



Supplementary Materials for

Redox stratification of an ancient lake in Gale crater, Mars

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Other Supplementary Materials for this manuscript includes the following:
(available at www.sciencemag.org/content/356/6341/eaah6849/suppl/DC1)

Data File S1 (Excel)

Supplementary Text

The effect of S-bearing phases on CIA: On **Fig. S2**, we illustrate the effect on CIA of adding important S-bearing phases discussed in the text to mudstones of the Sheepbed member and the Murray formation, which include Mg-sulfate, Ca-sulfate, and airfall dust (approximated by APXS analyses of soil at Gale crater), which can coat rock surfaces to varying degrees. As shown, Mg-sulfate has no effect on CIA, but likely plays a role in causing diagenetic and high-SO₃ targets from the Murray HP facies (red squares) to scatter horizontally to the right on **Fig. S2** relative to the rest of the Murray HP facies (black circles). Ca-sulfate imparts a strong negative trajectory on **Fig. S2** to any sample that has been impacted by anhydrite or gypsum addition (white squares). This effect is demonstrated by the APXS targets Mavor and Palmwag, which intentionally targeted Ca-sulfate veins in the Sheepbed member and Murray formation, respectively. In combination with Mg-sulfate, Ca-sulfate addition likely also plays a role in the offsetting the diagenetic and high-SO₃ HP facies targets from the rest of the Murray HP facies. The addition of dust coatings to the surfaces of rocks has the effect of drawing the samples towards total SO₃ values of ~5-10 wt. % and CIA of ~35%. The effects of dust coating are different for different mudstone compositions: for the Sheepbed member, dust contamination generates a horizontal vector on **Fig. S2**, because the CIA values of the Sheepbed mudstones are so similar to that of dust/soil. For the Murray formation (MS and HP facies), dust coatings will generate a vector with a negative slope, and therefore tend to mute the differences in CIA between the Murray formation and the Sheepbed member, while increasing total SO₃ concentration.

For both facies of the Murray formation some combination of Ca-sulfate addition (see also **Fig. 2**) and dust coating are implied, and the relationships displayed on **Fig. S2** indicate that the effects of these two processes would be difficult to disentangle from one another. Accordingly, rather than attempt an uncertain, and likely non-unique correction to CIA, we have chosen to compare “raw” CIA values. This represents a conservative approach to the treatment of CIA, as dust and/or Ca-sulfate removal calculations would only serve to amplify the differences in CIA between the mudstone suites. This is illustrated by a comparison of the difference between the average CIA of the Sheepbed member ($34 \pm 2\%$) and the Murray HP facies ($45 \pm 4\%$), which is 11%, and the difference between the CIA of the lowest SO₃ samples from the Sheepbed member (target Wernecke_brushed) and HP facies (target Topanga_DRT_Raster2), which is 16.1%.

Titanium and phosphorous distribution and mobility in the MS facies of the Murray fm.:

One of the more enigmatic features that the MS facies exhibits is high concentrations of SiO₂ and TiO₂ and correlated behavior between the two oxides (**Fig. S3A**, **Table S1**), which might be interpreted to reflect a phase of post-depositional chemical weathering or diagenesis under low-pH conditions. On the basis of terrestrial analogue studies, weathering processes at low-pH are known to produce dramatic residual enrichment in SiO₂ and TiO₂ [e.g., (87, 88)]. These enrichments result from the insoluble nature of TiO₂ relative to the other cations leached from the host rock, and because silica can effectively behave as an immobile element under low pH conditions when silica concentrations are maintained at saturation with respect to amorphous silica. However, if

the MS facies underwent a phase of post-depositional alteration under low pH conditions, then one would also predict that CIA values should increase in concert with increasing SiO₂ and TiO₂; after all, CIA compares the behavior of a relatively insoluble element (Al) against a suite of highly soluble elements (Ca, Na, K), so CIA should behave in the expected manner unless stoichiometric mineral dissolution under highly corrosive conditions is invoked. Indeed, an examination of data from weathering rinds generated under low-pH conditions on a basalt sample from Hawaii [sample HWSB820, (88)] indicates that as TiO₂ and SiO₂ increase, CIA also increases dramatically, to values as high as 85% . This is in stark contrast to the behavior displayed by the MS facies, which exhibits low to moderate CIA values (**Fig. S3B**).

There are a number of potential mechanisms that could explain the observed geochemical enrichment of TiO₂ in the MS facies. One possibility is that TiO₂ was enriched as a result of redistribution during non-acidic diagenesis. Diagenetic TiO₂ mobilization and precipitation under marine pH conditions has been reported from a Cretaceous sedimentary basin on Earth (89). Perhaps related, experimental studies indicate that titanium mobility increases in the presence of phosphorous (90, 91), particularly at alkaline pH, and the MS facies is notable for elevated phosphorous concentration, ranging between 1.2 and 1.5 wt. % (**Data S1**). Fluorine also enhances the mobility of titanium in aqueous systems, as demonstrated by high temperature experimental studies (92), and fluorine has been observed as a conspicuous component of sedimentary rocks in Gale Crater using the ChemCam instrument, with both clastic and diagenetic carriers of fluorine implicated (93).

