

1 Supplementary materials

2 0.1. The Matsui potential for silicate systems

3 The Matsui (1998) potential model in its original form can be represented by the following
4 equation:

$$V(r_{ij}) = \frac{q_i q_j}{r_{ij}} - \frac{C_j C_j}{r_{ij}^6} + f (B_i + B_j) \exp\left(\frac{A_i + A_j - r_{ij}}{B_i + B_j}\right) \quad (1)$$

5 From left to right, the three terms describes Coulomb interaction, van der Waals attraction
6 and Born-Mayer-Huggins repulsion, respectively. Born-Mayer-Huggins repulsion is a soft-
7 sphere model for closed-shell atoms and ions published by Gilbert (1968). The physical
8 meaning of f in this term is an arbitrary standard force by which the atom pairs are pushed
9 apart until the distance between them is the sum of their soft-core radius. Matsui set his
10 standard force to be $4.184 \text{ kJ } \text{\AA}^{-1} \text{ mol}^{-1}$. The soft-core radius of atom i and j are denoted
11 by A_i and A_j , respectively. B_i and B_j are called softness (or hardness) parameters.
12 To simplify the form of this pairwise potential, we can perform the following transformations
13 to equation 1.

$$C_{ij} = C_i * C_j \quad (2)$$

$$B_{ij} = B_i + B_j \quad (3)$$

$$A_{ij} = f B_{ij} \exp\left(\frac{A_i + A_j}{B_{ij}}\right) \quad (4)$$

14 And we can get a new form as follows.

$$V(r_{ij}) = \frac{q_i q_j e^2}{4\pi\epsilon_0 r_{ij}} + A_{ij} \exp\left(-\frac{r_{ij}}{B_{ij}}\right) - \frac{C_{ij}}{r_{ij}^6} \quad (5)$$

15 In this new form, A_{ij} and C_{ij} are called energy parameters for the pair ij and they describe
16 the repulsive and van der Waals attractive forces, respectively. B_{ij} is an e-folding length
17 characterizing the radially symmetric decay of electron repulsion energy between atom pair
18 ij . The second and third terms together are known as a Buckingham potential.

19 In LAMMPS, this pairwise potential can be described by the built-in buck/coul/long pair
20 style.

$$E_{\text{Buck}} = A e^{-r/\eta} - \frac{C}{r^6}, \quad r < r_c \quad (6)$$

$$E_{\text{Coul}} = \frac{C q_i q_j}{\epsilon r}, \quad r < r_c \quad (7)$$

$$E_{\text{total}} = E_{\text{Buck}} + E_{\text{Coul}} \quad (8)$$

21 In our force field optimizations, we started from the parameters given by Matsui (1998) and
22 modified the van der Waals part (E_{Buck}) only.

23 In all simulations, we used PPPM long-range interactions for the Coulomb potential and set
24 the cut-off radius for the Buckingham potential to be $r_c = 11 \text{ \AA}$.

25 *0.2. Transformation of Buckingham potential*

26 To understand the physical meanings of the optimization parameters and make more
 27 precise adjustments, we transformed the van der Waals part of the pairwise potential
 28 according to the exponential-6 (X6) form (Mayo, Olafson and Goddard, 1990).

29

$$E_{\text{vdw}}^{\text{X6}} = D_0 \left[\frac{6}{\zeta - 6} e^{\zeta(1-\rho)} - \frac{\zeta}{\zeta - 6} \rho^{-6} \right] \quad (9)$$

30 where D_0 is the van der Waals well depth, $\rho = R/R_0$ is relative bond length where R_0 is
 31 equilibrium bond length and ζ is a dimensionless scaling parameter. $\zeta = 12.0$ means the X6
 32 form and the LJ form have the same long range interactions and $\zeta = 13.772$ means they have
 33 the same force constant. The X6 form has advantages in describing short range interactions
 34 and is suitable for our force field where covalent and ionic bonds are dominant.

