

Article

Diboron- and Diaza-Doped Anthracenes and Phenanthrenes: Their Electronic Structures for Being Singlet Fission Chromophores

Ekadashi Pradhan, Seunghoon Lee, Cheol Ho Choi, and Tao Zeng

J. Phys. Chem. A, **Just Accepted Manuscript** • DOI: 10.1021/acs.jpca.0c06915 • Publication Date (Web): 09 Sep 2020

Downloaded from pubs.acs.org on September 10, 2020

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

1
2
3
4
5
6
7 **Diboron- and Diaza-doped Anthracenes and**
8 **Phenanthrenes: Their Electronic Structures for**
9
10 **Being Singlet Fission Chromophores**
11
12
13
14
15
16
17

18 Ekadashi Pradhan,[†] Seunghoon Lee,[‡] Cheol Ho Choi,[¶] and Tao Zeng^{*,†}
19
20

21 [†]*Department of Chemistry, York University, Toronto, Ontario, M3J1P3, Canada*
22

23 [‡]*Division of Chemistry and Chemical Engineering, California Institute of Technology,*
24 *Pasadena, California 91125, USA*
25
26

27 [¶]*Department of Chemistry, Kyungpook National University, Daegu 702-701, South Korea*
28
29

30 E-mail: tzeng@yorku.ca
31

32 Phone: +1 (416) 7362100 ext. 33673
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Abstract

We used quantum chemistry methods at the levels of mixed-reference spin-flipping time-dependent density functional theory and multireference perturbation theory to study diboron- and diaza-doped anthracenes and phenanthrenes. This class of structures recently surged as potential singlet fission chromophores. We studied electronic structures of their excited states and clarify the reasons why they satisfy or fail to satisfy the energy criteria for singlet fission chromophores. Many studied structures have their S_1 states not dominated by HOMO→LUMO excitation, so that they cannot be described using the conventional two sites model. This is attributed to frontier orbital energy shifts induced by the doping and different charge transfer energies in different one-electron singlet excitations, or in other words, different polarizations of hole and/or particle orbitals in their S_1 and T_1 states. There is a mirror relation between the orbital energy shifts induced by diboron- and diaza-dopings, which, together with alternant hydrocarbon pairings of occupied and unoccupied orbitals, leads to more mirror relations between the excited states of the two types of doped structures.

Introduction

Singlet fission (SF) is the process that one molecule at singlet excited state shares its energy with a neighboring molecule at ground state and result in two molecules in triplet excited states.¹⁻³ The number of excited states is doubled and the triplet excited states have longer lifetime. These two effects make SF a highly interesting process in photovoltaics. The process has the potential to enhance the photoelectric conversion efficiency to exceed the ~30% Shockley-Queisser limit⁴ and reach ~50% for a single junction photovoltaic device.⁵ One obstacle for the advancement of exploiting SF in photovoltaics is the small pool of chromophores that can undergo this process.⁶⁻⁹ Actually, SF research has been “overwhelmingly limited” to the two longer polyacenes and their derivatives.¹⁰ This fact motivates the thriving studies in proposing other SF chromophores.

In order to undergo SF, a chromophore molecule needs to satisfy the following criteria for its excited states' energies:¹

$$E(S_1) > 2E(T_1); \quad (1)$$

$$E(T_2) > 2E(T_1); \quad (2)$$

$$E(Q_1) > 2E(T_1). \quad (3)$$

“ S_i ”, “ T_i ” and “ Q_i ” stand for i -th excited states within the singlet, triplet, and quintet spin manifolds, respectively. Note that all energies in the three inequalities above are excitation energies of one chromophore molecule. The inequality in Eq. 1 ensures an exoergic fission of a singlet exciton to two triplet excitons. The other two inequalities ensure endoergic and thus unfavorable recombination of two triplet excitons on two molecules to one triplet exciton or one quintet exciton on one molecule. The recombination may occur between two independent triplet excitons, whose randomly oriented spins can add up to an overall singlet, triplet, or quintet spin.¹¹ The recombination is thus an internal conversion and does not involve spin flipping.¹ Since the Q_1 state is usually higher in energy than S_1 and T_2 , we can safely ignore Eq. 3 and focus on the other two criteria. If $E(T_2)$ is close to or even smaller than $2E(T_1)$, then the two SF-generated T_1 excitons may recombine to form one T_2 exciton on one molecule and the other partner in S_0 state. However, T_2 often involves HOMO→LUMO+1 or HOMO−1 →LUMO excitation, while T_1 involves HOMO→LUMO excitation. The difference in orbital energy gaps is usually large enough to guarantee the inequality in Eq. 2. Therefore, Eq. 2 is often considered as a secondary criterion compared to Eq. 1. However, this is not always the case, when there is orbital pseudo-degeneracy between HOMO−1 and HOMO, or between LUMO and LUMO+1. This situation does occur for some molecules considered in this work.

Eq. 1 is the primary criterion for SF chromophores and has been the focus of chromophores designing. Considering thermal activation, this criterion can be relaxed and

1
2
3 molecules with $E(S_1)$ slightly smaller than $2E(T_1)$ may also undergo SF.¹ Paci et al. first
4 pointed out that diradicaloids and large polycyclic aromatic hydrocarbons (PAHs) are likely
5 to satisfy the criterion.¹² Later, Nakano and coworkers used a symmetric two sites model to
6 associate the $E(S_1) \sim 2E(T_1)$ relation solely to diradical character;¹³⁻¹⁶ PAHs satisfy Eq. 1
7 when they are large enough to have adequate diradical character. This association explains
8 the rarity of SF chromophores. While too little diradical character makes $E(S_1) < 2E(T_1)$,
9 too much diradical character makes a molecule highly reactive and unstable. We hence need
10 to finely tune diradical character of a molecule to make it a stable SF chromophore, which
11 is not easier than “finding a few needles in a haystack.”¹⁷ Difficult yet possible. The beacon
12 of diradical character has guided the community of SF to design a few chromophores.^{7,9,17}
13 1,3-Diphenylisozofuran is a typical example and it has been synthesized and exhibits SF
14 behavior.¹⁸⁻²¹ For PAHs, diradical character is closely related to ground state antiaromatic-
15 ity, and so is the $E(S_1) \sim 2E(T_1)$ relation.¹⁴ Recently, more designs of SF chromophores
16 based on ground state antiaromaticity have been proposed.²²⁻²⁴ Using the two sites model,
17 the Nakano group also pointed out that the exchange integral $K_{a,b}$ is an extra knob for
18 tuning the $E(S_1) \sim 2E(T_1)$ relation, where a and b stand for the localized orbitals that are
19 transformed from HOMO and LUMO as $\sqrt{\frac{1}{2}}(\text{HOMO} \pm \text{LUMO})$.^{25,26}
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36

37 The essences of the two sites model are summarized as follows. (1) The HOMO and
38 LUMO are of different irreducible representations (IRREPs) of the chromophore’s point
39 group, e.g., of “ g ” and “ u ” I -parities and/or “ A ” and “ B ” C_2 -parities. They are viewed
40 as linear combinations of two “localized” “nonbonding” asymmetric orbitals a and b (*vide*
41 *supra*). These two **sites** are symmetry images of each other and hence have the same energy.
42 The quotation marks emphasize that a and b might themselves be delocalized over a few
43 atoms, and there may be bonding and/or antibonding characters within a and b . However,
44 it is the bonding and antibonding interactions between a and b that open the HOMO-LUMO
45 gap ($\Delta_{H,L}$). (2) With the symmetry, the 2 electrons and 2 orbitals configuration interaction
46 (2o2e CI) problem can be solved exactly. There are four configuration states: the singlet
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 diradical (SD) with one electron on a and one electron on b , its triplet counterpart (T), the
4 symmetric and antisymmetric charge-transfer states (SC and AC) with two electrons solely
5 on a or on b . S_0 (S_2) is the bonding (antibonding) mixture of SD and SC, S_1 is of pure
6 AC character, and T_1 is of pure T character. (3) $2K_{H,L} = (J_{a,a} - J_{a,b})$, where J denotes
7 Coulomb integral of the orbitals in subscript. $2K_{H,L}$ is thus the electrostatic energy cost
8 of charge transfer (CT) from the configuration with one electron on each site to that with
9 both electrons on one site. The energy difference between T and AC (i.e, the $E(T_1)-E(S_1)$
10 gap) is $2K_{H,L}$. A large $K_{H,L}$ suggests a higher $E(S_1)$ and a lower $E(T_1)$, and hence a
11 better propensity to satisfy Eq. 1. (4) $2K_{H,L}$ is also the energy gap between SD and SC,
12 whose interaction matrix element is $\frac{1}{2}\Delta_{H,L}$. Therefore, a small $\Delta_{H,L}/K_{H,L}$ ratio indicates a
13 smaller interaction between SD and SC, and thus a higher $E(S_0)$. Relatively, this implies
14 lower $E(T_1)$ and $E(S_1)$ together. Since the reduction of $E(T_1)$ counts twice in the Eq. 1
15 inequality, the together lowerings of $E(T_1)$ and $E(S_1)$ imply a better propensity to satisfy
16 the inequality. Considering all these effects of $K_{H,L}$ and $\Delta_{H,L}$, their ratio $K_{H,L}/\Delta_{H,L}$ (which
17 implies diradical character, and is also related to the MP2 perturbation amplitude for the
18 HOMO \Rightarrow LUMO excitation) is a determining factor for satisfying Eq. 1. (5) T and AC (T_1
19 and S_1) are $2K_{a,b}$ lower in energy than SD and SC, respectively. Therefore, a larger $K_{a,b}$
20 suggests lower $E(S_1)$ and $E(T_1)$ together. Again, the $E(T_1)$ lowering counts twice in Eq. 1,
21 and $K_{a,b}$ is also a factor that determines the propensity to satisfy Eq. 1.
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40

41 The mathematical expressions of the ground and excited states' energies taking $\Delta_{H,L}$,
42 $K_{H,L}$, and $K_{a,b}$ as variables are shown in Refs. 13–16,25,27,28. Since the impacts of the
43 three parameters on the energies are conceptually transferrable to calculations more accurate
44 than the 2o2e CI treatment, a qualitative description of the impacts given in the previous
45 paragraph is more helpful, we believe, for understanding why a molecule satisfies or does not
46 satisfy Eq. 1. However, if the two sites model becomes invalid, i.e., when the low-lying excited
47 states involve excitations beyond the 2o2e small space, diradical character and $K_{a,b}$ are not
48 the only determining factors for their energies. On the other hand, as long as the low-lying
49
50
51
52
53
54
55
56
57
58
59
60

excited states can be described using excitations within the 2o2e space, diradical character and $K_{a,b}$ are the main determining factors. For those molecules, we may also use ground state antiaromaticity to explain their $E(S_1) \sim 2E(T_1)$ relation. But this explanation must be consistent with the one based on diradical character and $K_{a,b}$. A few of the diboron- (2B-) and diaza-doped (2N-doped) molecules that we study in this work cannot be described using the two sites model. Although the 2B-doped molecules have been recently investigated as potential SF chromophores,²⁹ we see a necessity to look back at those molecules, and explain their unexpected properties of excited states.

The motivations of doping two B atoms into anthracene and phenanthrene are as follows. (1) Anthracene is historically related to SF. The first indirect observation of SF is through the delayed fluorescence of anthracene crystal.³⁰ (2) Unlike its longer analogues tetracene and pentacene,³¹⁻³⁵ anthracene and its isomer phenanthrene are small PAHs that do not satisfy Eq. 1. The indirectly observed SF of anthracene crystal is not thermodynamically favorable and cannot compete with other excited state dynamics channels. Chemical tailoring on the pristine molecules can be employed to design SF chromophores. (3) The 2B-dopings convert the two $4n + 2$ systems to $4n$ systems. The associated antiaromaticity implies high diradical character, low-lying T_1 state, and consequently the satisfying of Eq. 1. This is especially true if the C moiety(ies) of a doped molecule possesses (possess) a clear diradical structure.^{26,36,37} (4) Anthracene and phenanthrene are representatives of polyacenes and phenacenes, respectively. It is of interest to investigate their different responses to the dopings. (5) 2N-dopings also convert $4n + 2$ systems to $4n$ systems, and may also lead to SF chromophores. Also, it is of interest to study the differences and similarities between doping donors (N) and acceptors (B).

