# Angewandte manmone 

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## Practical and General Palladium-Catalyzed Synthesis of Ketones from Internal Olefins** <br> Bill Morandi, Zachary K. Wickens, and Robert H. Grubbs*

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## Supporting Information

## General procedure:

All olefin oxidation reactions were carried out under aerobic conditions. Commercial reagents were obtained from Aldrich and used without further purification.


#### Abstract

${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Varian 500 Mhz spectrometer and High resolution mass spectra were provided by the California Institute of Technology Mass Spectrometry Facility using JEOL JMS-600H High Resolution Mass Spectrometer.


Gas chromatography data was obtained using an Agilent 6850 FID gas chromatography system equipped with a HP-5 (5\%-phenyl)-methylpolysiloxane capillary column (Agilent). Response factors were collected for 4-octanone, 3-octanone, 2-octanone, cyclohexanone, dodecene, 2-dodecanone and lauric aldehyde following literature procedures. ${ }^{1}$

General Procedure 1 (Table 1): The corresponding palladium complex ( $0.01 \mathrm{mmol}, 5$ $\mathrm{mol} \%$ ) and benzoquinone ( $21.6 \mathrm{mg}, 0.2 \mathrm{mmol}, 1$ equiv) were charged in a resealable 4-mL vial under air. The corresponding solvent mixture was then added, followed by the addition of aqueous $\mathrm{HBF}_{4}$. After the addition of trans-4-octene ( $22.4 \mathrm{mg}, 0.2 \mathrm{mmol}$ ), the homogenous reaction mixture was stirred for 16 h at room temperature. The crude reaction mixture was then partitioned using a mixture of ether and water ( 10 mL each ), tridecane was added as a standard, and an aliquot of the organic phase was submitted to GC-analysis to determine the yield of 4-octanone, 3-octanone, 2-octanone.

[^0]General Procedure 2 (Table 2 and Scheme 2): Palladium acetate ( $11.5 \mathrm{mg}, 0.05 \mathrm{mmol}, 5$ $\mathrm{mol} \%$ ) and benzoquinone ( $108 \mathrm{mg}, 1.00 \mathrm{mmol}$ ) were charged in a resealable $20-\mathrm{mL}$ vial under air. A mixture of DMA ( 2.2 mL ) , MeCN $(2.2 \mathrm{~mL})$ and water $(0.63 \mathrm{~mL})$ was added, followed by the addition of aqueous $\mathrm{HBF}_{4}(0.18 \mathrm{~mL}, 48 \%$ in water, 1.38 mmol ). After the addition of the corresponding substrate $(1.00 \mathrm{mmol})$, the homogenous reaction mixture was stirred for 16 h at room temperature. The crude reaction mixture was then diluted with brine $(30 \mathrm{~mL})$ and ether $(30 \mathrm{~mL})$, the phases were separated and the aqueous phase was further extracted (2x) with ether. The combined organic phases were then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and evaporated in vacuo. In some cases, NMR-analysis of the crude mixture was performed to determine the regioselectivity of the process. The crude product was then further purified by column chromatography on silica gel using pentane/ether as eluent.

General Procedure 3 (Scheme 3): Palladium acetate ( $11.5 \mathrm{mg}, 0.05 \mathrm{mmol}, 5 \mathrm{~mol} \%$ ), benzoquinone ( $10.8 \mathrm{mg}, 0.10 \mathrm{mmol}, 10 \mathrm{~mol} \%$ ) and $\mathrm{Fe}($ phtalocyanin) ( $28.4 \mathrm{mg}, 0.05 \mathrm{mmol}, 5$ $\mathrm{mol} \%$ ) were charged in a resealable $20-\mathrm{mL}$ vial under air. A mixture of DMA ( 2.2 mL ), $\mathrm{MeCN}(2.2 \mathrm{~mL})$ and water $(0.63 \mathrm{~mL})$ was added, followed by the addition of aqueous $\mathrm{HBF}_{4}$ ( $0.18 \mathrm{~mL}, 48 \%$ in water, 1.38 mmol ). The mixture was then purged during 2 min using an oxygen balloon, and after the addition of the corresponding substrate ( 1 mmol ), the homogenous reaction mixture was stirred for 16 h at room temperature under an atmospheric pressure of oxygen (balloon). The crude reaction mixture was then diluted with brine ( 30 mL ) and ether ( 30 mL ), the phases were separated and the aqueous phase was further extracted (2x) with ether. The combined organic phases were then dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and evaporated in vacuo. In some cases, NMR-analysis of the crude mixture was performed to determine the regioselectivity of the process. The crude product was then further purified by column chromatography on silica gel using pentane/ether as eluent.

