

Genetically programmed chiral organoborane synthesis

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This PDF includes:

I.	Materials and Methods	2
II.	Kinetics Studies	3–4
III.	Inhibition Studies	4
IV.	Substrate Synthesis and Characterization	4–6
V.	Synthesis and Characterization of Authentic Organoborane Products	6–9
VI.	GC-MS Standard Curves for Organoborane Products	10–15
VII.	Determination of Enantioselectivity	16–25
VIII.	Preparative Scale Enzymatic Reactions	26–27
IX.	Derivatization of Enzymatic Borylation Product 18	28–30
X.	NMR Spectra	31–56
XI.	X-ray Crystallography and the Assignments of Absolute Configuration	57–63
XII.	Supplemental References	64–65

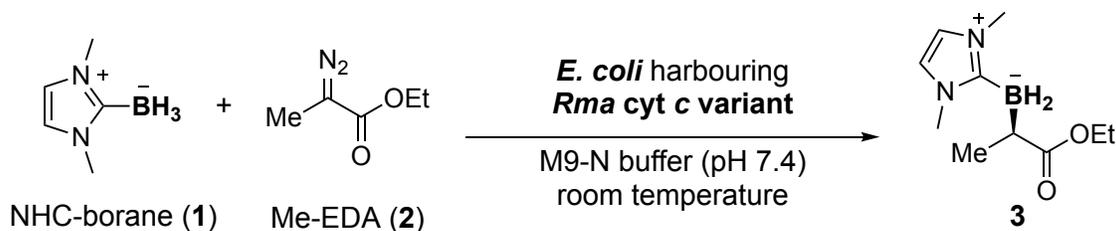
I. Materials and Methods

Unless otherwise noted, all chemicals and reagents were obtained from commercial suppliers (Sigma-Aldrich, VWR, Alfa Aesar, Acros) and used without further purification. Bovine serum albumin (BSA) was purchased from Sigma-Aldrich. Silica gel chromatography was carried out using AMD Silica Gel 60, 230-400 mesh. ^1H and ^{13}C NMR spectra were recorded on a Bruker Prodigy 400 MHz instrument (400 MHz for ^1H and 100 MHz for ^{13}C). Chemical shifts (δ) are reported in ppm downfield from tetramethylsilane, using the solvent resonance as the internal standard (^1H NMR: $\delta = 7.26$, ^{13}C NMR: $\delta = 77.36$ for CDCl_3). ^{19}F NMR and ^{11}B NMR data were collected on a VARIAN 300 MHz spectrometer (101 MHz for ^{19}F NMR) and a Bruker Prodigy 400 MHz instrument (128 MHz for ^{11}B NMR), respectively. Sonication was performed using a Qsonica Q500 sonicator. High-resolution mass spectra were obtained at the California Institute of Technology Mass Spectral Facility. Chemical reactions were monitored using thin layer chromatography (Merck 60 gel plates) using a UV-lamp for visualization. Gas chromatography (GC) analyses were carried out using a Shimadzu GC-17A gas chromatograph, a FID detector, and J&W HP-5 column (30 m x 0.32 mm, 0.25 μm film). Gas chromatography-mass spectrometry (GC-MS) analyses were carried out using Shimadzu GCMS-QP2010SE system and J&W HP-5ms column. Analytical chiral supercritical fluid chromatography (SFC) was performed with a JACSO 2000 series instrument using *i*-PrOH and supercritical CO_2 as the mobile phase. Chiral normal-phase HPLC analyses were performed using an Agilent 1200 series instrument with *i*-PrOH and hexanes as the mobile phase. Chiral GC was performed on an Agilent 6850 GC with FID detector using a Chiraldex GTA column (30.0 m \times 0.25 mm) at 1.0 mL/min He carrier gas flow.

Biological materials and methods are described in the Methods section of the manuscript.

II. Kinetic Studies

Comparison of carbon–boron bond forming rates of BOR^{WT} and BOR^{R1} as whole-cell catalysts, cell lysates, or purified proteins.



Biocatalysts	turnover frequency (TOF) / h ⁻¹
BOR ^{WT} purified protein	3 ± 2
BOR ^{WT} cell lysate	4 ± 1
BOR ^{WT} whole cell	410 ± 250
BOR ^{R1} purified protein	30 ± 2
BOR ^{R1} cell lysate	160 ± 100
BOR ^{R1} whole cell	6100 ± 700

TOFs reported represent mean values averaged over four experiments. Errors quoted indicate one standard deviation.

Whole cell-catalysed reaction: Experiments were performed using whole *E. coli* cells harbouring BOR^{WT} or BOR^{R1} (with the BOR protein concentration normalised to 10 μM), 10 mM borane, 10 mM diazo ester, 5 vol% MeCN, M9-N buffer at room temperature under anaerobic conditions for various time intervals.

Cell lysate-catalysed reaction: Experiments were performed using cell lysate of *E. coli* harbouring BOR^{WT} or BOR^{R1} (with the BOR protein concentration normalised to 10 μM), 10 mM borane, 10 mM diazo ester, 10 mM Na₂S₂O₄, 5 vol% MeCN, M9-N buffer at room temperature under anaerobic conditions for various time intervals. See Methods section of the manuscript for cell lysate preparation procedure.

Purified protein-catalysed reaction: Experiments were performed using purified BOR^{WT} or BOR^{R1} (10 μM), 10 mM borane, 10 mM diazo ester, 10 mM Na₂S₂O₄, 5 vol% MeCN, M9-N buffer at room temperature under anaerobic conditions for various time intervals. See Methods section of the manuscript for purified protein preparation procedure.

General procedure for carrying out timed experiments: In an anaerobic chamber, 3.8 mL of whole *E. coli* cells harboring BOR variant, or a solution of 3.4 mL of BOR variant cell lysate / purified protein and 0.4 mL Na₂S₂O₄ (100 mM in M9-N buffer), was added to a 10 mL glass vial. After charging NHC-borane **1** (100 μL, 400 mM in MeCN) and Me-EDA **2** (100 μL, 400 mM in MeCN), the vial was capped and the reaction was shaken at 600 rpm on an orbital shaker. At regular time intervals (see table below), 400 μL of the reaction mixture was removed from the vial and added to a 2 mL microcentrifuge tube containing 600 μL cyclohexane / EtOAc

(1:1 v/v) and internal standard (20 μ L, 20 mM 1,2,3-trimethoxybenzene in toluene). After vortexing for 20 seconds, 200 μ L of the organic layer was immediately removed for GC analysis.

Biocatalysts	Sampling time
BOR ^{WT} purified protein	Every hour from t = 1 to 4 h
BOR ^{WT} cell lysate	Every hour from t = 1 to 4 h
BOR ^{WT} whole cell	Every minute from t = 1 to 4 min
BOR ^{R1} purified protein	Every minute from t = 1 to 4 min
BOR ^{R1} cell lysate	Every minute from t = 0.5 to 3.5 min
BOR ^{R1} whole cell	Every minute from t = 0.5 to 3.5 min

Table above shows time points at which the biocatalytic reaction was sampled to determine the reaction initial rate.

III. Inactivation Studies

Inactivation studies of BOR^{R1} were carried out using purified protein or whole cell *E. coli* harbouring BOR^{R1}. Effects of NHC-borane **1**, Me-EDA **2**, or organoborane **3** were determined by preincubating the biocatalyst with either one of these reagents (10 mM) for 15 min before the catalyst was used for borylation, and by comparing the TTN of the resulting catalyst (TTN^{incub}) with that of an untreated biocatalyst (TTN^{control}), as described in Figure 2f.

Purified protein-catalysed reactions were performed using purified BOR^{R1} (10 μ M), 10 mM borane, 10 mM diazo ester, 10 mM Na₂S₂O₄, 5 vol% MeCN, M9-N buffer at room temperature under anaerobic conditions for 30 min. See Methods section of the manuscript for purified protein preparation procedure

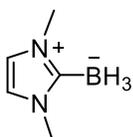
Whole cell-catalysed reactions were performed using whole *E. coli* cells harboring BOR^{R1} (with the BOR protein concentration normalised to 10 μ M), 10 mM borane, 10 mM diazo ester, 5 vol% MeCN, M9-N buffer at room temperature under anaerobic conditions for 30 min.

IV. Substrate Synthesis and Characterization

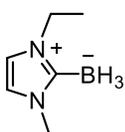
Picoline borane substrate was obtained from Sigma-Aldrich. Ethyl 2-diazopropanoate (Me-EDA) was obtained from Arch Bioscience. All commercially available reagents were used as received. The following diazo compounds are known and prepared according to literature procedures: methyl 2-diazopropanoate¹, isopropyl 2-diazopropanoate², benzyl 2-diazopropanoate³, ethyl 2-phenyldiazoacetate (Ph-EDA)⁴, ethyl 2-diazo-3,3,3-trifluoropropanoate (CF₃-EDA)⁵, and (1-diazo-2,2,2-trifluoroethyl)benzene (CF₃-DMB)⁶.

Other NHC-BH₃ substrates were synthesized from corresponding imidazolium iodide salts as reported⁷. Namely, imidazolium iodide salts (5 mmol) were resuspended in 5 mL THF. A solution of NaHMDS (1M in THF, 1.05 equiv.) was then added at -78 °C under Ar and shaken for 1 h at -78 °C. Afterwards, a solution of BH₃-THF (1M in THF, 1 equiv.) was added to the reaction

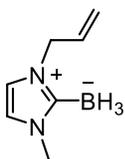
and the reaction mixture was allowed to warm from $-78\text{ }^{\circ}\text{C}$ to rt and stirred overnight. The solvent was removed under reduced pressure and the residue was purified by flash column chromatography to give the NHC-BH₃ complexes. The ¹H NMR resonances of the B-H protons are broad (due to geminal coupling with boron) and generally in the range of 0.4 – 1.6 ppm. The ¹³C NMR resonances of the boron-binding NHC quaternary carbons usually appear at around 170 ppm and are typically broad (due to geminal coupling with boron) and weak; these signals are sometimes not visible in the ¹³C NMR spectra.



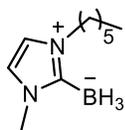
This compound is known⁸. ¹H NMR (400 MHz, Chloroform-*d*) δ 6.91 – 6.66 (m, 2H), 3.71 (s, 6H), 0.99 (dd, $J = 172.7, 86.3$ Hz, 3H).



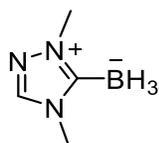
This compound is known⁹. ¹H NMR (400 MHz, Chloroform-*d*) δ 6.87 – 6.65 (m, 2H), 4.00 (q, $J = 7.3$ Hz, 2H), 3.57 (s, 3H), 1.22 (t, $J = 7.3$ Hz, 3H), 1.44 – 0.30 (m, 3H).



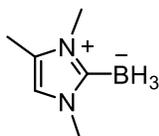
This compound is known⁸. ¹H NMR (400 MHz, Chloroform-*d*) δ 6.84 – 6.79 (m, 2H), 5.91 (ddt, $J = 17.1, 10.2, 6.1$ Hz, 1H), 5.30 – 5.06 (m, 2H), 4.71 (dt, $J = 6.1, 1.5$ Hz, 2H), 3.71 (s, 3H), 1.43 – 0.35 (m, 3H).



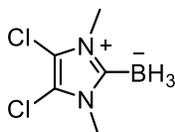
¹H NMR (400 MHz, Chloroform-*d*) δ 6.82 – 6.76 (m, 2H), 4.13 – 3.97 (m, 2H), 3.69 (s, 3H), 1.83 – 1.63 (m, 2H), 1.42 – 1.19 (m, 6H), 0.97 – 0.75 (m, 3H), 1.46 – 0.41 (m, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 171.0, 119.9, 118.7, 48.8, 35.8, 31.3, 30.1, 26.1, 22.5, 14.0; ¹¹B NMR (128 MHz, Chloroform-*d*) δ -37.4 (q, $J = 86$ Hz); MS (FAB) m/z [(M + H)⁺ - H₂] calcd for C₁₀H₂₀N₂B: 179.1720, found: 179.1707.



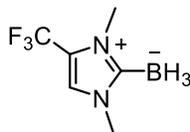
This compound is known⁸. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.85 (s, 1H), 3.94 (s, 3H), 3.74 (s, 3H), 1.45 – 0.42 (m, 3H).



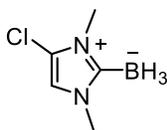
This compound is known¹⁰. ¹H NMR (400 MHz, Chloroform-*d*) δ 6.49 (q, $J = 1.2$ Hz, 1H), 3.56 (s, 3H), 3.50 (s, 3H), 2.07 (d, $J = 1.3$ Hz, 3H), 1.31 – 0.43 (m, 3H).



This compound is known¹¹. ¹H NMR (400 MHz, Chloroform-*d*) δ 3.72 (s, 6H), 1.44 – 0.41 (m, 3H).