The present study is inspired by the pioneering work of Stoycheva et al.²⁹ on 2B-doped anthracenes and phenanthrenes. The 2B- and 2N-doped phenanthrenes were also included in a more recent study by the Nakano group,²⁶ Most of our calculated results are consistent with those reported in Refs. 26 and 29. Still, the present study provides another perspective

1
2
3 on those molecules. Like in the pioneering work, we only consider symmetric dopings, which
4 allow us to view the molecules using the two sites model, and more importantly, to investigate
5 the validity of the model.
6
7
8
9

11 Computational methods

14 Since structural reorganization after excitation occurs at a similar or even faster time scale
15 than SF, we optimized structures for S_0 , S_1 , T_1 , and T_2 states of all studied molecules,
16 and used the minimum-to-minimum (i.e., adiabatic) $E(S_1)$ s, $E(T_1)$ s, and $E(T_2)$ s to judge
17 whether the molecules satisfy the two inequalities. Structural optimizations and subsequent
18 hessian calculations were performed using the mixed-reference spin-flipping time-dependent
19 density functional theory (MRSF-TDDFT^{38,39}) method, with the BHHLYP^{40,41} functional.
20 All the optimized structures have no imaginary frequency. At the optimized structures, n-
21 electron valence state perturbation theory⁴²⁻⁴⁴ (NEVPT2) calculations were performed to
22 determine the excitation energies. Four singlet and three triplet states are averaged equally
23 in the NEVPT2 calculations. Convergence of $E(S_1)$, $E(S_2)$, $E(T_1)$, and $E(T_2)$ with respect
24 to the number of states being averaged was reached, except for the **AB3B7** molecule, which
25 requires one more singlet state in the average. Both nondynamic and dynamic correlations
26 between electrons are treated by the MRSF-TDDFT and NEVPT2 methods. The NEVPT2
27 single point energies were corrected by the zero point energies (ZPEs) calculated at the
28 MRSF-TDDFT level. Without further specification, all reported excitation energies are
29 the ZPE-corrected 0-0 energies. All calculations were performed using the cc-pVDZ basis
30 set.⁴⁵ Test calculations using the cc-pVTZ basis set resulted in ≤ 0.02 eV differences in
31 excitation energies (Section S1 in the Supporting Information). Convergence with respect
32 to basis set was reached. We also performed general multi-configuration quasi-degenerate
33 perturbation theory⁴⁶⁻⁴⁸ (GMC-QDPT) calculations to obtain the wave functions of the
34 ground and excited states, and analyzed them to understand how the dopings modify the
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 electronic structures of the molecules.
4

5 A 12 valence π orbitals 12 valence electrons (12o12e) complete active space is used for
6 the NEVPT2 calculations. For the 2B-doped molecules, the two leftover valence π orbitals
7 feature the B empty p_z orbitals with the most substantial antibonding interaction with
8 surrounding C atoms and are highest in energy among the π valence. For the 2N-doped
9 molecules, the leftover π orbitals feature the N lone pair character with the most substantial
10 bonding character with surrounding C atoms and are lowest in energy. It is reasonable to ex-
11 clude them from the active space, when it is too costly to include them, and leave them to the
12 perturbation treatment of dynamic correlations. Actually, the same 12o12e active space was
13 successfully employed by Kawashima et al to calculate excitation energies of anthracene.⁴⁹
14 GAMESS-US^{50,51} was used for the MRSF-TDDFT and GMC-QDPT calculations. ORCA⁵²
15 was used for the NEVPT2 calculations. The GMC-QDPT wave functions are constructed
16 using state-averaged canonical orbitals, which are thus used for wave function analysis. The
17 energies of the active orbitals that correspond to unoccupied orbitals of ground state are
18 known to be much higher than those obtained from DFT calculations. However, the con-
19 sequent larger energy gaps between occupied and unoccupied orbitals do not impair our
20 qualitative, yet indicative analysis, which is focused on the overall trends of orbital energies'
21 changes. Natural orbitals obtained for ground states of the molecules are used to discuss
22 their diradical characters and assign Lewis structures. Most of the orbital plottings are given
23 in the Supporting Information, where the figures are labeled as Figure S1, Figure S2, etc.
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42

43 The C positions of the anthracene and phenanthrene are labeled with the IUPAC num-
44 berings. The numberings are shown in Figure 1. Following the convention in Ref. 29, the
45 doped molecules are labeled as **AB n B m** , **AN n N m** , **PB n B m** , and **PN n N m** . "A" and "P"
46 indicate the pristine anthracene and phenanthrene. n and m indicate the doping positions.
47 Throughout this paper, \rightarrow is used to represent single-electron excitations and \Rightarrow double-
48 electron excitations. Suppose that the HOMO \rightarrow LUMO configuration contributes $p_{HL}\%$ to
49 S_1 and another configuration that contributes most significantly to S_1 has the percentage of
50
51
52
53
54
55
56
57
58
59
60

$p_{other}\%$, the ratio of $p_{HL}\%$ / $p_{other}\%$ is used to estimate the dominance of HOMO→LUMO in S_1 . Please note that $p_{HL}\%$ may be smaller than $p_{other}\%$. This ratio is labeled as r_{HL} . Since all T_1 states of the discussed molecules are dominated by HOMO→LUMO, r_{HL} also indicates the applicability of the two sites model: the larger, the more applicable. The occupancy of the lowest unoccupied natural orbital (n_{LUNO}) is used to indicate diradical character of a molecule.^{53,54} $\Delta_{SF}^{(1)} = E(S_1) - 2E(T_1)$ and $\Delta_{SF}^{(2)} = E(T_2) - 2E(T_1)$ quantify the extent to which a molecule satisfies Eqs. 1 and 2.

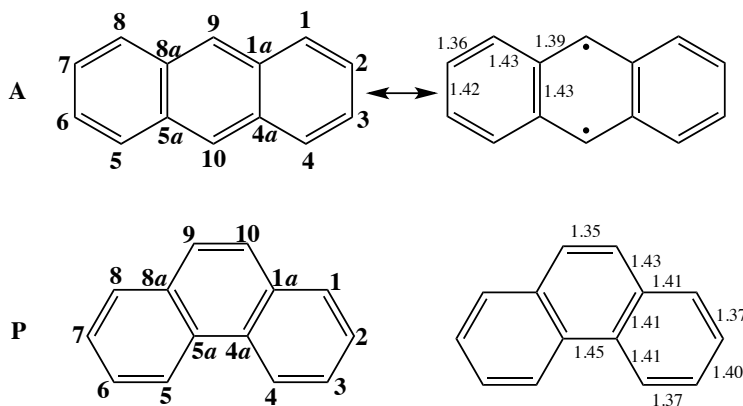


Figure 1: Anthracene (A) and phenanthrene (P), their atom numberings, and bond lengths (in Å). Shown for anthracene is also a diradical Lewis structure contained in the molecule's ground state.

Within the closed-shell Hartree-Fock approximation, the energy of a singlet excitation from an occupied orbital i to an unoccupied orbital j' is $\Delta_{ij'} - J_{i,j'} + 2K_{i,j'}$, and the triplet excitation energy is $\Delta_{i,j'} - J_{i,j'}$. $\Delta_{i,j'}$ indicates the energy cost for promoting an electron from an orbital with more bonding character to an orbital with more antibonding character. $J_{i,j'}$ is the Coulombic binding energy exerted by a hole in i onto an electron in j' . In a state-averaged CASSCF calculation, this binding energy is partially absorbed in orbital energies and thus $\Delta_{i,j'}$ as well. This is because the density matrix of the $i \rightarrow j'$ configuration is also included in the state-averaging for the construction of Fock operator. Therefore, we ignore this term in the following discussion. The $2K_{i,j'}$ reflects the different electron correlations in the singlet and triplet. In general, $2K_{i,j'} = \frac{1}{2}(J_{a,a} + J_{b,b}) - J_{a,b}$, with a and b being the localized orbitals $\frac{1}{\sqrt{2}}(i \pm j')$. $2K_{i,j'}$ hence reflects the averaged energy cost of cramming two electrons on one

1
2
3 site rather than having one electron on each site. The CT description of singlet excitation
4 holds even for two asymmetric sites. The aforementioned $2K_{H,L} = J_{a,a} - J_{a,b}$ is a special
5 case with the $J_{a,a} = J_{b,b}$ symmetry. Henceforth, Subscripts a and b in $K_{a,b}$ solely denote
6 localized orbitals transformed from HOMO and LUMO.
7
8
9

10 11 12 13 **Results and discussions**

14
15
16 We ran calculations for all the 2B-doped molecules considered in Ref. 29 except **AB9B10**
17 and **AB5B8**. These two molecules feature typical closed-shell characters in their ground
18 states. They are far from satisfying Eq. 1 and are not worth further studying in the con-
19 text of SF. The 2N-doped analogues **AN9N10** and **AN5N8** are also not considered. We
20 first present the calculated results of anthracene and phenanthrene in Supporting Infor-
21 mation (SI, Section S2), which are references for the discussions of the doped molecules.
22 In short, with not enough diradical characters, both molecules do not satisfy Eq. 1. An-
23 thracene can be described using the two sites model as both its S_1 and T_1 are dominated
24 by HOMO→LUMO excitation. Contrarily, the S_1 of phenanthrene is dominated by equal
25 mixing of HOMO→LUMO+1 and HOMO−1 →LUMO excitations; the two sites model is
26 inapplicable. The small $\Delta_{H-1,H}$ and $\Delta_{L,L+1}$ gaps, the smaller $K_{H-1,L}$ and $K_{H,L+1}$ than
27 the $K_{H,L}$ integral, and the configuration interaction between the HOMO→LUMO+1 and
28 HOMO−1 →LUMO configurations are responsible for the unconventional S_1 of phenan-
29 threne.
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45

46 **The 2B- and 2N-dopings on orbital energies**

47
48
49 The dopings significantly modify the orbital energies, which are shown in Figure 2. $\Delta_{H,LS}$ are
50 all substantially reduced compared to those of the undoped molecules. This is as expected
51 for the $4n + 2$ to $4n$ conversion and the associated enhancement of diradical character.
52
53 The general enhancement of diradical character is clearly shown by the n_{LUMO} plottings in
54
55
56
57
58
59
60

Figure 3(a). **AN1aN8a** is an exception in the orbital energy and n_{LUMO} diagrams and will be discussed later. The most significant changes in Figure 2 are the decrease of LUMO energies of the 2B-doped molecules and the increase of the HOMO energies of the 2N-doped molecules, which are mostly responsible for the $\Delta_{H,L}$ reductions. These general trends can be explained using the general knowledge of p - and n -type semiconductors: the doping of acceptor (B here) inserts empty orbitals in band gap and the doping of donor (N here) inserts occupied orbitals. Compared in Figure 3(b) are the decreases of LUMO energies of the 2B-doped molecules and the increases of HOMO energies of the 2N-doped molecules. There is an interesting mirror relation: the LUMO energy decreases of the 2B-dopings have similar magnitudes as the HOMO energy increases of the 2N-dopings. This indicates that the B and N dopants have similar intensities in modifying electronic structure, but in opposite directions.

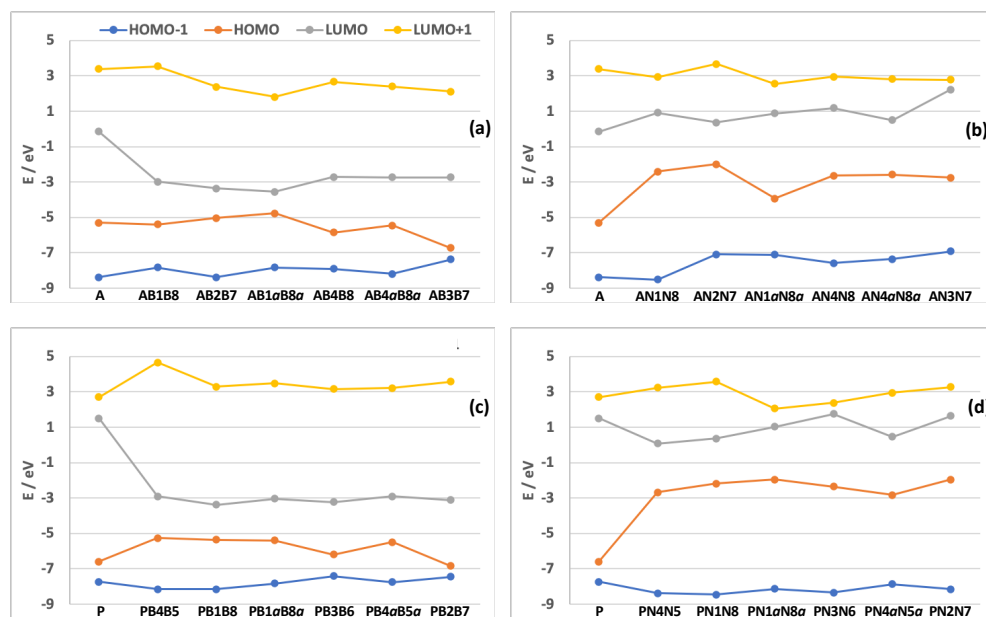


Figure 2: Canonical orbital energies of HOMO–1 to LUMO+1 of all molecules studied in this work.

