

Additional file 2-Protocol: Preparation of [¹³C₁₁,¹⁵N]Indole-3-pyruvic acid (IPyA)

► *CRITICAL*: For absolute quantification of IPyA in biological samples, a known amount of [¹³C₁₁,¹⁵N]IPyA is added to each sample as the internal standard, and [¹³C₁₁,¹⁵N₁]IPyA has to be synthesized just prior to or no more than 2 days before use.

Materials:

- 50 mM Phosphate buffer (pH 8.0), made from sodium phosphate monobasic (NaH₂PO₄, Sigma-Aldrich, cat. no. S8282) and sodium phosphate dibasic (Na₂HPO₄; Sigma-Aldrich, cat. no. S7907)
- 0.2 mg/ml Transaminase (Sigma-Aldrich, cat. no. T7684) in 50 mM phosphate buffer
 - *CRITICAL*: store as 100 µl aliquots at -20 °C. Use one aliquot for each reaction.
- [¹³C₁₁,¹⁵N₂]L- Trp (Cambridge Isotope Laboratories, cat. no. CNLM-2475)
- α-Ketoglutarate (Sigma-Aldrich, cat. no. K-1750)
- 5 mM Pyridoxal 5'-phosphate (PLP; Sigma-Aldrich, cat. no. 82870)
- 10 mM Ascorbic acid (Sigma-Aldrich, cat. no. A4544)
- 25% Phosphoric acid (PA) (ACS grade; Fisher, cat. no. A242)
- Indole-3-pyruvic acid (Sigma-Aldrich, cat. no. I7017)
 - *CRITICAL*: the compound is sensitive to oxygen. Keep container tightly closed and store in a dry place at -20 °C.
- Ethyl acetate
- 50% Isopropanol

Synthesis of [¹³C₁₁,¹⁵N]IPyA using an enzyme-catalyzed transamination reaction

1. Let one aliquot of transaminase solution thaw on ice.
2. In 1 ml of 50 mM phosphate buffer (pH 8.0), add 1 mg [¹³C₁₁,¹⁵N₂]Trp and 1 mg α-ketoglutarate.
3. Add 20 µl 5 mM PLP and one aliquot of transaminase solution.
4. Mix the solution gently by inverting the tube several times.
5. Incubate the reaction mixture at 37 °C in the dark for 3 h.

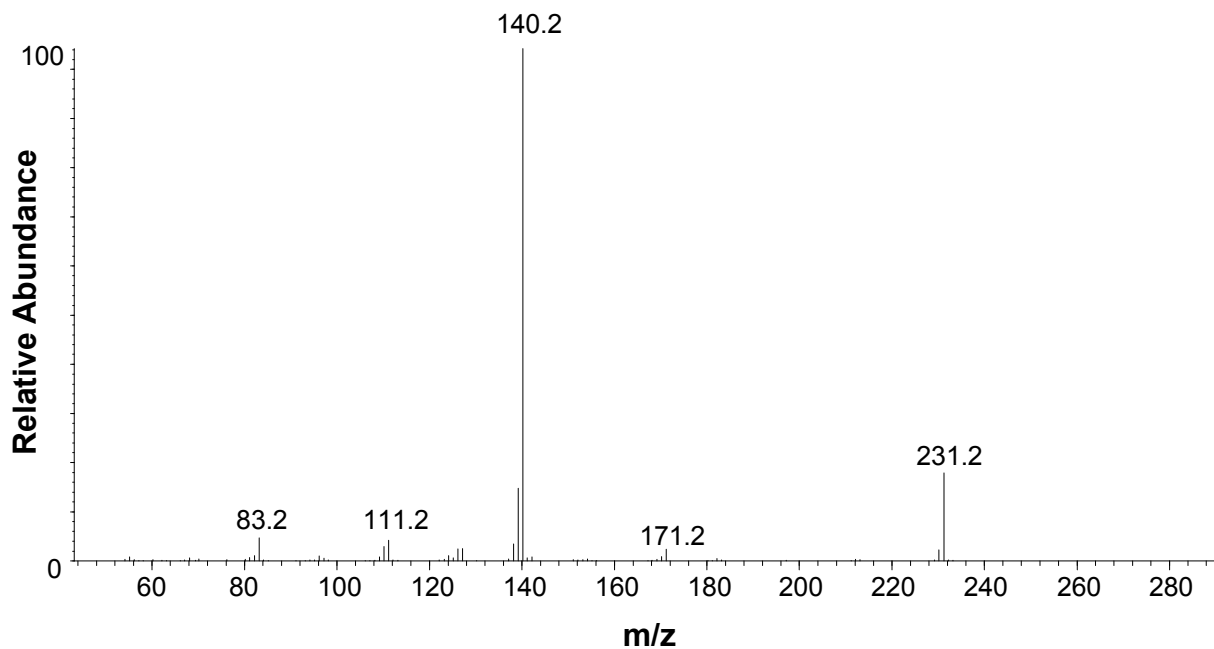
6. Add 1 ml 10 mM ascorbic acid and 50 μ l 25% PA to bring the pH to 2.5.
7. Add 600 μ l ethyl acetate to partition the synthesized [$^{13}\text{C}_{11}$, ^{15}N]IPyA.
8. Collect ethyl acetate (upper layer) into a 2-ml amber glass vial.
9. Add another 600 μ l ethyl acetate to partition, and collect the ethyl acetate into the same amber glass vial.
10. Evaporate ethyl acetate to complete dryness using a gentle N_2 gas stream in a sand bath heated to 55 $^\circ\text{C}$.
11. Re-suspend the [$^{13}\text{C}_{11}$, ^{15}N]IPyA in 1 ml 50% isopropanol. The product can be verified by GC-MS using a “Full Scan” mode after NaBH_4 derivatization (see the procedure in the main text). An example spectral scan is shown in **Additional Figure 2**, which illustrates the identity and purity of the labeled compound.
12. Cap the vial and store the solution at -20 $^\circ\text{C}$.
13. The yield of [$^{13}\text{C}_{11}$, ^{15}N]IPyA is about 0.1 mg.