^1H NMR (400 MHz, Chloroform-*d*) δ 7.23 (q, $J = 1.5$ Hz, 1H), 3.82 (s, 3H), 3.77 (s, 3H), 1.49 – 0.54 (m, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 177.1, 122.2, 121.8, 119.40 (q, $J = 267.3$ Hz), 36.5, 34.0. ^{11}B NMR (128 MHz, Chloroform-*d*) δ -37.4 (q, $J = 88$ Hz). ^{19}F NMR (282 MHz, Chloroform-*d*) δ -61.2 (d, $J = 3$ Hz); MS (FAB) m/z [(M+H) $^+$ -H $_2$] calcd for $\text{C}_6\text{H}_9\text{F}_3\text{N}_2\text{B}$: 177.0811, found: 177.0815.

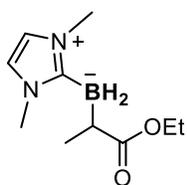


^1H NMR (400 MHz, Chloroform-*d*) δ 6.78 (s, 1H), 3.71 (s, 3H), 3.67 (s, 3H), 1.45 – 0.51 (m, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 173.5, 119.3, 116.9, 36.4, 33.0. ^{11}B NMR (128 MHz, Chloroform-*d*) δ -36.9 (q, $J = 87$ Hz); MS (FAB) m/z [(M+H) $^+$ -H $_2$] calcd for $\text{C}_5\text{H}_9\text{N}_2\text{BCl}$: 143.0547, found: 143.0547.

V. Synthesis and Characterization of Authentic Organoborane Products

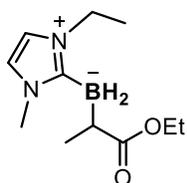
Racemic standard references of organoborane products were prepared *via* Rh-catalyzed B–H insertion reactions with procedures slightly modified from a previously reported method⁸. Namely, a 4 mL vial with screw cap and PTFE septum was charged with a borane substrate (1.0 mmol, 1 equiv.) and $\text{Rh}_2(\text{OAc})_4$ (11 mg, 2.5 mol%). The vial was evacuated and backfilled with Ar three times and 2 mL of anhydrous CH_2Cl_2 was added. The vial was placed in a 38 °C water bath. A CH_2Cl_2 solution (1 mL) of diazo compound (1.0 mmol) was slowly added to the reaction mixture over 4 hours. Afterwards, the reaction mixture was allowed to further react overnight. The crude reaction mixture was purified by flash chromatography (dry loading) using EtOAc and hexanes as eluents and afforded organoborane products in 30 - 75% yield. The ^1H NMR resonances of the B–H protons are broad (due to geminal coupling with boron) and generally in the range of 0.4 – 1.6 ppm. The ^{13}C NMR resonances of the boron-binding NHC quaternary carbons usually appear at around 170 ppm and are typically broad (due to geminal coupling with boron) and weak; these signals are sometimes not visible in the ^{13}C NMR spectra.

(1,3-Dimethyl-1H-imidazol-3-ium-2-yl)(1-ethoxy-1-oxopropan-2-yl)dihydroborate (3)



^1H NMR (400 MHz, Chloroform-*d*) δ 6.82 (s, 2H), 3.98 – 3.78 (m, 2H), 3.75 (s, 6H), 1.95 – 1.10 (m, 2H), 1.88 (br s, 1H), 1.10 (d, $J = 6.2$ Hz, 3H), 1.06 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.5, 120.4, 58.7, 36.2, 30.5, 17.8, 14.6. The boron-bound NHC quaternary carbon was not resolved; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.6 (t, $J = 90$ Hz); MS (FAB) m/z [(M + H) $^+$ - H $_2$] calcd for $\text{C}_{10}\text{H}_{18}\text{O}_2\text{N}_2\text{B}$: 209.1461, found: 209.1456.

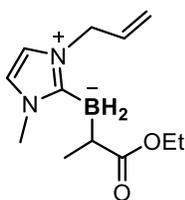
(1-Ethoxy-1-oxopropan-2-yl)(3-ethyl-1-methyl-1H-imidazol-3-ium-2-yl)dihydroborate (4)



^1H NMR (400 MHz, Chloroform-*d*) δ 6.91 – 6.82 (m, 2H), 4.28 – 3.97 (m, 2H), 3.93 – 3.73 (m, 2H), 3.70 (s, 3H), 1.84 (br s, 1H), 1.95 – 1.10 (br m, 2H), 1.34 (t, $J = 7.3$ Hz, 3H), 1.05 (d, $J = 6.7$ Hz, 3H), 0.98 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.3, 170.0, 120.7, 118.2, 58.4, 43.5, 35.9, 30.4, 17.6,

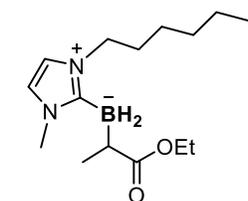
15.8, 14.4; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.5 (t, J = 89 Hz). MS (FAB) m/z [M^+] calcd for $\text{C}_{11}\text{H}_{21}\text{O}_2\text{N}_2\text{B}$: 224.1696, found: 224.1693.

(3-Allyl-1-methyl-1*H*-imidazol-3-ium-2-yl)(1-ethoxy-1-oxopropan-2-yl)dihydroborate (5)



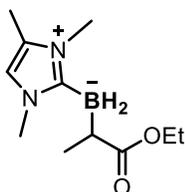
^1H NMR (400 MHz, Chloroform-*d*) δ 6.85 (AB q, J = 2.0 Hz, 2H), 5.94 (ddt, J = 17.1, 10.2, 6.1 Hz, 1H), 5.39 – 5.17 (m, 2H), 4.82 (ddt, J = 15.3, 6.0, 1.5 Hz, 1H), 4.68 (ddt, J = 15.3, 6.2, 1.4 Hz, 1H), 3.99 – 3.78 (m, 2H), 3.76 (s, 3H), 1.92 – 1.05 (m, 2H), 1.87 (br s, 1H), 1.09 (d, J = 6.6 Hz, 3H), 1.05 (t, J = 7.1 Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.6, 132.9, 120.9, 119.7, 119.0, 58.8, 51.4, 36.4, 32.0, 17.9, 14.8. The boron-bound NHC quaternary carbon was not resolved; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.6 (t, J = 90 Hz). MS (FAB) m/z [$\text{M} + \text{H}^+$] calcd for $\text{C}_{12}\text{H}_{22}\text{O}_2\text{N}_2\text{B}$: 237.1774, found: 237.1783.

(1-Ethoxy-1-oxopropan-2-yl)(3-hexyl-1-methyl-1*H*-imidazol-3-ium-2-yl)dihydroborate (6)



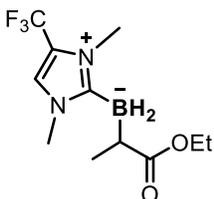
^1H NMR (400 MHz, Chloroform-*d*) δ 6.88 – 6.79 (m, 2H), 4.20 – 4.06 (m, 1H), 3.99 (m, 1H), 3.93 – 3.74 (m, 2H), 3.72 (s, 3H), 1.93 – 1.79 (m, 1H), 1.72 (dt, J = 13.8, 6.9 Hz, 2H), 1.71 – 1.20 (m, 8H), 1.12 – 1.05 (m, 3H), 1.01 (td, J = 7.2, 2.4 Hz, 3H), 0.90 – 0.80 (m, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.6, 120.7, 119.0, 58.7, 48.9, 36.2, 31.6, 30.8, 30.5, 26.5, 22.7, 17.9, 14.7, 14.2. The boron-bound NHC quaternary carbon was not resolved; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.5 (t, J = 90 Hz); MS (FAB) m/z [M^+] calcd for $\text{C}_{15}\text{H}_{29}\text{O}_2\text{N}_2\text{B}$: 280.2322, found: 280.2330.

(1-Ethoxy-1-oxopropan-2-yl)(1,3,4-trimethyl-1*H*-imidazol-3-ium-2-yl)dihydroborate (7)

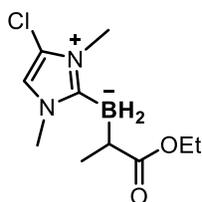


^1H NMR (400 MHz, Chloroform-*d*) δ 6.55 (q, J = 1.2 Hz, 1H), 3.95 – 3.77 (m, 2H), 3.66 (s, 3H), 3.61 (s, 3H), 2.16 (d, J = 1.1 Hz, 3H), 1.84 (br s, 1H), 1.93 – 1.10 (m, 2H), 1.07 (s, 3H), 1.12 – 1.02 (m, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.7, 170.0, 128.3, 117.6, 58.7, 35.9, 32.7, 32.0 – 29.5 (m), 17.9, 14.7, 9.7; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.2 (t, J = 89 Hz); MS (FAB) m/z [M^+] calcd for $\text{C}_{11}\text{H}_{21}\text{O}_2\text{N}_2\text{B}$: 224.1696, found: 224.1695.

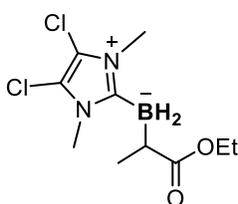
(1,3-Dimethyl-4-(trifluoromethyl)-1*H*-imidazol-3-ium-2-yl)(1-ethoxy-1-oxopropan-2-yl)dihydroborate (8)



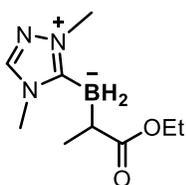
^1H NMR (400 MHz, Chloroform-*d*) δ 7.29 (q, J = 1.3 Hz, 1H), 4.00 – 3.70 (m, 2H), 3.85 (s, 3H), 3.81 (s, 3H), 1.88 (br s, 1H), 1.85 – 1.05 (m, 2H), 1.12 (d, J = 6.6 Hz, 3H), 1.05 (t, J = 7.1 Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.2, 123.6, 122.8 – 122.5 (m), 119.6 (q, J = 267.6 Hz), 59.0, 36.9, 34.4, 30.0, 17.9, 14.7. The boron-bound NHC quaternary carbon was not resolved; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.7 (t, J = 91 Hz). ^{19}F NMR (282 MHz, Chloroform-*d*) δ -61.1; MS (FAB) m/z [M^+] calcd for $\text{C}_{11}\text{H}_{18}\text{O}_2\text{N}_2\text{BF}_3$: 278.1414, found: 278.1405.

(4-Chloro-1,3-dimethyl-1*H*-imidazol-3-ium-2-yl)(1-ethoxy-1-oxopropan-2-yl)dihydroborate (9)

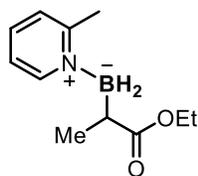
^1H NMR (400 MHz, Chloroform-*d*) δ 6.83 (s, 1H), 3.97 – 3.79 (m, 2H), 3.73 (s, 3H), 3.70 (s, 3H), 2.00 – 1.10 (m, 2H), 1.94 – 1.76 (m, 1H), 1.13 – 1.02 (m, 6H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.4, 173.0, 119.9, 117.3, 58.9, 36.6, 33.3, 30.3, 17.9, 14.7; ^{11}B NMR (128 MHz, Chloroform-*d*) δ –24.2 (t, J = 90 Hz); MS (FAB) m/z [M^+] calcd for $\text{C}_{10}\text{H}_{18}\text{O}_2\text{N}_2\text{BCl}$: 244.1150, found: 244.1154.

(4,5-Dichloro-1,3-dimethyl-1*H*-imidazol-3-ium-2-yl)(1-ethoxy-1-oxopropan-2-yl)dihydroborate (10)

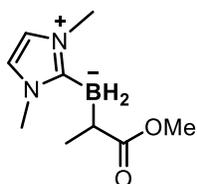
^1H NMR (400 MHz, Chloroform-*d*) δ 3.95 – 3.78 (m, 2H), 3.72 (s, 6H), 1.83 (br s, 1H), 1.99 – 1.05 (m, 2H), 1.10 (d, J = 11.9 Hz, 3H), 1.05 (t, J = 7.1 Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.1, 172.0, 116.6, 59.0, 34.1, 30.0, 17.8, 14.7; ^{11}B NMR (128 MHz, Chloroform-*d*) δ –23.9 (t, J = 91 Hz); MS (FAB) m/z [$\text{M} + \text{H}^+$] calcd for $\text{C}_{10}\text{H}_{18}\text{O}_2\text{N}_2\text{BCl}_2$: 279.0838, found: 279.0846.