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octan-4-one (Table 2, Entry 1)
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Was obtained as a clear oil ( $100 \mathrm{mg}, 0.78 \mathrm{mmol}, 78 \%$ ) following the general procedure 2. The yield obtained by GC-analysis of the crude was $87 \%$. The difference is attributed to the high volatility of the compound.
${ }^{1} \mathrm{H}$ NMR: $\delta 2.35(\mathrm{q}, J=7.0 \mathrm{~Hz}, 4 \mathrm{H}), 1.62-1.47(\mathrm{~m}, 4 \mathrm{H}), 1.33-1.22(\mathrm{~m}, 2 \mathrm{H}), 0.87(\mathrm{td}, J=$ $7.4,3.3 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 211.5,44.7,42.5,25.9,22.3,17.3,13.8,13.7$.

Spectral data were in accordance with a commercial sample.
octan-4-one (Table 2, Entry 2)


Cis-4-octene was reacted following the general procedure 2 . The mixture of crude products was analyzed by GC using tridecane as a standard. Yields of products: 3\% 2-octanone, 3\% 3octanone, $70 \%$ 4-octanone.
octan-2-one and octan-3-one (Table 2, Entry 3)


Trans-2-octene was reacted following the general procedure 2 . The mixture of crude products was analyzed by GC using tridecane as a standard. Yields of products: 62\% 2-octanone, $25 \%$ 3-octanone, 3\% 4-octanone.

## cyclohexanone (Table 2, Entry 4)



Cyclohexene was reacted following the general procedure 2. The mixture of crude products was analyzed by GC using tridecane as a standard. $75 \%$ yield was obtained. Around $9 \%$ cyclohexenone was observed by NMR spectroscopy using mesitylene as an internal standard.

## 1-(4-methoxyphenyl)propan-1-one (Table 2, Entry 5)



Was obtained as a solid ( $137 \mathrm{mg}, 0.84 \mathrm{mmol}, 84 \%$ ) following the general procedure 2.
${ }^{1} \mathrm{H}$ NMR: $\delta 7.92(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}), 2.92(\mathrm{q}, J=7.3$ $\mathrm{Hz}, 2 \mathrm{H}), 1.18(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 199.4,163.3,130.2,130.0,113.6,55.4,31.4$, 8.4.

Values were in accordance with a commercial sample.

## Propiophenone and phenyl acetone (Table 2, Entry 6)



Were obtained from trans- $\beta$-methyl styrene following a modified general procedure 2 using $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(4.4 \mathrm{~mL} / 0.63 \mathrm{~mL})$ as the solvent. Crude ratio by NMR was $1: 1$. The products could be separated by column chromatography, giving two clear oils (A: $62 \mathrm{mg}, 0.46 \mathrm{mmol}$, $46 \%$ and B: $60 \mathrm{mg}, 0.45 \mathrm{mmol}, 45 \%)$.

A: ${ }^{1} \mathrm{H}$ NMR: $7.98-7.94(\mathrm{~m}, 2 \mathrm{H}), 7.57-7.52(\mathrm{~m}, 1 \mathrm{H}), 7.48-7.43(\mathrm{~m}, 2 \mathrm{H}), 3.00(\mathrm{q}, J=7.2$ $\mathrm{Hz}, 2 \mathrm{H}), 1.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 200.8,136.9,132.9,128.5,128.0,31.8,8.2$.

B: ${ }^{1} \mathrm{H}$ NMR: $\delta 7.36-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.30-7.25(\mathrm{~m}, 1 \mathrm{H}), 7.23-7.19(\mathrm{~m}, 2 \mathrm{H}), 3.70(\mathrm{~s}, 2 \mathrm{H})$, 2.15 (s, 3H). ${ }^{13} \mathrm{C}$ NMR: $\delta 206.3,134.2,129.4,128.8,127.1,51.0,29.3$.

Values were in accordance with a commercial sample.

## Propiophenone and phenyl acetone (Table 2, Entry 7)



Were obtained from cis- $\beta$-methyl styrene following a modified general procedure 2 using $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(4.4 \mathrm{~mL} / 0.63 \mathrm{~mL})$ as the solvent. Crude ratio by NMR was 1.4:1 (A:B). The products could be separated by column chromatography, giving two clear oils (A: 75 mg , $0.56 \mathrm{mmol}, 56 \%$ and B: $47 \mathrm{mg}, 0.35 \mathrm{mmol}, 35 \%)$.

