

Observations of large reductions in the NO/NO_y ratio near the mid-latitude tropopause and the role of heterogeneous chemistry

E. R. Keim¹, D. W. Fahey¹, L. A. Del Negro^{1,2}, E. L. Woodbridge^{1,2},
R. S. Gao^{1,2}, P. O. Wennberg³, R. C. Cohen³, R. M. Stimpfle³,
K. K. Kelly¹, E. J. Hintsas³, J. C. Wilson⁴, H. H. Jonsson⁴, J. E. Dye⁵,
D. Baumgardner⁵, S. R. Kawa⁶, R. J. Salawitch⁷, M. H. Proffitt^{1,2},
M. Loewenstein⁸, J. R. Podolske⁸, and K. R. Chan⁸

Abstract. During the 1993 NASA Stratospheric Photochemistry, Aerosols and Dynamics Expedition (SPADE), anomalously low nitric oxide (NO) was found in a distinct sunlit layer located above the mid-latitude tropopause. The presence of a significant amount of reactive nitrogen (NO_y) in the layer implies the systematic removal of NO, which is without precedent in stratospheric *in situ* observations. Large increases in measured chlorine monoxide (ClO) and the hydroperoxyl radical (HO₂) also were observed in the layer. Heterogeneous reaction rate constants of chlorine nitrate (ClONO₂) with hydrogen chloride (HCl) and H₂O to form nitric acid (HNO₃) on sulfate aerosol are enhanced in the NO removal layer by local increases in H₂O and aerosol surface area. The associated conversion of NO_x (= NO + NO₂) to HNO₃ is the most likely cause of the observed low NO and NO_x/NO_y values and high ClO values.

Introduction

Reactive nitrogen (NO_y = NO + NO₂ + NO₃ + 2N₂O₅ + HONO + HO₂NO₂ + HNO₃ + PAN + ClONO₂ + aerosol nitrate + ...) species play an important role in the photochemistry of the troposphere and stratosphere. NO is an important indicator of the partitioning within the NO_y reservoir and can be used to infer and limit the abundance of the higher nitrogen oxides [Kawa *et al.*, 1992]. During the day, NO is produced by the photolysis of nitrogen dioxide (NO₂) and other NO_y species. The NO/NO₂ ratio requires only minutes in the sunlit atmosphere to reach steady state [Kawa *et al.*, 1992]. We describe here a distinct layer encountered at mid-latitudes above the sunlit tropopause during the spring of 1993 in which NO approached values near zero (< 0.02 parts per billion by volume, ppbv) and NO_y values were large (~2 ppbv). This partitioning is consistent with the effects of heterogeneous reactions in the layer involving the inorganic chlorine species HCl and ClONO₂. These reactions form HNO₃ at rates signifi-

cantly increased by a combination of high aerosol loading following the eruption of Mt. Pinatubo, high H₂O mixing ratios, and low temperatures near the tropopause.

Observations of Radical Species

The observations reported here were obtained with instruments onboard the NASA ER-2 high-altitude aircraft during its descent into Moffett Field, CA (38°N, 122°W) on 30 April 1993. The ER-2 payload included a suite of instruments designed for *in situ* measurements of a wide variety of reactive and long-lived species and aerosol parameters [Fahey *et al.*, 1995a; b]. Figure 1 displays vertical profiles of observations and calculated parameters for the near-tropopause portion of the descent. The profile range represents ~250 s of flight time close to 1500 hr local time. Based on the minimum in the temperature profile, the tropopause is located near 12.5 km at a temperature of ~205 K. The NO removal layer is defined by the region above the tropopause between 12.7 and 13.2 km where NO mixing ratios are near the detection limit of 0.01 to 0.02 ppbv. Below this layer and spanning the tropopause, NO has higher values of 0.20 ± 0.05 ppbv. Based on the ER-2 observation dataset, NO values of 0.10 to 0.30 ppbv are typically found just above the tropopause, smoothly decreasing with altitude to values < 0.10 ppbv in the upper troposphere. The removal layer also was observed during ascent on this flight (~1300 hr), and a less distinct removal layer was observed during ascent on 1 May (~0900 hr).