(1,4-Dimethyl-4*H*-1,2,4-triazol-1-ium-5-yl)(1-ethoxy-1-oxopropan-2-yl)dihydroborate (11)

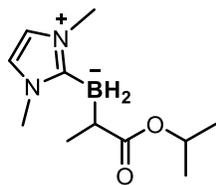
^1H NMR (400 MHz, Chloroform-*d*) δ 7.92 (s, 1H), 3.95 (s, 3H), 3.94 – 3.79 (m, 2H), 3.78 (s, 3H), 1.89 (br s, 1H), 2.00 – 1.05 (m, 2H), 1.13 – 1.09 (m, 3H), 1.05 (t, J = 7.1 Hz, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.2, 141.7, 59.0 (d, J = 8.0 Hz), 38.6, 34.1, 30.0, 17.9, 14.7. The boron-bound NHC quaternary carbon was not resolved; ^{11}B NMR (128 MHz, Chloroform-*d*) δ –25.0 (t, J = 91 Hz). MS (FAB) m/z [$\text{M} + \text{H}^+$] calcd for $\text{C}_9\text{H}_{19}\text{O}_2\text{N}_3\text{B}$: 212.1570, found: 212.1570.

Ethyl 2-((2-methyl-pyridin-1-yl)boraneyl)propanoate (12)

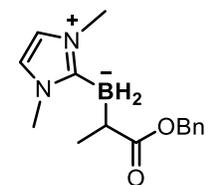
^1H NMR (400 MHz, Chloroform-*d*) δ 8.53 (dd, J = 6.0, 1.6 Hz, 1H), 7.84 (td, J = 7.7, 1.7 Hz, 1H), 7.42 – 7.36 (m, 1H), 7.33 – 7.28 (m, 1H), 3.79 (AB qq, J = 10.8, 7.1 Hz, 2H), 3.30 – 2.15 (m, 2H), 2.77 (s, 3H), 2.05 – 1.92 (m, 1H), 1.05 (d, J = 6.8 Hz, 3H), 0.94 (t, J = 7.1 Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 182.1, 157.9, 149.4, 140.2, 127.7, 122.6, 58.8, 32.8, 22.8, 15.2, 14.6; ^{11}B NMR (128 MHz, Chloroform-*d*) δ –5.1 (t, J = 103 Hz); MS (FAB) m/z [M^+] calcd for $\text{C}_{11}\text{H}_{18}\text{O}_2\text{NB}$: 207.1431, found: 207.1431.

(1,3-Dimethyl-1*H*-imidazol-3-ium-2-yl)(1-methoxy-1-oxopropan-2-yl)dihydroborate (13)

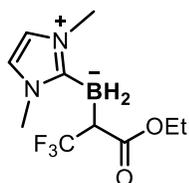
^1H NMR (400 MHz, Chloroform-*d*) δ 6.82 (s, 1H), 3.72 (s, 6H), 3.43 (s, 2H), 1.99 – 1.08 (m, 3H), 1.06 (d, J = 6.8 Hz, 2H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.9, 170.0, 120.6, 50.7, 36.2, 30.5, 17.8; ^{11}B NMR (128 MHz, Chloroform-*d*) δ –24.6 (t, J = 90 Hz); MS (FAB) m/z [M^+] calcd for $\text{C}_9\text{H}_{17}\text{O}_2\text{N}_2\text{B}$: 196.1383, found: 196.1388.

(1,3-Dimethyl-1*H*-imidazol-3-ium-2-yl)(1-isopropoxy-1-oxopropan-2-yl)dihydroborate (14)

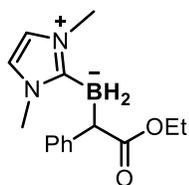
^1H NMR (400 MHz, Chloroform-*d*) δ 6.81 (s, 2H), 4.76 (hept, $J = 6.2$ Hz, 1H), 3.75 (s, 6H), 1.86 (br s, 1H), 2.00 – 1.10 (m, 2H), 1.09 (d, $J = 6.2$ Hz, 6H), 0.94 (d, $J = 6.3$ Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.3, 170.0, 120.6, 65.1, 36.4, 30.7, 22.5, 22.2, 18.1; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.5 (t, $J = 90$ Hz); MS (FAB) m/z [M^+] calcd for $\text{C}_{11}\text{H}_{21}\text{O}_2\text{N}_2\text{B}$: 224.1696, found: 224.1703.

(1-(Benzyloxy)-1-oxopropan-2-yl)(1,3-dimethyl-1*H*-imidazol-3-ium-2-yl)dihydroborate (15)

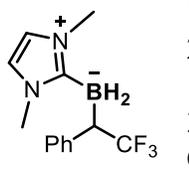
^1H NMR (400 MHz, Chloroform-*d*) δ 7.36 – 7.17 (m, 5H), 6.71 (s, 2H), 4.92 (s, 2H), 3.62 (s, 6H), 2.10 – 1.15 (m, 2H), 1.97 (br s, 1H), 1.16 (d, $J = 6.5$ Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 183.2, 170.0, 137.6, 128.4, 128.1, 127.7, 120.5, 64.7, 36.1, 30.8, 17.9; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -24.5 (t, $J = 88$ Hz); MS (FAB) m/z [$(\text{M} + \text{H})^+ - \text{H}_2$] calcd for $\text{C}_{15}\text{H}_{20}\text{O}_2\text{N}_2\text{B}$: 271.1618, found: 271.1616.

(1,3-Dimethyl-1*H*-imidazol-3-ium-2-yl)(3-ethoxy-1,1,1-trifluoro-3-oxopropan-2-yl)dihydroborate (16)

^1H NMR (400 MHz, Chloroform-*d*) δ 6.88 (s, 2H), 4.15 – 3.97 (m, 2H), 3.76 (s, 6H), 2.65 (s, 1H), 2.10 – 1.25 (m, 2H), 1.18 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 174.2 (d, $J = 5.2$ Hz), 168.0, 128.7 (q, $J = 276.2$ Hz), 121.2, 60.0, 42.6, 36.3, 14.6; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -28.6 (t, $J = 92$ Hz). ^{19}F NMR (282 MHz, Chloroform-*d*) δ -62.5 (d, $J = 10$ Hz); MS (FAB) m/z [$(\text{M} + \text{H})^+ - \text{H}_2$] calcd for $\text{C}_{10}\text{H}_{15}\text{O}_2\text{N}_2\text{BF}_3$: 263.1179, found: 263.1167.

(1,3-Dimethyl-1*H*-imidazol-3-ium-2-yl)(2-ethoxy-2-oxo-1-phenylethyl)dihydroborate (17)

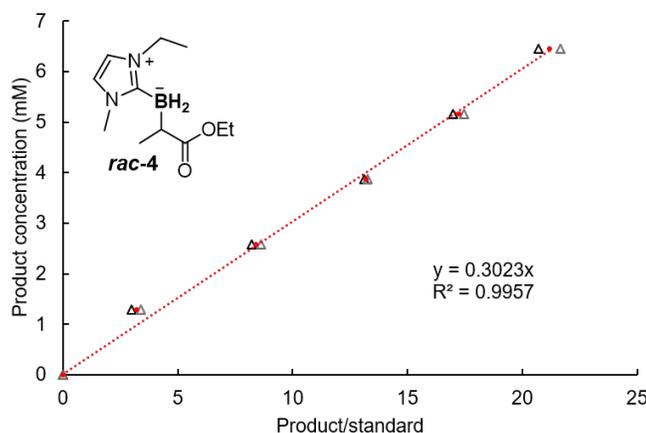
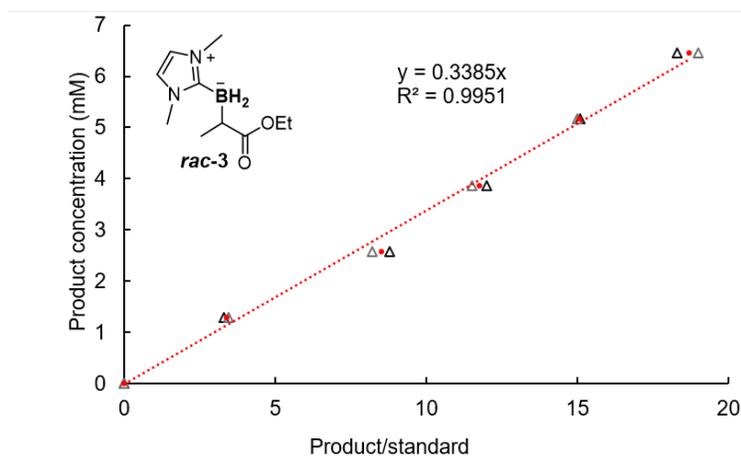
^1H NMR (400 MHz, Chloroform-*d*) δ 7.35 – 7.24 (m, 2H), 7.19 – 7.11 (m, 2H), 7.07 – 6.99 (m, 1H), 6.77 (s, 2H), 4.24 – 3.93 (m, 2H), 3.46 (s, 6H), 3.35 – 3.22 (m, 1H), 2.34 – 1.41 (m, 2H), 1.21 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 179.7, 145.8, 127.9, 127.8, 124.1, 120.7, 120.2, 59.3, 45.6 (d, $J = 44.3$ Hz), 36.0, 14.8, (the NHC quaternary carbon was too broad to be visible due to coupling with B); ^{11}B NMR (128 MHz, Chloroform-*d*) δ -23.2 (t, $J = 93$ Hz); MS (FAB) m/z [M^+] calcd for $\text{C}_{15}\text{H}_{21}\text{O}_2\text{N}_2\text{B}$: 272.1696, found: 272.1687.

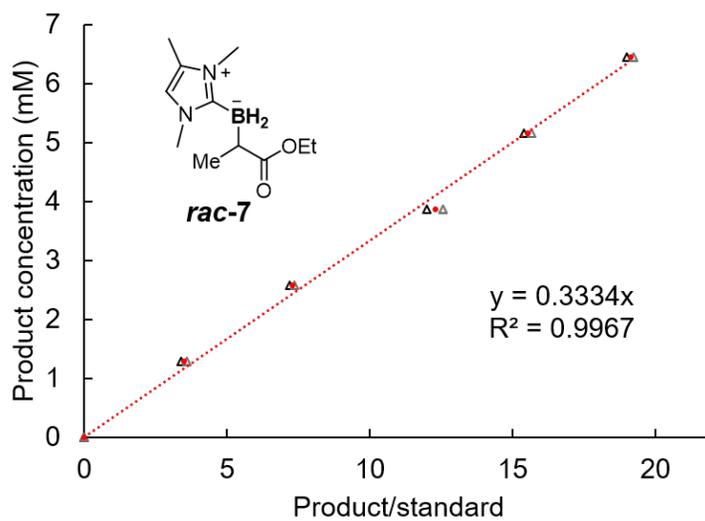
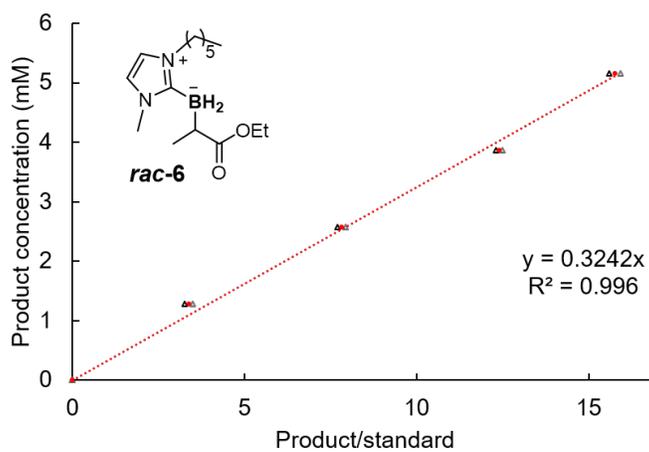
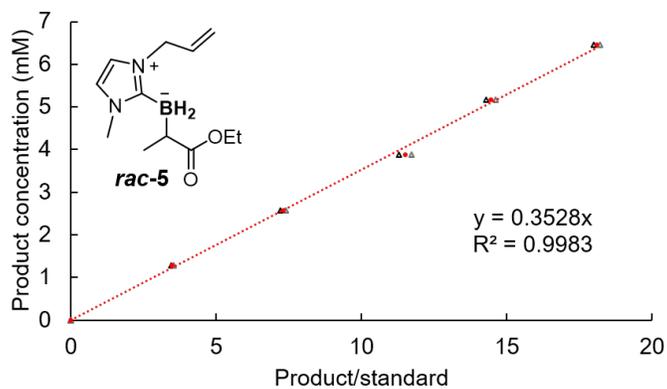
(1,3-Dimethyl-1*H*-imidazol-3-ium-2-yl)(2,2,2-trifluoro-1-phenylethyl)dihydroborate (18)