Another possibility is that the elevated TiO₂ observed in the MS facies is due to a provenance effect. There are no crystalline, high-TiO₂ content minerals that form an obvious repository to which elevated TiO₂ can be assigned in the MS facies (e.g., ilmenite, rutile, anatase), though if one of these phases were present just below the CheMin detection limit of ~1 wt. %, then they could account for some of the elevated TiO₂ (43). In addition, experimental studies indicate that synthetic cristobalite can incorporate 10 wt. % TiO₂ by solid solution (94), while synthetic tridymite has been observed to incorporate up to 1.25 wt. % TiO₂ as a substitutional impurity (95). Studies on terrestrial and extraterrestrial samples report that TiO₂ can approach 0.5-1.0 wt. % in natural tridymite and cristobalite (96-98). On this basis, we suggest that elevated TiO₂ in the MS facies may be explained by a combination of diagenetic titania redistribution in the presence of phosphate and fluorine, low abundance titanium-bearing detrital oxides, and substitutional incorporation of TiO₂ in the crystalline silica phases (tridymite, cristobalite) that are present in high abundance in this part of the Murray fm.

Lamination thickness measurements for Figure S5: The thickness and variability of lamination near targets investigated with the APXS instrument was characterized based on visual inspection of Mars Hand Lens Imager (MAHLI) and Mastcam images (7, 99), which are publically available on the Planetary Data System Catography and Imaging Sciences Node (img.pds.nasa.gov). APXS targets in the Murray Fm. with nearby well-expressed lamination were identified using Mastcam images. Although useful for identifying lamination, Mastcam images are often of insufficient resolution for accurate measurements of the mm-scale laminae in the Murray fm. because the stereoscopically derived scale of Mastcam images is on the order of cm. Instead, laminae were identified

visually in MAHLI images. Laminae thicknesses were calculated from the pixel separation between laminae centers along a transect drawn orthogonally through the mapped laminae. Thicknesses were measured orthogonally to bedding to account for the structural orientation of exposed blocks of Murray. The resolution of each image was approximated from the working distance of the MAHLI instrument, and computed thicknesses were corrected for instrument viewing angle when necessary. The arithmetic mean and one standard deviation in lamination thickness were computed for each target for comparison with FeO and SiO₂ weight percentages measured in-situ with the APXS instrument.

Another approach to examining the laminae is through wavelet analysis of grayscale pixel intensity in regions orthogonal to bedding. Due to the relatively small number (as low as 20-30) of countable laminae in some MAHLI images, the difficulty of distinguishing laminae from other mm-scale textures in some images via wavelet analysis, and artifacts resulting from image compression and stitching, only visual estimates of laminae were used to compute thicknesses. There are some additional caveats in the visual lamination thickness analysis. Thickness measurements were not corrected for surface variability such as undulations or recessed surfaces, which are typically negligible over the small field of view of MAHLI (5.0 x 3.7 cm at the nominal 5.0 cm standoff distance). Because lateral variations in lamination thickness were negligible, thicknesses were not averaged over multiple transects of the same laminae. In some images, sets of countable laminae are offset spatially by regions where laminae are not expressed. In these cases, different transects of each region contributed to the lamination thickness distribution for the same target.

