

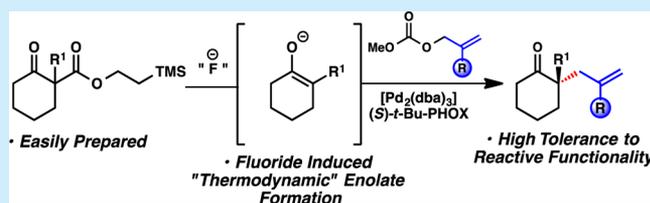
Development of (Trimethylsilyl)ethyl Ester Protected Enolates and Applications in Palladium-Catalyzed Enantioselective Allylic Alkylation: Intermolecular Cross-Coupling of Functionalized Electrophiles

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S Supporting Information

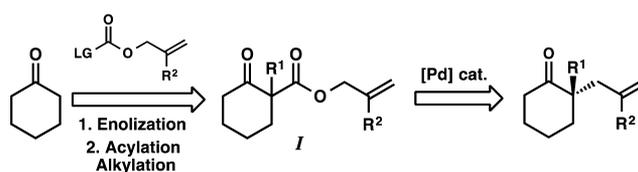
ABSTRACT: The development of (trimethylsilyl)ethyl ester protected enolates is reported. The application of this class of compounds in palladium-catalyzed asymmetric allylic alkylation is explored, yielding a variety of α -quaternary six- and seven-membered ketones and lactams. Independent coupling partner synthesis engenders enhanced allyl substrate scope relative to traditional β -ketoester substrates; highly functionalized α -quaternary ketones generated by the union of (trimethylsilyl)ethyl β -ketoesters and sensitive allylic alkylation coupling partners serve to demonstrate the utility of this method for complex fragment coupling.



Latent or protected enolates such as silyl enol ethers, silyl ketene acetals, allyl enol carbonates, allyl β -keto esters, and others have found broad use in organic synthesis owing to their mild release and ease of use.¹ Perhaps the most well studied class of protected enolates employ oxygen-bound protecting groups (i.e., silyl enol ethers). Unfortunately, the utility of this class of compounds is often limited by poor regioselectivity when forming fully substituted enol derivatives.² Although much effort has been devoted to the identification of conditions that allow for selective generation of so-called “thermodynamic” enolate isomers, selectivity often drops precipitously when sterically demanding α -substitution is introduced.³ This problem would be solved, ideally, by the development of enolate precursors that are readily prepared and, when triggered, release the “thermodynamic” enolate under kinetic control.

In the context of allylic alkylation reactions, carboxylate-protected enolates (i.e., allyl β -ketoesters, e.g., **I**, Figure 1A) represent a significant advance toward such a solution. Allyl β -ketoesters enjoy relatively uncomplicated, selective synthesis⁴ and mild deprotection, resulting in enolate formation following decarboxylation.⁵ Despite these advantages, facile nucleophilic attack of the incipient enolate at the transition metal-allyl species generated during deprotection often precludes applications that do not involve allylic alkylation.⁶ Moreover, with traditional carboxylate-protected enolates any functionality borne by the allyl fragment (R^2 , Figure 1A) must be compatible with the conditions required for substrate synthesis (i.e., strong base and reactive electrophiles, Figure 1A). Tunge and co-workers have demonstrated the utility of acyl-protected

A. Previous Reports: R^2 Subjected to Acylation/Alkylation Sequence



B. Current Research: Expanded Functional Group Tolerance

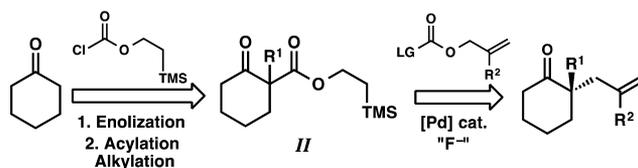


Figure 1. New substrate design enables broader functional group (R^2) scope in allylic alkylation reactions.

enolates, which may undergo deprotection via a retro Claisen condensation to reveal fully substituted enolates that participate in catalysis.⁷ However, these reactions often require the use of elevated temperatures and alkoxide base to proceed.

