ is only a noncompetitive antagonist, with a large excess of “spare receptors.” We performed a few experiments to show that maximal responses could in fact be obtained with carbachol even in the presence of 1 \( \mu \)M 2BQ (Krouse, 1984); more complete studies done previously under the same conditions (Sheridan and Lester, 1977; Lester et al., 1978) showed that the test concentrations of carbachol used were indeed high enough to reveal any non-competitive blockade.

In addition, 2BQ does exhibit a form of noncompetitive antagonism, open channel blockade. However, this complication was not serious, because the rate of blockade was slow under our conditions and this effect could be subtracted from the measured agonist-induced currents. If the local anesthetic effect were significant, light flashes would cause photodissociation of bound 2BQ molecules from their binding sites within channels; this would lead to increases in the current on a submillisecond time scale. Such step increases have been observed with the light-sensitive acetylcholine channel blocker EW-1 (Lester et al., 1979; Krouse, M. E., unpublished observations). However, no such step was observed in our experiments, even at the highest 2BQ concentration (5 \( \mu \)M). It can thus be concluded that the light-flash relaxations were not distorted by the effects of open channel blockade. Experiments like that of Fig. 6 suggest, however, that it would be possible to study this open channel blockade directly by delivering flashes several seconds after a step to a hyperpolarized voltage.

MODELS FOR THE BINDING REACTIONS

The equilibrium binding constant \( K_i \) for the 2BQ-receptor interaction was calculated from dose-ratio studies. This method was originally formulated for a case where (a) the receptor R can be
completely inhibited by the binding of a single antagonist molecule I, and (b) agonist molecules A bind noncooperatively (see, for instance, Jenkinson, 1960). However, it is well known that the nicotinic receptor binds two or more agonist molecules per channel; presumably, each agonist site can bind antagonist molecules as well. A more general scheme, then, would be written:

\[
2I + R + 2A \xrightleftharpoons{k_{+1}} I + AR + A \xrightleftharpoons{k_{+2}} \frac{\beta}{\alpha} A_2R \xrightarrow{k_{+3}} A_2R^* \\
I + R + 2A \xrightarrow{k_{+4}} \frac{k_{-4}}{k_{-5}} \xrightarrow{k_{-6}} \text{AIR} \\
1_2R
\]

\(A_2R^*\) state has the open channel. Each of the three possible antagonist binding reactions would influence the experimentally measured inhibition. If all three antagonist binding reactions have the same equilibrium binding constant \(K_i\), the dose ratio is simply

\[D = 1 + [I]/K_i;\]

i.e., the dose-ratio plots are linear with a slope equal to \(1/K_i\).

For other combinations of dissociation constants, the dose-response curves are no longer shifted in a parallel fashion by the antagonist (cf. Krouse, 1984); the complete expression is

\[D = \frac{[A]/K_1(1 + [I]/K_3) + ([A]/K_1)^2(1 + [I]/K_3)^2 + (2[A]/K_1 + 1)(1 + 2[I]/K_4 + [I]^2/K_4K_6)^{1/2}}{2[A]/K_1 + 1}.\]

Even if one confines the measurements to a particular response amplitude, the dose-ratio plots are nonlinear. The plot is upwardly concave (i.e., has a sigmoid start) if the affinity constant increases for the second antagonist \((1/K_6)\); the concavity has the opposite sense if the second binding is weaker than the first. However, these nonlinearities decrease as the level of activation increases and they would be quite small for our experimental results, which were taken near the half-maximal response (Colquhoun, 1973). The relationship between antagonist binding sites and the shape of the dose-ratio curve has also been addressed by Pennefather and Quastel (1981). The departures from linearity in our equilibrium data do suggest a slight upward concavity (cf. Fig. 4), so that one would suspect a tighter second binding. This observation does not agree with the suggestion of Sine and Taylor (1981) that there are pre-existing sites of high and low affinity for antagonists on BC3-H1 cells. However, our data are insufficiently precise for a more detailed comparison on this topic. The important point can be summarized as follows: the \(K_i\) estimated from the dose-ratio plot is
determined by the smaller of the two equilibrium dissociation constants, $K_4$ or $K_5$.

**Kinetic Constants for the Interaction Between cis-2BQ and the Receptor**

The light-flash relaxations are attributed to the binding and dissociation of 2BQ. We now consider the molecular interpretation for such data in terms of Scheme 2. Fig. 12 shows simulated light-flash relaxations using the protocol of Fig. 6 (a UV flash producing a relaxation from a low to a high level of inhibition). The relaxations are approximately exponential and they are dominated by the antagonist-receptor binding step with the largest rate constants. It can therefore be stated that the relaxation rate constants of Fig. 7 do refer to the faster step; however, the data do not allow for a choice as to which step ($k_{-4}$ or $k_{-5}$) is the faster.