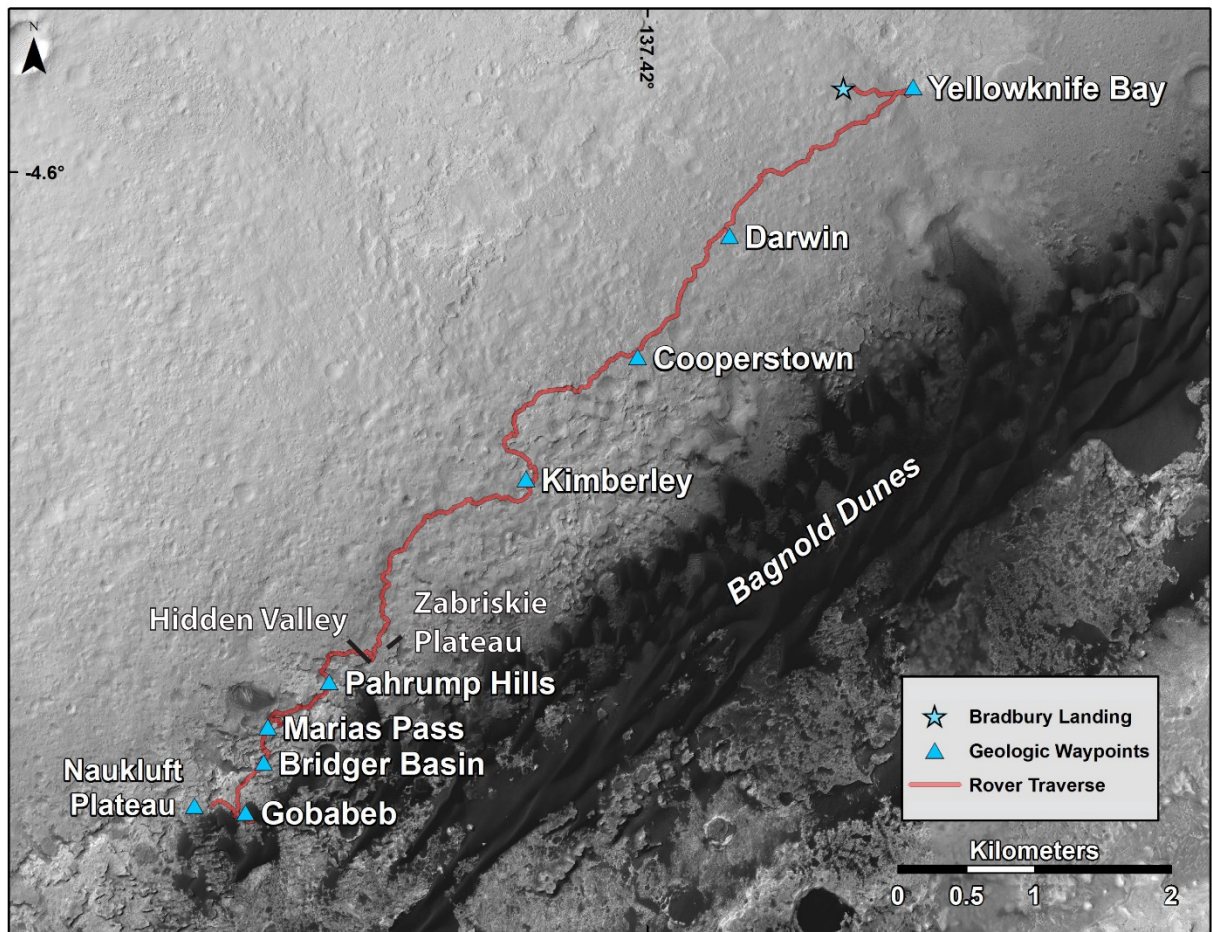


Figure S1: *Curiosity*'s traverse across the floor of Gale crater. Major geological waypoints along the traverse between sols 1-1310 are identified. The Sheepbed member was investigated at the Yellowknife Bay waypoint. The investigation of the Murray fm. began at Hidden Valley, just southwest of the Zabriskie Plateau, and continued through Pahrump Hills, Marias Pass, Bridger Basin, Gobabeb, and up to the Naukluft Plateau, where the Stimson fm. unconformably overlies the Murray fm. Along the traverse, the Hematite-Phyllosilicate (HP) facies of the Murray fm. occurs primarily between Pahrump Hills and Marias Pass, and between Bridger Basin and the Naukluft Plateau. The Magnetite-Silica (MS) facies occurs at Hidden Valley, and between Marias Pass and Bridger Basin. The base image from the map is from the High Resolution Imaging Science Experiment Camera on NASA's Mars Reconnaissance Orbiter (Image Credit: NASA/JPL-Caltech/Univ. of Arizona).

