

Supplementary Information

**Onsite Graywater Treatment in a Two-Stage Electro-Peroxone Reactor with a Partial
Recycle of Treated Effluent**

Léopold Dobelle^{1,†}, Seungkyeum Kim^{2,†}, Axl X Levan³, Hugo Léandri¹,

Michael R Hoffmann¹, Clément A Cid^{1,*}

¹Dept. of Environmental Science and Engineering, California Institute of Technology, 1200 E
California Blvd, Pasadena, CA 91125 USA

²Dept. of Chemical Engineering, California Institute of Technology, 1200 E California Blvd,
Pasadena, CA 91125 USA

³Dept. of Chemistry, California Institute of Technology, 1200 E California Blvd, Pasadena, CA
91125 USA

*ccid@caltech.edu

[†] the authors contributed equally to the work.

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24 **Tables:**

25 **Table S1.** Details about carbon-based electrode materials.

Name	Description	Supplier	Thickness (mm)	Bulk density (g/cm³)
CP75T	Carbon paper with PTFE coating (13%)	AvCarb Material Solutions	0.255	0.33
G100	Soft Graphite Battery Felt	AvCarb Material Solutions	3.2	0.08
C100	Soft Carbon Battery Felt	AvCarb Material Solutions	3.2	0.09
RVC 80 PPI	Duocel Reticulated Vitreous Carbon (RVC) Foam	Duocel Foam	3.175	not provided
MGL 190	Molded Graphite Laminate	AvCarb Material Solutions	0.19	0.44

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27 **Table S2.** Synthetic graywater recipe adapted from NSF/ANSI 350

Components	Description	Quantity in 10 L of tap water
Body Wash	Johnson's Gentle Baby Body Moisture Wash	1.59 g
Toothpaste	Colgate® Cavity Protection Toothpaste	0.159 g
Deodorant	Gillette® Clinical Clear Gel Cool Wave	0.106 g
Shampoo	Suave Essentials® Daily Clarifying Shampoo	1.007 g
Conditioner	Suave Essentials® Waterfall Mist Conditioner	1.113 g
Bath Cleaner	Lysol® Disinfectant Bathroom Cleaner	0.53 g
Hand Soap	Dial® Gold Antibacterial Hand Soap	1.219 g
Laundry Detergent	Ultra Tide® Stain Release Laundry Detergent	1.88 mL
Fabric Softener	Ultra Downy® April Fresh Liquid Fabric Softener	0.987 mL
Secondary Effluent	Obtained from the San Jose Creek Water Reclamation Plant	200 mL
L-Lactic Acid	Alfa Aesar, anhydrous, 98%	0.159 g
Na ₂ SO ₄	Macron Fine Chemicals, anhydrous, granular, ACS grade	0.188 g
NaHCO ₃	Macron Fine Chemicals, ACS grade	0.094 g
Na ₃ PO ₄	Alfa Aesar, tribasic, anhydrous, technical grade	0.188 g

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 29 **Table S3.** Synthetic graywater characteristics and graywater test water parameters defined by the
 30 NSF 350/350-1 standard.

Parameter	Synthetic Graywater	NSF 350/350-1 Required Range
pH	7-8	6.5-8
Temperature (°C)	20-22	25-35
Chemical oxygen demand (mg O ₂ /L)	355-362	250-400
Five-day biochemical oxygen demand (mg O ₂ /L)	155-162	130-180
Total organic carbon (ppm)	83-102	50-100
Total nitrogen (mg N/L)	5.10	3-5*

Total phosphorus (mg P/L)	3.93	1-3
Turbidity (NTU)	11.2 ^{\$}	50-100
<i>E. coli</i> (CFU/100 mL)	10 ^{6,#}	10 ² -10 ³
Electrical conductivity (μ S/cm)	700	-
Alkalinity (mg CaCO ₃ /L)	142	-

31 * In total Kjeldahl nitrogen.