A: ${ }^{1} \mathrm{H}$ NMR: $7.98-7.94(\mathrm{~m}, 2 \mathrm{H}), 7.57-7.52(\mathrm{~m}, 1 \mathrm{H}), 7.48-7.43(\mathrm{~m}, 2 \mathrm{H}), 3.00(\mathrm{q}, J=7.2$ $\mathrm{Hz}, 2 \mathrm{H}), 1.22(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 200.8,136.9,132.9,128.5,128.0,31.8,8.2$.

B: ${ }^{1} \mathrm{H}$ NMR: $\delta 7.36-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.30-7.25(\mathrm{~m}, 1 \mathrm{H}), 7.23-7.19(\mathrm{~m}, 2 \mathrm{H}), 3.70(\mathrm{~s}, 2 \mathrm{H})$, $2.15(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 206.3,134.2,129.4,128.8,127.1,51.0,29.3$.

Values were in accordance with a commercial sample.

## 3-oxo-3-phenylpropyl acetate (Table 2, Entry 8)



Was obtained a as clear oil ( $153 \mathrm{mg}, 0.80 \mathrm{mmol}, 80 \%$ ) following a modified general procedure 2 using $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(4.4 \mathrm{~mL} / 0.63 \mathrm{~mL})$ as the solvent and $10 \mathrm{~mol} \%$ palladium acetate.
${ }^{1} \mathrm{H}$ NMR: $\delta 7.97-7.93(\mathrm{~m}, 2 \mathrm{H}), 7.60-7.55(\mathrm{~m}, 1 \mathrm{H}), 7.49-7.44(\mathrm{~m}, 2 \mathrm{H}), 4.51(\mathrm{t}, J=6.4$ $\mathrm{Hz}, 2 \mathrm{H}), 3.31(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.02(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 197.0,171.0,136.5,133.4,128.7$, 128.0, 59.6, 37.3, 20.9.

Values are in accordance with literature. ${ }^{2}$

## 4-oxohexyl 2-hydroxybenzoate (Table 2, Entry 9)



Was obtained as an oil ( $176 \mathrm{mg}, 0.75 \mathrm{mmol}, 75 \%$ ) following the general procedure 2. Crude NMR analysis showed the formation of a $4: 1$ mixture of regioisomers. Only the major product was isolated by column chromatography.

[^1][^2]
## 4-oxohexyl benzoate (Table 2, Entry 10)



Was obtained a as clear oil ( $200 \mathrm{mg}, 0.91 \mathrm{mmol}, 91 \%, 4: 1$ mixture) following the general procedure 2.
${ }^{1} \mathrm{H}$ NMR: $\delta 8.04-7.95(\mathrm{~m}, 2 \mathrm{H}), 7.57-7.50(\mathrm{~m}, 1 \mathrm{H}), 7.46-7.38(\mathrm{~m}, 2 \mathrm{H}), 4.58(\mathrm{t}, J=6.4$ $\mathrm{Hz}, 2 \mathrm{H}$, minor), $4.31(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.86(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}$, minor), $2.56(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $2 \mathrm{H}), 2.44(\mathrm{q}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.12-1.97(\mathrm{~m}, 2 \mathrm{H}), 1.70-1.51(\mathrm{~m}, 2 \mathrm{H}$, minor), $1.04(\mathrm{t}, J=$ $7.3 \mathrm{~Hz}, 3 \mathrm{H}$ ), $0.91\left(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}\right.$, minor). ${ }^{13} \mathrm{C}$ NMR: $\delta 210.3$, 207.9 (minor), 166.5, 166.4 (minor), 133.0 (minor), 132.9, 130.2 (minor), 129.5 (minor), 129.5, 128.3, 128.3 (minor), 64.2, 60.0 (minor), 45.1 (minor), 41.4 (minor), 38.6, 36.0, 22.9, 17.1 (minor), 13.7 (minor), 7.8 .

Values are in accordance with literature. ${ }^{3}$

## 1,4-bis(benzyloxy)butan-2-one (Table 2, Entry 11)



Was obtained a as clear oil ( $150 \mathrm{mg}, 0.53 \mathrm{mmol}, 53 \%$ ) following a modified general procedure 2 using $10 \mathrm{~mol} \%$ palladium acetate.
${ }^{1} \mathrm{H}$ NMR: $\delta 7.38-7.28(\mathrm{~m}, 10 \mathrm{H}), 4.59(\mathrm{~s}, 2 \mathrm{H}), 4.50(\mathrm{~s}, 2 \mathrm{H}), 4.11(\mathrm{~s}, 2 \mathrm{H}), 3.77(\mathrm{t}, J=6.2 \mathrm{~Hz}$, 2H), $2.75(\mathrm{t}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 207.0,138.0,137.2,128.5,128.4,128.0,127.9$, 127.7, 127.7, 75.4, 73.3, 73.3, 65.0, 39.4 .