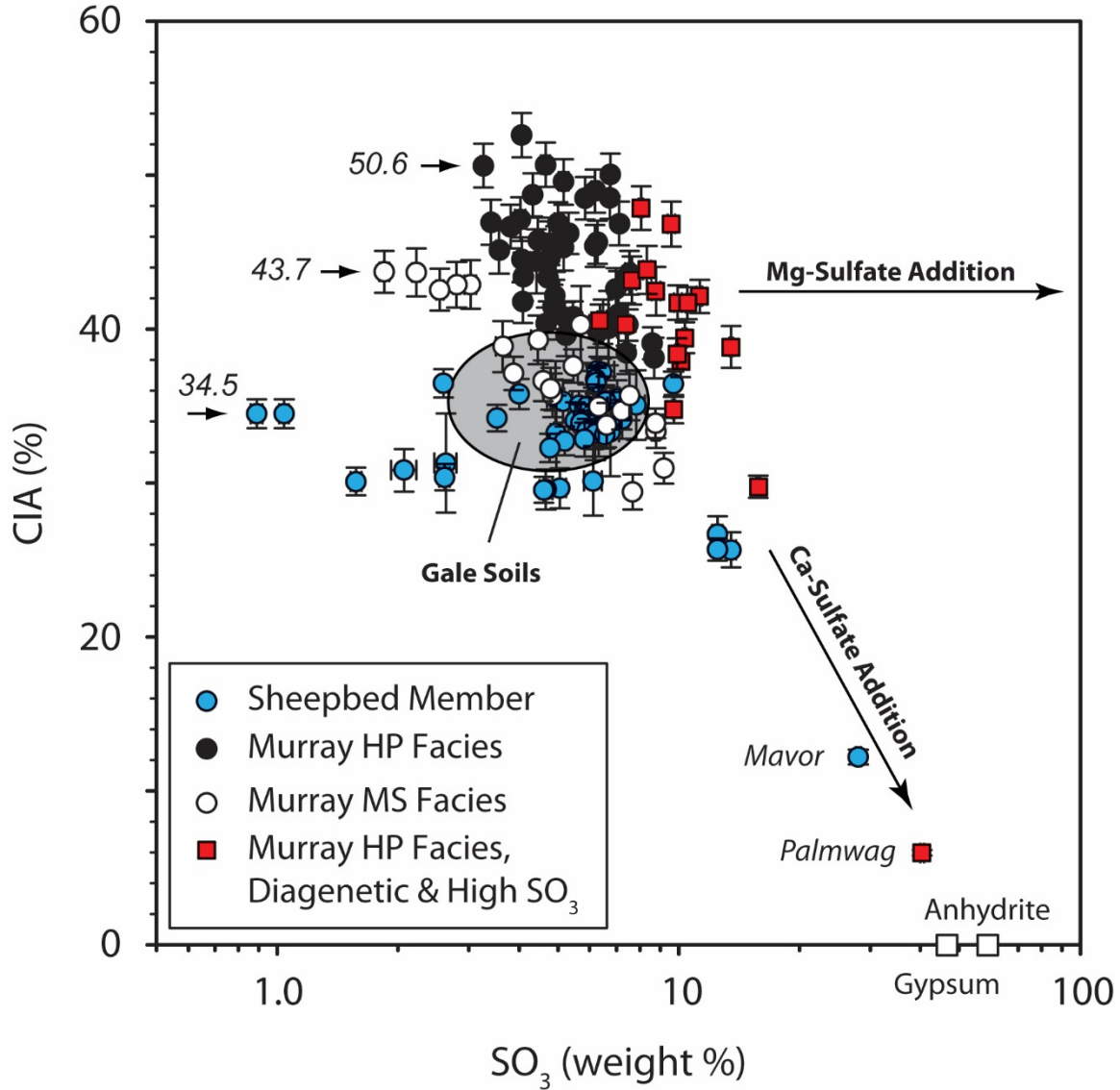


Figure S2: The effects of S-bearing dust and minerals on CIA. Vectors for the addition of Mg- and Ca-sulfate to mudstone samples shown on a plot of CIA (%) versus SO_3 (wt. %). The gray field represents the plotted positions of Gale Crater soils analyzed between sols 58-1184, used here as a proxy for the composition of airfall dust [see also (39)]. Arrowheaded lines with italicized CIA values reference the lowest SO_3 samples from the Sheepbed member, HP facies, and MS facies, which display a maximum difference of 16.1 %.