The measurements shown in Figure 1 were acquired in sunlight (solar zenith angle of 31°), allowing the steady-state NO₂ mixing ratio (NO_{2,ss}) to be calculated using measured ozone (O₃), ClO, HO₂, temperature, pressure, and a calculated NO₂ photolysis rate [Kawa *et al.*, 1992]. These steady-state NO₂ values are less than the observed NO throughout the profile, and thus very low (< ~0.02 ppbv) in the removal layer. Because NO_y exceeds 2 ppbv in the removal layer, the NO_x/NO_y ratio is much less than typical values of 0.1 [Fahey *et al.*, 1993], signaling an unexpectedly large repartitioning within the lower stratospheric NO_y reservoir. Expected NO_y, denoted NO_y^{*} in Figure 1, is determined from the well-established, highly compact correlation between NO_y and nitrous oxide (N₂O) observed in the lower stratosphere [Loewenstein *et al.*, 1993]. Variations in NO_y not associated with N₂O changes are expected near the tropopause, due in part to lightning and other sources of NO_y in the troposphere [Murphy *et al.*, 1993]. NO_y in excess of NO_y^{*} (~1 ppbv in the removal layer) could be caused at least in part by enhanced sampling of sulfate aerosol particles containing a significant fraction of condensed NO_y species [Fahey *et al.*, 1989].

Repartitioning of the NO_y reservoir is likely to have an impact on other species observed in the removal layer because of the strong coupling between reactive nitrogen, chlorine,

¹NOAA Aeronomy Laboratory, Boulder, CO

²Cooperative Institute for Research in Environmental Science, University of Colorado, Boulder

³Department of Chemistry, Harvard University, Cambridge, MA

⁴Department of Engineering, University of Denver, CO

⁵National Center for Atmospheric Research, Boulder, CO

⁶NASA Goddard Space Flight Center, Greenbelt, MD

⁷Jet Propulsion Laboratory, California Institute of Technology, Pasadena

⁸NASA Ames Research Center, Moffett Field, CA

Copyright 1996 by the American Geophysical Union.

Paper number 96GL02593.

0094-8534/96/96GL-02593\$05.00

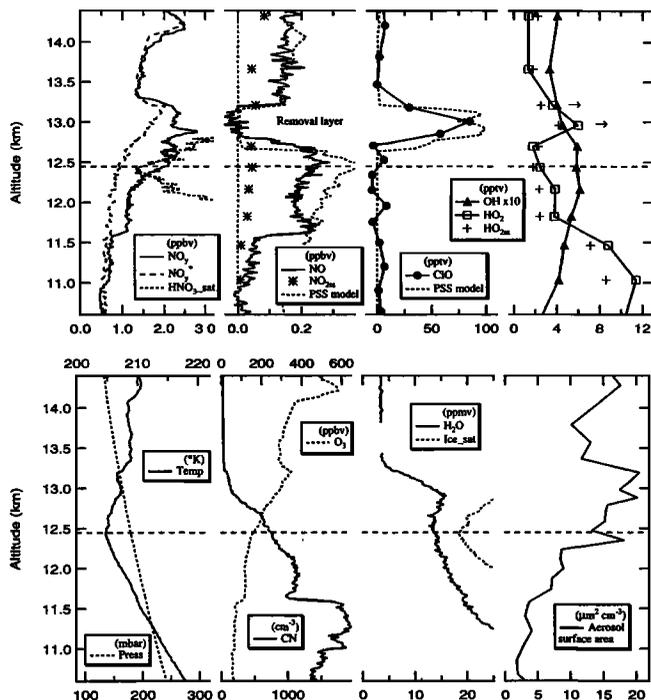


Figure 1. Observations and calculated parameters during descent on the flight of 30 April 1993. Units are noted in the legends. Measurement uncertainties are included in Fahey *et al.* [1995 a; b]. The tropopause is shown as a horizontal dashed line. NO_{2ss} and HO_{2ss} are calculated using steady-state relations. Ice_{sat} and HNO₃_{sat} are calculated equilibrium saturation mixing ratios of H₂O over ice and HNO₃ over NAT, respectively. NO_y* is the expected amount of stratospheric NO_y determined from observed N₂O and the NO_y/N₂O correlation. NO_y* values in the troposphere reflect the intercept of the fit to stratospheric observations. The arrows indicate lower limit HO₂ values.