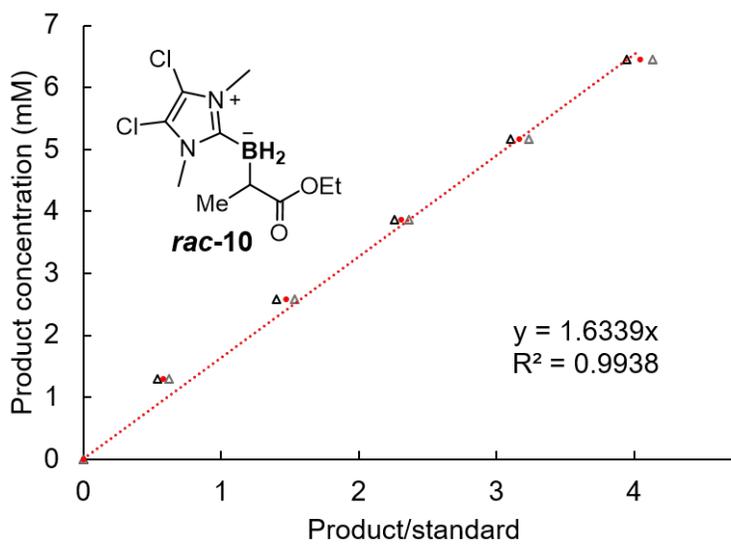
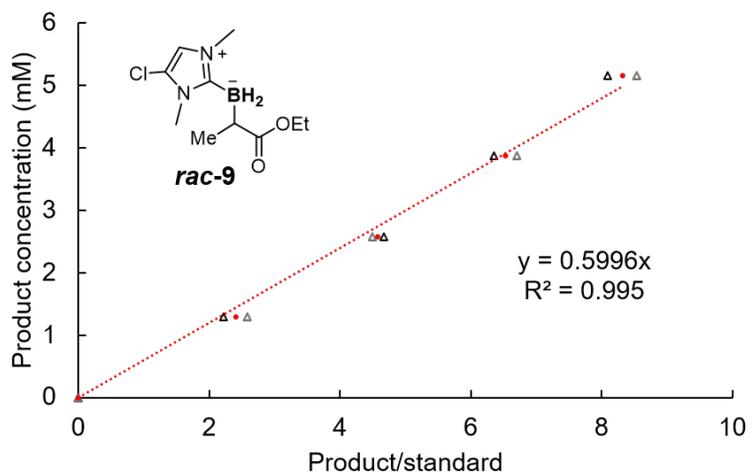
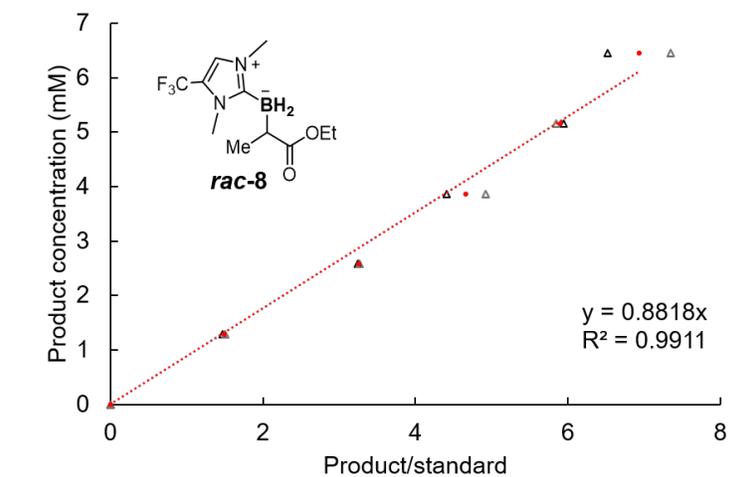
^1H NMR (400 MHz, Chloroform-*d*) δ 7.23 – 7.05 (m, 5H), 6.76 (s, 2H), 3.52 (s, 6H), 2.90 – 2.60 (m, 1H), 2.25 – 1.40 (m, 2H); ^{13}C NMR (101 MHz, Chloroform-*d*) δ 169.1, 143.7 (d, $J = 3.5$ Hz), 131.4 (q, $J = 278.0$ Hz), 128.4, 128.3, 125.2, 120.8, 43.5, 36.0; ^{11}B NMR (128 MHz, Chloroform-*d*) δ -26.7 (t, $J = 90$ Hz); ^{19}F NMR (282 MHz, Chloroform-*d*) δ -61.8 (d, $J = 13$ Hz); MS (ESI) m/z [$\text{M} + \text{H}^+$] calcd for $\text{C}_{13}\text{H}_{17}\text{N}_2\text{BF}_3$: 269.1437, found: 269.1440.

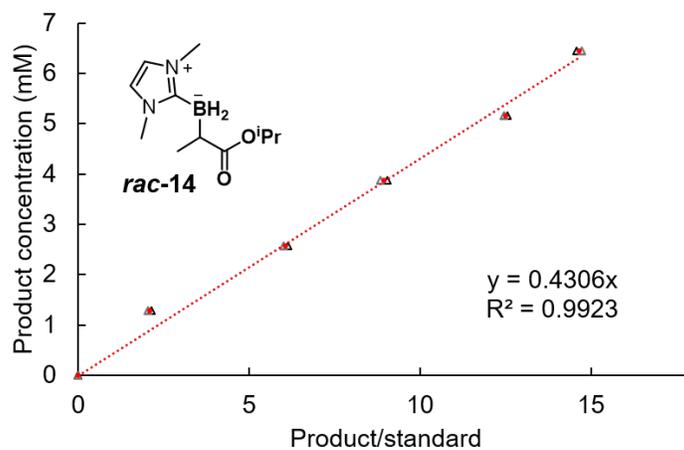
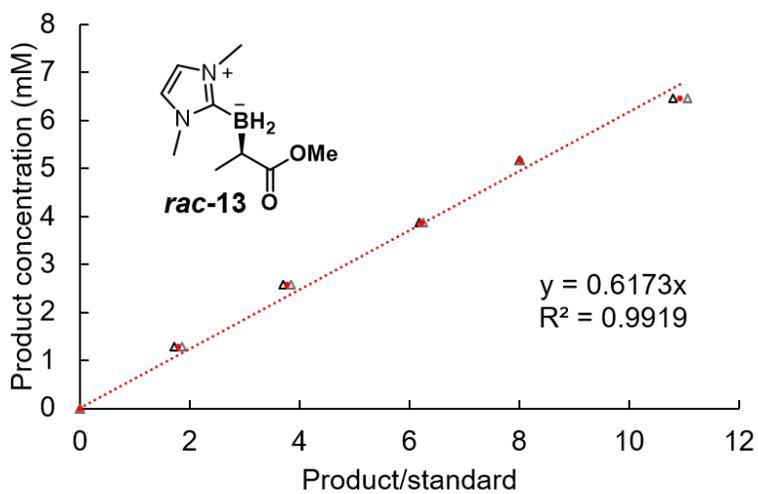
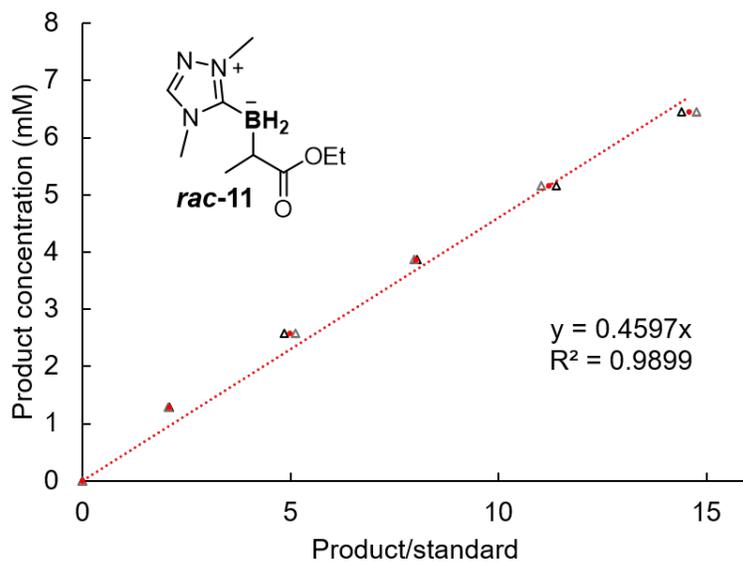
VI. GC-MS Standard Curves for Organoborane Products

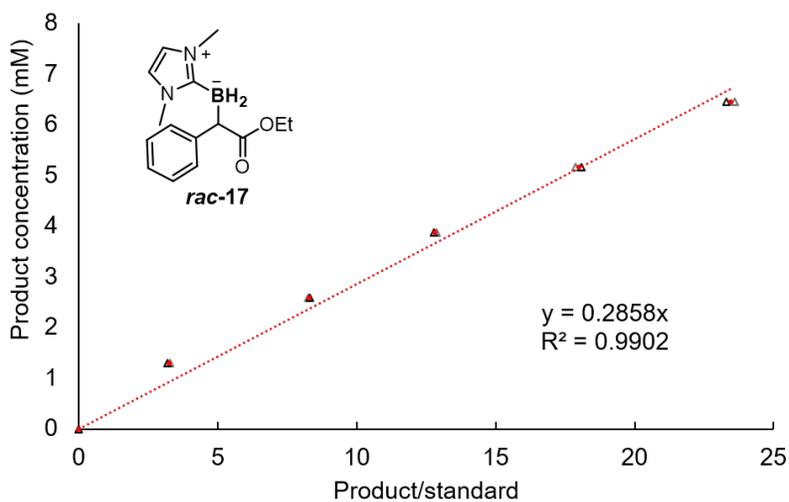
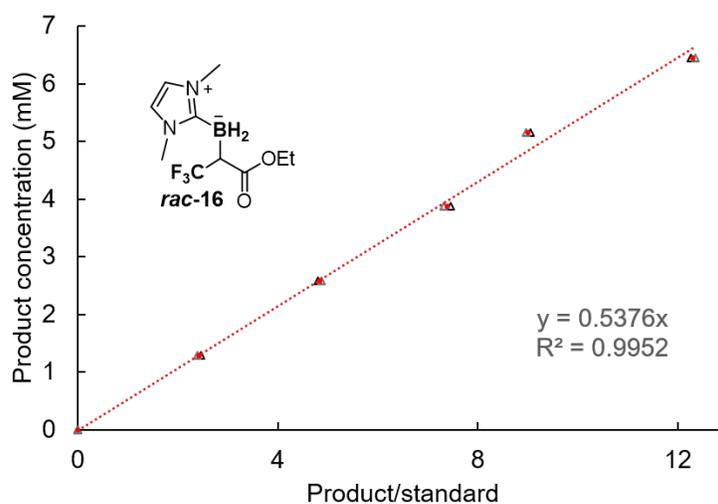
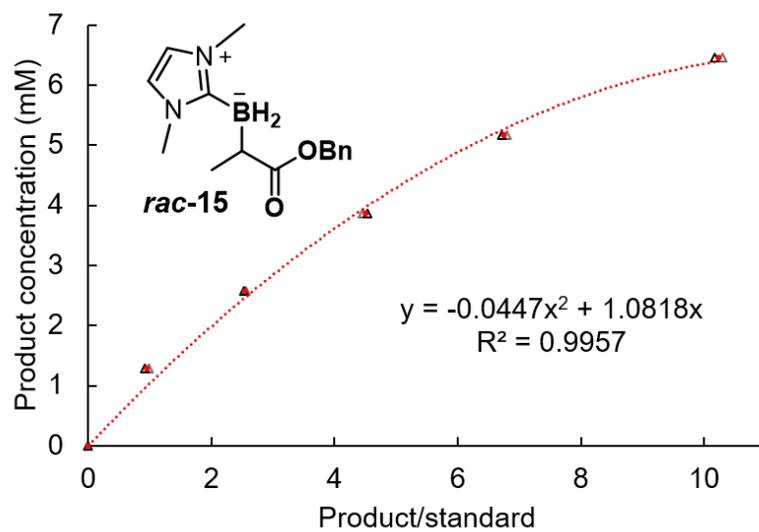
Product formation in enzymatic reactions was quantified by GC-MS based on standard curves. To determine the standard calibration curves, stock solutions of chemically synthesized organoborane products were prepared at various concentrations (1 - 7 mM in 4:6 hexanes/EtOAc) with added internal standard 1,2,3-trimethoxybenzene with a final concentration of 6.45 mM in the stock solutions of organoborane products. Individual data point for each duplicate run is marked as triangle, the average of duplicate runs is marked as red dot. The standard curves plot product concentration in mM (y-axis) against the average ratio of product area to internal standard area on GC-MS (x-axis). The quantification of organoborane **12** was determined by preparative scale reactions as this compound cannot be identified by GC-MS.