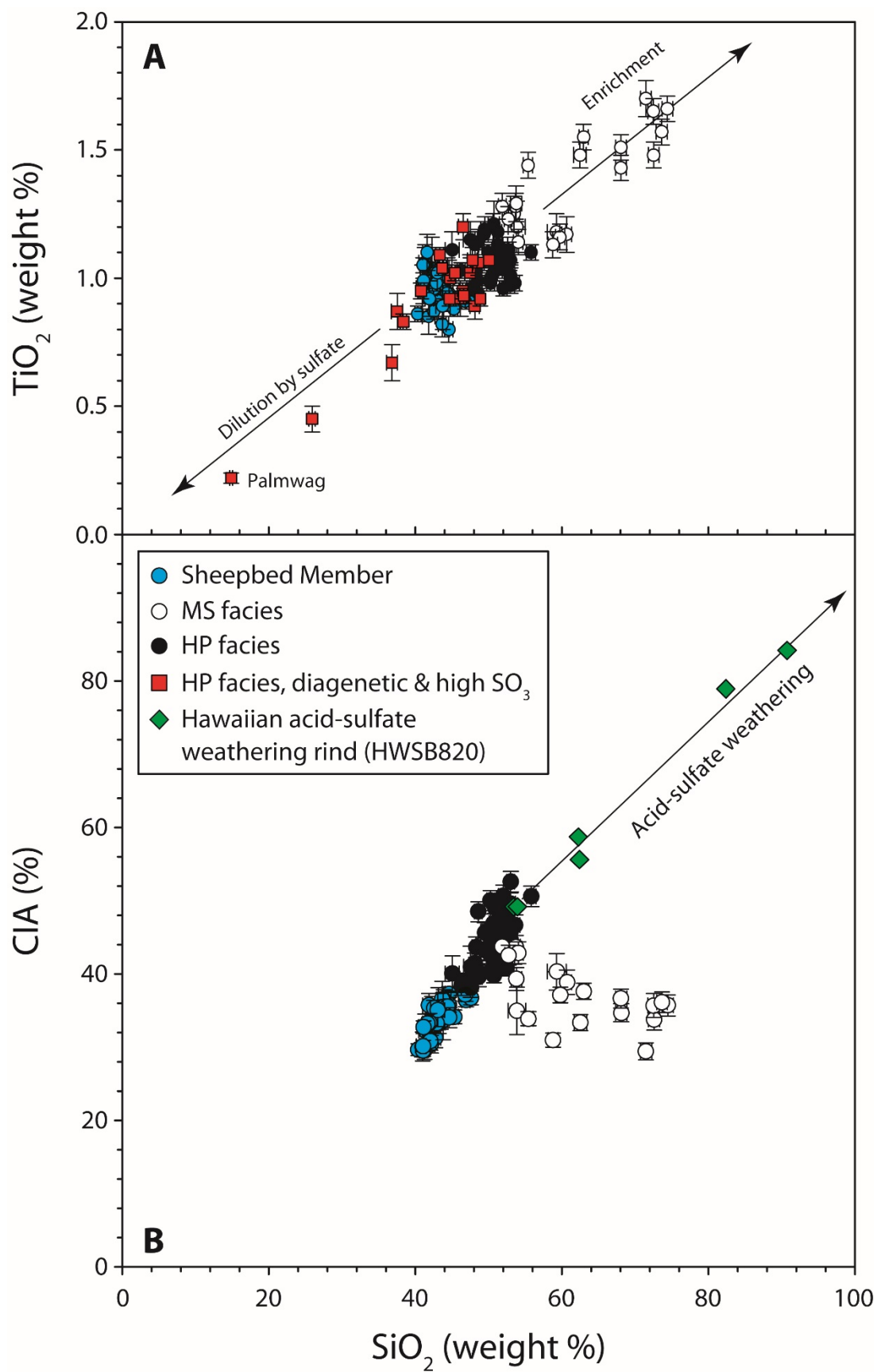


Figure S3 (previous page): Redistribution of Si and Ti in the Murray formation. (A) Plot of TiO_2 (wt. %) versus SiO_2 (wt. %) showing vectors for enrichment in these two elements in the MS facies, and dilution in diagenetic, high SO_3 , and vein targets in the HP facies (see especially target Palmwag). Symbols as described for **Fig. 3** and **Fig. 4**. **(B)** Plot of CIA (%) versus SiO_2 (wt. %) demonstrating differences between the MS facies, affected by processes of silica and Ca-sulfate addition (also see **Fig. 2**), and acid-sulfate weathering of terrestrial basalts (88).

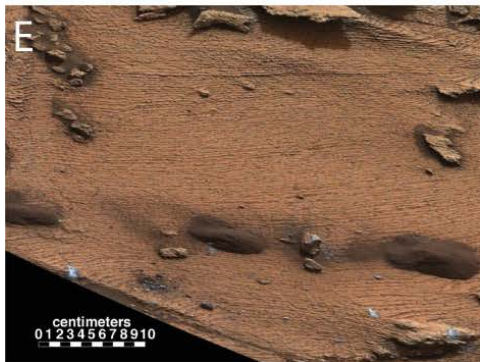
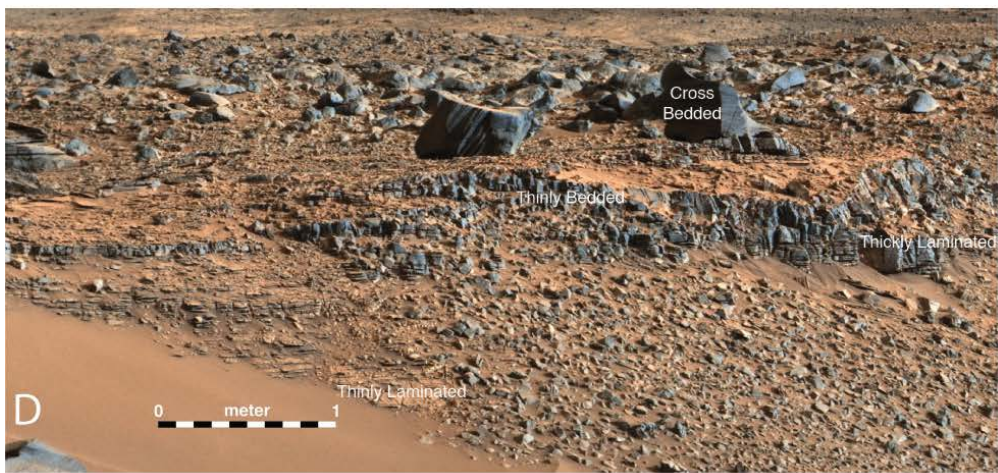
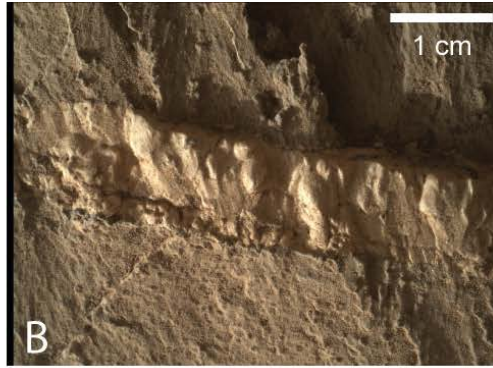
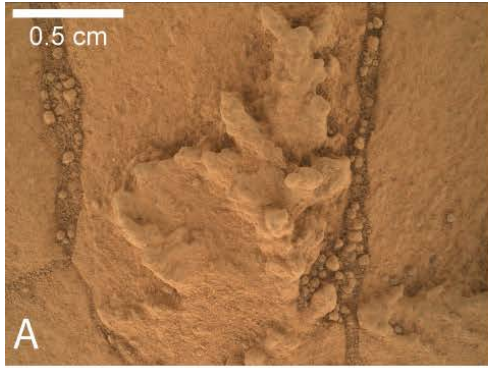


Figure S4 (previous page): Sedimentological features of the Murray formation. (A) MAHLI image of an MgO and SO₃-rich diagenetic feature at the base of the Murray fm. taken on sol 758 (image ID: 0758MH0001520010204633C00_DXXX). (B) MAHLI image of a Ca-sulfate vein (target Palmwag) cross-cutting the Murray fm., taken on sol 1,275 (image ID: 1275MH0005820010404440C00_DXXX). (C) MAHLI image of MgO and SO₃-rich diagenetic nodules in the Stimson fm., at the contact between the Murray and the Stimson, taken on sol 1,277 (image ID: 1277MH0001970010404560C00_DXXX). (D) Mastcam image taken of the walls of Hidden Valley on sol 703 showing cross-bedded fluvio-deltaic sandstones typical of the Zabriskie Plateau and Murray fm. mudstones displaying variable lamination and bedding thicknesses. Image reproduced from Fig. 6 of Grotzinger et al., (10), who interpret this image to represent a lacustrine facies overlain by laterally prograding fluvial-deltaic deposits. (E) Mastcam image of laminated Murray fm. mudstone from the HP facies, taken on sol 798 at Pahrump Hills. Image reproduced from Fig. 7B of Grotzinger et al. (10). Laminae are estimated to be ~2mm thick. (F) MAHLI image of the MS facies target “LaMoose”, taken on sol 1041 (image ID: 1041MH0001900010400213C00). Laminae are estimated to be ~0.5mm thick. All images credited to: NASA/JPL-Caltech/MSSS.