**DISSOCIATION RATE** The theoretical concentration dependence of these rate constants is shown in Fig. 13A for the simplest possible case, i.e., all three antagonist-receptor binding constants are equal. The calculations confirm that the zero-concentration intercept is the rate constant for dissociation of the antagonist-receptor complex. The plots have a sigmoid start, however, so that extrapolations from the linear portion would give underestimates of the true dissociation rate constant. Therefore, the estimated dissociation rate constants for cis-2BQ ($60 \text{ s}^{-1}$ at $20^\circ\text{C}$ and $11 \text{ s}^{-1}$ at $6^\circ\text{C}$) may be too small by 10–20%.

**ASSOCIATION RATE** The model relaxations also predict the variation in relaxation rate constants with agonist concentration. Here the point of interest
FIGURE 13. (A) Calculated relation between drug concentrations and the rate constant of light-flash relaxations. For each agonist concentration, 10 relaxations like those of Fig. 11 were simulated at varying antagonist concentrations [1]. The plots should be compared with the experimental data in Fig. 7. The calculations assume that the three antagonist-receptor interactions have equal rate and equilibrium constants ($k_4 = k_5 = k_6 = 5 \times 10^7 \text{M}^{-1} \text{s}^{-1}$; $k_{-4} = k_{-5} = k_{-6} = 50/\text{s}$). The other assumed parameters equal those of Fig. 11. (B) Data are from experiments, including that of Fig. 7B, at 20°C on four cells from a single fish. The straight line is a least-squares fit, weighted for the errors at each point; the intercept at zero agonist concentration is $2 \times 10^8 \text{M}^{-1} \text{s}^{-1}$. See text for discussion.

is the factor $f$, the fractional forward binding rate that results from the rapidly equilibrating occupation of the antagonist binding site by the agonist. The kinetic simulations summarized in Fig. 13 reveal that $f$ can be approximated well by a simple hyperbolic relation, $f = 1/(1 + [A]/K')$, where $K'$ depends on $k_{\pm 1}, k_{\pm 2}, \alpha$,
and $\beta$. (An explicit formula is not available for $K'$, which does not in general equal the agonist concentration giving a half-maximal response.) Therefore, simple reciprocal plots were used to extrapolate $f$ to unity at zero agonist concentration; an example of such a plot suggests that the largest measured slopes, in experiments like those of Fig. 7, are a factor of 1.5–2 less than the actual forward binding rate constant, $k_{+4} = k_{+5} = k_{+6}$ (Fig. 13B). (This estimate includes a correction for the fact that the extrapolated value is actually $1/2k_{+4}$, because we assume two equivalent binding sites.) We conclude that $k_{+4}$ is $\sim 1 \times 10^8 \text{M}^{-1} \text{s}^{-1}$ at 20°C and $\sim 0.2 \times 10^8 \text{M}^{-1} \text{s}^{-1}$ at 6°C.

**Comparison with Equilibrium Measurements** With these estimates of the binding rate constants for cis-2BQ, we can calculate the ratio $k_{-}/k_{+}$ to obtain an equilibrium value of 0.6 μM. This estimate should be considered correct only to within a factor of 3 because of (a) uncertainties about the actual molecular model, (b) errors in the extrapolation to zero agonist concentration, and (c) scatter in the data themselves. Therefore, the calculated equilibrium value, based on the kinetic measurements, certainly shows satisfactory agreement with the equilibrium measurements ($K_i = 0.3$ μM).

**Rates of Drug-Receptor Binding**

The present data encourage us to make a few general comments about the rate constants for ligand bindings at the nicotinic acetylcholine receptor. These rates have been investigated over the past decade with voltage-jump and concentration-jump relaxations, as well as with fluorescence and flow-quench methods. There are still some uncertainties: the biochemical studies might measure binding to the desensitized rather than to the active state (Heidmann and Changeux, 1979; Boyd and Cohen, 1980; Prinz and Maelicke, 1984), and the electrophysiological studies could be measuring a rate-limiting conformational change rather than a binding (Sheridan and Lester, 1977; Lester et al., 1980; Ogden and Colquhoun, 1983). Nonetheless, there is now general agreement that the agonist-receptor binding proceeds with a forward rate constant of $10^8$–$10^9 \text{M}^{-1} \text{s}^{-1}$ at 20°C, with the bis-quaternary compounds falling in the high end of this range. The present data show that antagonist-receptor binding proceeds with rates about an order of magnitude smaller. The most direct comparisons involve the concentration-jump experiments using the photoisomerizable bis-quaternary azobenzene derivatives. The agonist Bis-Q binds with rates of $\sim 10^8 \text{M}^{-1} \text{s}^{-1}$ at 20°C (Sheridan and Lester, 1977; Lester et al., 1980) vs. the value of $10^8 \text{M}^{-1} \text{s}^{-1}$ determined for 2BQ in the present study. These differences in binding rate are probably small enough to agree with the generally held view that (a) antagonists and agonists have similar initial steric and coulombic interactions with the receptor’s binding site, but (b) only the latter allow the subsequent conformational changes that open the channel.