There are more changes beyond the $\Delta_{H,L}$ reductions. The lower LUMO energies of the 2B-doped molecules also reduce their $\Delta_{H-1,L,S}$. For the 2B-doped anthracenes, such a gap change may lower the energy of the HOMO–1 \rightarrow LUMO configuration and, given ap-

1
2
3 appropriate symmetry, significantly enhance its contribution to the S_1 state. Consequently,
4 it may invalidate the two sites model, which is applicable for anthracene. Similarly, the
5 2N-dopings reduce $\Delta_{H,L+1}$ s and may raise the percentage of the HOMO \rightarrow LUMO+1 config-
6 uration in S_1 , invalidating the two sites model for the 2N-doped anthracenes. An opposite
7 situation may be seen for the doped phenanthrenes. The $\Delta_{H,L}$ reduction is likely to make
8 the HOMO \rightarrow LUMO configuration dominates the S_1 states of the doped phenanthrenes.
9 Also, the $\Delta_{L,L+1}$ s are enlarged from 1.2 eV of phenanthrenes to > 6.5 eV in the 2B-doped
10 analogues. The HOMO \rightarrow LUMO+1 configuration must contribute less to S_1 . For the 2N-
11 doped phenanthrenes, the $\Delta_{H-1,H}$ s are enlarged from 1.1 eV to > 5.0 eV, diminishing the
12 HOMO-1 \rightarrow LUMO contribution to S_1 . With these considerations, the two sites model,
13 which is inapplicable for phenanthrene, may be more applicable for the doped phenanthrenes.
14

15 The orbital energy gap changes are not the only determining factors for excited states,
16 especially for one-electron singlet excited states. The CT energy is another factor. Therefore,
17 the spatial overlaps between the relevant localized orbitals also determine the percentages of
18 those non-HOMO \rightarrow LUMO configurations in S_1 . This is a focus of the following discussion.
19 On the contrary, an $i \rightarrow j'$ triplet excited configuration is of pure diradical character, and its
20 energy is more determined by the orbital energy gap $\Delta_{ij'}$. As a matter of fact, all T_2 states of
21 the 2B-doped (2N-doped) molecules have the largest contributions from HOMO-1 \rightarrow LUMO
22 (HOMO \rightarrow LUMO+1) excitation. The mirror relation on orbital energy changes between 2B-
23 and 2N-dopings results in a mirror relation on the T_2 characters.
24

25 **Doped anthracenes with large diradical characters**

26 All calculated results of the 2B-doped anthracenes' excited states are summarized in Ta-
27 ble 1. Figure 3(a) shows that **AB1B8**, **AB2B7**, and **AB1aB8a** possess exceptionally large
28 diradical characters. The three make the Group III chromophores in Ref. 29. As shown by
29 the results in Table 1, they satisfy Eqs. 1 and 2. The 2B-dopings result in clear diradical
30 Lewis structures in their ground state resonances. Those Lewis structures are shown in
31
32
33
34
35
36
37
38
39
40
41
42
43
44



Figure 3: (a) n_{LUNOS} , (b) changes of molecular orbital energies with respect to undoped molecules, and (c) HOMO-LUMO gaps of all molecules studied in this work. “A” and “P” in the legends indicate doped anthracenes and phenanthrenes. “B” and “N” indicate 2B- and 2N-dopings. In each panel, the doping positions are given by the AB and PB markers, and they are applicable for the corresponding AN and PN markers. The leftmost markers are for the undoped anthracene and phenanthrene.

Figure 4. They are assigned based on natural orbitals of the molecules. For instance, the HONO (highest occupied natural orbital) and LUNO (lowest unoccupied natural orbital) of **AB1B8** (Figure S4(a) and (b)) are mainly antisymmetrized and symmetrized combinations of the singly occupied molecular orbitals (SOMOs) of the two allyl moieties on the two terminal rings, pointing to the Lewis structure shown in Figure 4(a). The HONO and LUNO of **AB2B7** (Figure S4(e) and (f)) resemble the SOMOs of *m*-quinodimethane (MQDM), and

Table 1: Calculated quantities for the 2B- and 2N-doped anthracenes discussed in this paper. All energies are given in eV. Excitation energies are relative to $E(S_0)$ of respective molecules.

	$E_v(S_1)^a$	$E(S_1)$	$E(T_1)$	$E(T_2)$	f_{S_1}	$\Delta E_{SF}^{(1)}$	$\Delta E_{SF}^{(2)}$	r_{HL}
A^b	3.43	3.11	1.89	3.52	0.090	-0.66	-0.25	28.44
AB1B8	1.63	1.38	0.36	1.88	0.015	0.66	1.16	0.79
AB2B7	0.68	0.58	-0.03	1.57	0.002	0.64	1.63	5.25
AB1aB8a	1.26	0.59	0.11	1.35	0.002	0.37	1.13	7.29
AB4B8	2.41	2.08	0.70	2.13	0	0.68	0.73	0
AB4aB8a	1.76	1.22	0.80	2.20	0	-0.38	0.60	22.74
AB3B7	2.52	1.93	1.20	1.94	0	-0.47	-0.46	0
AN1N8	2.00	1.81	0.74	1.95	0.015	0.33	0.47	0.33
AN2N7	1.02	0.85	0.45	1.76	0.002	-0.05	0.86	4.47
AN1aN8a	1.81	1.04	0.56	1.91	0.0004	-0.08	0.79	18.98
AN4N8	2.56	2.18	0.93	2.10	0	0.32	0.24	0
AN4aN8a	1.30	0.74	0.73	2.00	0	-0.72	0.54	37.97
AN3N7	2.79	2.17	1.40	2.24	0	-0.63	-0.56	0

^a The subscript v indicates vertical $E(S_1)$, which was calculated at the S_0 structure and with ZPE-correction. ^b The undoped anthracene.

the corresponding Lewis structure is assigned in Figure 4(c).

The HOMO and LUMO of **AB1B8** are shown in Figure S5(b) and (c). The HOMO-LUMO gap opening is induced by the bonding interaction between the allyl SOMOs and the empty B p_π orbitals. A similar situation is seen for **AB2B7** (Figure S6(c) and (d)). The gap openings are so small that the two molecules have low-lying T_1 states. Actually, after the T_1 geometry optimization, the T_1 state is found to be the ground state of **AB2B7**. While both molecules have low $E(T_1)$ s, their $E(S_1)$ s differ by ~ 1 eV. The configuration that contributes the largest portion to the S_1 state of **AB1B8** is HOMO-1 \rightarrow LUMO, followed by HOMO \rightarrow LUMO, giving a low $r_{HL} = 0.79$. The two configurations also make the largest contributions to the S_1 of **AB2B7**, but with $r_{HL} = 5.25$. While the two sites model is still applicable for **AB2B7**, it is invalid for **AB1B8**. This difference is explained using the localized orbitals of the relevant CTs. One of the HOMO-LUMO localized orbitals of **AB1B8**

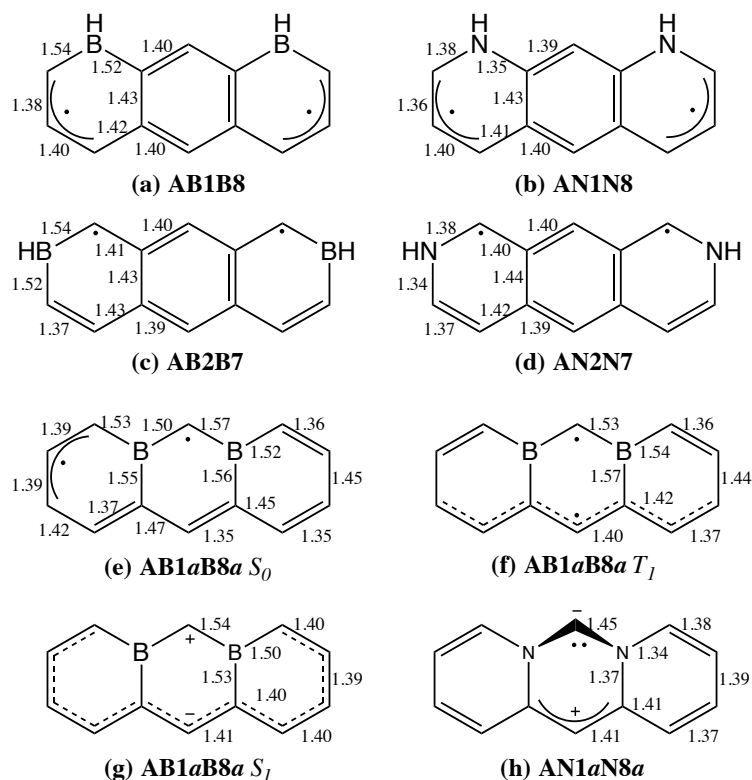


Figure 4: Dominant Lewis structures of 2B-doped anthracenes with large diradical characters, and their corresponding 2N-doped molecules. The bond lengths are given in Å. The Lewis structures and bond lengths are given for ground states of the molecules, unless further specified.

is shown in Figure S5(e). The HOMO→LUMO excitation corresponds to a CT between this orbital and its symmetry image. This CT can be viewed as between the two allyl fragments in Figure 4(a). It is across the whole molecule along the long axis and needs a high energy ($2K_{H,L} = 3.54$ eV). The HOMO–1-LUMO localized orbitals are highly delocalized along the long axis (Figure S5(f)) and have a much better spatial overlap, substantially lowering the CT energy to $2K_{H-1,L} = 0.96$ eV. The difference in CT energies compensates the difference between $\Delta_{H,L}$ and $\Delta_{H-1,L}$, so that the HOMO–1 →LUMO configuration has a lower energy than the HOMO→LUMO configuration, and dominates in the S_1 state.

The HOMO-LUMO and HOMO–1-LUMO localized orbitals of **AB2B7** are shown in Figure S6(g) and (h). Compared to the **AB1B8** counterpart, the HOMO-LUMO localized orbital is more evenly distributed over the left and central ring. Therefore, the CT between

1
2
3 this orbital and its symmetry image consists of more portion of short distance transfer
4 between the terminal and the central ring, with $2K_{H,L} = 2.00$ eV. A lower energy for
5 the HOMO→LUMO excitation is expected. The orbital in Figure S6(h) is obviously more
6 localized than the one in Figure S6(g), suggesting a slightly higher CT energy ($2K_{H-1,L} =$
7 2.06 eV) for the HOMO-1 →LUMO excitation. The similar CT energies cannot compensate
8 the difference between $\Delta_{H,L}$ and $\Delta_{H-1,L}$, and therefore, less HOMO-1 →LUMO percentage
9 in the S_1 state of **AB2B7** is seen. Although the S_1 state is dominated by HOMO→LUMO,
10 the dominance is way less than in the S_1 state of anthracene ($r_{HL} = 5.25$ vs. 28.44). This is
11 a reflection of the $\Delta_{H-1,L}$ reduction on the 2B-doping.
12
13
14
15
16
17
18
19
20

21 The transition dipole moments between the molecular orbitals are substantial: 3.83
22 and 1.52 a.u. for the HOMO-LUMO and HOMO-1-LUMO pairs of **AB1B8**; 2.35 and
23 2.42 a.u. for the corresponding pairs of **AB2B7**. However, there is a significant de-
24 structive interference of those transition dipoles in the mixing of the HOMO→LUMO and
25 HOMO-1 →LUMO configurations, resulting in the small $f_{S_1} = 0.015$ and 0.002 of the two
26 molecules. The destructive interference is rooted in the minimization of the CT energy in the
27 S_0 - S_1 excitation. Shown in Figure S7(a) and (b) are the HONO and LUNO of the S_1 state of
28 **AB1B8**. Their close-to-1 occupancies indicate a good one-electron excitation scheme. The
29 mixing of HOMO→LUMO and HOMO-1 →LUMO configurations is essentially to mix the
30 HOMO and HOMO-1 to result in a hole orbital that resembles the HONO. Clearly, the
31 HONO and LUNO have their amplitudes on different sets of atoms, and their $2K_{\text{HONO,LUNO}}$
32 is minimized to 0.31 eV. The CT energy in the S_0 - S_1 excitation has thus been minimized.
33 This mixing occurs whenever the CT energy lowering overcomes the cost of the larger or-
34 bital energy gap $\Delta_{H-1,L}$ involved in the excitation. The LUNO resembles the LUMO. The
35 LUMO has a less tendency to mix with the LUMO+1 to achieve the lower CT energy than
36 the HOMO-1-HOMO mixing because $\Delta_{L,L+1} > \Delta_{H-1,H}$. Now with the hole and particle
37 orbitals (HONO and LUNO) on different sets of atoms, their transition dipole integral must
38 be small, explaining the small $f_{S_1} = 0.015$.
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 It is interesting to inspect the T_1 HONO and LUNO, which are shown in Figure S7(c)
4 and (d). They resemble more the HOMO and LUMO. There is some discernible difference
5 between the HONO and HOMO (Figure S5(b)), which arises from mixing a small portion of
6 the HOMO-1 to the HOMO so that the amplitudes on C1a and C8a are reduced and those
7 on C2 and C7 are enhanced. The purpose is to enhance the spatial overlap of the hole and
8 particle orbitals and thus increase the $2K_{\text{HONO,LUNO}}$ to 5.02 eV, since this exchange integral
9 is a stabilizing factor for the triplet excitation. The arguments in these two paragraphs also
10 apply to **AB2B7**, whose S_1 and T_1 HONOs and LUNOs are shown in Figure S7(e)-(h).