35 By comparing Eqn. 6 and Eqn. 9, we can find a set of solutions that represents D_0 , R_0 and
 36 ζ using A , η and C . The transformation of force field parameters is automated by Python
 37 and done in Mathematica (Wolfram Research, Inc., 2020).

38 *0.3. Optimization details*

39 The parameters that are tuned in our optimization are the transformed Buckingham pa-
 40 rameters D_0 , R_0 and ζ for all Cat-O pairs (tuned simultaneously) and for O-O pairs. There
 41 are six parameters in total.

42 To avoid cancelling of different parts, we used the sum of log (e-based) losses as our opti-
 43 mization goal.

$$L = \log(\alpha_P L_P + \alpha_{C_V} L_{C_V} + \alpha_\gamma L_\gamma) \quad (10)$$

44 where L is the total loss, α 's are the scaling parameters for each kind of loss and L_P , L_{C_V}
 45 and L_γ are the losses corresponding to pressure, heat capacity and Grüneisen parameter,
 46 respectively. For a simulation containing 4 reference points (which yields 4 pressures, 2
 47 heat capacities and 2 Grüneisen parameters because the latter two are calculated by finite
 48 difference), each part of the loss function is defined as:

$$L_P = \frac{1}{4} \sum_{i=1}^4 (P_i - P_{i,ref})^2 \quad (11)$$

$$L_{C_V} = \frac{1}{2} \sum_{i=1}^2 (C_{V_i} - C_{V_i,ref})^2 \quad (12)$$

$$L_\gamma = \frac{1}{2} \sum_{i=1}^2 (\gamma_i - \gamma_{i,ref})^2 \quad (13)$$

49 The choice of scaling parameters considers both the relative magnitude of each physical
 50 quantity and their degree of fluctuation. For our final optimizations, the choices are:

$$\alpha_P = 0.01 \tag{14}$$

$$\alpha_{C_V} = 100 \tag{15}$$

$$\alpha_\gamma = 70000000 \tag{16}$$

51 Both BFGS and gradient descent (GD) algorithms were used in the optimization and al-
 52 though BFGS reduced the loss function faster at the beginning of the optimization, the best
 53 result was produced by GD, which reduced the loss function from 3.90 to 3.59 and achieved
 54 satisfying accuracy for C_V and γ .

55 0.4. The force field parameters

56 Compared to the original Matsui force field, the optimized force field scales each param-
 57 eter differently (Table S1).

The van der Waals well depth parameter (D_0) has the largest change of all, which can

Table S1: Final scaling factors of the optimized force field

Optimization parameter	Scaling=new/original
$D_0(\text{Cat-O})$	0.478192
$R_0(\text{Cat-O})$	0.771891
$\zeta(\text{Cat-O})$	1.044594
$D_0(\text{O-O})$	0.684991
$R_0(\text{O-O})$	0.990607
$\zeta(\text{O-O})$	0.804164

58 be attributed to the fact that high temperature and pressure conditions reduce the bond
 59 strength. For all of the oxygen-containing ion pairs, the change of equilibrium bond length
 60 (R_0) is very small, indicating that the structure of silicates can be relatively accurately
 61 described by the original force field. For Cation-Oxygen and Oxygen-Oxygen pairs, the cur-
 62 vature parameter (ζ) changes in opposite directions, which could be related to the difference
 63 in long range attraction between cation-anion pairs and anion-anion pairs.

64 These parameters are then transformed to input parameters for LAMMPS and can be found
 65 in section 0.9.

67 0.5. Calculating Hugoniot from Equation of State

68 The original Hugoniot data shown in Figure 5 were presented by Asimow and Ahrens
 69 (2010). The measured quantities in each experiment are shock velocity U_S , particle velocity
 70 u_p , and initial sample density ρ_o . These results are converted to the Pressure (P)-Density
 71 (ρ) plane using the first two Rankine-Hugoniot equations (i.e., conservation of mass and