and hydrogen photochemistry in the lower stratosphere [Wennberg *et al.*, 1994]. Because ClONO₂ is formed by the reaction of NO₂ with ClO, the removal of NO, and thus NO_x, is likely to be associated with increased ClO. In Figure 1, ClO values < 10 parts per trillion by volume (pptv) near the removal layer are typical for midday conditions at mid-latitudes [Salawitch *et al.*, 1994]. However, ClO increases to over 90 pptv in the removal layer, corresponding to ~25% of the available inorganic chlorine (Cl_y) as calculated using the observed correlation with N₂O [Woodbridge *et al.*, 1995]. These unexpectedly high values of ClO corroborate the NO removal layer as a region in the mid-latitude stratosphere with an unexpected photochemical balance.

The hydroxyl radical (OH) shows a largely invariant vertical profile across the tropopause in Figure 1. In contrast, HO₂ increases significantly upon entering the upper troposphere, increasing the HO₂/OH ratio accordingly [Wennberg *et al.*, 1995]. In the NO removal layer, lower-limit HO₂ values are significantly larger than values above the layer. Figure 1 includes steady-state HO₂ (HO_{2ss}) values calculated with the relation:

$$[\text{HO}_2]_{\text{ss}} = \frac{[\text{OH}](k_{\text{OH}+\text{O}_3}[\text{O}_3] + k_{\text{OH}+\text{CO}}[\text{CO}])}{k_{\text{HO}_2+\text{NO}}[\text{NO}] + k_{\text{HO}_2+\text{O}_3}[\text{O}_3] + k_{\text{HO}_2+\text{ClO}}[\text{ClO}] + k_{\text{HO}_2+\text{BrO}}[\text{BrO}]} \quad (1)$$

where *k* denotes the associated rate constant [Cohen *et al.*, 1994; Wennberg *et al.*, 1994]. In the near-tropopause region, the bromine monoxide (BrO) term can be neglected. Average

carbon monoxide (CO) values, as estimated using the correlation observed with O₃ [Murphy *et al.*, 1993], are 34 ppbv in the layer. The steady-state HO₂ values, which increase in the removal layer primarily because of the reduction in NO, are comparable to the lower-limit observed values.

Observations of Tracers, Aerosols, and Meteorological Parameters

Large and opposing vertical gradients of condensation nuclei (CN) and O₃ exist across the tropopause region. Stratospheric CN and O₃ values are 30 cm⁻³ and > 350 ppbv, respectively, while troposphere values are > 950 cm⁻³ and < 100 ppbv. Because of the CN and O₃ values observed in the removal layer (180 cm⁻³, 290 ppbv) and the long lifetimes of O₃ and CN (weeks to months), only a small fraction of the air in the removal layer could have been recently transported from the free troposphere where NO values are also very low.

H₂O also shows a strong gradient across the tropopause, with a local maximum (~16 parts per million by volume, ppmv) occurring in the center of the NO removal region. A similar maximum was observed in all three profiles showing evidence of an NO removal layer. Values of H₂O significantly greater than 4 ppmv are not unusual just above the tropopause at mid-latitudes [Foot, 1984]. These high values are attributed to air parcels crossing the tropopause in the extratropics [Dessler *et al.*, 1995]. The ice saturation mixing ratio, which reaches a minimum near 20 ppmv at the tropopause, remains above ambient H₂O values throughout the profile in Figure 1. Saturation mixing ratios of HNO₃ over the nitric acid trihydrate (NAT) phase (HNO₃_{sat} in Figure 1) approach available HNO₃ (or NO_y) only in a narrow layer centered at the tropopause [Fahey *et al.*, 1989]. This suggests that NAT particles are not present in the removal layer and are unlikely to be involved in its formation.

In addition to CN, aerosol spectrometer measurements provided surface area as a sum over discrete particles in sizes ranging from 0.1 to 20 µm in diameter. Based on a derived size distribution, the area associated with particles outside this size range is negligible. Surface area values in the lower stratosphere are near 1 µm² cm⁻³ during nonvolcanic periods [Hofmann and Solomon, 1989]. Larger values in Figure 1 reflect aerosol particles produced in the stratosphere after the eruption of Mt. Pinatubo in June 1991 [Trepte *et al.*, 1993]. Near the NO removal layer, the surface area shows a corresponding layer feature with maximum values of 20 µm² cm⁻³, primarily associated with particle diameters between 1 and 4 µm.