In a recent communication,⁸ we disclosed a novel class of substrates for enolate alkylation chemistry,⁹ (trimethylsilyl)ethyl β -ketoesters (TMSE β -ketoester), that undergo mild deprotection upon treatment with a fluoride source, and we demonstrated their use in the diastereoselective allylic alkylation of cyclic β -ketoesters. The TMSE β -ketoester

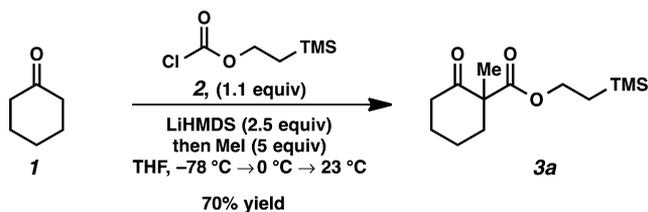
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substrate class (i.e., II, Figure 1B) boasts similar ease of preparation as compared with allyl β -ketoesters but is not susceptible to transition metal-mediated deprotection, a benefit that enabled sequential transition metal-catalyzed allylic alkylation events in our previous work.⁸ We hypothesized that use of TMSE β -ketoesters may enhance the breadth of functional group tolerance at the allyl coupling partner in asymmetric allylic alkylations, relative to allyl β -ketoesters, by virtue of the fact that the allyl fragment is not subjected to the conditions of substrate synthesis (Figure 1B). In this report, we describe the preparation and development of this substrate class and the evaluation thereof in the enantioselective palladium-catalyzed allylic alkylation of six- and seven-membered ketone and lactam scaffolds. Furthermore, we go on to show how the use of these substrates can enable the union of complex fragments bearing functionality that would be incompatible with incorporation into traditional allyl β -ketoester substrates.

In considering novel carboxylate protected enolates, our design criteria called for a substrate that could be synthesized efficiently and deprotected under mild conditions and facilitate the convergent union of complex fragments in a synthetic setting. To address these concerns we chose to explore 2-(trimethylsilyl)ethyl 1-alkyl-2-oxocyclohexane-1-carboxylates or TMSE β -ketoesters. We were pleased to find that α -methyl TMSE β -ketoester (**3a**) could be prepared in a single synthetic operation from commercially available cyclohexanone (**1**), 2-(trimethylsilyl)ethyl chloroformate (**2**), and methyl iodide (MeI) in good overall yield (Scheme 1).

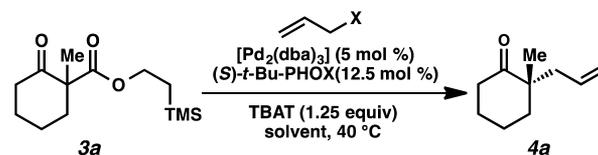
Scheme 1. Synthesis of TMSE β -Ketoester



With TMSE β -ketoester **3a** in hand, our investigation into this substrate class commenced in the context of palladium-catalyzed allylic alkylation. We were pleased to find that exposure of β -ketoester **3a** to allyl bromide, tetrabutylammonium difluorotriphenylsilicate (TBAT), $[\text{Pd}_2(\text{dba})_3]$, and (*S*)-*t*-Bu-PHOX¹⁰ in toluene at 40 °C generated the desired α -quaternary ketone **4a** in modest yield and good enantioselectivity (entry 1, Table 1). We next explored the scope of allyl sources that could be used in the reaction and found that a variety of diverse allyl sources were competent in the chemistry, including allyl sulfonates, allyl acetates, and allyl carbonates (entries 2–5). Allyl methyl carbonate proved to be the most efficient, selective, and prudent allyl source, in particular with respect to the number of the allyl equivalents required for optimal reactivity (entry 6). Reaction parameters including relative stoichiometry (entries 7–9), solvent (entries 10–13), and temperature (entry 14) were all subsequently explored and, ultimately, we found that a slight excess of mixed carbonate in THF at 25 °C proved optimal, delivering the desired ketone in 81% yield and 86% enantioselectivity (entry 14).

Having identified optimal reaction conditions, we turned our attention to exploring reaction scope, beginning with tolerance of variability with respect to the nucleophile's α -substitution, ring size, and carbonyl functionality (Figure 2). Simple α -alkyl

Table 1. Optimization of Reaction Parameters¹¹



| entry | X | equiv allyl | solvent | yield (%) ^a | ee (%) ^b |
|-----------------|------------------------|-------------|-------------|------------------------|---------------------|
| 1 | Br | 1.0 | toluene | 55 | 83 |
| 2 | OTs | 1.0 | 1,4-dioxane | 43 | 77 |
| 3 | OMs | 1.0 | 1,4-dioxane | 45 | 84 |
| 4 | OAc | 1.0 | 1,4-dioxane | 15 | 82 |
| 5 | OCO ₂ Allyl | 1.0 | 1,4-dioxane | 78 | 83 |
| 6 | OCO ₂ Me | 1.0 | 1,4-dioxane | 78 | 84 |
| 7 | OCO ₂ Me | 0.75 | 1,4-dioxane | 51 | 82 |
| 8 | OCO ₂ Me | 1.5 | 1,4-dioxane | 74 | 82 |
| 9 | OCO ₂ Me | 2.0 | 1,4-dioxane | 73 | 84 |
| 10 | OCO ₂ Me | 1.1 | toluene | 33 | 82 |
| 11 | OCO ₂ Me | 1.1 | MTBE | 65 | 84 |
| 12 | OCO ₂ Me | 1.1 | THF | 83 | 83 |
| 13 | OCO ₂ Me | 1.1 | tol/hex | 45 | 93 |
| 14 ^c | OCO ₂ Me | 1.1 | THF | 81 | 86 |

^aYield determined by comparison to tridecane internal standard.