72 momentum) for a single shock driven into a sample initially at zero pressure and at rest in
 73 the laboratory reference frame:

$$\rho_o U_S = \rho(U_S - u_p) \quad (17)$$

$$P = \rho_o U_S u_p. \quad (18)$$

74 In most of the experiments plotted, the initial state of the sample is liquid at 0.1 MPa (i.e.,
 75 approximately zero) pressure at 1673 K, which is also the reference state for the equation of
 76 state adopted. One experiment begins from the initially solid state at 300 K. Interpreting
 77 this experiment requires knowing the initial density of the isochemical mixture of solids at
 78 300 K (ρ_{oS}), as well the internal energy difference between the 1673 K liquid and 300 K
 79 solid states, ΔE_{tr} , which is the obtained from the integral of the solid heat capacity from
 80 300 K to the eutectic temperature (1547 K), the enthalpy of fusion, and the integral of the
 81 liquid heat capacity from 1547 K to 1673 K.

82 These data are compared to Hugoniot computed from the thermal equation of state as
 83 follows. The independent variable is ρ . At each value of ρ , we obtain the Eulerian finite
 84 strain parameter f relative to the reference volume ρ_o :

$$f = \left\{ \left(\frac{\rho}{\rho_o} \right)^{\frac{2}{3}} - 1 \right\}. \quad (19)$$

85 and then the pressure on the reference isentrope is calculated from the 3rd order Birch-
 86 Murnaghan equation of state:

$$P_S = 3K_{oS} f (2f + 1)^{\frac{5}{2}} \left[1 + \frac{3}{2} (K'_S - 4) f \right], \quad (20)$$

87 In which K_{oS} is the isentropic bulk modulus at the reference state (which is known precisely
 88 from ultrasonic measurements and therefore is held constant during fitting) and K'_S is the
 89 first derivative of the isentropic bulk modulus along the isentrope, evaluated at the reference
 90 state (which is not known from ambient pressure data and is therefore a fitted parameter).
 91 To obtain the pressure on the Hugoniot at shock density ρ , we take advantage of the fact
 92 that internal energy E is a path-independent state variable and therefore the difference in
 93 E between the initial reference state and the shock state is the same whether the change of
 94 state is accomplished by passage of the shock (ΔE_H) or by decomposition of the path into
 95 a series of three processes: transformation from the initial state to the reference state at 1
 96 bar (ΔE_{tr} , if necessary), isentropic compression from the reference state at ρ_o to the shock
 97 density ρ (ΔE_S) and finally isochoric heating from the isentrope to the corresponding state
 98 on the Hugoniot (ΔE_V). That is:

$$\Delta E_H = \Delta E_{tr} + \Delta E_S + \Delta E_V. \quad (21)$$

99 The increase in internal energy imparted by passage of a shock is obtained from the 3rd
 100 Rankine-Hugoniot jump condition:

$$\Delta E_H = \frac{1}{2} (P + P_o) \left(\frac{1}{\rho_{oo}} - \frac{1}{\rho} \right), \quad (22)$$

101 in which ambient pressure P_o is negligible compared to shock pressure P and ρ_{oo} is the
 102 density of the initial state, which may be equal to ρ_o if the sample begins in the reference
 103 state, ρ_{os} for the case where the initial state is a cold solid assemblage, or any other value
 104 as appropriate (e.g. for a porous target).

105 From the first and second laws of thermodynamics, the total derivative of specific internal
 106 energy $dE = TdS - PdV$, where T is temperature, S is specific entropy, and V is specific
 107 volume ($V = 1/\rho$). Hence, the increase in internal energy along an isentropic path (for
 108 which $dS = 0$) is obtained by integrating equation (20):

$$\Delta E_S = \frac{9}{2} \frac{K_{oS}}{\rho_o} [f^2 + (K'_S - 4)f^3]. \quad (23)$$

109 The increase in energy along the isochore from the isentrope to the Hugoniot is obtained
 110 from the selected formulation of the Grüneisen parameter, since $\gamma \equiv V(\partial P/\partial E)_V$. For the
 111 conventional Mie-Grüneisen approximation, in which γ is constant along an isochore, this is
 112 a simple finite difference:

$$\Delta E_V = \frac{V}{\gamma(V)}(P - P_S). \quad (24)$$

113 In this work, when using the Mie-Grüneisen approximation we have assumed that $\gamma(V)$ is
 114 given by

$$\frac{\gamma(V)}{\gamma_o} = \left(\frac{V}{V_o}\right)^q, \quad (25)$$

115 In which the Grüneisen parameter at the reference state γ_o is known precisely from ultra-
 116 sonic, thermal expansion, and heat capacity data and is a fixed parameter during fitting,
 117 whereas q is unknown and is a fitted parameter.