11
12 The S_0 state of **AB1aB8a** underwent a distortion to a C_s planar structure during ge-
13 ometry optimization. The 2B-doping isolates two odd-numbered C fragments, C9 and the
14 leftover 11-membered C chain. The distortion concentrates the radical center on one side of
15 the C chain, which interacts more effectively with the C9 radical through B8a. The bond
16 lengths (Figure 4(e)) indicate that it is an allyl moiety that contains the radical center on
17 the C₁₁ chain, which is corroborated by the HONO and LUNO (Figure S4(i) and (j)). We
18 assigned the Lewis structure in Figure 4(e). The molecule has the smallest $\Delta_{H,L} = 1.22$ eV
19 among all studied molecules. The HOMO-1 to LUMO+1, and the relevant localized orbitals
20 are shown in Figure S8(a)-(d). The HOMO-LUMO localized orbitals (Figure S8(e) and (f))
21 are concentrated at the central ring, with small extensions on the side rings. With their
22 good spatial overlap, the CT energy of HOMO→LUMO is relatively low ($2K_{H,L} = 2.02$ eV).
23 The HOMO-1-LUMO localized orbitals are shown in Figure S8(g) and (h), which also sug-
24 gest a short distance CT ($2K_{H-1,L} = 1.64$ eV). The not much larger $2K_{H,L}$ and the small
25 $\Delta_{H,L}$ are responsible for the dominance of HOMO→LUMO in S_1 , with $r_{HL} = 7.29$. The
26 HOMO-1-LUMO configuration makes the second largest contribution to S_1 . The HOMO-
27 LUMO transition dipole moment has a small magnitude of 0.99 a.u, and it has a destructive
28 interference with the 2.48 a.u. large HOMO-1-LUMO transition dipole, resulting in the
29 small $f_{S_1} = 0.002$.

30
31 Both the S_1 and T_1 states reorganize to C_{2v} structures, with 0.67 and 0.27 eV reorgani-
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54

1
2
3 zation energies, respectively. The two SOMOs of the T_1 state are shown in Figure S8(i) and
4 (j). They resemble the HOMO-LUMO localized orbitals in Figure S8(e) and (f), as expected.
5
6 The SOMOs do not interact in T_1 , and therefore there is no tendency to distort the structure
7 to concentrate them on one side of the molecule and enhance their interaction, as in the S_0
8 state. Actually, the S_0 HOMO and LUMO clearly arise from the bonding and antibonding
9 mixings of the the SOMOs of A_2 and B_2 IRREPs, respectively, which is turned on by the
10 pseudo-Jahn-Teller distortion⁵⁵ along a b_1 mode from C_{2v} to C_s . The S_1 state at the opti-
11 mized structure features a CT from the SOMO in Figure S8(j) to the SOMO in Figure S8(i),
12 corresponding to the Lewis structure in Figure 4(e). Compared to the S_0 structure, the S_1
13 state features a bond length equalization of the undecapentenyl anion fragment, and thus
14 the 0.67 eV large reorganization energy.
15
16
17
18
19
20
21
22
23
24

25 While all the three 2B-doped anthracenes have large positive $\Delta_{SF}^{(1)}$ s and $\Delta_{SF}^{(2)}$ s, their
26 too low $E(T_1)$ s imply instabilities. Also, not many acceptors can harvest the low-lying T_1
27 excitons, and the low energy excitons tend to lead to a low open-circuit voltage.⁵⁶ The 2B-
28 doped anthracenes are hence not good SF chromophore candidates. The $K_{a,b}$ s of the three
29 2B-doped molecules are larger than that of anthracene (Figure 5(a)). When we remove the B
30 atoms, the leftover C fragments are all diradicals with ferromagnetic couplings, and therefore
31 their SOMOs are not AH paired, resulting in nonnegligible exchange integrals between each
32 pair of the SOMOs. These integrals are carried onto the 2B-doped molecules, where the
33 SOMOs become HOMOs and LUMOs. The large $K_{a,b} = 0.56$ and 0.66 eV of **AB2B7** and
34 **AB1aB8a** contribute to their exceptionally low $E(T_1)$ s. The unpaired electrons densities
35 of **AB1B8**, **AB2B7**, and **AB1aB8a** are shown in Figure S9. They are consistent with the
36 diradical Lewis structures in Figure 4. Since the unpaired electrons densities deliver the
37 same information as the HONO and LUNO, they are not shown for the molecules discussed
38 below.
39
40
41
42
43
44
45
46
47
48
49
50
51
52

53 Figure 3(a) shows that all 2N-doped molecules have lower diradical characters than their
54 2B-doped analogues. This is consistent with the larger $\Delta_{H,L}$ s of all 2N-doped molecules
55
56
57
58
59
60

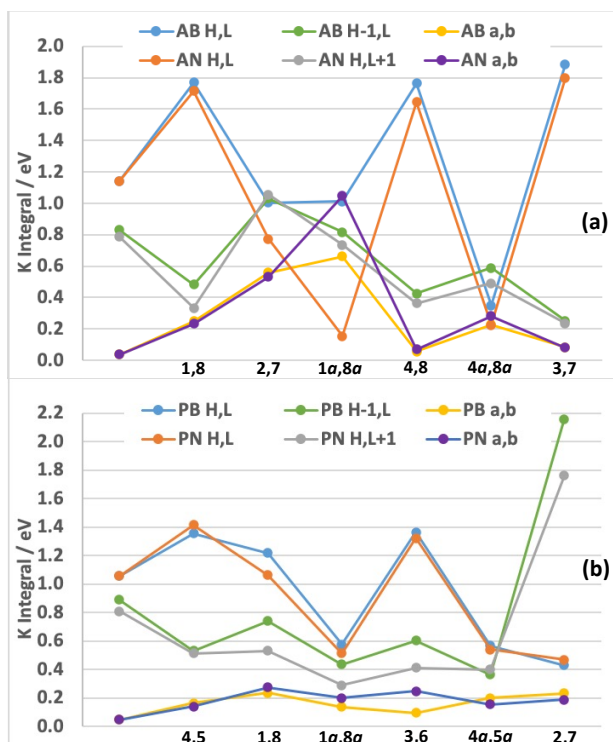


Figure 5: The calculated exchange integrals for canonical orbitals and localized orbitals of the 2B- and 2N-doped anthracenes (a) and phenanthrenes (b). The doping positions are shown at the bottom of each panel, beneath the corresponding markers. The first markers on the left are for the undoped molecules. The “H,L”, “a,b” etc. in the legends denote the subscripts of the K integrals. For **PB2B7**, the integrals are for natural orbitals and their localized orbitals (see the relevant discussion in text).

as shown in Figure 3(c). The $K_{H,L}$ s of the 2N-doped molecules, except **AB1aB8a**, are close to the 2B analogues (Figure 5(a) and (b)). With the smaller diradical characters, the 2N-doped molecules are anticipated to have higher $E(T_1)$ s. However, it is necessary to point out that this anticipation is not always met, especially after considering possibly different T_1 reorganization energies of the 2B- and 2N-doped molecules and their different $K_{a,b}$ s. **AN1N8**, **AN2N7**, and **AN1aN8a** have their $E(T_1)$ s higher by ~ 0.4 eV than their 2B analogues. Their $E(S_1)$ s are also higher. All three have $\Delta_{SF}^{(2)} > 0$, and $\Delta_{SF}^{(1)} > 0$ or just slightly negative. Of special interest is **AN1N8**, whose $E(T_1)$ is 0.74 eV. It is not much lower than the 0.86 eV $E(T_1)$ of pentacene, which is the most intensely investigated SF chromophore.

Like **AB1B8**, the S_1 state of **AN1N8** is not dominated by the HOMO \rightarrow LUMO config-

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
uration, but by the HOMO→LUMO+1 configuration, with $r_{HL} = 0.33$. The mirror relation holds between **AB1B8** and **AN1N8**, and as shown below, for most pairs of 2B- and 2N-dopings. The correlations between the canonical orbitals of **AB1B8** and **AN1N8** are shown in Figure S5(m). The HOMO and LUMO of **AN1N8** (Figure S5(h) and (i)) correlate to the LUMO and HOMO of **AB1B8** (Figure S5(c) and (b)). The bonding interaction with B in the **AB1B8** HOMO becomes antibonding interaction with N in **AN1N8**. This is another mirror relation anticipated based on the acceptor and donor characters of the two dopants and the first order perturbation theory. Overall, the dopings can be viewed as introducing perturbations to the π system of the leftover C_{12} system. Bonding perturbations of introducing acceptors must correspond to antibonding perturbations of introducing donors. The nonbonding LUMO of **AB1B8** remains nonbonding in **AN1N8** and thus becomes HOMO. Given the similar HOMO-LUMO pairs, the HOMO-LUMO localized orbitals of **AN1N8** looks similar to those of **AB1B8** (Figure S5(e) vs. (k)), except replacing the B-C bonding by N-C antibonding. A similar long distance and high energy CT ($2K_{HL} = 3.42$ eV) between the two terminal rings occurs in the HOMO→LUMO excitation, diminishing its percentage in S_1 of **AN1N8**.

35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
The HOMO-LUMO+1 localized orbital is shown in Figure S5(l), and it is evenly distributed across the molecule, just like the HOMO−1-LUMO localized orbital of **AB1B8** (Figure S5(f)). The obvious topological similarity between the two localized orbitals is not accidental. In both molecules, the HOMO−1 and LUMO+1 orbitals are not perturbed by the dopants since they have antisymmetric phases around each dopant (Figure S5(a),(d),(g),(i)). Therefore, there are correlations between the HOMO−1 orbitals and between the LUMO−1 orbitals in terms of their C-C bonding patterns. Within each molecule, the HOMO−1 and LUMO+1 orbitals are paired by AH symmetry. The correlations and pairings together determine that the HOMO−1 of **AB1B8** and the LUMO+1 of **AN1N8** are distributed on the same set of C atoms, and can be viewed as a bonding-antibonding pair (Figure S5(a) vs. (j)). Together with the correlation between the HOMO of **AN1N8** and the LUMO of

AB1B8, the HOMO-LUMO+1 localized orbitals of the former molecule must spread over the same set of atoms as the HOMO-1-LUMO localized orbitals of the latter molecule, along with the opposite bonding/antibonding characters between adjacent atoms (Figure S5(f) vs. (l)). A short distance low energy CT ($2K_{H,L+1} = 0.66$ eV) is hence expected for the HOMO→LUMO+1 excitation in **AN1N8**, similar to the HOMO-1 →LUMO excitation in **AB1B8**. Consequently, the HOMO→LUMO+1 excitation dominates the S_1 state of **AN1N8**.

Overall, the mirror relation between the S_1 states' compositions of **AB1B8** and **AN1N8** is a result of the following: (1) the mirror relation between LUMO energy decrease of the 2B-doping and HOMO energy increase of the 2N-doping; (2) the HOMO-to-LUMO and LUMO-to-HOMO correlations between the two molecules; (3) the AH pairing of doubly occupied and unoccupied orbitals of the C_{12} fragment. All these apply to the **AB2B7-AN2N7** pair. Therefore, the S_1 of **AN2N7** is dominated by HOMO→LUMO, and supplemented by HOMO→LUMO+1, with $r_{HL} = 4.47$. The orbital correlation diagram between **AB2B7** and **AN2N7** appears to be different from that between **AB1B8** and **AN1N8** (Figure S6(q) vs. Figure S5(m)). But the differences in the C-C bonding pattern correlations are counterpoised by the differences in the AH pairing correlations. Therefore, the HOMO-1-LUMO localized orbitals of **AB2B7** and the HOMO-LUMO+1 localized orbitals of **AN2N7** are similarly correlated as the localized orbitals of **AB1B8** and **AN1N8**: (1) B-C bonding is replaced by N-C antibonding; (2) C-C bonding (antibonding) is replaced by C-C antibonding (bonding).