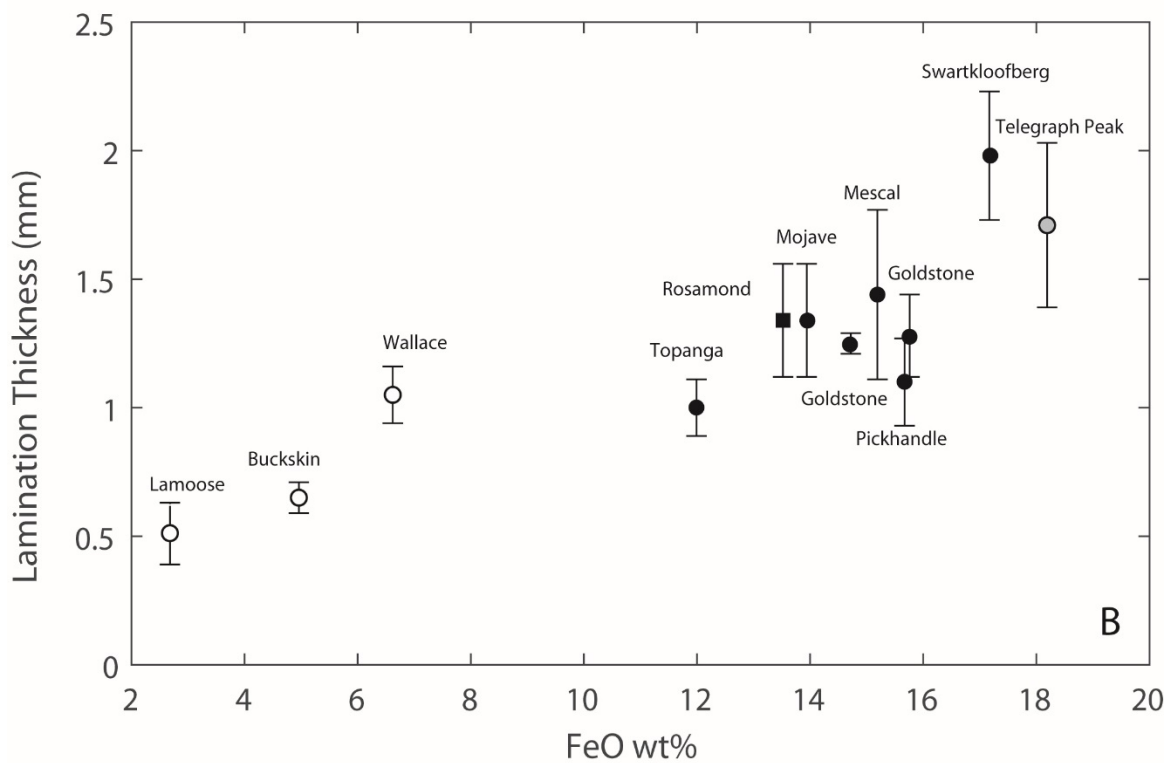
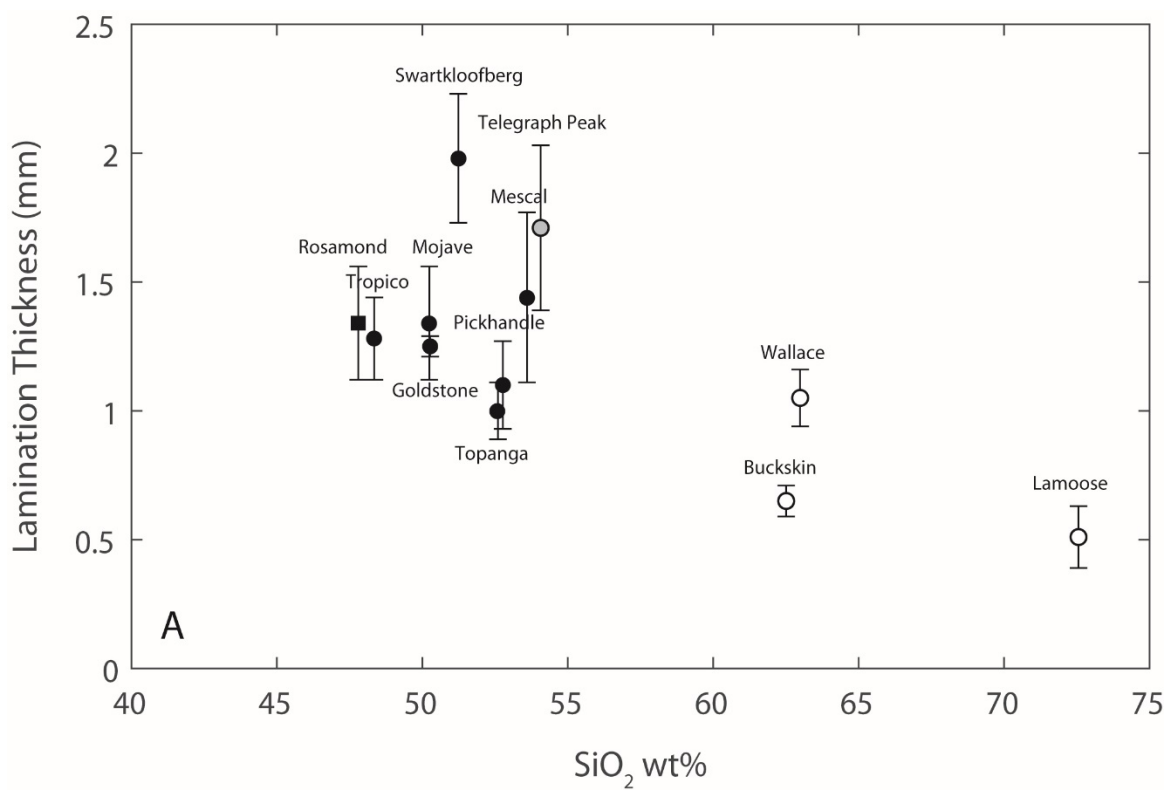