118 The solution of these equations for Hugoniot pressure at a given density, for the Mie-
 119 Grüneisen case, can be compactly written

$$P_H = \frac{P_S - \rho\gamma(\Delta E_S + \Delta E_{tr})}{1 - \frac{\gamma}{2} \left(\frac{\rho}{\rho_{oo}} - 1\right)}, \quad (26)$$

120 If the sample begins in the reference state, $\Delta E_{tr} = 0$. Hence equation (26) is used to
 121 compute both the hot liquid Hugoniot and the cold solid Hugoniot curves for the Mie-
 122 Grüneisen model.

123 Note that an alternative expression for γ is $\gamma = (\partial \ln T/\partial \ln V)_S$. Hence the temperature
 124 along an isentrope can be obtained by integrating this expression. For the Mie-Grüneisen
 125 case and the form of equation (23), we have

$$T_S = T_o \exp \left\{ \frac{\gamma_o}{q} \left(1 - \left(\frac{\rho_o}{\rho} \right)^q \right) \right\}. \quad (27)$$

126 For the model introduced in this paper, $\gamma(E, V)$ is not a constant along the isochore from
 127 the isentrope to the Hugoniot and so we cannot use equation (24). In this model, we have
 128 adopted the form

$$\gamma(E, V) = \left(\frac{a}{V} + b \right) (E - E^*) + \gamma^*, \quad (28)$$

129 where a , b , E^* , and γ^* are the adjustable parameters. Note that the reference datum

$$\gamma(E_o, V_o) = \gamma_o \quad (29)$$

130 is added to the fit and assigned an extremely high weight, in order to enforce the known
 131 reference value of the Grüneisen parameter.

132 The relationship between pressure increase and internal energy increase along the isochore
 133 from the isentrope to the Hugoniot state is given in this case by

$$P - P_S = \frac{1}{V} \int_{\Delta E_S + \Delta E_{tr}}^{\Delta E_H} \gamma(E', V) dE'. \quad (30)$$

134 In combination with equations (21) and (22), integrating the linear dependence on E from
 135 (28) yields a quadratic expression for P , of which one root is physical and one root is non-
 136 physical. With appropriate choice of ρ_{oo} and ΔE_{tr} , this quadratic solution yields either the
 137 Hugoniot from the hot liquid initial state or from the cold solid initial state, for the $\gamma(E, V)$
 138 model.

139 Since the changes in both E and P with changes in V along the isentrope are known from
 140 equations (20) and (23), it is straightforward to obtain $\gamma(E, V)$ along the isentrope and
 141 hence to compute the temperature increase along the isentrope in this model as well.

142 The final equation that is relevant to fitting the model form to the experimental dataset for
 143 AnDi liquid incorporates the measurement of bulk sound velocity C_B in the shock state. As
 144 with any bulk sound wave, $C_B = \sqrt{V K_S}$, where again K_S is the isentropic bulk modulus.
 145 If the slope of the Hugoniot $(\partial P / \partial V)_H$ is obtained from the above equations by finite
 146 difference, then it can be shown that

$$K_S = \frac{1}{2} \left(\gamma P - \left(\frac{\partial P}{\partial V} \right)_H \left[2V - \gamma \left(\frac{1}{\rho_{oo}} - V \right) \right] \right). \quad (31)$$

147 The inversion for fitted parameters $\{K'_S, a, b, E^*, \gamma^*\}$ is therefore based on γ_o , all
 148 twelve available preheated liquid Hugoniot shots, one cold solid aggregate Hugo-
 149 niot shot, and one sound speed datum. All the constants $\{\rho_o, \rho_{os}, K_{oS}, \gamma_o, \Delta E_{tr}\}$
 150 are given by Asimow and Ahrens (2010) but are repeated here for convenience:
 151 $\{2618 \text{ kg m}^{-3}, 3078 \text{ kg m}^{-3}, 22.98 \text{ GPa}, 0.356, 1.954106 \text{ MJ kg}^{-1}\}$.