As mentioned above, the mixings of the HOMO-1 →LUMO and HOMO→LUMO configurations and their destructive interferences of transition dipoles in the S_1 states of **AB1B8** and **AB2B7** are driven by the hole orbital polarizations that minimize the CT energies in the S_0 - S_1 excitation. A similar argument applies to the HOMO→LUMO and HOMO→LUMO+1 configurations in **AN1N8** and **AN2N7**, but instead with the particle orbital polarizations realized by mixing LUMO and LUMO+1. The hole and particle orbitals (HONOs and LUNOs) of the S_1 state of **AN1N8** are shown in Figure S7(i) and (j). The

LUNO is obviously a mixture of LUMO and LUMO+1. With the mixture, the LUNO and HONO have amplitudes on different sets of atoms (as much as possible), so that both the CT energy and transition dipole are reduced. The f_{S_1} s of **AN1N8** and **AN2N7** are hence as small as those of the 2B counterparts.

As an exception to the caption of this section, **AN1aN8a** has the lowest diradical character among all studied molecules (Figure 3(a)). The molecule is placed in this section to be compared with **AB1aB8a**. The mirror relation in orbital energy changes does not hold for the **AB1aB8a-AN1aN8a** pair (Figure 3(b)). The $\Delta_{H,L}$ of **AN1aN8a** is exceptionally large (Figure 3(c)). The molecule possess a C_s puckered S_0 structure. The HOMO and LUMO are shown in Figure S8(l) and (m). They correlate with the LOMO and HOMO of the S_1 state of **AB1aB8a** (Figure S8(j) and (i)). Therefore, a similar ionic Lewis structure is assigned to **AN1aN8a** (Figure 4(h)), but with opposite charge localizations. The two butadienyl fragments are assigned to the two sides based on the HONO-1 to LUNO+1 (Figure S4(k)-(n)), which are symmetrized and antisymmetrized combinations of HONOs and LUNOs of the two butadienyls. The nonplanar structure reduces the conjugation between the terminal and the central ring, resulting in the localized π systems. C9 is sandwiched by two electronegative N atoms and is thus capable of accommodating another electron. The cramming of six electrons in the N-C9-N π system maximizes the π antibonding interaction among the atoms. A pyramidalization of C9 ensues. The low adiabatic $E(S_1) = 0.59$ eV of **AB1aB8a** hints at the possibility of swapping diradical and ionic characters of S_0 and S_1 , which is indeed the case in **AN1aN8a**. The strong ionic character of **AN1aN8a** explains its low diradical character. The different canonical orbitals also lead to the different exchange integrals between this pair of 2B- and 2N-doping in Figure 5(a).

The low diradical character is not incompatible with a low $E(T_1)$. The antibonding interaction within the N-C9-N moiety is diminished in the HOMO \rightarrow LUMO excitation, so that both S_1 and T_1 undergo reorganization back to C_{2v} planar structure. The associated 0.95 eV large reorganization energy lowers $E(T_1)$ to 0.56 eV. The S_1 reorganization energy

1
2
3 is 0.76 eV. This smaller S_1 reorganization energy is important for the molecule to have just
4 a slightly negative $\Delta_{SF}^{(1)}$. The CT in the HOMO→LUMO excitation is of low energy; when
5 HOMO and LUMO are localized on different sets of atoms, their “localized” orbitals are
6 delocalized and have good spatial overlap, which speaks for a small $2K_{H,L} = 0.30$ eV and
7 also a large $K_{a,b} = 1.05$ eV. The large $K_{a,b}$ facilitates to lower $E(T_1)$ to satisfy Eq. 1. The
8 localized orbitals of HOMO-LUMO and HOMO-LUMO+1 pairs are shown in Figure S8(o)
9 and (p). The latter pair of localized orbitals span mainly over one terminal ring and the
10 central ring, giving $2K_{H,L+1} = 1.46$ eV. Without the benefit of a shorter distance CT,
11 the HOMO→LUMO+1 configuration is not competitive, and the S_1 state is dominated by
12 HOMO→LUMO, with $r_{HL} = 18.98$. The comparison between **AN1aN8a** and **AB1aB8a**
13 cautions that the mirror relations between the 2B- and 2N-dopings shall be used carefully
14 in future studies of proposing SF chromophores, and more broadly, other optoelectronic
15 materials.
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30

31 Doped anthracenes with intermediate diradical characters

32
33 The other three 2B-doped anthracenes, **AB4B8**, **AB4aB8a**, and **AB3B7**, possess far
34 less diradical characters (Figure 3(a)). Correspondingly, their $E(T_1)$ s are higher. These
35 molecules are of C_{2h} symmetry. They belong to Group II chromophores classified by Stoy-
36 cheva et al.²⁹ Among them, **AB4B8** has the largest diradical character and is the only
37 one that satisfies both Eqs. 1 and 2. The HONO and LUNO (Figure S10(a) and (b))
38 of the molecule show clear characters of *p*-quinodipropane (PQDP), with the two allyls
39 at the two terminal rings interacting with the central phenyl. The corresponding Lewis
40 structure is assigned in Figure 6(a). The canonical orbitals and their localized orbitals
41 are shown in Figure S11(a)-(i). As expected, the HOMO-LUMO localized orbitals (Fig-
42 ure S11(g)) are concentrated on the terminal allyls that are separated along the diagonal
43 direction of the molecular plane. The localization of the orbitals is well reflected by the
44 small $K_{a,b} = 0.06$ eV. A long distance CT is involved in the HOMO→LUMO excitation
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

($2K_{H,L} = 3.52$ eV). The HOMO–1–LUMO localized orbitals are shown in Figure S11(h) and (i) and exhibit a better spatial overlap. A short distance CT ($2K_{H-1,L} = 0.84$ eV) is involved in the HOMO–1 \rightarrow LUMO excitation. The S_1 state is hence dominated by HOMO–1 \rightarrow LUMO. Also, since the HOMO–1 \rightarrow LUMO and HOMO \rightarrow LUMO configurations are of A_g and B_u IRREPs, respectively, they do not interact and mix. The S_1 state does not contain HOMO \rightarrow LUMO, and because of symmetry, it is a dark state with $f_{S_1} = 0$. The HOMO \rightarrow LUMO configuration makes the largest contribution to S_2 , with vertical $E(S_2) = 2.92$ eV and $f_{S_2} = 0.196$.

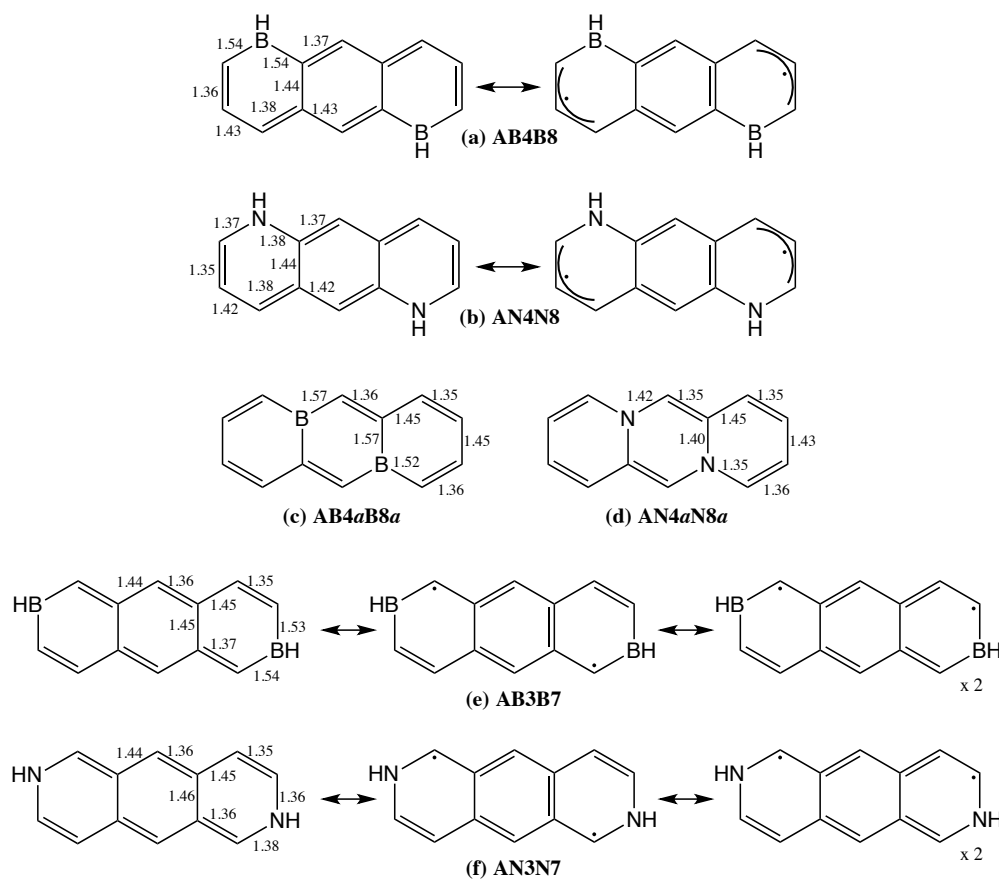


Figure 6: Dominant Lewis structures of 2B-doped anthracenes with intermediate diradical characters, and their corresponding 2N-doped molecules. The bond lengths are given in Å. The Lewis structures and bond lengths are given for ground states of the molecules. The “ $\times 2$ ” in rightmost Lewis structures in (e) and (f) indicate that there are two such structures that are symmetry images of each other.

It is interesting to compare **AB4B8** and **AB1B8**, whose C_{12} fragments are PQDP and

1
2
3 MQDP, respectively. According to Ovchinnikov's rule,⁵⁷ MQDP has a ferromagnetic cou-
4 pling of the two radical centers, while PQDP has an anti-ferromagnetic coupling and hence
5 **AB4B8** possesses a lower diradical character.²⁹ Also, the HOMO and LUMO of **AB4B8** are
6 AH paired, while those of **AB1B8** are not. The similar terminal allyls Lewis structures lead
7 to long distance CT between the terminal rings in the respective HOMO→LUMO excitations.
8 Therefore, both molecules have their S_1 states being dominated by HOMO-1 →LUMO ex-
9 citation. Because of their different symmetries, the **AB1B8** S_1 consists of HOMO→LUMO
10 configuration, which is absent in the **AB4B8** S_1 . The hole orbital polarization through
11 mixing HOMO-1 and HOMO does not occur in **AB4B8**'s S_1 state because of the different
12 IRREPs of the canonical orbitals. **AB4B8** has a not too low $E(T_1)$ and its vertical absorp-
13 tion to S_2 occurs at the blue region of solar spectrum with fairly high irradiance.⁵⁸ It is a
14 candidate of SF chromophore.

15
16
17
18
19
20
21
22
23
24
25
26
27 The natural orbitals (Figure S12(a)-(d)) and bond lengths of **AB4aB8a** indicate a Lewis
28 structure with two hexatrienyl fragments (Figure 6(c)). The pseudo-degenerate $n_{\text{HONO}-1} =$
29 1.88 and $n_{\text{HONO}} = 1.86$ suggest that the localized orbitals of the HONO-1-HONO pair
30 reflect the ground state wave function to a similar extent. They (Figure S12)(e) and (f)
31 show characters of the HONOs of the two hexatrienyls. The pseudo-degenerate $n_{\text{LUNO}} = 0.16$
32 and $n_{\text{LUNO}+1} = 0.13$ likewise allow us to use the localized orbitals of the LUNO-LUNO+1 pair
33 to view the ground state, and they (Figure S12)(g) and (h)) resemble the two LUNOs of the
34 hexatrienyls. The short polyenyl fragments feature low diradical character, speaking for the
35 smallest n_{LUNO} among the 2B-doped anthracenes. The canonical orbitals and their localized
36 orbitals of the molecule are shown in Figure S13(a)-(g). The HOMO-LUMO localized orbitals
37 show a better spatial overlap than those of the HOMO-1-LUMO pair ($2K_{H,L} = 0.70$ eV
38 vs. $2K_{H-1,L} = 1.18$ eV). Naturally, the S_1 state is dominated by HOMO→LUMO, with
39 $r_{HL} = 22.74$. It is a dark state because of symmetry. The short distance CT explains the
40 low $E(S_1)$ and the failure to satisfy Eq. 1.