[^3]Values are in accordance with literature. ${ }^{4}$

## 10-oxooctadecanoic acid and 9-oxooctadecanoic acid (Table 2, Entry 12)



Were obtained as white solids ( $245 \mathrm{mg}, 0.82 \mathrm{mmol}, 82 \%, 1: 1$ ) following the general procedure 2
${ }^{1} \mathrm{H}$ NMR: $\delta 2.37-2.33(\mathrm{~m}, 6 \mathrm{H}), 1.66-1.49(\mathrm{~m}, 6 \mathrm{H}), 1.35-1.20(\mathrm{~m}, 18 \mathrm{H}), 0.86(\mathrm{t}, J=7.0$ $\mathrm{Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 211.8,211.8,180.0,178.0,42.8,42.8,42.7,42.7,34.0,34.0,31.9,31.8$, 29.4, 29.4, 29.4, 29.3, 29.2, 29.2, 29.1, 29.0, 29.0, 29.0, 28.8, 24.6, 24.6, 23.9, 23.8, 23.7, 22.7, 22.6, 14.1, 14.1. HRMS (EI): calcd $\mathrm{C}_{18} \mathrm{H}_{34} \mathrm{O}_{3}\left(\mathrm{M}^{+}\right):$298.2508; measured: 298.2499.

Values are in accordance with literature. ${ }^{5}$

## methyl 10-oxooctadecanoate and methyl 9-oxooctadecanoate (Table 2, Entry 13)



Were obtained as white solids ( $261 \mathrm{mg}, 0.84 \mathrm{mmol}, 84 \%, 1: 1$ ) following the general procedure 2

[^4][^5]29.2, 29.2, 29.1, 29.0, 29.0, 28.9, 24.9, 24.8, 23.9, 23.8, 23.7, 14.1, 14.1. HRMS (EI): calcd $\mathrm{C}_{18} \mathrm{H}_{36} \mathrm{O}_{3}\left(\mathrm{M}^{+}\right)$: 312.2664 ; measured: 312.2674.

Values are in accordance with literature. ${ }^{6}$

## 18-hydroxyoctadecan-9-one and 1-hydroxyoctadecan-9-one (Table 2, Entry 14)



Were obtained as white solids ( $215 \mathrm{mg}, 0.76 \mathrm{mmol}, 76 \%, 1: 1$ ) following the general procedure 2.
${ }^{1} \mathrm{H}$ NMR: $\delta 3.62(\mathrm{t}, J=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.36(\mathrm{t}, J=7.5 \mathrm{~Hz}, 4 \mathrm{H}), 1.60-1.40(\mathrm{~m}, 7 \mathrm{H}), 1.36-$ $1.18(\mathrm{~m}, 20 \mathrm{H}), 0.86(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 211.8,211.7,63.0,62.9,42.8,42.8$, $42.8,42.7,32.7,32.7,31.8,31.8,29.4,29.4,29.4,29.4,29.3,29.3,29.3,29.2,29.2,29.1$, 29.1, 25.7, 25.6, 23.9, 23.8, 23.8, 22.6, 22.6, 14.1, 14.1. HRMS (EI): calcd $\mathrm{C}_{18} \mathrm{H}_{36} \mathrm{O}_{2}\left(\mathrm{M}^{+}\right)$: 284.2715; measured: 284.2721.

Values are in accordance with literature. ${ }^{7}$
dodecan-2-one (Table 2, Entry 15)


Was obtained a as clear oil ( $158 \mathrm{mg}, 0.86 \mathrm{mmol}, 86 \%$ ) following the general procedure 2. GC-analysis of the crude sample showed $97.5 \%$ selectivity for ketone formation ( $2.5 \%$ for the aldehyde).

[^6]${ }^{1} \mathrm{H}$ NMR: $\delta 2.39(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.11(\mathrm{~s}, 3 \mathrm{H}), 1.54(\mathrm{p}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 1.30-1.15(\mathrm{~m}$, $14 \mathrm{H}), 0.86(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 209.3,43.8,31.9,29.8,29.5,29.4,29.4,29.3$, 29.2, 23.8, 22.6, 14.1.