Figure S5 (previous page): Relationships between bedding thickness and chemistry. (A) Plot of SiO_2 versus bedding thickness and (B) Total iron as FeO (FeO_T) versus bedding thickness, with target names identified. Symbols are the same as in **Figs. 3 and 4**. In (B), the sample Telegraph Peak is displaced to high FeO_T concentration relative to the other Murray MS samples plotted, possibly a consequence of the high concentration of magnetite (**Table S2**) of probable authigenic origin in this sample.

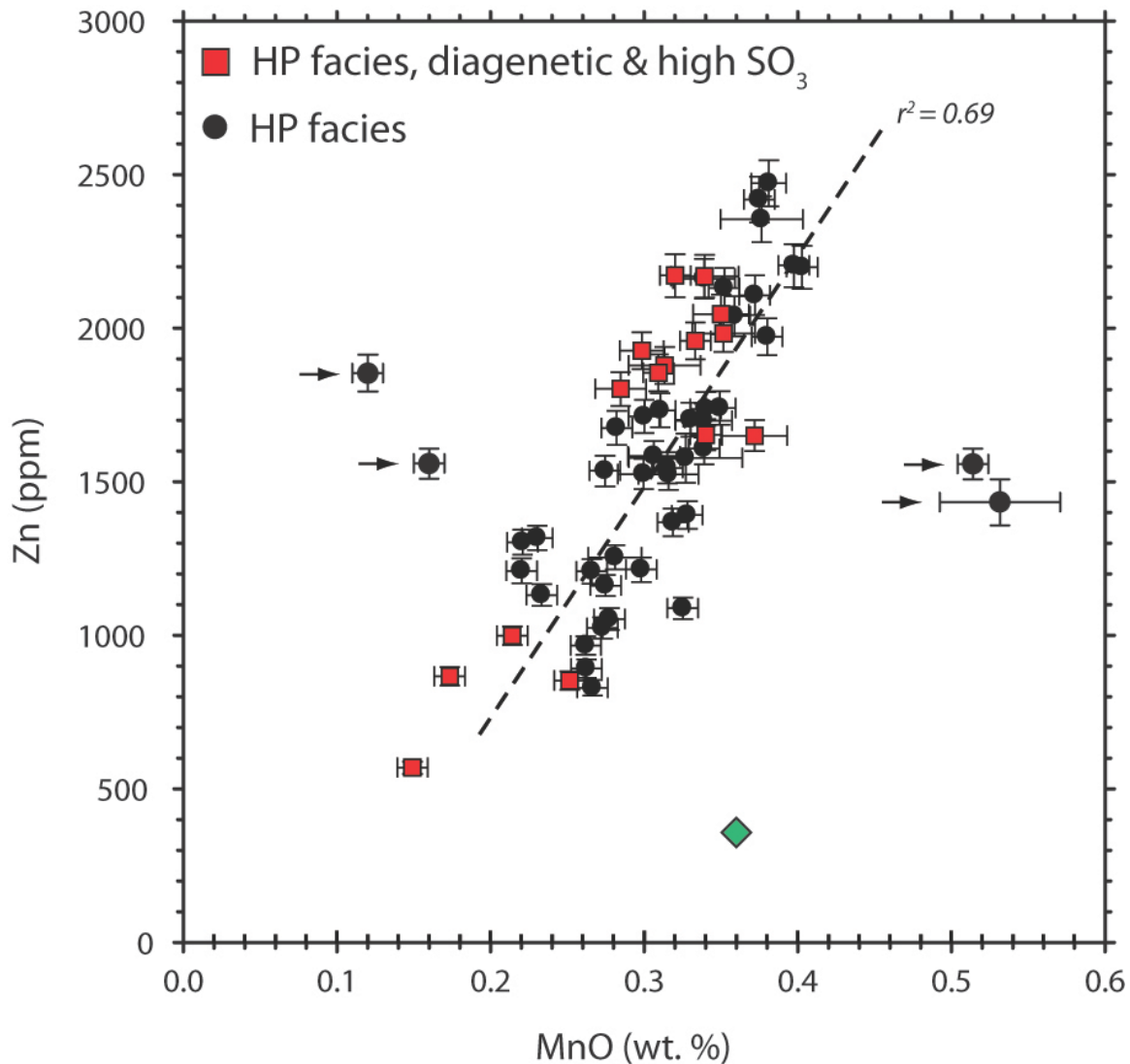


Figure S6: Mn-oxidation and Zn-concentration in diagenetic and high- SO_3 samples of the Murray HP facies. Zn (ppm) vs. MnO (wt. %) showing a linear regression for the HP facies that results from Mn-oxidation and scavenging of aqueous Zn by Mn-oxides (the outlier samples Ricardo_Raster_1, Ricardo_DRT_1, Kleinberg, and Schwarzrand, are excluded from the regression, labelled with arrowhead lines). Not included in the regression, but falling along the same trend, are diagenetic and high SO_3 samples from the HP facies. For these samples, either Mn-oxidation and Zn-scavenging continued during the later saline diagenesis, or MnO-Zn relationships in these targets were developed as a result of primary oxidative processes in the lake, but were not impacted by later saline diagenesis. Green diamond is average Martian crust.

Table S1. Mineralogical data from Gale Crater Mudstones

Mineral abundances with reported analytical uncertainties derived from CheMin XRD data. NR = not reported

	Sheepbed Member		Murray Fm			
Mineral (abun, %)	John Klein	Cumberland	Confidence Hills	Mojave2	Telegraph Peak	Buckskin
	Salts					
Anhydrite	2.3 (0.8)	0.9 (0.7)	0.0	0.0	0.0	0.7 (0.2)
Bassanite	1.2 (0.4)	1.3 (0.6)	0.0	0.0	0.0	0.0
Halite	0.4 (0.2)	0.1 (0.1)	0.0	0.0	0.0	0.0
<i>Sum</i>	<i>3.9</i>	<i>2.3</i>	<i>0.0</i>	<i>0.0</i>	<i>0.0</i>	<i>0.7</i>
	Amorphous					
Amorphous	30 (18)	30 (18)	39 (15)	53 (NR)	27 (15)	54 (NR)
	Clastic					
Plagioclase	23.7 (1.2)	22 (1.4)	20.4 (2.3)	23.5 (1.6)	27.1 (2.8)	17.1 (1.2)
Alkali feldspar	0.0	1.7 (0.9)	5.0 (0.7)	0.0	5.2 (2.2)	3.4 (0.7)
Augite	12.9 (1.1)	3.3 (1.2)	6.4 (2.2)	2.2 (1.1)	0.0	0.0
Orthopyroxene	0.0	5.7 (2.1)	2.1 (3.1)	0.0	3.4 (2.6)	0.0
Pigeonite	0.0	6.8 (1.8)	5.3 (1.7)	4.6 (0.7)	4.2 (0.1)	0.0
Forsterite	3.5 (0.7)	0.9 (0.8)	1.2 (0.7)	0.2 (0.8)	1.1 (1.2)	0.0
Ilmenite	0.4 (0.4)	0.3 (0.3)	0.0	0.0	0.0	0.0
Fluorapatite	0.0	0.0	1.3 (1.5)	1.8 (1.0)	1.9 (0.5)	0.0
Cristobalite	0.0	0.0	0.0	0.0	7.3 (1.7)	2.4 (0.3)
Quartz	0.4 (0.4)	0.0	0.7 (0.5)	0.8 (0.3)	0.9 (0.4)	0.0
Tridymite	0.0	0.0	0.0	0.0	0.0	13.6 (0.8)
<i>Sum</i>	<i>40.9</i>	<i>40.7</i>	<i>42.4</i>	<i>33.1</i>	<i>51.1</i>	<i>36.9</i>
	Secondary/Redox Sensitive					

Magnetite	3.3 (0.9)	4.4 (1.0)	3 (0.7)	3.0 (0.6)	8.2 (0.9)	2.8 (0.3)
Hematite	0.5 (0.5)	0.6 (0.5)	6.8 (1.5)	3.0 (0.6)	1.1 (0.5)	0.0
Akaganeite	1.1 (0.6)	1.8 (0.7)	0.0	0.0	0.0	0.0
Pyrite	0.4 (0.3)	0.0	0.0	0.0	0.0	0.0
Pyrrhotite	1 (0.5)	1.1 (0.6)	0.0	0.0	0.0	0.0
Jarosite	0.0	0.0	1.1 (0.7)	3.1 (1.6)	1.5 (1.8)	0.0
Opal-CT	0.0	0.0	0.0	0.0	10.9 (NR)	6.0 (NR)
Phyllosilicate	19 (9)	19 (9)	7.6 (NR)	4.7 (NR)	0.0	0.0
<i>Sum</i>	<i>25.3</i>	<i>26.9</i>	<i>19.3</i>	<i>13.8</i>	<i>21.7</i>	<i>8.8</i>

Plag: Pyx + Olivine	1.4	1.4	1.7	3.4	3.7	---
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Data S1 (separate file). Geochemical Data from Gale Crater Mudstones and Soils.

Chemical abundances and CIA values with reported analytical uncertainties derived from APXS XRF data. Samples are subdivided into their appropriate formation, member, or facies and listed by target name with the sol number on which the data were collected.

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