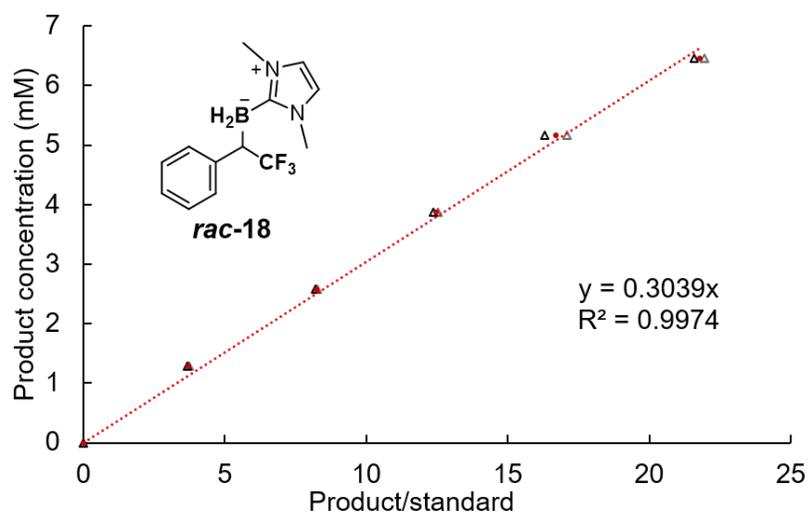






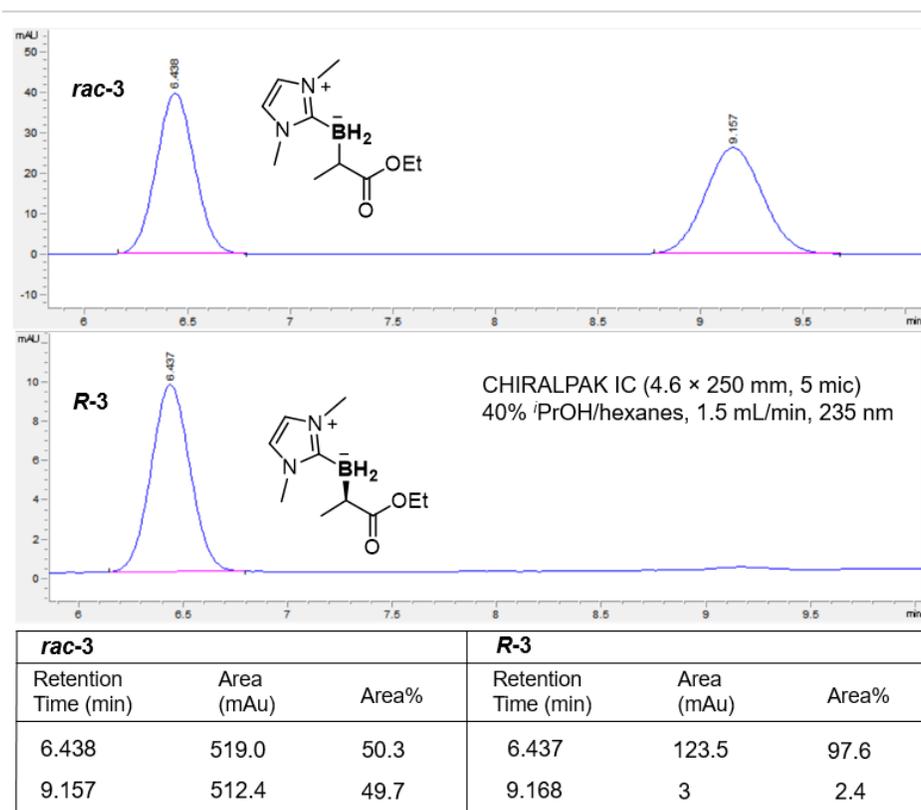


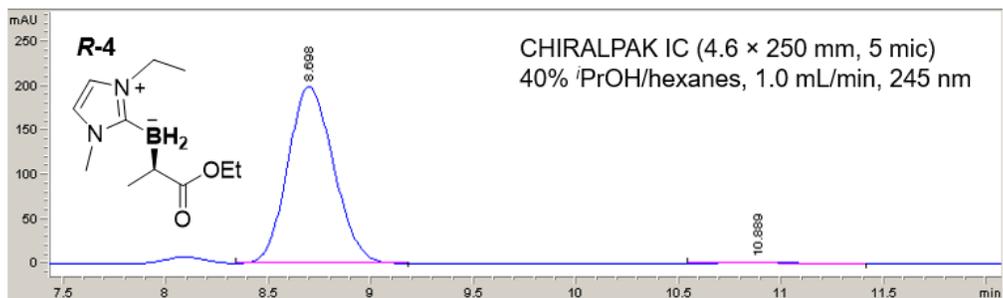
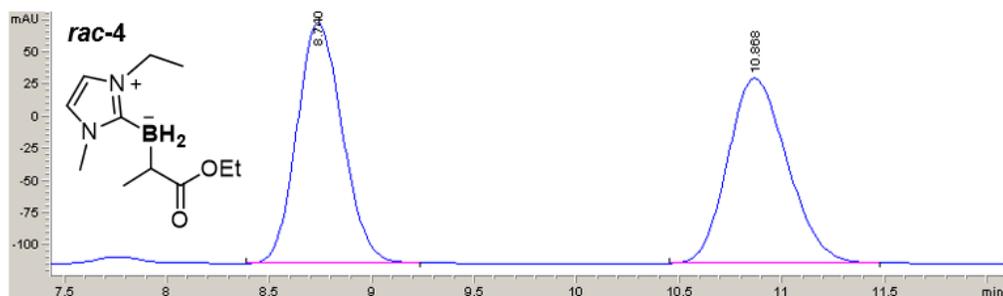




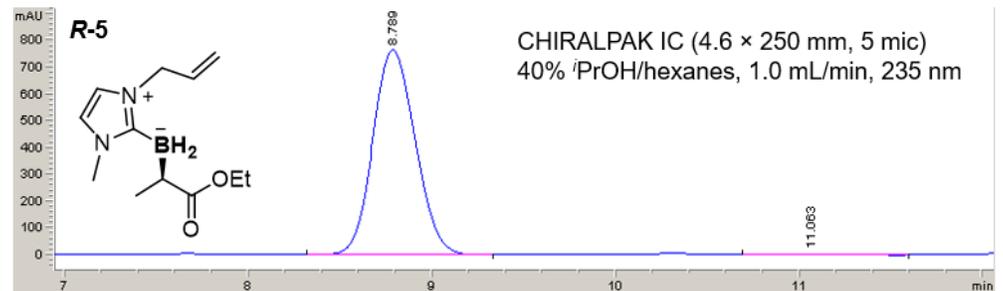
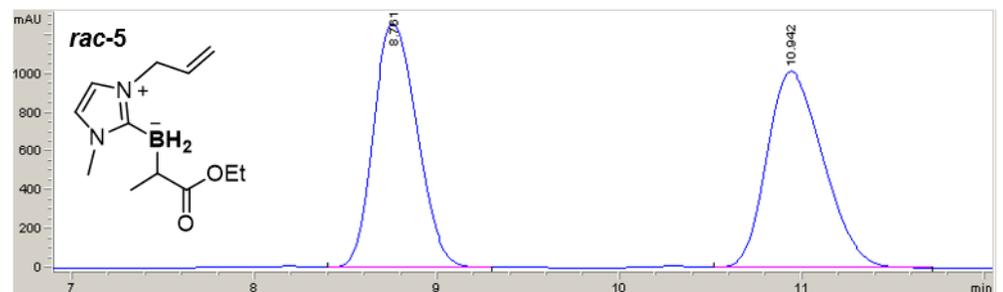
VII. Determination of Enantioselectivity

All e.r. values of enzymatically synthesized borane products were determined using chiral SFC or normal-phase chiral HPLC. The absolute configurations of enzymatically synthesized borane products **3**, **12**, and **18** were determined to be *R* via X-ray crystallography. The absolute configurations of organoborane products **4-11**, **13-16** were inferred by analogy, assuming the facial selectivity of the diazo reagents from which these products were made remains the same as that of Me-EDA.

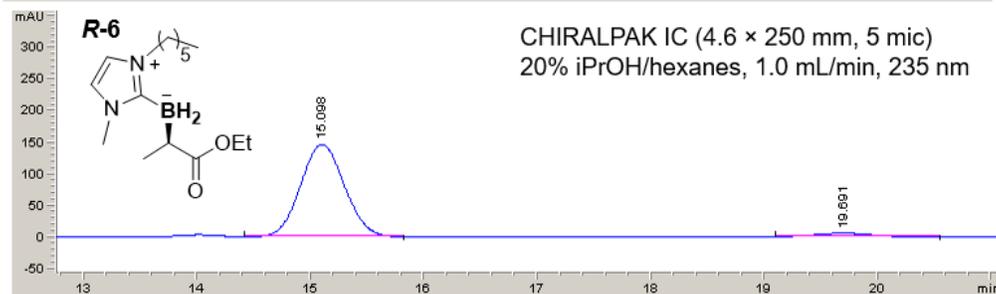
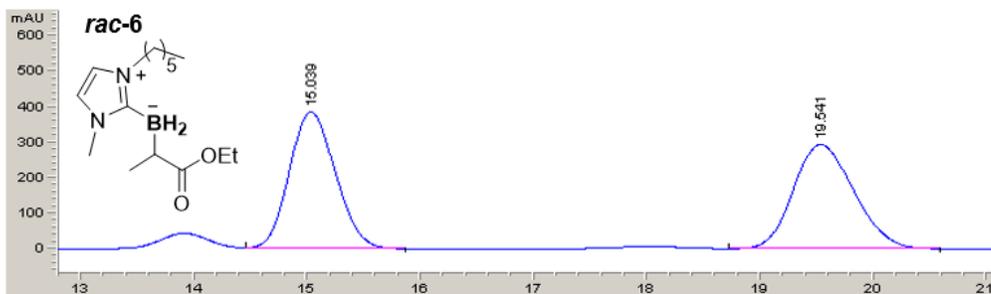




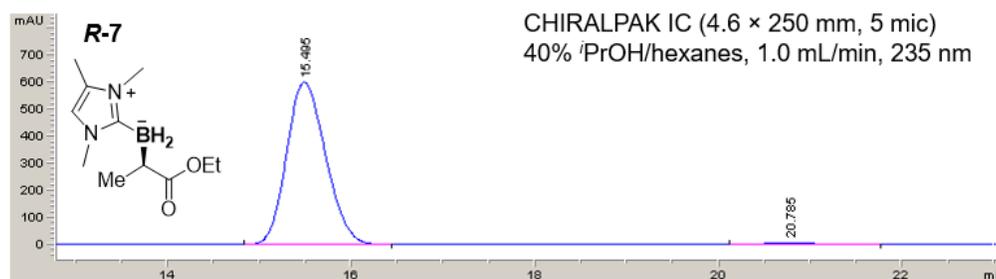
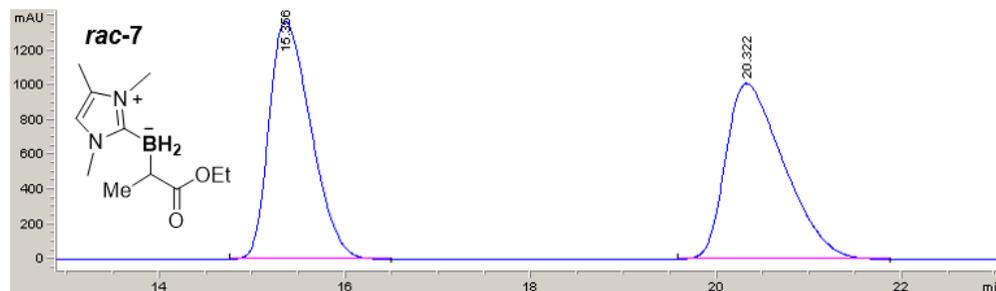
rac-4			R-4		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
8.74	2890.3	49.8	8.698	3105.2	98.6
10.868	2910.5	50.2	10.889	42.7	1.3



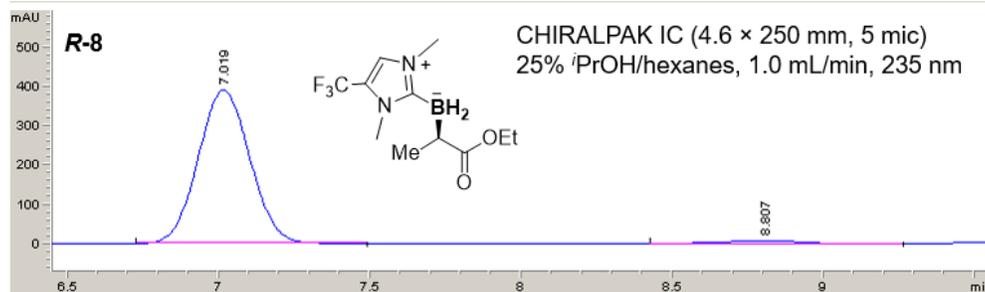
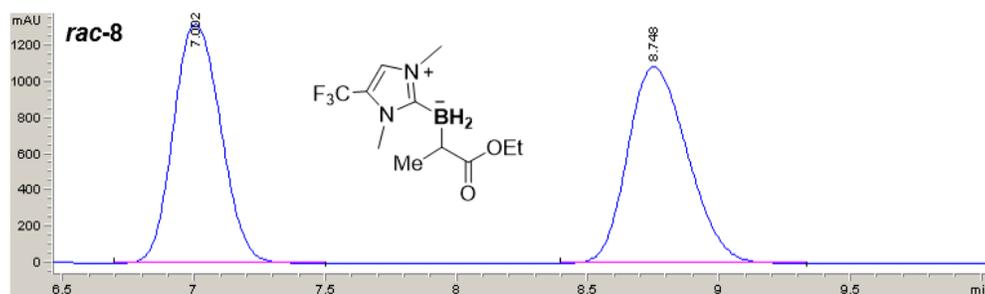
rac-5			R-5		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
8.761	21036	49.2	8.789	12279.6	99.4
10.942	21752	50.8	11.063	81.3	0.6