Values were in accordance with a commercial sample

## 3-(1,3-dioxoisoindolin-2-yl)butanal (Table 2, Entry 16)



Was obtained as a white solid ( $188 \mathrm{mg}, 0.87 \mathrm{mmol}, 87 \%$ ) following the general procedure 2.
${ }^{1} \mathrm{H}$ NMR: $\delta 9.74(\mathrm{~s}, 1 \mathrm{H}), 7.80(\mathrm{dd}, J=5.5,3.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.70(\mathrm{dd}, J=5.5,3.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.94-$ $4.86(\mathrm{~m}, 1 \mathrm{H}), 3.29$ (ddd, $J=18.0,8.2,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.00(\mathrm{ddd}, J=18.0,6.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.49$ (d, $J=7.0 \mathrm{~Hz}, 3 \mathrm{H}$ ). ${ }^{13} \mathrm{C}$ NMR: $\delta 199.3,168.1,134.0,131.8,123.2,47.3,41.4,18.8$.

Values are in accordance with literature. ${ }^{8}$

N -(4-hydroxy-3-methoxybenzyl)-8-methyl-7-oxononanamide and N -(4-hydroxy-3-methoxybenzyl)-8-methyl-6-oxononanamide (Scheme 2)


[^7]Was obtained as a clear oil $(128 \mathrm{mg}, 0.40 \mathrm{mmol}, 80 \%, 5: 1)$ from a mixture of capsaicin and dehydrocapsaicin (TCI, $60 \%$ capsaicin) following a modified general procedure 2 on a 0.5 mmol substrate and using $10 \mathrm{~mol} \%$ palladium acetate.
${ }^{1} \mathrm{H}$ NMR: $\delta 6.83(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.78(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{dd}, J=8.0,1.9 \mathrm{~Hz}, 1 \mathrm{H})$, $5.96-5.89(\mathrm{~m}, 2 \mathrm{H}), 4.31(\mathrm{~d}, J=5.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.84(\mathrm{~s}, 3 \mathrm{H}), 2.55$ (hept, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.41$ (t, $J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.24$ (d, $J=7.0 \mathrm{~Hz}, 2 \mathrm{H}$, minor), $2.18(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.09(\mathrm{~m}, 2 \mathrm{H}$, minor), 1.67-1.58 (m, 2H), 1.57-1.50 (m, 2H), $1.33-1.24(\mathrm{~m}, 2 \mathrm{H}), 1.05(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 6 \mathrm{H})$, 0.88 (d, $J=6.6 \mathrm{~Hz}, 6 \mathrm{H}$, minor). ${ }^{13} \mathrm{C}$ NMR: $\delta 215.0,210.9$ (minor), $172.8,172.5$ (minor), 146.7, 145.1, 130.3, 130.2 (minor), 120.7, 114.4, 110.7, 55.9, 51.8 (minor), 43.5, 42.8 (minor), 40.8, 39.9, 36.4, 36.4 (minor), 28.7, 25.5, 25.1 (minor), 24.6 (minor), 23.2, 23.0 (minor), 22.5 (minor), 18.2. HRMS (EI): calcd for $\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{NO}_{4}$ (M+): 321.1940; found: 321.1951.

## octan-4-one (Scheme 3)



Was obtained following the general procedure 3. A yield of $83 \%$ was obtained by GCanalysis of the crude.

## 1-(4-methoxyphenyl)propan-1-one (Scheme 3)



Was obtained as a solid ( $1.59 \mathrm{~g}, 9.7 \mathrm{mmol}, 72 \%$ ) on a 2 g -scale following general procedure 3. In that case a washing of the ethereal phase with aq. LiCl was necessary to remove DMA prior to chromatography.

## 10-oxooctadecanoic acid and 9-oxooctadecanoic acid (Scheme 3)



Were obtained as white solids ( $235 \mathrm{mg}, 0.79 \mathrm{mmol}, 79 \%$, $1: 1$ ) following the general procedure 3 .
dodecan-2-one (Scheme 3)


Was obtained as a clear oil ( $140 \mathrm{mg}, 0.76 \mathrm{mmol}, 76 \%$ ) following the general procedure 3 .

## Reaction profile (Figure 1)

Each profile was generated in triplicate and the values were averaged and graphed using Microsoft Excel to produce the final curves.