32 \$ Test dust was not used in synthetic graywater recipe as the system did not have pretreatment
33 steps such as filtration to remove mineral particles in suspension.

34 # The effluent used in synthetic graywater was obtained from the secondary clarifier of the San
35 Jose Creek Water Reclamation Plant instead of the final clarifier as suggested by NSF/ANSI
36 350.

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Table S4. Concentration of each H₂O₂ solution injected during the peroxone treatment from cycle 1 to cycle 4 and the conductivity of their related electrolyte before electro-generation.

	[H₂O₂] (mM)	Conductivity (mS/cm²)
Cycle 1	4.81	12.09
Cycle 2	4.80	6.52
Cycle 3	4.21	3.97
Cycle 4	3.88	2.67

44 **Table S5.** Decrease of H₂O₂ in presence of untreated graywater over an hour.

	T (min)				
	0	15	30	45	60
Measured [H₂O₂] (mM)	4.8	2.073	2.051	1.931	1.591
Adjusted (mM)*	4.8	4.146	4.101	3.862	3.181
Decrease (%)	0	13.79%	14.74%	19.70%	33.86%

45 * H₂O₂ concentration in solution was adjusted for dilution to project the decrease on the initial

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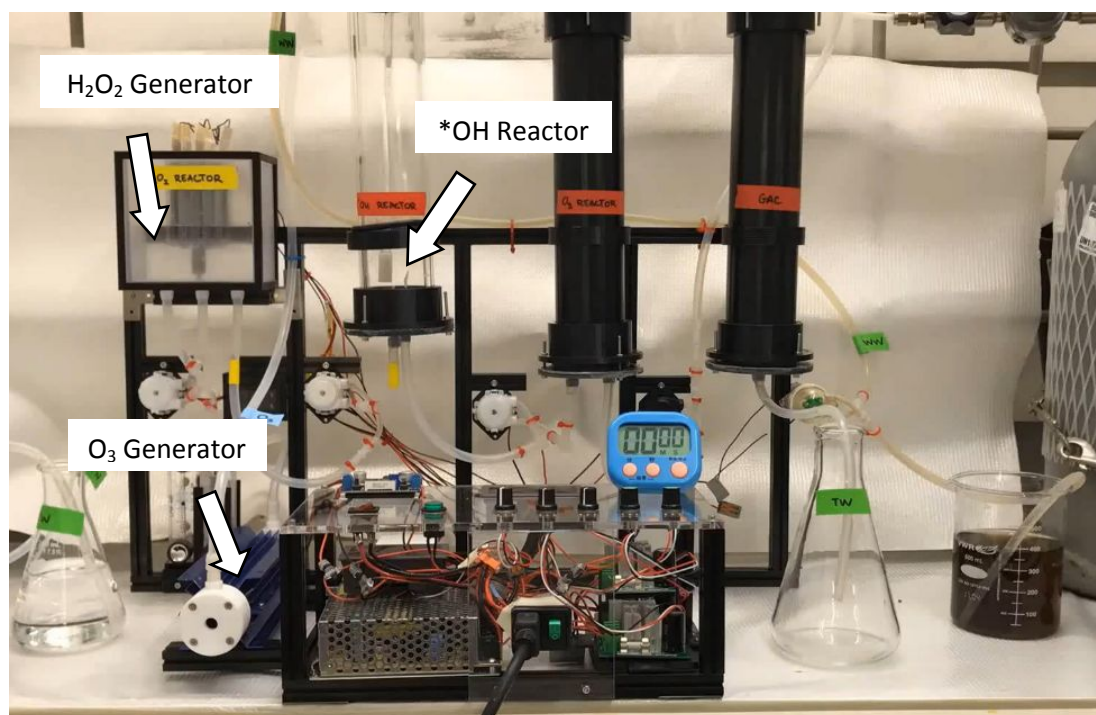
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48 **Table S5.** Decrease of H₂O₂ in presence of HOCl over an hour