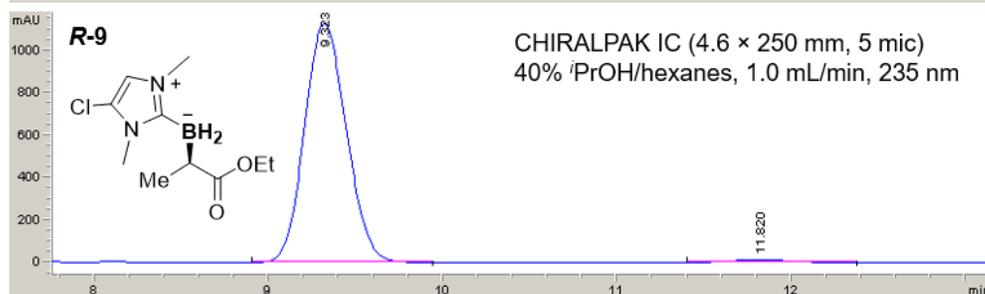
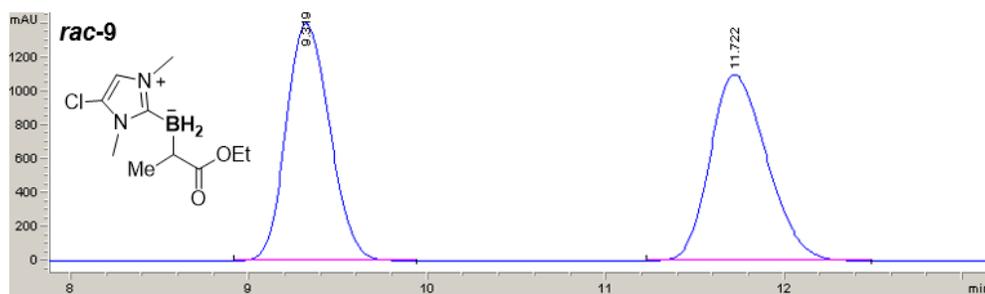
rac-6			R-6		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
15.039	10916.7	50.0	15.098	3940.4	95.6
19.541	10905.3	50.0	19.691	181.6	4.4



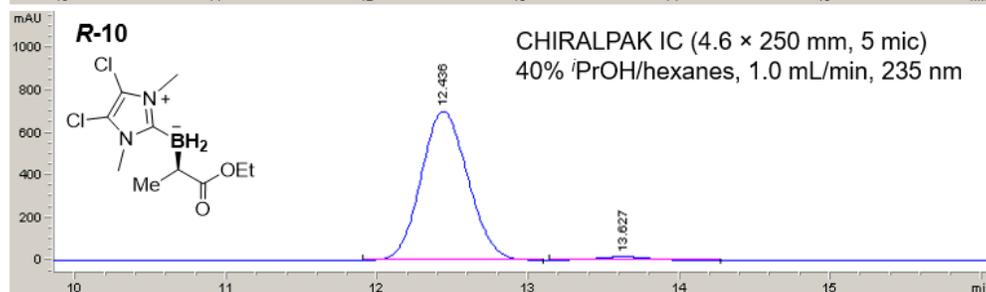
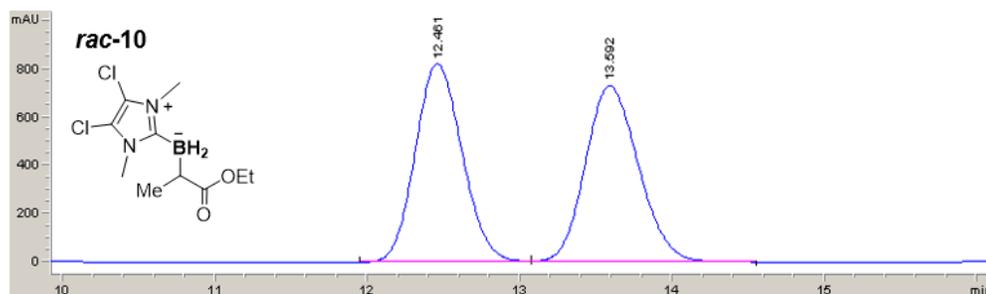
rac-7			R-7		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
15.356	42847.9	49.3	15.495	17552.6	98.6
20.322	44076.6	50.7	20.785	252.7	1.4



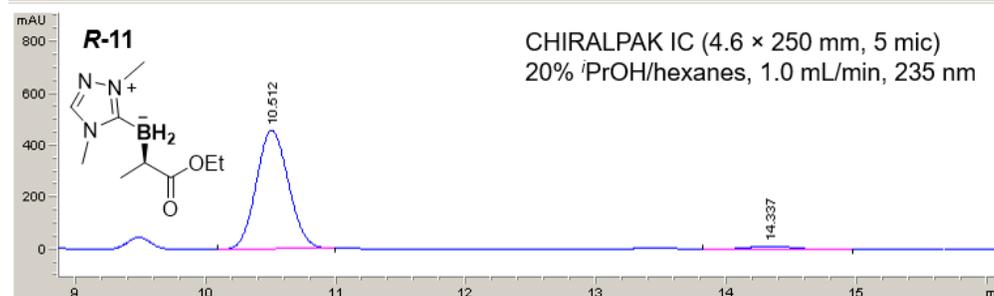
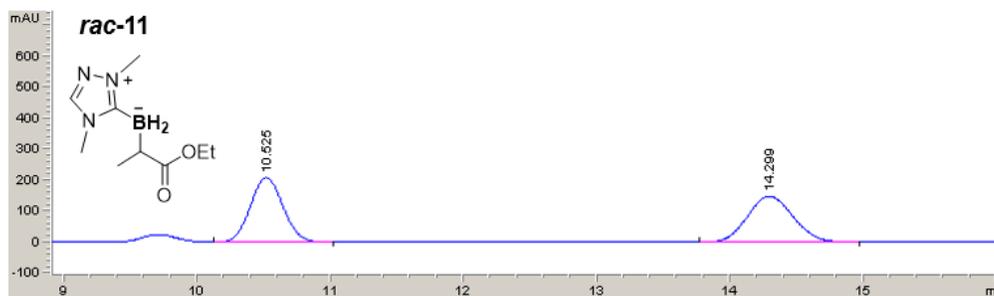
rac-8			R-8		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
7.002	16485.6	49.0	7.019	4675.3	97.1
8.748	17138.5	51.0	8.807	140.6	2.9



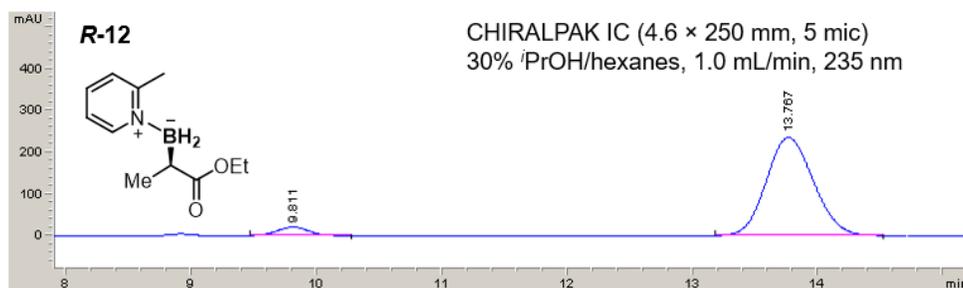
rac-9			R-9		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
9.319	23716.2	49.4	9.323	18767.6	98.7
11.722	24243.2	50.5	11.82	245.9	1.3



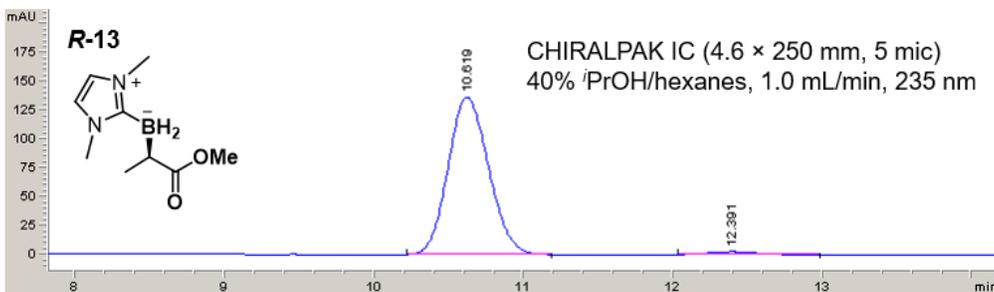
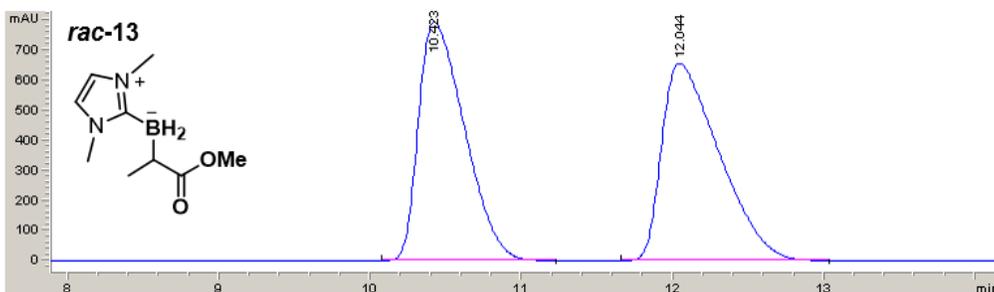
rac-10			R-10		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
12.461	17593.3	49.8	12.436	15102.1	97.5
13.592	17704.5	50.2	13.627	394.2	2.5



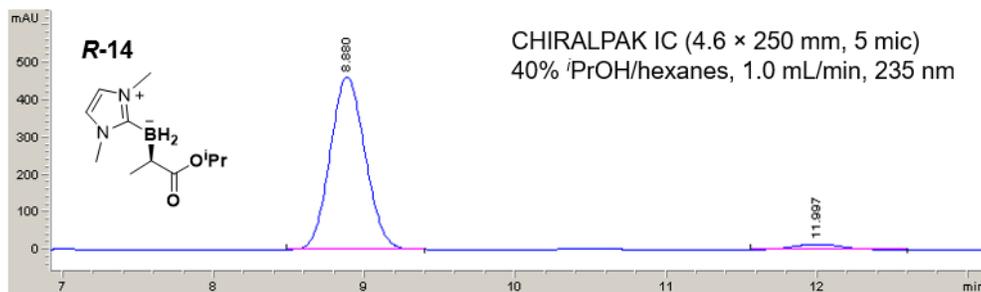
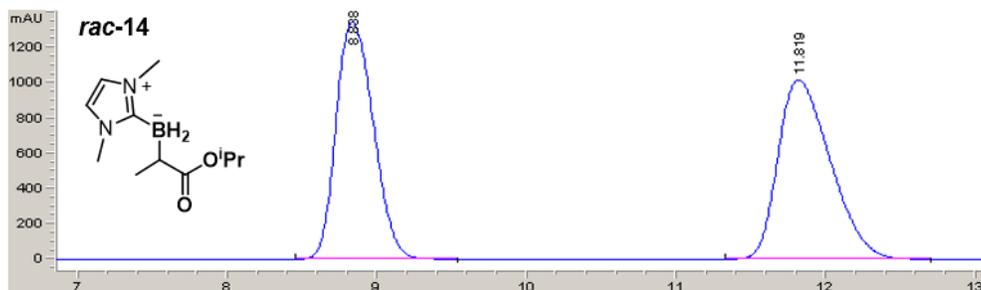
rac-11			R-11		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
10.525	3548.2	50.0	10.512	7851.9	96.3
14.299	3548	50.0	14.377	303.5	3.7