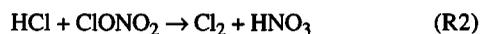
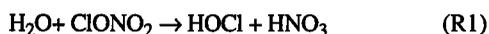
The recent history of air parcels near the tropopause was examined using isentropic trajectory calculations based on assimilated winds and temperatures from the National Meteorological Center [Newman *et al.*, 1993]. The 10-day back trajectories show that air parcels just above, in, and just below the removal layer had very similar trajectories and experienced no significant latitude excursions after the first three days. The lowest temperature (~208 K) for the removal layer parcel occurred on 30 April, while the 10-day temperature ranges for parcels above and below the removal layer were 206 to 216 K and 208 to 225 K, respectively.

Discussion

Previous *in situ* measurements have established the role of heterogeneous reactions in the partitioning of the stratospheric NO_y reservoir [Fahey *et al.*, 1993]. In particular, the dinitrogen pentoxide (N₂O₅) hydrolysis reaction on sulfate aerosol reduces the NO_x/NO_y ratio in the mid-latitude stratosphere through conversion of NO_x to HNO₃. The accommoda-

tion coefficient of this reaction is large and nearly independent of temperature and aerosol composition in the lower stratosphere. However, in these previous *in situ* measurements, the NO_x/NO_y ratio remained above 0.07 as the surface area increased to over 20 μm² cm⁻³. This nonlinear or saturation effect is the result of limitations in N₂O₅ production, which occurs primarily at night. Therefore, a process other than N₂O₅ hydrolysis must be the cause of low NO and NO_x/NO_y in the removal layer in Figure 1. This additional process must compete with NO_x production, which occurs through HNO₃ photolysis and the reaction of OH with HNO₃. For conditions in the NO removal layer, calculations of the production rate for NO_x using a photochemical steady-state (PSS) model [Salawitch *et al.*, 1994] yield a value of 8.8 × 10³ molecules cm⁻³ s⁻¹, corresponding to an HNO₃ lifetime in daylight of ~30 hr.

Two heterogeneous reactions that can affect the partitioning for both the NO_y and Cl_y reservoirs in the stratosphere are:



These reactions can take place on solid and liquid stratospheric particles such as NAT, ice, and sulfate aerosols. Normally, the partitioning in the mid-latitude stratosphere is not affected by these reactions due to their large temperature dependence, with rates that only become comparable to the N₂O₅ hydrolysis reaction at temperatures below 200 K. However, the rate constants also depend very strongly on aerosol composition, increasing exponentially with H₂O weight percent (wt%). Above the removal layer, the aerosol H₂O fraction calculated using an analytic ternary solution model [Carslaw *et al.*, 1995; Del Negro *et al.*, 1996] is 35 wt%, a value that is typical for the stratosphere. In the removal layer, the value increases to over 55 wt% due to the local increase in H₂O. This change in aerosol composition corresponds to an increase in the water activity of ~10 [Ravishankara and Hanson, 1996], with a corresponding 50-fold increase in the reactive uptake coefficient, γ, for (R1) [Hanson and Ravishankara, 1994]. The lifetime of ClONO₂ for the combination of (R1) and (R2) decreases from more than 30 days above the removal layer to 0.5 day or less in the layer. Thus, the conversion rate of NO_x to HNO₃ by (R1) and (R2) exceeds the production rate of NO_x from HNO₃, thereby largely removing NO_x within a diurnal cycle. NO removal requires both (R1) and (R2): (R2) to increase the Cl_y fraction available to form ClONO₂ (HCl reformation is slow) and (R1) to remove ClONO₂ formed with ClO and available NO₂. Since HOCl in (R1) reforms ClO through photolysis and reaction with O₃, ClONO₂ will continue to form and undergo conversion provided NO₂ is available. This cycle will maintain low levels of NO_x as long as the rate constants of (R1) and (R2) are sufficiently enhanced.