41
42
43
44
45
46
47
48
49
50
51
52
53
54
55 The HONO and LUNO of **AB3B7** (Figure S14(b) and (c)) and the bond lengths indicate

1
2
3 the Lewis structure in Figure 6(e). The diradical character of this molecule is associated to
4 the central *p*-quinodimethane (PQDM) fragment. The HONO and HONO−1 are pseudo-
5 degenerate ($n_{\text{HONO}-1} = 1.88$ and $n_{\text{HONO}} = 1.82$). Their localized orbitals (Figure S14(e))
6 resemble the HONOs of the two hexatrienyls in Figure 6(e). The LUNO and LUNO+1
7 are less pseudo-degenerate ($n_{\text{LUNO}} = 0.21$ and $n_{\text{LUNO}+1} = 0.12$). Their localized orbitals
8 (Figure S14(f)) resemble LUNOs of the hexatrienyls. The localized natural orbitals indicate
9 multiple diradical characters in addition to PQDM, i.e., those involving the hexatrienyl
10 diradical structures.
11
12

13
14
15 The relation between **AB4B8** and **AB3B7** is similar to the relation between **AB1B8**
16 and **AB2B7**. Therefore, **AB3B7** shares a few common features with **AB4B8**, $r_{HL} = 0$,
17 and an S_1 dominated by HOMO−1 →LUMO being dark because of symmetry. However, the
18 **AB3B7** does not satisfy Eq. 1, a result of its smaller diradical character. Shorter exocyclic
19 odd-numbered C fragments at *p* positions have larger SOMO lobes in contact with the
20 phenyl frontier orbitals, leading to a stronger anti-ferromagnetic coupling and lower diradical
21 character. Correspondingly, the $\Delta_{H,L}$ of **AB3B7** is larger than the **AB4B8** counterpart
22 (3.99 vs 3.13 eV), while their $K_{H,LS}$ are similar (1.88 vs 1.76 eV). The diradical character of
23 the former is hence smaller. This molecule features a low $E(T_2)$ and does not satisfy Eq. 2
24 either. As mentioned above, this is attributed to the multiple diradical characters of PQDM
25 + hexatrienyls. The pseudo-degeneracy between HONO−1 and HONO is associated with the
26 0.64 eV small $\Delta_{H-1,H}$ (Figure 2(a)), which is responsible for the low-lying T_2 characterized by
27 HOMO−1 →LUMO excitation. It is necessary to point out the difference between multiple
28 diradical characters and tetraradical character. A substantial tetraradical character in the
29 ground state of a molecule can lead to a low-lying T_2 state with the same tetraradical
30 character,¹⁵ which is disadvantageous for satisfying Eq. 2. In the multiple diradical characters
31 case, the low-lying T_2 is still a diradical state.
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52

53 The mirror relations hold for the 2N-doped counterparts. **AN4N8**, **AN4aN8a**, and
54 **AN3N7** share similar natural orbitals (Figures S10, S12, S14) and adopt similar Lewis
55
56
57
58
59
60

1
2
3 structures as the 2B analogues (Figure 6(b),(d),(f)). The 2N-doped molecules have similar S_1
4 characters as the 2B species, except that the roles of HOMO-1 \rightarrow LUMO in the 2B analogues
5 are taken by HOMO \rightarrow LUMO+1. Note that the small $\Delta_{H-1,H}$ of **AB3B7** is translated to
6 the small $\Delta_{L,L+1}$ of **AN3N7** (Figure 2(a) and (b)), which also features multiple diradical
7 characters, and consequently a low-lying T_2 and the failure to satisfy Eq. 2. The topological
8 correlations between frontier canonical orbitals of the 2B- and 2N-doped molecules are given
9 in Figures S11(s), S13(o), and S15(o). These correlations underly good parallel variations
10 of the key exchange integrals (Figure 5(a)) and the mirror relations between the two sets of
11 molecules. **AN4N8** is the only one among the three that satisfy Eqs. 1 and 2. The generally
12 smaller diradical character induced by 2N-doping gives **AN4N8** a higher $E(T_1) = 0.93$ eV,
13 which is beneficial for harvesting T_1 excitons and high open-circuit voltage.
14
15
16
17
18
19
20
21
22
23
24
25
26

27 28 2B-doped phenanthrenes

29
30 All the studied 2B-doped phenanthrenes have relatively mild diradical characters (Fig-
31 ure 3(a)). Yet, five of the six (except **PB2B7**) have low enough $E(T_1)$ s to satisfy Eq. 1.
32 All the calculated results for the excited states of the 2B- and 2N-doped phenanthrenes are
33 summarized in Table 2. The detailed analyses of the electronic structures of the 2B-doped
34 phenanthrenes are given in SI (Section S3). Similar factors determine the S_1 compositions
35 and the applicability of the two sites model: $\Delta_{H-1,H}$, $K_{H,L}$ vs. $K_{H-1,L}$, and hole orbital
36 polarization. The hole orbital polarization underlies small f_{S_1} . Regardless of the applica-
37 bility of the two sites model, large diradical character and $K_{a,b}$ favor low $E(T_1)$ and the
38 satisfaction of Eq. 1. The different hole particle polarizations in S_1 and T_1 can explain the
39 different reorganization energies of the two states. The satisfaction of Eq. 1 can be reached
40 by the larger T_1 reorganization energy. Small $\Delta_{H-1,H}$ may result in multiple diradical char-
41 acter, which speaks for low $E(T_2)$, and the consequent possible crossing of T_1 and T_2 PESs,
42 as well as the propensity of not satisfying Eq. 2. $\Delta_{H-1,H} \approx 0$ also leads to ambiguity of
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

r_{HL} . For such cases, it is more reasonable to use $r_{HONO,LUNO}$ to characterize S_1 . Among the 2B-doped phenanthrenes that satisfy Eqs. 1 and 2, **PB1aB8a** is the best SF chromophore candidate, as it has the highest 0.62 eV $E(T_1)$. The disadvantage of its 0.018 low f_{S_1} can be counterpoised by the high irradiance at the ~ 1.90 eV (vertical $E(S_1)$) region of the solar spectrum.

Table 2: Calculated quantities for the 2B- and 2N-doped phenanthrenes discussed in this paper. All energies are given in eV. Excitation energies are relative to $E(S_0)$ of respective molecules.

	$E_v(S_1)^a$	$E(S_1)$	$E(T_1)$	$E(T_2)$	f_{S_1}	$\Delta E_{SF}^{(1)}$	$\Delta E_{SF}^{(2)}$	r_{HL}
P ^b	3.98	3.81	2.82	3.60	0.0003	-1.83	-2.03	0
PB4B5	1.78	1.50	0.34	2.09	0.054	0.82	1.41	6.61
PB1B8	1.87	1.59	0.37	2.04	0.067	0.85	1.30	4.33
PB1aB8a	1.90	1.41	0.62	1.80	0.018	0.17	0.56	11.86
PB3B6	2.51	2.03	0.86	1.80 ^c	0.009	0.31	0.08	0.48
PB4aB5a	1.37	1.18	0.50	1.66	0.029	0.18	0.66	14.06
PB2B7	2.37	1.94	1.05	1.73	0.004	-0.16	-0.37	3.8 ^d
PN4N5	1.96	1.44	0.49	2.13	0.061	0.46	1.15	7.84
PN1N8	1.39	1.04	0.23	1.89	0.044	0.58	1.43	9.32
PN1aN8a	1.32	0.89	0.46	1.48	0.006	-0.03	0.56	9.00
PN3N6	1.62	0.96	0.16	0.76	0.002	0.64	0.44	0.10
PN4aN5a	1.53	1.12	0.61	1.85	0.020	-0.10	0.63	14.53
PN2N7	2.38	1.93	1.10	1.99	0.002	-0.27	-0.21	2.97

^a The subscript v indicates vertical $E(S_1)$, which was calculated at the S_0 structure and with ZPE-correction. ^b The undoped phenanthrene. ^c This is the $E(T_2)$ calculated at the T_1 structure. ^d This r_{HL} is for the S_1 wave function expressed using the natural orbitals of the state.

2N-doped phenanthrenes

Four (except **PN1N8** and **PN1aN8a**) of the six investigated 2N-doped phenanthrenes undergo pyramidalizations at their N centers and distort their structures to C_2 symmetry. This is different from the 2B-doped analogues, but not unexpected, since the doping of N atoms

comes with filling antibonding orbitals of phenanthrene. **PN4aN5a** has the most severe distortion because of the adjacency of the two N lone pairs. The C1a-N4a-N5a-C8a dihedral angle is 41° large. In general, none of the 2N-doped phenanthrenes are good SF chromophore candidates. **PN4aN5a** is the only one that has not too low $E(T_1)$ (0.61 eV). However, it has a negative $\Delta_{SF}^{(1)} = -0.10$ eV. We will not go over the molecules in details as for the 2B-doped analogues in SI. Instead, we focus more on the comparison of the two dopings.

Overall, the mirror relations between 2B- and 2N-dopings maintain. Figure 5(b) shows that the $K_{a,bS}$, $K_{H,LS}$, and $K_{H,L+1S}$ integrals of the 2N-doped phenanthrenes closely follow the same trends as the corresponding integrals of the 2B analogues. Note that $K_{H,L+1}$ of 2N-doping corresponds to $K_{H-1,L}$ of 2B-doping. The AH pairing of occupied and unoccupied orbitals underlies the parallel changes between the two sets of integrals. The arguments on the CT energies of the HOMO→LUMO and HOMO−1 →LUMO excitations of the 2B-doped phenanthrenes can be transplanted to the CT energies of the HOMO→LUMO and HOMO→LUMO+1 excitations of the 2N-doped molecules. The characters of the S_1 states of the 2N-doped molecules are in good correspondence to those of the 2B analogues, i.e., the r_{HLS} and f_{S_1S} of each pair of 2N- and 2B-dopings are qualitatively similar. Again, the role of HOMO−1 →LUMO in the S_1 states of the 2B species is taken by HOMO→LUMO+1. Natural orbitals (Figures S17, S21, S24, and S27) and bond lengths indicate that the 2N-doped species share similar resonance structures in their ground states with the 2B analogues (see Figure S18(b), (d), (f), (h), (j), and (l)). The good correlations between the molecular orbitals of the 2B- and 2N-doped species in Figures S16(o), S19(q), S20(o), S22(q), S25(o), and S28(s) underlie the mirror relations. Note that in Figure S28(s), the natural orbitals of **PB2B7** are correlated with the canonical orbitals of **PN2N7**, as the HOMO−1 and HOMO of the former molecule are ambiguous.

Conclusions and further discussions

So far, most of the theoretical studies on designing singlet fission chromophores are based on the two sites model. Despite its broad applicability, this model has its limit, that S_1 and T_1 shall arise from the same HOMO→LUMO excitation. This requirement cannot always be met, especially for molecules that possess close-lying HOMO−1 and HOMO, and/or close-lying LUMO and LUMO+1. The investigated 2B- and 2N-doped anthracenes and phenanthrenes feature such close-lying orbitals. Among the 24 molecules studied in this work, 8 cannot be described using the two sites model: **AB1B8**, **AB4B8**, **AB4aB8a**, and **PB3B6**, and their 2N-doped analogues. Actually, taking the undoped anthracene as the reference, then only for those molecules with their $r_{HL} > 20$, we may conclude that the model is applicable. For the others, the model becomes less and less valid as r_{HL} decreases. As shown in the two tables of this work, there is a continuous spectrum of r_{HL} , and there is no clear-cut threshold to signify the validity of the two sites model. We have used $r_{HL} > 1$ as a validity indicator. But this simply means that the molecules with $r_{HL} > 1$ are better (but not necessarily adequately) described using the two sites model than those with $r_{HL} < 1$.

Similar to the doping of acceptors to generate *p*-type semiconductors, the 2B-dopings systematically lower LUMO energies and reduce both HOMO-LUMO and HOMO−1-LUMO gaps. This reduced HOMO−1-LUMO gaps promote the HOMO−1 →LUMO configurations to compete with the HOMO→LUMO configurations to dominate the S_1 states. Triplet one-electron excitation is of pure diradical character and the excitation energy is more dependent on the orbital energy gap involved in the excitation. Therefore, all T_1 states of the doped molecules have the largest contributions from HOMO→LUMO excitations. Singlet one-electron excitation involves an extra charge-transfer energy. Once the HOMO−1 →LUMO excitation involves a shorter distance and then lower energy charge transfer (CT) than the HOMO→LUMO excitation, and this difference in CT energies counterpoises the difference in orbital energy gaps, S_1 is dominated by HOMO−1 →LUMO. The two sites model then becomes inapplicable. Even if the HOMO−1 →LUMO configuration does not dominate S_1 ,

1
2
3 it may have low enough energy to effectively interact with the HOMO→LUMO configu-
4 ration and lower the S_1 energy. In the language of hole and particle orbitals, this mixing
5 of configurations reflects the polarization of the hole orbital to minimize its exchange inte-
6 gral with the particle orbital. This polarization is realized through mixing HOMO−1 and
7 HOMO, and it unfavorably reduces transition dipole between the hole and particle orbitals,
8 and consequently the S_1 oscillator strength. The low $E(S_1)$ is disadvantageous for satisfying
9 the primary thermodynamic criterion for singlet fission, Eq. 1. On the other hand, the low
10 S_1 energy may be advantageous since the solar spectrum has higher irradiance at the low
11 energy region of red visible and near infrared photons. The smaller HOMO−1-LUMO gap
12 may also induce multiple diradical characters and a low-lying T_2 state, so that Eq. 2 is not
13 satisfied. **PB3B6** and **PB2B7** are two typical examples.