152 *0.6. Ergodicity check by mean squared displacement (MSD)*

153 In liquid phase, particles follow the laws of Brownian motion and their MSD is linear
 154 in time (namely, $\text{MSD} = 6Dt$ where D is the diffusion coefficient.) However, in glass phase,

155 the motion of the particles are confined by their neighbors and MSD increases more slowly
 156 with time. This phenomenon is called sub-diffusion and can be characterized by a slope
 157 lower than unity in a log-log plot of MSD *vs.* time.
 158 To ensure the accuracy of the results, only ergodic MD simulations should be used in our
 159 thermodynamics calculations. We checked the MSD- t relationship for all simulations and
 160 excluded subdiffusive runs. Figure S1 shows two segments from our An-Di simulations
 161 highlighting the difference between liquid (diffusive) and glass (sub-diffusive) behavior. To
 162 avoid the superdiffusive region at the beginning of each simulation, both plots are based on
 the last 2.5 ns of 5-ns runs.

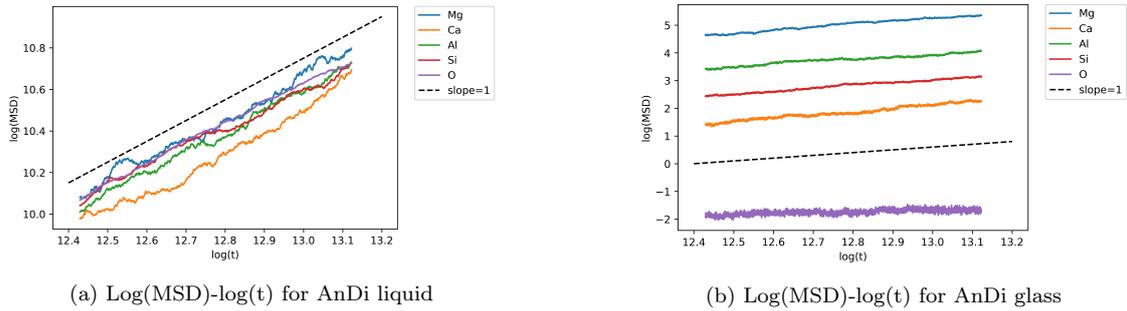


Figure S1: Comparison of MSD plots of liquid-like and glass-like runs. (a) 6500 K, 1.5 GPa. Liquid-like behavior is indicated by slopes approximately equal to unity. (b) 2000 K, 150 GPa. Glass-like behavior is indicated by slopes much smaller than unity.

163

164 0.7. Thermal pressure coefficient

165 The thermal pressure coefficient $(\partial T/\partial P)_V = \alpha K_T$ is related to the Grüneisen parameter,
 166 which can be written $\gamma = V\alpha K_T/C_V$. When working with shock wave data, it is more
 167 convenient to use the Grüneisen parameter because data can be fit in energy space without
 168 reference to the heat capacity; when working with PVT data it is more convenient to use the
 169 thermal pressure coefficient. In any case, the Grüneisen parameter and the thermal pressure
 170 coefficient can each be functions only of volume, but only in the specific case that C_V is also
 171 a function of volume only. This is very unlikely to be the case; certainly it is not expected
 172 for models of heat capacity such as the Debye or Einstein models. Our simulations indeed
 173 show that C_V is a function of volume and energy. One might wonder whether all the energy
 174 dependence that we observe in gamma is then attributable to the energy dependence of C_V ,
 175 but we find that C_V decreases with increasing energy at all volumes and so cannot explain
 176 the reversal in the slope of the energy dependence of gamma. Figure S2 shows the thermal
 177 pressure coefficient in our models. We find that the thermal pressure coefficient does not
 178 reduce to a simple function of volume only, especially at small volume (high pressure). The
 179 rate of decrease in thermal pressure coefficient with increasing temperature becomes more
 180 pronounced as volume decreases.