Figure 1 shows that observed profiles of ClO and NO are well reproduced by the PSS model [Salawitch *et al.*, 1994] when constrained by measured values of long-lived radical precursors (e.g., O₃, H₂O, CH₄, NO_y). The Ravishankara and Hanson [1996] formulation was used for the rate constants of (R1) and (R2), and an uptake coefficient of 0.1 was used for N₂O₅ hydrolysis. The Cl_y profile was determined from its observed correlation with N₂O in a manner similar to Woodbridge *et al.* [1995], except that Cl_y was set to zero for tropospheric N₂O values. The upper edge of the NO removal layer is caused by the sharp increase in ClONO₂ reactivity as the H₂O mixing ratio increases, and the lower edge is caused by concentrations of available Cl_y that approach zero at the tropopause.

Summary and Implications

Anomalously low NO was observed in sunlight in a 0.5-km layer located above the tropopause at mid-latitudes. The NO_x/NO_y partitioning is shifted to significantly lower values than normally observed in the mid-latitude stratosphere. Simultaneous measurements of ClO and HO₂ are enhanced when NO is low. In the layer, temperatures are low (~206 K) and aerosol surface area and H₂O values are enhanced above stratospheric values found away from the tropopause. Consequently, rate constants increase for heterogeneous reactions involving ClONO₂ which effectively convert NO_x to HNO₃. The distinct layers in NO and ClO are well simulated by the PSS model in which the upper and lower edges of the layer are controlled primarily by the availability of H₂O and Cl_y, respectively. The combination of observations and interpretation presented here provide strong evidence that the roles of (R1) and (R2) in the stratosphere are consistent with laboratory measurements. Because of low NO and enhanced ClO and HO₂ values, 24-hr average removal rates of O₃ in the removal layer increase to values of ~0.2 ppbv hr⁻¹, much larger than in adjacent regions. These observations show that NO_y and Cl_y partitioning and O₃ destruction in the lower stratosphere near the tropopause are sensitive to increases in H₂O concentrations, particularly during periods of high aerosol loading. Consequently, the heterogeneous processes and controlling parameters that cause these important photochemical changes should be represented accurately in models of global change. Of specific interest will be assessments of the global effects of continued aircraft emissions of NO_x, H₂O, and sulfur species.

Acknowledgments. This work is supported by the High-Speed Research Program and Upper Atmosphere Research Program of NASA. The authors appreciate discussions with H. S. Johnston. ERK acknowledges postdoctoral support from the National Research Council.

References

- Carslaw, K. S., B. P. Luo, and T. Peter, An analytic expression for the composition of aqueous HNO₃-H₂SO₄ stratospheric aerosols including gas phase removal of HNO₃, *Geophys. Res. Lett.*, **22**, 1877-1880, 1995.
- Cohen, R. C., *et al.*, Are models of catalytic removal of O₃ by HO_x accurate? Constraints from *in situ* measurements of the OH to HO₂ ratio, *Geophys. Res. Lett.*, **21**, 2539-2542, 1994.
- Del Negro, L. A., *et al.*, Evaluating the role of NAT, NAD, and liquid H₂SO₄/H₂O/HNO₂ solutions in Antarctic polar stratospheric cloud aerosol: Observations and implications, *J. Geophys. Res.*, in preparation, 1996a.
- Dessler, A. E., E. J. Hints, E. M. Weinstock, J. G. Anderson, and K. R. Chan, Mechanisms controlling water vapor in the lower stratosphere: "A tale of two stratospheres," *J. Geophys. Res.*, **100**, 23167-23172, 1995.
- Fahey, D. W., K. K. Kelly, G. V. Ferry, L. R. Poole, J. C. Wilson, D. M. Murphy, M. Loewenstein, and K. R. Chan, *In situ* measurements of total reactive nitrogen, total water, and aerosol in a polar stratospheric cloud in the Antarctic, *J. Geophys. Res.*, **94**, 11299-11315, 1989.
- Fahey, D. W., *et al.*, *In situ* measurements constraining the role of sulphate aerosols in mid-latitude ozone depletion, *Nature*, **363**, 509-514, 1993.
- Fahey, D. W., *et al.*, Emission measurements of the Concorde supersonic aircraft in the lower stratosphere, *Science*, **270**, 70-74, 1995a.
- Fahey, D. W., *et al.*, *In situ* observations of aircraft exhaust plumes in the lower stratosphere at midlatitudes, *J. Geophys. Res.*, **100**, 3065-3074, 1995b.
- Foot, J. S., Aircraft measurements of the humidity in the lower stratosphere from 1977 to 1980 between 45°N and 65°N, *Quart. J. R. Met. Soc.*, **110**, 303-319, 1984.
- Hanson, D. R., and A. R. Ravishankara, Reactive uptake of ClONO₂ onto sulfuric acid due to reaction with HCl and H₂O, *J. Phys. Chem.*, **98**, 5728-5735, 1994.