24
25 Similar to the doping of donors to generate n -type semiconductors, the 2N-dopings
26 systematically generate high-lying HOMOs, and reduce both HOMO-LUMO and HOMO-
27 LUMO+1 gaps. There is an interesting mirror relation in orbital energy changing: the
28 decreases of LUMO energy induced by 2B-dopings are similar to the increases of HOMO en-
29 ergy induced by 2N-dopings. The reductions of HOMO-LUMO and HOMO−1-LUMO gaps
30 of the 2B-doped molecules are hence similar to the reductions of HOMO-LUMO and HOMO-
31 LUMO+1 gaps of the corresponding 2N-doped molecules. Considering the alternant hydro-
32 carbon pairing theorem, the CT energies of the HOMO→LUMO configurations of a pair of
33 2B- and 2N-doped molecules are comparable. The CT energies of the HOMO−1 →LUMO
34 configuration of a 2B-doped molecule and the HOMO→LUMO+1 configuration of the cor-
35 responding 2N-doped molecule are also comparable. Consequently, the conclusions about
36 the 2B-doped molecules in the previous paragraph are generally transferrable to the 2N-
37 doped molecules, with the role of HOMO−1 →LUMO being taken by HOMO→LUMO+1,
38 and likewise, the polarization of hole orbital being replaced by the polarization of particle
39 orbital. The polarization is realized through mixing LUMO and LUMO+1. There are hence
40 additional mirror relations between 2B- and 2N-dopings in S_0 character (e.g., diradical char-
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

acter), S_1 character (e.g., r_{HL} and f_{S_1}), T_2 character (e.g., low $E(T_2)$ induced by multiple diradical characters), and key exchange integrals ($K_{a,b}$, $K_{H,L}$, and $K_{H-1,L}$ -vs- $K_{H,L+1}$). Despite the general mirror relations, there must be exceptions. **AN1aN8a** is such one. The high electronegativity of the N-C-N fragment gives an ionic ground state, vs. the diradicaloid ground state of **AB1aB8a**. This difference makes the pair incomparable.

Before this study, we tended to think that in general, a large $K_{H,L}$ corresponds to a small $K_{a,b}$, and vice versa. This is because when HOMO and LUMO are distributed on the same atoms, their localized orbitals, through cancelation of amplitudes in $\frac{1}{\sqrt{2}}$ (HOMO \pm LUMO), are distributed on different sets of atoms (e.g., starred and unstarred atoms of alternant hydrocarbons with antiferromagnetic coupling), and vice versa. This anticorrelation holds well in Figure 5(a) for the 2B- and 2N-doped anthracenes. The together increases of $K_{H,L}$ and $K_{a,b}$ from the undoped anthracene to **AB1B8** and **ABNBN** result from the loosening of the AH orbital pairing in the doped molecules. The anticorrelation, however, does not hold in Figure 5(b) for the doped phenanthrenes. **PB1aB8a** is a typical example with both small $K_{H,L}$ and $K_{a,b}$. Actually, when HOMO and LUMO are evenly distributed in a large number of atoms, even if they have amplitudes on the same atoms, their exchange integral is attenuated to a small quantity. This situation of having small $K_{H,L}$ and $K_{a,b}$ shall be avoided in designing singlet fission chromophores. The relation between $K_{H,L}$ and $K_{a,b}$ is a good subject for a future study.

All the 2B-doped molecules have been previously studied in the context of designing singlet fission chromophores. Most of the 2N-doped molecules investigated here are not good chromophore candidates, except only **AN1N8** and **AN4N8**. And the synthesizabilities of these molecules are unknown. More interesting results of this work, we believe, are the relations between 2B- and 2N-dopings, which can be generalized to di-acceptor and di-donor dopings, and the discussion of the applicability of the two sites model. Actually, the findings of this work are applicable for all systems with (pseudo)-degenerate HOMO and HOMO-1, and/or (pseudo)-degenerate LUMO and LUMO+1, regardless of the origin of the (pseudo-

1
2
3)degeneracy. Since the early studies of designing singlet fission chromophores, the two sites
4 model has been the most useful interpretive and predictive tool. The association of singlet
5 fission chromophore to diradical character has inspired the strategies of using captodative
6 effects^{36,59-64} and antiaromaticity^{14,22-24,65} to design the chromophores. A fundamental as-
7 sumption of the two sites model is that S_1 and T_1 share the same hole and particle orbitals,
8 and hence the same $K_{H,L}$ and $K_{a,b}$. The model certainly becomes invalid when the hole
9 and/or particle orbitals are polarized differently in S_1 and T_1 . And there is a strong mo-
10 tivation for the different polarizations, owing to the intrinsically different natures (charge
11 transfer vs. diradical) of singlet and triplet excitations. As more and more chromophores are
12 proposed, we have reached the boundary of the effective range of the two sites model and the
13 associated diradical character descriptor. While the diradical character beacon still sheds
14 light to the area of low T_1 energy, the S_1 energy, composition, and oscillator strength require
15 thoughts beyond the two sites model. Or, the two sites model needs to be amended with
16 consideration of different hole and/or particle orbitals in S_1 and T_1 . The knowledge that
17 we learn in this work will be of value in future studies of the unconventional chromophores,
18 whose S_1 is not dominated by HOMO→LUMO excitation.
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36

37 Acknowledgements

38
39
40 We thank York University for the start-up grant (TZ: 481333) and the Natural Sciences and
41 Engineering Research Council (NSERC) of Canada (TZ: RGPIN-2016-06276) for financial
42 support. We thank the Neese group at the University of Bonn for their continuous devel-
43 opment of the ORCA program package. We thank the Gordon group at the Iowa State
44 University for their continuous development of the GAMESS-US program package. This
45 research was enabled in part by support provided by WestGrid (www.westgrid.ca), SciNet
46 (www.scinethpc.ca), and Compute Canada (www.computecanada.ca).
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Supporting Information

See Supporting Information for comparison of calculations using cc-pVDZ and cc-pVTZ basis sets, calculated results of undoped anthracene and phenanthrene, detailed analyses of 2B-doped phenanthrenes, all plottings of orbitals, and coordinates of the investigated molecules.

References

- (1) Smith, M. B.; Michl, J. Singlet Fission. *Chem. Rev.* **2010**, *110*, 6891–6936.
- (2) Smith, M. B.; Michl, J. Recent Advances in Singlet Fission. *Annu. Rev. Phys. Chem.* **2013**, *64*, 361–386.
- (3) Michl, J. Unconventional Solar Energy: Singlet Fission. *Mol. Front. J.* **2019**, *3*, 84–91.
- (4) Shockley, W.; Queisser, H. J. Detailed Balance Limit of Efficiency of P-N Junction Solar Cells. *J. Appl. Phys.* **1961**, *32*, 510–519.
- (5) Hanna, M. C.; Nozik, A. J. Solar Conversion Efficiency of Photovoltaic and Photoelectrolysis Cells with Carrier Multiplication Absorbers. *J. Appl. Phys.* **2006**, *100*, 074510.
- (6) Krishnapriya, K. C.; Musser, A. J.; Patil, S. Molecular Design Strategies for Efficient Intramolecular Singlet Exciton Fission. *ACS Energy Lett.* **2019**, *4*, 192–202.
- (7) Japahuge, A.; Zeng, T. Theoretical Studies of Singlet Fission: Searching for Materials and Exploring Mechanisms. *ChemPlusChem* **2018**, *83*, 146–182.
- (8) Casanova, D. Theoretical Modeling of Singlet Fission. *Chem. Rev.* **2018**, *118*, 7164–7207.
- (9) Ito, S.; Nagami, T.; Nakano, M. Molecular Design for Efficient Singlet Fission. *J. Photochem. Photobiol.* **2018**, *34*, 85–120.
- (10) Xia, J.; Sanders, S. N.; Cheng, W.; Low, J. Z.; Liu, J.; Campos, L. M.; Sun, T. Singlet Fission: Progress and Prospects in Solar Cells. *Adv. Mater.* **2017**, *29*, 1601652.
- (11) Scholes, G. D. Correlated Pair States Formed by Singlet Fission and Exciton–Exciton Annihilation. *J. Phys. Chem. A* **2015**, *119*, 12699–12705.

- 1
2
3 (12) Paci, I.; Johnson, J. C.; Chen, X. D.; Rana, G.; Popović, D.; David, D. E.; Nozik, A. J.;
4 Ratner, M. A.; Michl, J. Singlet Fission for Dye-Sensitized Solar Cells: Can a Suitable
5 Sensitizer be Found? *J. Am. Chem. Soc.* **2006**, *128*, 16546–16553.
6
7
8
9
10 (13) Minami, T.; Ito, S.; Nakano, M. Theoretical Study of Singlet Fission in Oligorylenes.
11 *J. Phys. Chem. Lett.* **2012**, *3*, 2719–2723.
12
13
14 (14) Ito, S.; Minami, T.; Nakano, M. Diradical Character Based Design for Singlet Fission
15 of Condensed-Ring Systems with $4n$ Electrons. *J. Phys. Chem. C* **2012**, *116*, 19729–
16 19736.
17
18
19
20
21 (15) Minami, T.; Nakano, M. Diradical Character View of Singlet Fission. *J. Phys. Chem.*
22 *Lett.* **2012**, *3*, 145–150.
23
24
25
26 (16) Minami, T.; Ito, S.; Nakano, M. Fundamental of Diradical-Character-Based Molecular
27 Design for Singlet Fission. *J. Phys. Chem. Lett.* **2013**, *4*, 2133–2137.
28
29
30
31 (17) Padula, D.; Omar, Ö. H.; Nematiram, T.; Troisi, A. Singlet Fission Molecules among
32 Known Compounds: Finding a Few Needles in a Hay Stack. *Energy Environ. Sci.* **2019**,
33 *12*, 2412–2416.
34
35
36
37
38 (18) Schwerin, A. F.; Johnson, J. C.; Smith, M. B.; Sreearunothai, P.; Popović, D.; Černý, J.;
39 Havlas, Z.; Paci, I.; Akdag, A.; MacLeod, M. K. et al. Toward Designed Singlet Fission:
40 Electronic States and Photophysics of 1,3-Diphenylisobenzofuran. *J. Phys. Chem. A*
41 **2010**, *114*, 1457–1473.
42
43
44
45
46 (19) Ryerson, J. L.; Schrauben, J. N.; Ferguson, A. J.; Sahoo, S. C.; Naumov, P.; Havlas, Z.;
47 Michl, J.; Nozik, A. J.; Johnson, J. C. Two Thin Film Polymorphs of the Singlet Fission
48 Compound 1,3-Diphenylisobenzofuran. *J. Phys. Chem. C* **2014**, *118*, 12121–12132.
49
50
51
52
53 (20) Johnson, J. C.; Nozik, A.; Michl, J. High Triplet Yield from Singlet Fission in a Thin
54 Film of 1,3-Diphenylisobenzofuran. *J. Am. Chem. Soc.* **2010**, *132*, 16302–16303.
55
56
57
58
59
60

- 1
2
3 (21) Johnson, J. C.; Michl, J. 1,3-Diphenylisobenzofuran: a Model Chromophore for Singlet
4 Fission. *Top. Curr. Chem.* **2017**, *375*, 80.
5
6
7
8 (22) El Bakouri, O.; Smith, J. R.; Ottosson, H. Strategies for Design of Potential Singlet
9 Fission Chromophores Utilizing a Combination of Ground-State and Excited-State Aro-
10 maticity Rules. *J. Am. Chem. Soc.* **2020**, *142*, 5602–5617.
11
12
13
14 (23) Baranac-Stojanović, M. Triplet-State Structures, Energies, and Antiaromaticity of BN
15 Analogues of Benzene and Their Benzo-Fused Derivatives. *J. Org. Chem.* **2019**, *84*,
16 13582–13594.
17
18
19
20
21 (24) Fallon, K. J.; Budden, P.; Salvadori, E.; Ganose, A. M.; Savory, C. N.; Eyre, L.;
22 Dowland, S.; Ai, Q.; Goodlett, S.; Risko, C. et al. Exploiting Excited-State Aromaticity
23 to Design Highly Stable Singlet Fission Materials. *J. Am. Chem. Soc.* **2019**, *141*, 13867–
24 13876.
25
26
27
28
29
30 (25) Nagami, T.; Okada, K.; Miyamoto, H.; Yoshida, W.; Tonami, T.; Nakano, M. Molec-
31 ular Design Principle for Efficient Singlet Fission Based on Diradical Character and
32 Exchange Integral: Molecular Design Principle for Efficient Singlet Fission Based on
33 Diradical Character and Exchange Integral: Multiple Heteroatom Substitution Effect
34 on Anthracene. *J. Phys. Chem. C* **2020**, *124*, 11800–11809.
35
36
37
38
39
40
41 (26) Nagami, T.; Miyamoto, H.; Yoshida, W.; Okada, K.; Tonami, T.; Nakano, M. Theoreti-
42 cal Molecular Design of Phenanthrenes for Singlet Fission by Diazadibora-Substitution.
43 *J. Phys. Chem. A* **2020**, *124*, 6778–6789.
44
45
46
47
48 (27) Malrieux, J. P.; Caballol, R.; Calzado, C. J.; de Graaf, C.; Guihéry, N. Magnetic In-
49 teractions in Molecules and Highly Correlated Materials: Physical Content, Analytical
50 Derivation, and Rigorous Extraction of Magnetic Hamiltonians. *Chem. Rev.* **2014**, *114*,
51 429–492.
52
53
54
55
56
57
58
59
60

