Support Information:

First Principles Investigation of Anistropic Hole Mobilities in Organic Semiconductor

Shu-Hao Wen^{1,2}, An Li¹, Junling Song¹, Wei-Qiao Deng¹*, Ke-Li Han²*, William A. Goddard III³*

E-mail: wqdeng@ntu.edu.sg, klhan@dicp.ac.cn, wag@wag.caltech.edu

Documentary statement

Figure S1 Illustrations of the identical surrounding of random selected molecules in the rubrene crystals.

The derivation of eq. 8

Table S1 Computation details of internal reorganization energy (λ). E(neutral in neutral geometry), E*(neutral in ion geometry), E₊(ion in ion geometry) and E₊* (ion in neutral geometry) are in Hartree, and the VIP(vertical ionization potential) and λ are in eV.

Table S2 Calculated electronic coupling elements for different dimer types in tetracene, pentacene, rubrene, DCT, HTP, and Bracelet-14 organic semiconductor. (values from reference⁵ are shown in parentheses)



Figure S1. Illustrations of the identical surrounding of random selected molecules in the rubrene crystals.

The derivation of eq. 8

$$\begin{split} \mu'(\Phi) &= \left(\frac{e}{2k_BT} \sum_{i} \rho_i W_i r_i^2 \cos^2 \gamma_i \cos^2(\theta_i - \Phi)\right)' \\ &= \frac{e}{2k_BT} \sum_{i} \rho_i W_i r_i^2 \cos^2 \gamma_i (2\cos(\theta_i - \Phi))(\sin(\theta_i - \Phi))) \\ &= \frac{e}{2k_BT} \sum_{i} \rho_i W_i r_i^2 \cos^2 \gamma_i \sin 2(\theta_i - \Phi) \\ &= \frac{e}{2k_BT} \sum_{i} (\rho_i W_i r_i^2 \cos^2 \gamma_i (\sin(2\theta_i)) \cos(2\Phi) - \cos(2\theta_i) \sin(2\Phi)))) \\ \mu'(\Phi) &= 0 \Rightarrow \\ \left(\frac{e}{2k_BT} \sum_{i} \rho_i W_i r_i^2 \cos^2 \gamma_i \sin(2\theta_i)\right) \cos(2\Phi) &= \left(\frac{e}{2k_BT} \sum_{i} \rho_i W_i r_i^2 \cos^2 \gamma_i \cos(2\theta_i)\right) \sin(2\Phi) \\ &\Rightarrow \tan(2\Phi) &= \frac{\sum_{i} \rho_i W_i r_i^2 \cos^2 \gamma_i \sin(2\theta_i)}{\sum_{i} \rho_i W_i r_i^2 \cos^2 \gamma_i \cos(2\theta_i)} \\ \because W_i &= \frac{V_i^2}{\hbar} \left(\frac{\pi}{\lambda k_BT}\right)^{1/2} \exp\left(-\frac{\lambda}{4k_BT}\right) \\ &\Rightarrow \Phi_{externa} &= \frac{n\pi}{2} + \frac{1}{2} \arctan\left(\frac{\sum_{i} P_i V_i^2 r_i^2 \cos^2 \gamma_i \cos 2\theta_i}{\sum_{i} P_i V_i^2 r_i^2 \cos^2 \gamma_i \cos 2\theta_i}\right), n = 0, \pm 1, \pm 2, \pm 3, \cdots \end{split}$$

Table S2. Computation details of internal reorganization energy (λ). E(neutral in neutral geometry), E*(neutral in ion geometry), E₊(ion in ion geometry) and E₊* (ion in neutral geometry) are in Hartree, and the VIP(vertical ionization potential) and λ are in eV.

	TETR	PENT	RUBR	DCT	НТР	1CT	2CT	Bracelet-14
E	-693.32185	-846.98821	-1617.69581	-1612.55784	-3232.68949	-785.58749	-877.85116	-2150.77471
E*	-693.31982	-846.98649	-1617.69301	-1612.55516	-3232.68692	-785.58570	-877.84955	-2150.77451
E+	-693.08279	-846.76312	-1617.47425	-1612.31024	-3232.44351	-785.33387	-877.58360	-2150.59643
E+*	-693.08056	-846.76125	-1617.47148	1612.30751	-3232.44088	-785.33193	-877.58197	-2150.59533
VIP	6.5658	6.1758	6.1032	6.8117	6.7649	6.9542	7.3249	4.8849
	6.97 ^{<i>a</i>}	6.63 ^{<i>a</i>}	6.41 ^{<i>a</i>}	6.505 ^d				
λ	0.1157	0.0976	0.1521	0.1472	0.1428	0.1017	0.0881	0.0357
	0.1176	0.0992	0.1590 ^c	0.1410 ^d				

^{*a*} NIST database. ^{*b*} Reference 2 ^{*c*} Reference 3 ^{*d*} Reference 4

2 (a) Podzorov, V., Sysoev, S.E., Loginova, E., Pudalov, V.M. & Gershenson, M.E. Appl. Phys.
Lett. 2003, 83, 3504-3506. (b) Podzorov, V., Pudalov, V.M. & Gershenson, M.E. Appl. Phys. Lett. 2003, 82, 1739-1741.

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4 Kuo, M. Y.; Chen, H. Y.; Chao, I. *Chemistry-a European Journal* 2007, 13, 4750-4758.

Table S3Calculated electronic coupling elements for different dimer types in
tetracene, pentacene, rubrene, DCT, HTP, 1CT, 2CT and bracelet-14 organic
semiconductor. (values from reference⁵ are shown in parentheses)

	TETR	PENT	RUBR	Bracelet-14
V _{T1} (eV)	-0.017 (-0.023)	0.085 (0.085)	0.019 (0.015)	0.054
$R_{T1}(Å)$	5.11	4.73	7.97	14.3
V _{T2} (eV)	0.070 (0.070)	-0.048 (-0.051)	0.019 (0.015)	0.053
$R_{T2}(Å)$	4.82	5.19	7.97	14.3
V _p (eV)	-0.001 (-0.004/16)	0.036 (0.037/34)	0.089 (0.083)	0.052
$R_{P}(Å)$	6.08	6.26	7.15	14.3
V _L (eV)	0.0000 (0.0000)	0.001 (0.0000)	0.0023 (0.0000)	-0.005
$R_L(Å)$	13.38	14.12	13.73	7.131

DCT

$V_{T1} (eV)$	-0.006
\mathbf{R}_{T1} (Å)	9.822
$V_{T2}(eV)$	-0.0055
$\mathbf{R}_{\mathrm{T1}}(\mathrm{\AA})$	10.435
V _P (eV)	-0.0863
$R_P(A)$	3.848
$V_{L1} (eV)$	-0.0017
$R_L(Å)$	8.652
$V_{L2} (eV)$	0.00049
$R_L(Å)$	12.717

HTP

V_{T1} (eV)	0.0035
R _L (Å)	12.477
V_{T2} (eV)	0.0031
R _L (Å)	12.749
$V_{T3}(eV)$	0.0026
R _L (Å)	9.323
$V_{T4} (eV)$	0.003
R _L (Å)	9.567
$V_{\rm p} (eV)$	0.0225
R _L (Å)	3.934
V _L (eV)	-0.00066
R _L (Å)	14.311

⁵ Coropceanu, V., Cornil, J., da Silva, D.A., Olivier, Y., Silbey, R. & Bredas, J.L. Charge transport in organic semiconductors. *Chem. Rev.* **107**, 2165-2165 (2007).