- Hofmann, D. J., and S. Solomon, Ozone destruction through heterogeneous chemistry following the eruption of El Chichón, *J. Geophys. Res.*, **94**, 5029-5041, 1989.
- Kawa, S. R., *et al.*, Photochemical partitioning of the reactive nitrogen and chlorine reservoirs in the high-latitude stratosphere, *J. Geophys. Res.*, **97**, 7905-7923, 1992.
- Loewenstein, M., *et al.*, New observations of the NO_y/N₂O correlation in the lower stratosphere, *Geophys. Res. Lett.*, **20**, 2531-2534, 1993.
- Murphy, D. M., D. W. Fahey, M. H. Proffitt, S. C. Liu, K. R. Chan, C. S. Eubank, S. R. Kawa, and K. K. Kelly, Reactive nitrogen and its correlation with ozone in the lower stratosphere and upper troposphere, *J. Geophys. Res.*, **98**, 8751-8774, 1993.
- Newman, P. A., *et al.*, Stratospheric meteorological conditions in the Arctic polar vortex, 1991 to 1992, *Science*, **261**, 1143-1146, 1993.
- Ravishankara, A. R., and D. R. Hanson, Differences in the reactivity of type I polar stratospheric clouds depending on their phase, *J. Geophys. Res.*, **101**, 3885-3890, 1996.
- Salawitch, R. J., *et al.*, The distribution of hydrogen, nitrogen, and chlorine radicals in the lower stratosphere: Implications for changes in O₃ due to emission of NO_y from supersonic aircraft, *Geophys. Res. Lett.*, **21**, 2547-2550, 1994.
- Trepte, C. R., R. E. Veiga, and M. P. McCormick, The poleward dispersal of Mount Pinatubo volcanic aerosol, *J. Geophys. Res.*, **98**, 18563-18573, 1993.
- Wennberg, P. O., *et al.*, Removal of stratospheric O₃ by radicals: *In situ* measurements of OH, HO₂, NO, NO₂, ClO, and BrO, *Science*, **266**, 398-404, 1994.
- Wennberg, P. O., T. F. Hanisco, R. C. Cohen, R. M. Stimpfle, L. B. Lapson, and J. G. Anderson, *In situ* measurements of OH and HO₂ in the upper troposphere and stratosphere, *J. Atmos. Sci.*, **52**, 3413-3420, 1995.
- Woodbridge, E. L., *et al.*, Estimates of total organic and inorganic chlorine in the lower stratosphere from *in situ* and flask measurements during AASE II, *J. Geophys. Res.*, **100**, 3057-3064, 1995.
-
- E. R. Keim, D. W. Fahey, K. K. Kelly, NOAA Aeronomy Laboratory, Boulder, CO 80303
- L. A. Del Negro, E. L. Woodbridge, R. S. Gao, M. H. Proffitt, NOAA Aeronomy Laboratory and Cooperative Institute for Research in Environmental Science (CIRES), University of Colorado, Boulder 80309
- P. O. Wennberg, R. C. Cohen, R. M. Stimpfle, E. J. Hints, Department of Chemistry, Harvard University, Cambridge, MA 02138
- J. C. Wilson, H. H. Jonsson, Department of Engineering, University of Denver, Denver, CO 80208
- J. E. Dye, D. Baumgardner, National Center for Atmospheric Research, Boulder, CO 80307
- S. R. Kawa, NASA Goddard Space Flight Center, Greenbelt, MD 20771
- R. J. Salawitch, Jet Propulsion Laboratory, California Institute of Technology, Pasadena, CA 91109
- M. Loewenstein, J. R. Podolske, K. R. Chan, NASA Ames Research Center, Moffett Field, CA 94035

(Received June 10, 1996; revised August 14, 1996; accepted August 19, 1996)