- 1
2
3 (28) Stuyver, T.; Chen, B.; Zeng, T.; Geerlings, P.; De Proft, F.; Hoffmann, R. Do Diradicals
4 Behave like Radicals? *Chem. Rev.* **2019**, *119*, 11291–11351.
5
6
7
8 (29) Stoycheva, J.; Tadjer, A.; Garavelli, M.; Spassova, M.; Nenov, A.; Romanova, J. Boron-
9 Doped Polycyclic Aromatic Hydrocarbons: a Molecular Set Revealing the Interplay
10 between Topology and Singlet Fission Propensity. *J. Phys. Chem. Lett.* **2020**, *11*, 1390–
11 1396.
12
13
14
15
16 (30) Singh, S.; Jones, W. J.; Siebrand, W.; Stoicheff, B. P.; Schneider, W. G. Laser Gener-
17 ation of Excitons and Fluorescence in Anthracene Crystals. *J. Chem. Phys.* **1965**, *42*,
18 330–342.
19
20
21
22
23 (31) Petrenko, T.; Krykova, O.; Neese, F.; Sokolowski, M. Optical Absorption and Emission
24 Properties of Rubrene: Insight from a Combined Experimental and Theoretical Study.
25 *New. J. Phys.* **2009**, *11*, 015001.
26
27
28
29
30 (32) McGlynn, S. P.; Azumi, T.; Kasha, M. External Heavy Atom Spin-Orbital Coupling
31 Effect. V. Absorption Studies of Triplet States. *J. Chem. Phys.* **1964**, *40*, 507–515.
32
33
34
35 (33) Heinecke, E.; Hartmann, D.; Muller, R.; Hese, A. Laser Spectroscopy of Free Pentacene
36 Molecules (I): the Rotational Structure of the Vibrationless $S_1 \rightarrow S_0$ Transition. *J.*
37 *Chem. Phys.* **1998**, *109*, 906–911.
38
39
40
41
42 (34) Zeng, T.; Hoffmann, R.; Ananth, N. The Low-Lying Electronic States of Pentacene and
43 Their Roles in Singlet Fission. *J. Am. Chem. Soc.* **2014**, *136*, 5755–5764.
44
45
46
47 (35) Fuemmeler, E. G.; Sanders, S. N.; Pun, A. B.; Kumarasamy, E.; Zeng, T.; Miyata, K.;
48 Steigerwald, M. L.; Zhu, X.-Y.; Sfeir, M. Y.; Campos, L. M. et al. A Direct Mechanism
49 of Ultrafast Intramolecular Singlet Fission in Pentacene Dimers. *ACS Cent. Sci.* **2016**,
50 *2*, 316–324.
51
52
53
54
55
56
57
58
59
60

- 1
2
3 (36) Zeng, T.; Mellerup, S. K.; Yang, D.; Wang, X.; Wang, S.; Stampelcoskie, K. Identifying (BN)₂pyrenes as a New Class of Singlet Fission Chromophores: Significance of
4
5
6
7
8
9
10
11 (37) Pradhan, E.; Bentley, J. N.; Caputo, C. B.; Zeng, T. Designs of Singlet Fission Chromophores with a Diazadiborinine Framework. *ChemPhotoChem* **2020**,
12
13
14
15
16
17 (38) Lee, S.; Filatov, M.; Lee, S.; Choi, C. H. Eliminating Spin-Contamination of Spin-Flip
18
19
20
21
22
23
24
25
26 (39) Lee, S.; Kim, E. E.; Nakata, H.; Lee, S.; Choi, C. H. Efficient Implementations of Analytic Energy Gradient for Mixed-Reference Spin-Flip Time-Dependent Density Functional Theory (MRSF-TDDFT). *J. Chem. Phys.* **2019**, *150*, 184111.
27
28
29
30
31
32
33 (40) Becke, A. D. Density-Functional Thermochemistry. III. The Role of Exact Exchange. *J. Chem. Phys.* **1993**, *98*, 5648–5652.
34
35
36
37
38 (41) Lee, C.; Yang, W.; Parr, R. G. Development of the Colle-Salvetti Correlation-Energy Formula into a Functional of the Electron Density. *Phys. Rev. B* **1988**, *37*, 785–789.
39
40
41
42 (42) Angeli, C.; Cimiraglia, R.; Evangelisti, S.; Leininger, T.; Malrieu, J.-P. Introduction of N-Electron Valence States for Multireference Perturbation Theory. *J. Chem. Phys.* **2001**, *114*, 10252–10264.
43
44
45
46
47
48
49 (43) Angeli, C.; Cimiraglia, R.; Malrieu, J.-P. N-Electron Valence State Perturbation Theory: a Fast Implementation of the Strongly Contracted Variant. *Chem. Phys. Lett.* **2001**, *328*, 297–305.
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3 (44) Angeli, C.; Cimiraglia, R.; Malrieu, J.-P. N-Electron Valence State Perturbation The-
4 ory: A Spinless Formulation and an Efficient Implementation of the Strongly Con-
5 tracted and of the Partially Contracted Variants. *J. Chem. Phys.* **2002**, *117*, 9138–
6 9153.
7
8
9
10
11
12 (45) Dunning, T. H. Gaussian Basis Sets for Use in Correlated Molecular Calculations. I.
13 The Atoms Boron Through Neon and Hydrogen. *J. Chem. Phys.* **1989**, *90*, 1007–1023.
14
15
16
17 (46) Nakano, H.; Uchiyama, R.; Hirao, K. Quasi-Degenerate Perturbation Theory with Gen-
18 eral Multiconfiguration Self-Consistent Field Reference Functions. *J. Comput. Chem.*
19 **2002**, *23*, 1166–1175.
20
21
22
23
24 (47) Miyajima, M.; Watanabe, Y.; Nakano, H. Relativistic Quasidegenerate Perturba-
25 tion Theory with Four-Component General Multiconfiguration Reference Functions.
26 *J. Chem. Phys.* **2006**, *124*, 044101.
27
28
29
30
31 (48) Ebisuzaki, R.; Watanabe, Y.; Nakano, H. Efficient Implementation of Relativistic and
32 Non-Relativistic Quasidegenerate Perturbation Theory with General Multiconfigura-
33 tional Reference Functions. *Chem. Phys. Lett.* **2007**, *442*, 164–169.
34
35
36
37
38 (49) Kawashima, Y.; Hashimoto, T.; Nakano, H.; Hirao, K. Theoretical Study of the Valence
39 $\pi \rightarrow \pi^*$ Excited States of Polyacenes: Anthracene and Naphthacene. *Theor. Chem. Acc.*
40 **1999**, *102*, 49–64.
41
42
43
44 (50) Schmidt, M. W.; Baldrige, K. K.; Boatz, J. A.; Elbert, S. T.; Gordon, M. S.;
45 Jensen, J. H.; Koseki, S.; Matsunaga, N.; Nguyen, K. A.; Su, S. et al. General Atomic
46 and Molecular Electronic Structure System. *J. Comput. Chem.* **1993**, *14*, 1347–1363.
47
48
49
50
51 (51) Gordon, M. S.; Schmidt, M. W. In *Theory and Applications of Computational Chem-*
52 *istry: The First Forty Years*; Dykstra, C. E., Frenking, G., Kim, K. S., Scuseria, G. E.,
53 Eds.; Elsevier: Amsterdam, 2005; Chapter 41, pp 1167–1189.
54
55
56
57
58
59
60

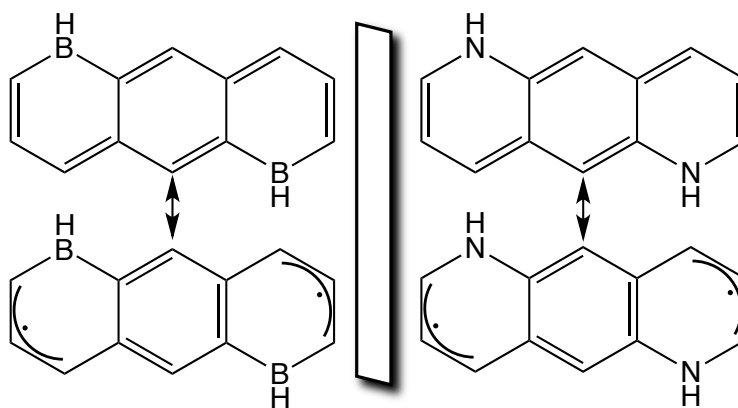
- 1
2
3 (52) Neese, F. The ORCA Program System. *WIREs Comput. Mol. Sci.* **2012**, *2*, 73–78.
4
5
6 (53) Yamaguchi, K. In *Self-Consistent Field: Theory and Applications*; Carbó, R.,
7 Klobukowski, M., Eds.; Elsevier B. V.: The Netherlands, 1990; Chapter 20, pp 727–823.
8
9
10 (54) Flynn, C. R.; Michl, J. π,π -Biradicaloid Hydrocarbons. o-Xylylene. Photochemical
11 Preparation from 1,4-Dihydrophthalazine in Rigid Glass, Electric Spectroscopy, and
12 Calculations. *J. Am. Chem. Soc.* **1974**, *96*, 3280–3288.
13
14
15
16
17 (55) Bersuker, I. B. *The Jahn-Teller Effect*; Cambridge University Press: Cambridge, UK,
18 2006.
19
20
21
22 (56) Lee, J.; Jadhav, P.; Reusswig, P. D.; Yost, S. R.; Thompson, N. J.; Congreve, D. N.;
23 Hontz, E.; Van Voorhis, T.; Baldo, M. A. Singlet Exciton Fission Photovoltaics. *Acc.*
24 *Chem. Res.* **2013**, *46*, 1300–1311.
25
26
27
28
29 (57) Ovchinnikov, A. A. Multiplicity of the Ground State of Large Alternant Organic
30 Molecules with Conjugated Bonds. *Theor. Chem. Acc.* **1978**, *47*, 297–304.
31
32
33
34 (58) Würfel, P. *Physics of Solar Cells: From Basic Principles to Advanced Concepts*;
35 WILEY-VCH Verlag: Weinheim, 2009.
36
37
38
39 (59) Wen, J.; Havlas, Z.; Michl, J. Captodatively Stabilized Biradicaloids as Chromophores
40 for Singlet Fission. *J. Am. Chem. Soc.* **2015**, *137*, 165–172.
41
42
43 (60) Akdag, A.; Havlas, Z.; Michl, J. Search for a Small Chromophore with Efficient Singlet
44 Fission: Biradicaloid Heterocycles. *J. Am. Chem. Soc.* **2012**, *134*, 14624–14631.
45
46
47
48 (61) Zeng, T.; Ananth, N.; Hoffmann, R. Seeking Small Molecules for Singlet Fission: a
49 Heteroatom Substitution Strategy. *J. Am. Chem. Soc.* **2014**, *136*, 12638–12647.
50
51
52 (62) Zeng, T.; Goel, P. Design of Small Intramolecular Singlet Fission Chromophores: An
53 Azaborine Candidate and General Small Size Effects. *J. Phys. Chem. Lett.* **2016**, *7*,
54 1351–1358.
55
56
57
58
59
60

- 1
2
3 (63) Zeng, T. Through-Linker Intramolecular Singlet Fission: General Mechanism and De-
4 signing Small Chromophores. *J. Phys. Chem. Lett.* **2016**, *7*, 4405–4412.
5
6
7
8 (64) Japahuge, A.; Lee, S.; Choi, C. H.; Zeng, T. Design of Singlet Fission Chromophores
9 with Cyclic (Alkyl)(Amino) Carbene Building Blocks. *J. Chem. Phys.* **2019**, *150*,
10 234306.
11
12
13
14
15 (65) Ito, S.; Nakano, M. Theoretical Molecular Design of Heteroacenes for Singlet Fission:
16 Tuning the Diradical Character by Modifying π -Conjugation Length and Aromaticity.
17 *J. Phys. Chem. C* **2015**, *119*, 148–157.
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

TOC Graphic

TOC Graphic

For Table of Contents use only



Diboron- and diaza-dopings in anthracene and phenanthrene exhibit a series of mirror relations.