181 0.8. Fitting EOS to empirical simulation data

182 In the main text, equations of state based on a Birch-Murnaghan isentrope and either a
 183 Mie-Grüneisen or a $\gamma(E, V)$ thermal pressure model are fitted to experimental shock wave

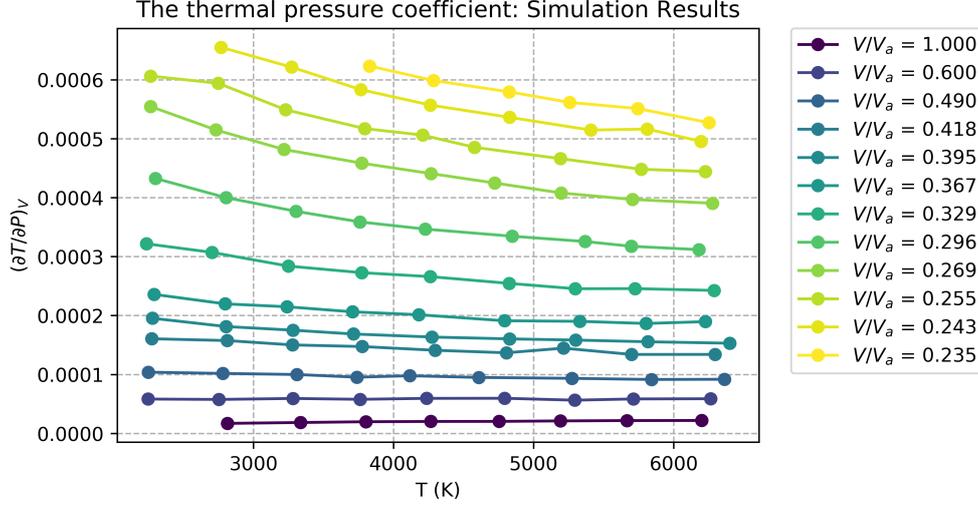


Figure S2: Thermal pressure coefficient isochores from the simulations of the present work.

184 data. Although the main purpose of our simulation campaign is to identify the general form
 185 of the EOS, here we also fit these functional forms to the EPMD simulation data themselves.
 186 We used the EPMD data at $V/V_a < 0.395$, i.e. $P > 10$ GPa, which are best tuned to the
 187 AIMD simulations. V_o was kept fixed in both models. For the Birch-Murnaghan plus Mie-
 188 Grüneisen model, we adjusted K_{S0} , K'_S , γ_0 , and q to minimize the squared sum of residuals
 189 in pressure between the simulation points and the model at equal E and V . For the Birch-
 190 Murnaghan plus $\gamma(E, V)$ model, we similarly adjusted K_{S0} , K'_S , and the four parameters
 191 of the $\gamma(E, V)$ model. The results are shown in Figure S3 and Table S2. Although the
 192 differences are subtle, the $\gamma(E, V)$ model is a slightly better fit, with RMS error of 0.15 GPa
 193 vs. 0.17 GPa for the Mie-Grüneisen model.

Table S2: Comparison of the Birch-Murnaghan plus Mie-Grüneisen fit and the $\gamma(V, E)$ fit to EPMD simulation results.

Parameters	Mie-Grüneisen model	$\gamma(E, V)$ model
V_0 (m ³ /kg)	0.0003820	0.0003820
K_{S0} (GPa)	21.95	24.68
K'_S	6.118	5.344
Model specific Parameters	$\gamma_0 = 0.521$ (at V_0) $q = -1.5$	$\gamma^* = 0.412$ $E^* = -2.04 \times 10^7$ J/kg $a = -1.95 \times 10^{-11}$ m ³ /J $b = 4.43 \times 10^{-8}$ kg/J ($\gamma_0 = 0.548$ at V_0, E_0)

V_0 is the reference volume at 1 bar, 1673 K.