► **CRITICAL:** The concentration of the [$^{13}\text{C}_{11}$, ^{15}N]IPyA solution has to be determined immediately before use (Step 2 in “Procedure”).

Determine the concentration of [$^{13}\text{C}_{11}$, ^{15}N]IPyA solution by reverse isotope dilution

14. Freshly prepare 0.1 mg/ml unlabeled IPyA in 50% isopropanol.
15. Freshly prepare 20 mg/ml NaB^2H_4 in 0.3 N NaOH.
16. In a 0.5-ml microcentrifuge tube, accurately add 5 μ l each of unlabeled IPyA and synthesized [$^{13}\text{C}_{11}$, ^{15}N]IPyA solution.
- **CRITICAL STEP:** for the ease of calculation in Step 15, it is recommended to add equal volumes of IPyA and [$^{13}\text{C}_{11}$, ^{15}N]IPyA solutions.
17. Add 8 μ l NaB^2H_4 solution and mix well.
18. Incubate the tube at 37 $^\circ\text{C}$ for 30 min.
19. Add 5 μ l 25% PA to acidify the mixture and consume the residual NaB^2H_4 .
20. Add 200 μ l water and mix well.

21. Extract the reduced IPyA (now ILA) by adding 100 μ l ethyl acetate to partition.
22. Collect the ethyl acetate extract in a 2-ml clear glass vial.
23. Add 100 μ l methanol to the vial.
24. Methylate the sample by filling the vial with ethereal diazomethane.
25. Evaporate the solvents until complete dryness using a N_2 gas stream.
26. Re-suspend the compounds in 1 ml of ethyl acetate.
27. Analyze the sample using GC-MS/MS under SRM mode. For unlabeled methyl-ILA, the transition of parent ion m/z 220 to the product ion m/z 130 is monitored; for methyl- $[^{13}C_{11}, ^{15}N]$ ILA, the transition of parent ion m/z 232 to the product ion m/z 140 is monitored.
28. The ion abundance ratio of m/z 130 over m/z 140 equals the ratio of IPyA concentration over $[^{13}C_{11}, ^{15}N]$ IPyA concentration in the initial sample.



Additional Figure 2 Analyses of $[^{13}\text{C}_{11}, ^{15}\text{N}]$ IPyA internal standards. A full-scan mass spectrum showing ions generated from Me- $[^{13}\text{C}_{11}, ^{15}\text{N}]$ ILA.