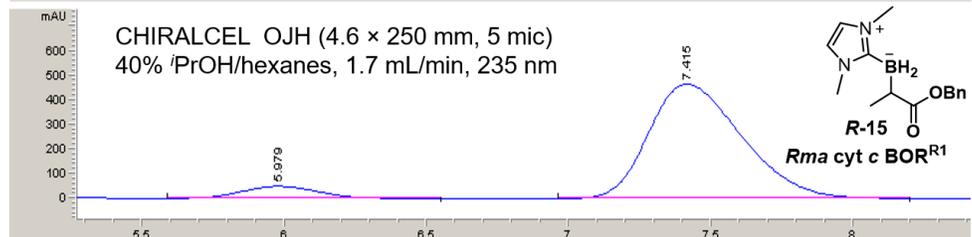
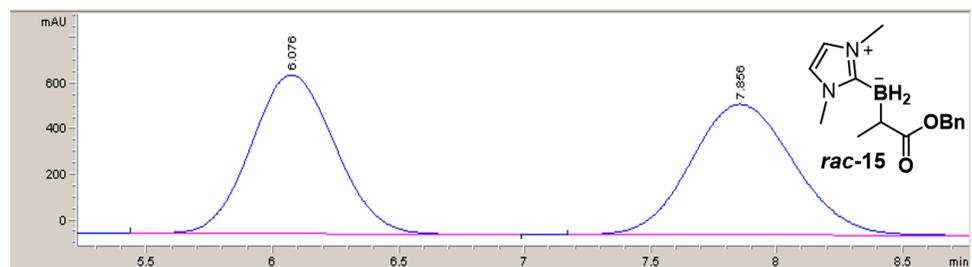
rac-12			R-12		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
9.751	3855	50.0	9.811	300.4	4.4
13.748	3856.4	50.0	13.767	6600.1	95.6



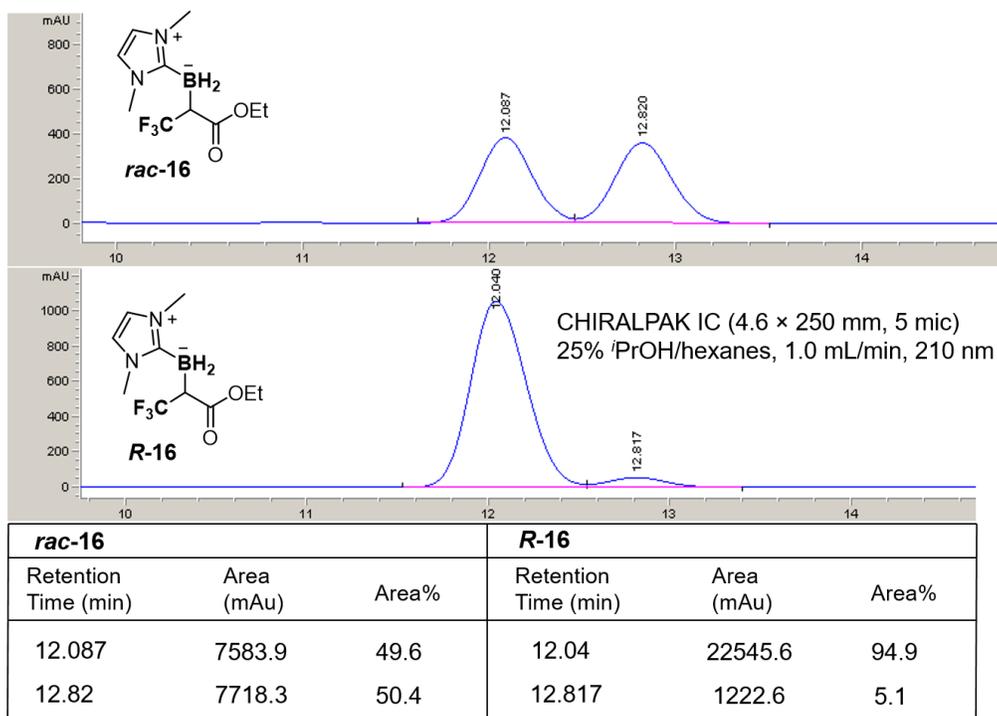
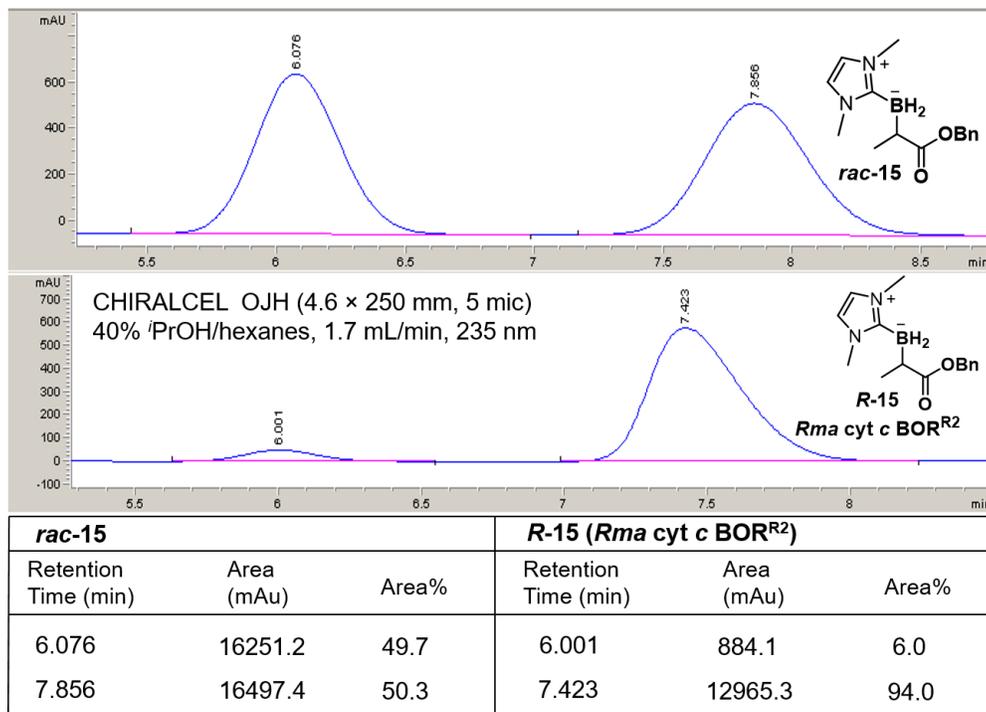
rac-13			R-13		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
10.423	792.9	49.2	10.619	2569.9	98.0
12.044	655.8	50.8	12.391	54.3	2.0

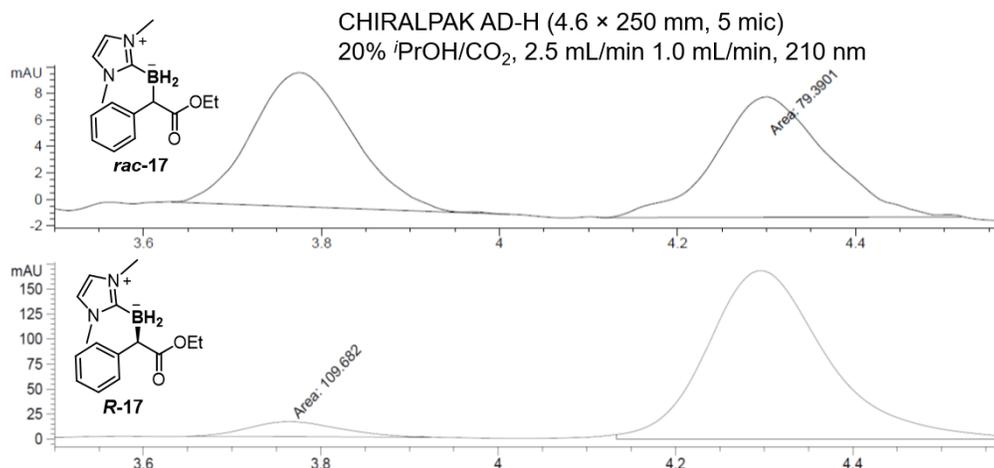


rac-14			R-14		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
8.838	23352.5	49.0	8.88	7453.1	95.9
11.819	24689.3	51.0	11.997	323.1	4.1

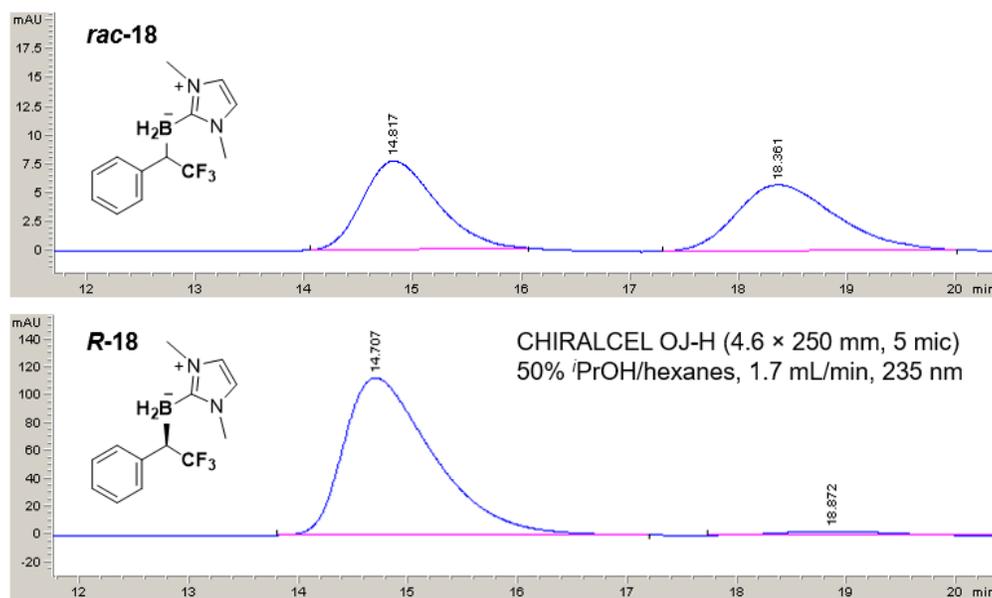


rac-15			R-15 (<i>Rma cyt c BOR^{R1}</i>)		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
6.076	16251.2	49.7	5.979	920	8.0
7.856	16497.4	50.3	7.415	10567.3	92.0

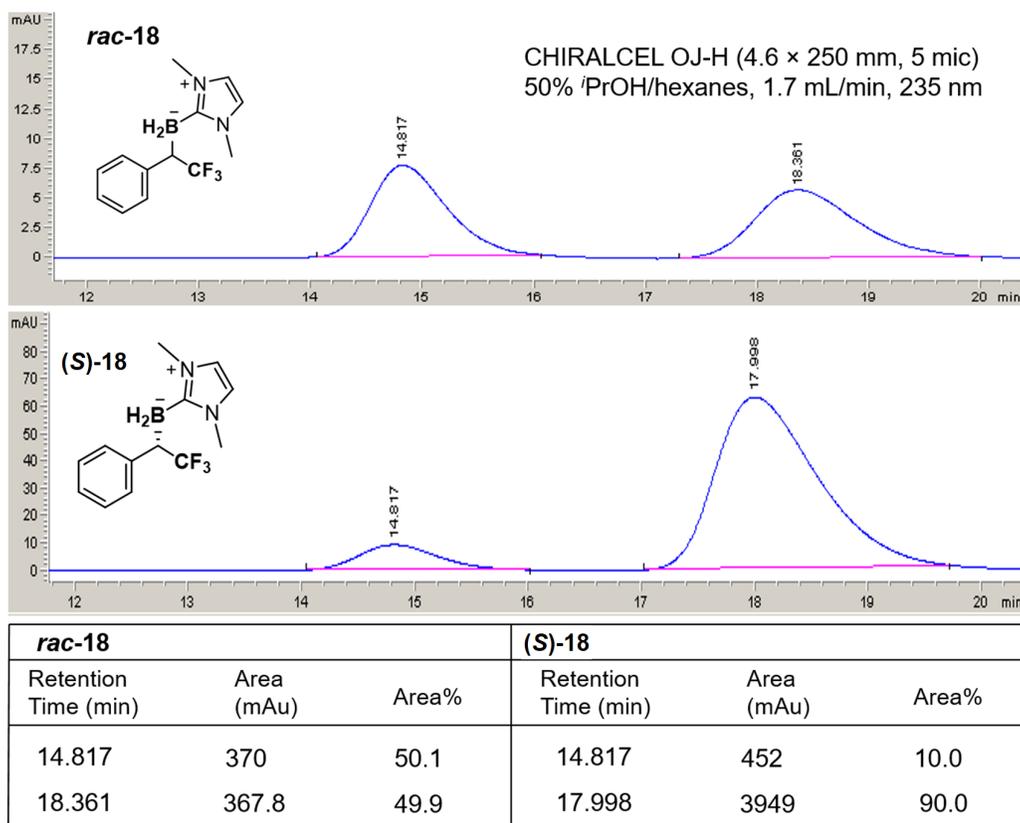




rac-17			R-17		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
3.775	79.9	50.1	3.765	109	6.0
4.299	79.4	49.9	4.296	1583	94.0

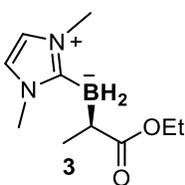


rac-18			R-18		
Retention Time (min)	Area (mAu)	Area%	Retention Time (min)	Area (mAu)	Area%
14.817	370	50.1	14.707	6420.3	96.7
18.361	367.8	49.9	18.872	215.8	3.3



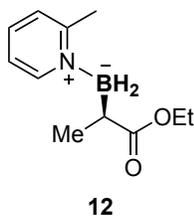
VIII. Preparative Scale Enzymatic Reactions

Preparation of whole-cell suspensions for borylation reactions: HB_{amp/chlor} (200 mL) in a 1 L flask was inoculated with an overnight culture (4 mL, LB_{amp/chlor}) of recombinant *E. coli*[®] EXPRESS BL21(DE3) cells containing a pET22b(+) plasmid encoding *Rma* cyt *c* variant, and the pEC86 plasmid. The culture was shaken at 37 °C and 250 rpm (no humidity control) until the OD₆₀₀ was 0.7 (typically 2 - 3 hours). The culture was placed on ice for 30 minutes, and IPTG and ALA were added to final concentrations of 20 μM and 200 μM, respectively. The incubator temperature was reduced to 20 °C, and the culture was allowed to shake for 22 hours at 140 rpm. Cells were pelleted by centrifugation (4 °C, 5 min, 4,000xg), resuspended in M9-N buffer and adjusted to OD₆₀₀ = 30. The whole-cell suspension was placed on ice and bubbled with Ar for 30 min.