Palladium acetate ( $11.5 \mathrm{mg}, 0.05 \mathrm{mmol}, 5 \mathrm{~mol} \%$ ) and benzoquinone ( $108 \mathrm{mg}, 1.00 \mathrm{mmol}$ ) were charged into $8-\mathrm{mL}$ vials with permeable septum caps under air. 5.4 mL of a stock solution consisting of all of the liquid components was added (stock solution: $9 \mathrm{~mL} \mathrm{MeCN}, 9$ mL DMA, $2 \mathrm{~mL} \mathrm{H}_{2} \mathrm{O}, 0.72 \mathrm{~mL} \mathrm{HBF}_{4}$ ( $48 \%$ in water), $250 \mu \mathrm{~L} \mathrm{PhNO}_{2}$ (to be used as an internal standard) and $628 \mu \mathrm{~L}$ of either trans-4-octene or cis-4-octene (for $\mathbf{A}$ or $\mathbf{B}$ respectively)). Time points were taken at the given times and quenched with a 3:1 mixture of EtOAc and $\mathrm{Et}_{3} \mathrm{~N}$, followed by analysis with GC.






CARBONO1
BM-193_CC



PROTONO1
BM-192_CC




CARBONO1 BM-192 CC




CARBONO1
BM-203_CC


BM-203_CC

${ }^{\text {PROTONOI }}$
BM-248B_CC



CARBONO1
CARBONO1
BM-248B_CC



PROTONO1
BM-238 CC
BM-238_CC



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CARBONO1
BM-238_CC



## PROTONOI




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CARBONOI
BM-233B_CC






[^8]PROTONOL
BM-207A_CC



CARBON01
BM-207A_CC





CARBONOI
BM-208_CC




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CARBONO1
BM-215_CC



PROTONO1
BM-200_CC



CARBONO1
BM-200_CC



PROTONO1
BM-242 CC
BM-242_CC



CARBONO1
BM-242_CC






CARBONOI
CARBONOI
BM-247 CC



[^9]
[^0]:    ${ }^{1}$ Ritter, T.; Hejl, A.; Wenzel, A. G.; Funk, T. W.; Grubbs, R. H. Organometallics 2006, 25, 5740.

[^1]:    ${ }^{1} \mathrm{H}$ NMR: $\delta 10.77(\mathrm{~s}, 1 \mathrm{H}), 7.80(\mathrm{dd}, J=8.0,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.45(\mathrm{ddd}, J=8.6,7.2,1.7 \mathrm{~Hz}, 1 \mathrm{H})$, 6.97 (dd, $J=8.4,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.87$ (ddd, $J=8.2,7.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.35(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H})$, $2.57(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.45(\mathrm{q}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.13-2.01(\mathrm{~m}, 2 \mathrm{H}), 1.06(\mathrm{t}, J=7.3 \mathrm{~Hz}$, 3H). ${ }^{13} \mathrm{C}$ NMR: $\delta 210.1,170.1,161.7,135.7,129.8,119.1,117.6,112.4,64.6,38.3,36.1$, 22.7, 7.8. HRMS (EI): calcd (M+): 236.2049; measured: 236.2046.

[^2]:    ${ }^{2}$ Org. Lett 2012, 14, 2414.

[^3]:    ${ }^{3}$ Org Lett 2011, 13, 4308.

[^4]:    ${ }^{1} \mathrm{H}$ NMR: $\delta 3.65$ (s, 3H), 2.36 (t, $\left.J=7.5 \mathrm{~Hz}, 4 \mathrm{H}\right), 2.28(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.60-1.50(\mathrm{~m}, J$ $=28.6,7.5 \mathrm{~Hz}, 6 \mathrm{H}), 1.33-1.19(\mathrm{~m}, 18 \mathrm{H}), 0.86(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR: $\delta 211.6,211.6$, 174.2, 174.2, 51.4, 51.4, 42.8, 42.8, 42.7, 42.7, 34.0, 34.0, 31.8, 31.8, 29.4, 29.4, 29.4, 29.3,

[^5]:    ${ }^{4}$ Bull. Chem. Soc. Jap. 1981, 54, 3100.
    ${ }^{5}$ Biosci. Biotechnol. Biochem 2007, 71, 1120. Phytochemistry 1990, 29, 2323.

[^6]:    ${ }^{6}$ Biosci. Biotechnol. Biochem 2007, 71, 1120. Phytochemistry 1996, 42, 889.
    ${ }^{7}$ Tetrahedron 1995, 51, 11863.

[^7]:    ${ }^{8}$ J. Am. Chem. Soc. 2009, 131, 9473.

[^8]:    

[^9]:    