	T (min)				
	0	15	30	45	60
Measured [H₂O₂] (mM)	4.81	1.797	1.861	1.733	1.690
Adjusted (mM)*	4.81	3.593	3.722	3.465	3.379
Decrease (%)	0	25.33%	22.67%	28.00%	29.78%

49 * H₂O₂ concentration in solution was adjusted for dilution to project the decrease on the initial
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52 **Figures:**

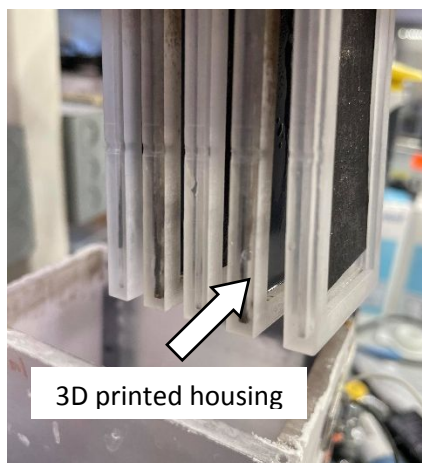
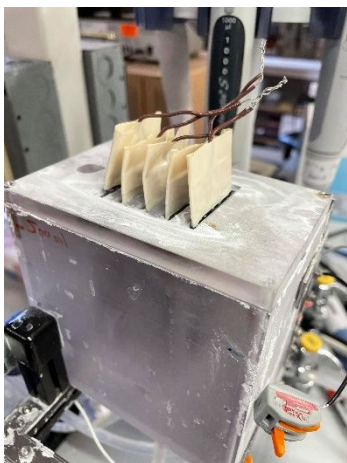


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54 **Figure S1.** Laboratory-scale prototype.

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58 **Figure S2.** 3D-printed electrodes housing.

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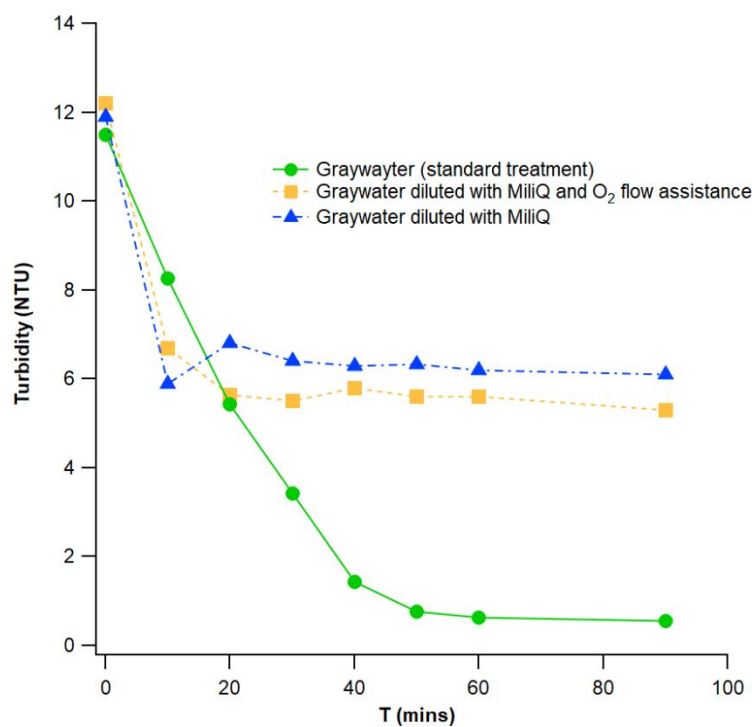


Figure S3. Turbidity trend during single-pass E-peroxone treatment of synthetic graywater. As comparison, synthetic graywater was mixed and diluted with Milli-Q water with and without O₂ sparging. Milli-Q water was added to the graywater at the same dosage as the H₂O₂-containing Na₂SO₄ solution was introduced to the *OH reactor.