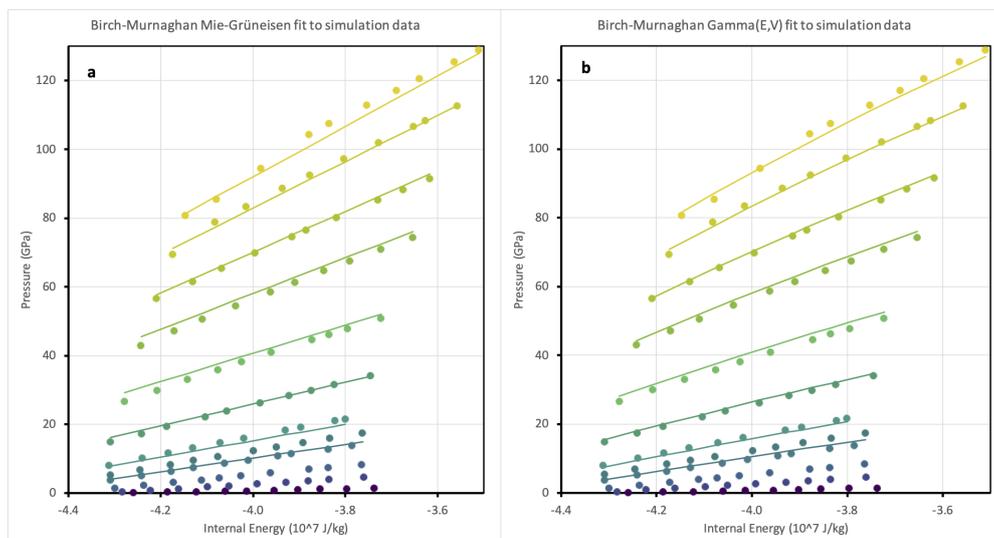


Figure S3: Fits of two EOS forms to the high-pressure subset of EPMD data, in pressure vs. internal energy space. (a) A Birch-Murnaghan isentrope plus a Mie-Grüneisen thermal pressure. (b) A Birch-Murnaghan isentrope plus a $\gamma(E, V)$ thermal pressure model. Colors of simulation points and EOS fit lines correspond to volumes as in Figures 1 and 3 in the main text.

194 0.9. Sample LAMMPS input files

195 Below is a sample input file showing how we used the new force field to calculate one
 196 NVT sample point.

```

1971 # lammps input file for NCMAS melts simulation
1982
1993 # VARIABLES
2004 variable      i          index    3000
2015 variable      restart_name index    nvt.3500.restart
2026 variable      settings_name index    scaled_matsui.in.settings
2037 variable      log_name    index    3000.log
2048 variable      nSteps_ramp equal    1000000 # time for each annealing
205   period. slow=1ns
2069 variable      avg_freq    index    200
2070 variable      coords_freq index    200
2081 variable      dump4avg    index    10
2092 variable      T1          equal    3500 # Temperature during the
210   initial anneal
2113 variable      T2          equal    3000 # Temperature during the
212   final anneal
2134
2145 # Change the name of the log output #
2156 log ${log_name}
2167
2178 #=====
2189 # GENERAL PROCEDURES
2190 #=====
2201 units      real # g/mol, angstroms, fs, kcal/mol, K, atm, charge*angstrom