There are exceptions to this pleasing concordance. First, in experiments analogous to that of Fig. 2, d-tubocurarine decreased both the steady state activation and the relaxation rate constant for voltage-jump relaxations (Sheridan and Lester, 1977; Colquhoun and Sheridan, 1982). The usual interpretation of the d-tubocurarine result is that it equilibrates rapidly with receptors (severalfold
more rapidly than the agonist itself). This admittedly indirect evidence leads to the conclusion that d-tubocurarine binds with rates that push the limits of physical possibility. We now think that this apparent anomaly can be explained by a special combination of equilibrium and kinetic properties (Krouse, M. E., and H. A. Lester, manuscript in preparation). Second, the so-called "open channel blockers" seem to bind one or two orders of magnitude more slowly, even when this binding is accelerated by the membrane field (Adams, 1976; Neher and Steinbach, 1978). Presumably, the binding site is different from the one under consideration here—it may be directly within the open channel.

**What Are the Rates for trans-2BQ?**

In contrast to the consistent picture obtained for cis-2BQ, our data do not allow clear conclusions about the kinetics of the interaction between cis-2BQ and the receptor. One set of experiments involved component I_c, the rapid increase observed after cis → trans photoisomerization of bound 2BQ molecules. To obtain relaxations that were dominated by the dissociation of trans-2BQ, it was necessary to employ the most rapidly equilibrating agonist, carbachol, and to use laser flashes (Fig. 9). From these data, one might infer that the interaction between trans-2BQ and the receptor proceeds with characteristic times on the order of 1 ms. However, another set of experiments measured the time required for the relaxation back to equilibrium after unfiltered flashes at the trans-photostationary state (Fig. 10). From these data, one might infer that hundreds of milliseconds are involved. We suspect that both inferences are misleading and that the actual rates lie between these two extremes. Component I_c probably does not directly reflect the lifetime of newly created trans-2BQ on the receptor because (a) the trans-2BQ-receptor complex has received a large amount of energy from the photon and (b) the ligand may be bound in a different configuration (or range of configurations) than if it had entered normally from the free solution. Thus, the kinetics of this relaxation may not reflect the same molecular events as the equilibrium studies with trans-2BQ.

We have considered two possible mechanisms for the long relaxations of Fig. 10. (a) In the trans-photostationary solution, the proportion of trans is ~80%. Although the cis isomer exhibits a threefold-greater binding constant, trans-2BQ-receptor complexes are slightly more numerous. The relaxations are thus slowed under those conditions where the trans-2BQ-receptor interaction would be expected to dominate the kinetics. To examine this possibility, we extended our kinetic model (Scheme 2) to include trans-2BQ-receptor interactions at the same binding sites that bind agonist and cis-2BQ. We were at first surprised that the model could not reproduce the observed 3.5-fold slowing, even with the many free parameters offered by nine states and even if the kinetics for the trans isomer were 30-fold slower than for the cis isomer. We then realized that the decrease in conductance is still dominated by the fastest inhibition, that caused by the cis isomer. Thus, the slow relaxations are not explained by a slow trans-2BQ-receptor interaction.

(b) Buffered diffusion could play a role in the slowing. The unfiltered flashes, with a large cis → trans flux, markedly deplete the synaptic cleft of cis-2BQ, most
of it previously bound to receptors. In order to re-establish equilibrium, cis-2BQ must therefore diffuse in from the external solution (steps e and f in Fig. 11), and this diffusion would be slowed by multiple binding.

Rough calculations suggest that the buffering model can easily account for the slow relaxations. If one begins with the data of Armstrong and Lester (1979, Fig. 8A) and corrects for the slightly higher affinity of cis-2BQ for the receptor as compared with d-tubocurarine, one concludes that a relaxation rate constant as slow as 2 s\(^{-1}\) would still be explained by buffered diffusion. The fact that the relaxations were about threefold faster may arise because (a) some cis-2BQ molecules do remain within the synaptic cleft during the flash and (b) buffering is reduced by the agonist-receptor interactions. Thus, we tentatively suggest that the slow relaxations of Fig. 10 reflect buffered diffusion of cis-2BQ molecules as they enter the synaptic cleft.

We did not notice the same slowing phenomenon in preliminary experiments at 20°C. However, if this very slow phase has a rather low temperature dependence, its rate would be near that of the noncompetitive channel block effect. Thus, the temperature test could not be cleanly applied to decide whether the slow relaxations are governed by the trans-2BQ-receptor interaction or by buffered diffusion of cis-2BQ as it re-enters the synaptic cleft.

It would be helpful to have comparable data from a preparation not subject to buffered diffusion. Because cultured myoballs lack synapses and have a low receptor density, buffered diffusion is not expected to play a role in the kinetics of drug action. Light-flash experiments are under way with patch-clamped rat myoballs (Chabala et al., 1982, 1984; Lester and Chabala, 1984).

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