^bPercent ee determined by chiral GC analysis of the crude reaction mixture. ^cReaction performed at 25 °C.

substitutions, such as α -benzyl substituted β -ketoester **3b** ($R^1 = \text{Bn}$, $X = Y = \text{CH}_2$, $n = 1$, Figure 2), functioned consistently well in the chemistry; the desired benzyl substituted α -quaternary ketone **4b** was obtained in high yield and enantioselectivity. In addition to simple α -alkyl substrates (i.e., compounds **3a** and **3b**), heteroatom-substituted substrate **3c** ($R^1 = \text{F}$, $X = Y = \text{CH}_2$, $n = 1$) proved to be a viable coupling partner and provided the corresponding α -fluoro-allylic alkylation product **4c** in good yield and excellent ee. Subjecting methyl ester bearing substrate **3d** ($R^1 = \text{CH}_2\text{CH}_2\text{CO}_2\text{Me}$, $X = Y = \text{CH}_2$, $n = 1$) to our optimized conditions resulted in an efficient and selective reaction, furnishing enantioenriched ketone **4d** in 93% yield and 89% ee. Substrates constituted from seven-membered rings, including ketone **3e** ($R^1 = \text{Me}$, $X = Y = \text{CH}_2$, $n = 2$) and vinylogous ester **3f** ($R^1 = \text{Me}$, $X = \text{CH}$, $Y = \text{CO}(i\text{-Bu})$, $n = 2$), were shown to be suitable coupling partners, affording α -quaternary ketone **4e** and α -quaternary vinylogous ester **4f** products in 95% and 89% yield and 87% and 92% ee, respectively. Finally, six- and seven-membered lactams were investigated. We were pleased to find that under slightly modified reaction conditions (40 °C), the desired α -functionalized lactam products **4g** and **4h** were obtained in good to excellent yields and excellent ee's.

Having surveyed the scope of reaction with respect to nucleophile α -substitution and scaffold type, we next probed the reaction with respect to substitution at the 2-allyl position. We were pleased to find that a variety of functional groups could be introduced through the use of variously substituted allyl carbonates (**5**, $R^2 \neq \text{H}$, Figure 2). Simple alkyl substitution at the internal allyl position was well tolerated as 2-methylallyl ketone **4i** was obtained in 89% yield and 89% ee. 2-Chloroallyl

efficient and diastereoselective reaction giving the desired amino ester **8** in 95% yield and greater than 25:1 dr, with no detectable epimerization at the amino ester side chain (entry 2). The inherent diastereoselectivity could be completely reversed under catalyst control by using (*R*)-*t*-Bu-PHOX (**L3**), without significant loss in selectivity or reactivity (entry 3). Likewise, upon exposing carbonate **7** and ketoester **3b** to slightly modified allylic alkylation conditions (40 °C) with an achiral ligand, we again observed an efficient reaction and slight inherent diastereoselectivity (entry 4, Table 2B). This bias could be enhanced by using ligand **L2** to obtain α -quaternary ketone **10** in 6:1 dr and 87% yield or overturned by use of **L3** to obtain **11** in 5:1 dr and 77% yield (entries 5 and 6).

In conclusion, we have developed a new class of substrates for enolate alkylation chemistry that benefit from ease of preparation and mild deprotection conditions that are orthogonal to those used with traditional allyl β -ketoesters. We examined the application of these compounds in palladium-catalyzed asymmetric allylic alkylation chemistry and found that a wide range of functional groups and substrate scaffolds are well tolerated, including six- and seven-membered ketones and lactams. We have further demonstrated the value of these compounds for uniting complex coupling partners that would be incompatible to preparation via standard allyl β -ketoester based allylic alkylation. We envision this technology will also enable the convergent cross-coupling of synthetically challenging fragments for complex molecule synthesis. Further studies exploring the application of TMSE β -ketoesters in diverse reaction methodologies and complex natural product synthesis are ongoing in our laboratory.

■ ASSOCIATED CONTENT

Supporting Information

This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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