```

```

2222 dimension 3 # 3 dimensional simulation
2223 newton off # use Newton's 3rd law
2224 boundary p p p # periodic boundary conditions
2245 atom_style charge # charge (for ions: coordinates, velocities, atom
225 IDs, types, charges)
2266
2277 #=====
2288 # FORCE FIELD DEFINITION
2299 #=====
2300 pair_style buck/coul/long 11.0 # outer_Coul (cutoff values, see
231 LAMMPS Doc)
2321 kspace_style ppm 0.0001 # long-range electrostatics sum method
2332
2343 #=====
2354 # SETUP SIMULATIONS
2365 #=====
2376
2387 # READ IN COEFFICIENTS/COORDINATES/TOPOLOGY
2398 read_restart ${restart_name}
2409 include ${settings_name}
2410
2421 # SET RUN PARAMETERS
2432 timestep 1.0 # fs
2443 run_style verlet # Velocity-Verlet integrator
2454 neigh_modify every 1 delay 0 check no # More relaxed rebuild criteria can
246 be used
2475
2486 #=====
2497 # THERMO
2508 #=====
2519 # DECLARE RELEVANT OUTPUT VARIABLES
2520 variable my_step equal step
2531 variable my_temp equal temp
2542 variable my_vol equal vol
2553 variable my_rho equal density
2564 variable my_etot equal etotal
2575 variable my_pe equal pe
2586 variable my_coul equal ecoul
2597 variable my_long equal elong
2608 variable my_evdw equal evdwl
2619 variable my_ent equal enthalpy
2620 variable my_P equal press
2631
2642 fix averages all ave/time ${dump4avg} $(v_avg_freq/v_dump4avg) ${avg_freq
265 } v_my_temp v_my_vol v_my_rho v_my_etot v_my_pe v_my_coul v_my_long
266 v_my_evdw v_my_ent v_my_P file thermo.$i.avg
2673
2684 #=====
2695 # EQUILIBRATE THE BOX BY NPT RUN
2706 #=====

```

```

2757 dump equil all custom ${coords_freq} equil.$i.lammpstrj id type xu yu zu
272   vx vy vz
2738 dump_modify equil sort id
2749 fix equil all nvt temp ${T1} ${T2} 100.0
2750
2761 run   ${nSteps_ramp}
2772
2783 unfix equil
2794 undump equil
2805 unfix averages
2816
2827 # WRITE RESTART FILES
2838 write_restart nvt.$i.restart
2849 write_data   nvt.$i.data

```

285 Below is the the input file describing our new force field. (*scaled_matsui.in.settings*
286 mentioned in line 6 in the previous file)

```

2871 # Non-bonded interactions (pair-wise)
2882 pair_coeff 1 1 314506743771.52 0.08 400.0067
2893 pair_coeff 1 2 11313599738.4 0.08 284.0429
2904 pair_coeff 1 3 365404704482.73 0.08 439.9976
2915 pair_coeff 1 4 19288912785.37 0.074 360.0158
2926 pair_coeff 1 5 578091787566.03 0.063 482.0418
2937 pair_coeff 1 6 52943434.3862237 0.109451274814982 78.9992138379116
2948 pair_coeff 2 2 406978678.76 0.08 201.6975
2959 pair_coeff 2 3 13144527584.6 0.08 312.4402
2960 pair_coeff 2 4 529901504.2 0.074 255.6456
2971 pair_coeff 2 5 8478304051.37 0.063 342.2957
2982 pair_coeff 2 6 5030841.14729292 0.111200388183525 54.4266683653554
2993 pair_coeff 3 3 424539698122.04 0.08 483.9866
3004 pair_coeff 3 4 22684743951.5 0.074 396.0086
3015 pair_coeff 3 5 699390225204.21 0.063 530.2342
3026 pair_coeff 3 6 55601206.5371633 0.109479300137684 86.8482972553961
3037 pair_coeff 4 4 728042565.29 0.068 324.023
3048 pair_coeff 4 5 16878915502.56 0.057 433.8494
3059 pair_coeff 4 6 2499266.0433064 0.10926898033415 67.374133749643
3060 pair_coeff 5 5 1843386062908.58 0.046 580.9011
3071 pair_coeff 5 6 3821560.4434135 0.102832132904819 89.6631707406477
3082 pair_coeff 6 6 351953.268667045 0.24517533166497 1159.41353486781

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309 References

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