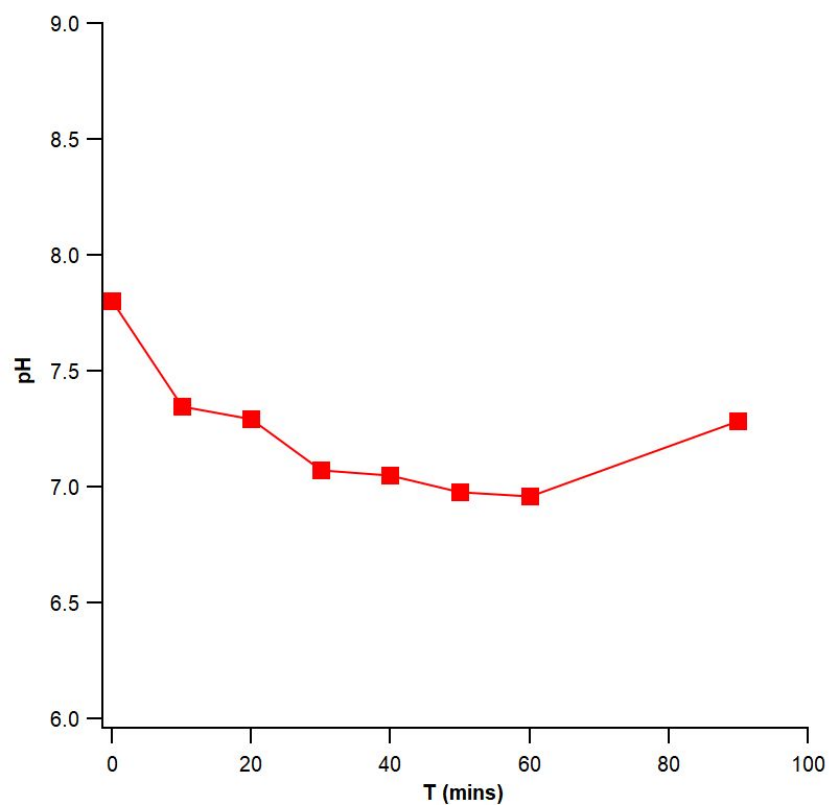
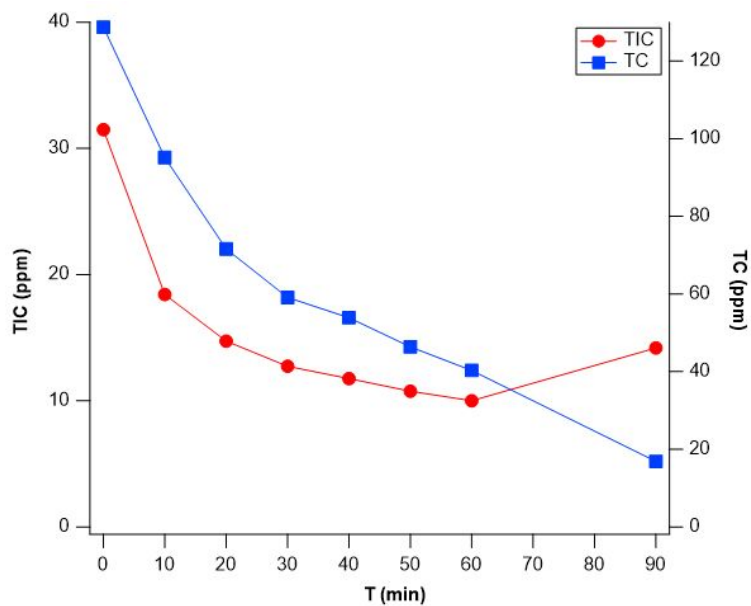


Figure S4. Evolution of pH with time during single-pass E-peroxone treatment of synthetic graywater.



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75 **Figure S5.** Total inorganic carbon (TIC) and total carbon (TC) trends during single-pass E-
 76 peroxone treatment of synthetic graywater. The TIC increased along with the pH from 60-minute
 77 to the end of the treatment while TC kept decreasing.