Biocatalytic synthesis of (1,3-dimethyl-1*H*-imidazol-3-ium-2-yl)(1-ethoxy-1-oxopropan-2-yl)dihydroborate (**3**) (0.5 mmol scale reaction).

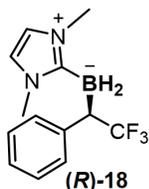
Under anaerobic conditions, to a 40 mL vial were added 12 mL *Rma* cyt *c* BOR^{R1} whole-cell suspension (OD₆₀₀ = 30), 3 mL glucose solution (250 mM), *di*MeNHC-BH₃ solution (125 μL, 2 M in MeCN) and Me-EDA (125 μL, 2 M in MeCN). The vial was capped and shaken at 520 rpm in an anaerobic chamber at room temperature. After 4 hours, another portion of *di*MeNHC-BH₃ (125 μL, 2 M in MeCN) and Me-EDA (125 μL, 2 M in MeCN) were added and the vial was shaken for 8 more hours at 520 rpm. The reaction mixture was then transferred to a 50 mL Falcon tube and extracted by 30 mL 3:7 hexanes/EtOAc *via* vortexing (30 s for three times). Centrifugation (5,000xg, 5 min) was used to completely separate the organic and aqueous layers. After removal of the organic layers, two additional rounds of extraction were performed. The combined organic extracts were dried over anhydrous Na₂SO₄, concentrated, and purified by flash chromatography (dry loading) with EtOAc/hexanes (5% to 60% EtOAc/hexanes gradient) to afford pure organoborane product **3** (79 mg, 0.376 mmol, 75% yield). The protein concentration of OD₆₀₀ = 30 whole-cell solution was determined to be 10.41 μM by hemochrome assay after cell lysis by sonication. The total turnover number for this reaction was 3000. The stereoselectivity of the product was determined as 97.5:2.5 e.r. by normal-phase chiral HPLC. [α]_D²³ = + 114.5 (*c* 0.19, EtOAc). ¹H NMR (400 MHz, Chloroform-*d*) δ 6.82 (s, 2H), 3.98 – 3.78 (m, 2H), 3.75 (s, 6H), 1.95 – 1.10 (m, 2H), 1.88 (br s, 1H), 1.10 (d, *J* = 6.2 Hz, 3H), 1.06 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (101 MHz, Chloroform-*d*) δ 183.5, 120.4, 58.7, 36.2, 30.5, 17.8, 14.6. The boron-bound NHC quaternary carbon was not resolved; ¹¹B NMR (128 MHz, Chloroform-*d*) δ -24.55 (t, *J* = 90 Hz).



Biocatalytic synthesis of ethyl 2-((2-methyl-pyridin-1-yl)boraneyl)propanoate (**12**) on gram scale.

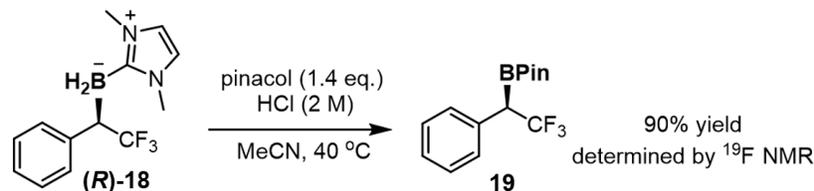
Under anaerobic conditions, to a 250 mL conical flask were added 50 mL *Rma* cyt *c* BOR^{R1} whole-cell solution (OD₆₀₀ = 30), glucose (2.6 mL, 1 M), picoline borane (1.4 mL, 2 M in MeCN) and Me-EDA (1.4 M in MeCN). The flask was shaken at 240 rpm in an anaerobic chamber. At 3 h intervals, two additional batches of whole-cell solution (50 mL), glucose (2.6 mL, 1 M), picoline borane (1.4 mL, 2 M in MeCN) and Me-EDA (1.4 M in MeCN) were added. The reaction mixture was shaken for a total of 24 hours and then divided between six 50 mL Falcon tubes. 25 mL 3:7 hexanes/EtOAc was added to each tube to

extract the borylation product via vortexing (30 s for three times) and centrifugation (5,000xg, 5 min). After removal of the organic layers, two additional rounds of extraction were performed. The combined organic extracts were dried over anhydrous Na₂SO₄, concentrated, and purified by flash chromatography (dry loading) with EtOAc/hexanes (5% to 40% EtOAc/hexanes gradient) to afford pure organoborane product **12** (0.74 g, 3.57 mmol, 42% yield). The protein concentration of OD₆₀₀ = 30 whole-cell solution was determined to be 8.18 μM by hemochrome assay after cell lysis by sonication. The total turnover number for this reaction was 2910. The stereoselectivity of the product was determined as 96:4 e.r. by normal-phase chiral HPLC. [α]_D²³ = + 117.2 (*c* 0.37, EtOAc). ¹H NMR (400 MHz, Chloroform-*d*) δ 8.53 (dd, *J* = 6.0, 1.6 Hz, 1H), 7.84 (td, *J* = 7.7, 1.7 Hz, 1H), 7.42 – 7.36 (m, 1H), 7.33 – 7.28 (m, 1H), 3.79 (AB qq, *J* = 10.8, 7.1 Hz, 2H), 3.30 – 2.15 (m, 2H), 2.77 (s, 3H), 2.05 – 1.92 (m, 1H), 1.05 (d, *J* = 6.8 Hz, 3H), 0.94 (t, *J* = 7.1 Hz, 3H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 182.13, 157.92, 149.35, 140.26, 127.67, 122.60, 58.83, 32.78, 22.76, 15.16, 14.61. ¹¹B NMR (128 MHz, Chloroform-*d*) δ –5.10 (t, *J* = 103 Hz).

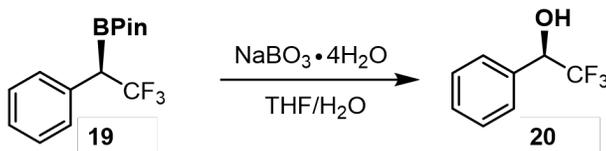


Biocatalytic synthesis of (1,3-Dimethyl-1H-imidazol-3-ium-2-yl)(2,2,2-trifluoro-1-phenylethyl)dihydroborate ((*R*)-18) at (1 mmol scale reaction).

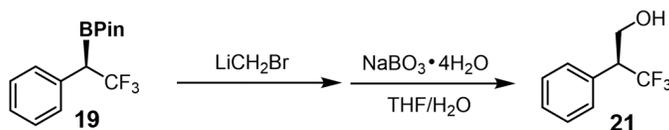
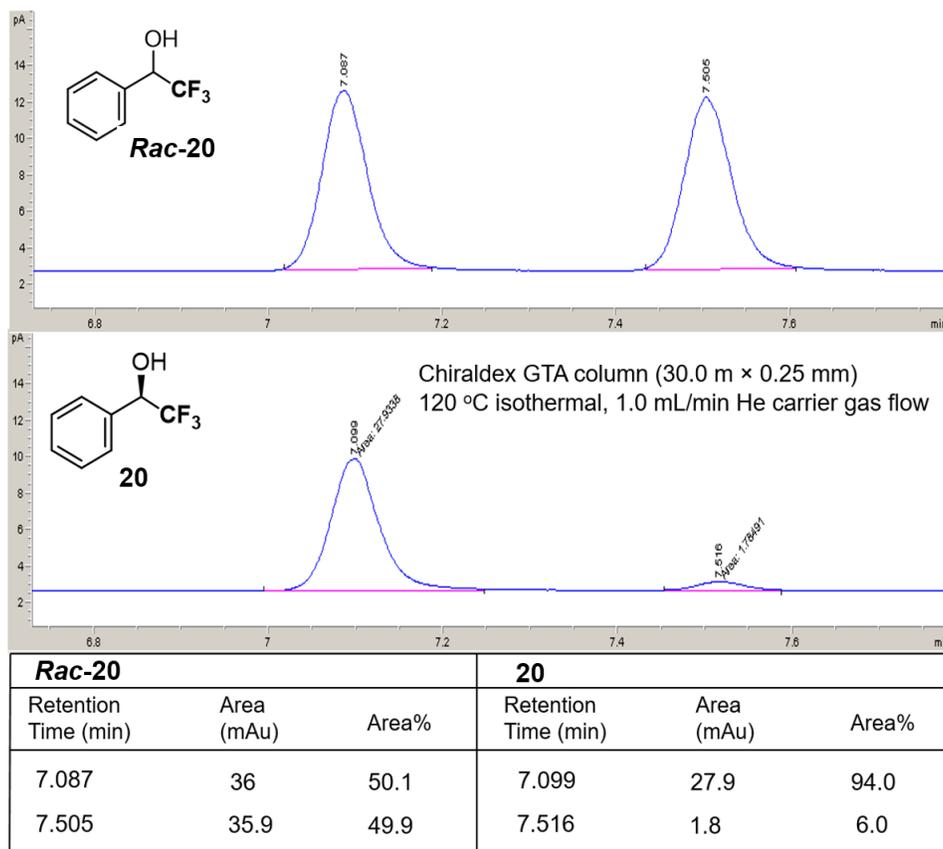
To a 250 mL conical flask were added 40 mL *Rma* cyt *c* BOR^{P1} whole-cell solution (OD₆₀₀ = 30), glucose (2.6 mL, 1 M), *di*MeNHC-BH₃ (1.2 mL, 0.6 M in MeCN) and CF₃-DMB (1.0 mL, 0.6 M in MeCN). The flask was shaken at 240 rpm in the anaerobic chamber. After 6 hours, another batch of whole-cell solution (40 mL, OD₆₀₀ = 30), glucose (2.6 mL, 1 M), *di*MeNHC-BH₃ (1.2 mL, 0.6 M in MeCN) and CF₃-DMB (1.0 mL, 0.6 M in MeCN) were added to the reaction mixture. The reaction mixture was shaken for a total of 30 hours and then divided between four 50 mL Falcon tubes. 25 mL 3:7 hexanes/EtOAc was added to each tube to extract the borylation product via vortexing (30 s for three times) and centrifugation (5,000xg, 5 min). After removal of the organic layers, two additional rounds of extraction were performed. The combined organic extracts were dried over anhydrous Na₂SO₄, concentrated, and purified by silica column chromatography (dry loading) with EtOAc/hexanes (5% to 50% EtOAc/hexanes gradient) to afford pure organoborane product (*R*)-**18** (130 mg, 0.485 mmol, 40% yield). Recovered borane starting material is 82 mg. The yield based on consumed starting material is 70%. The protein concentration of OD₆₀₀ = 30 whole-cell solution was determined to be 6.06 μM by hemochrome assay after cell lysis by sonication. The total turnover number (TTN) for this reaction was 1000. The stereoselectivity of the product was determined as 96:4 e.r. by normal-phase chiral HPLC. [α]_D²³ = – 81.3 (*c* 0.67, EtOAc). ¹H NMR (400 MHz, Chloroform-*d*) δ 7.23 – 7.05 (m, 5H), 6.76 (s, 2H), 3.52 (s, 6H), 2.90 – 2.60 (m, 1H), 2.25 – 1.40 (m, 2H); ¹³C NMR (101 MHz, Chloroform-*d*) δ 169.1, 143.7 (d, *J* = 3.5 Hz), 131.4 (q, *J* = 278.0 Hz), 128.4, 128.3, 125.2, 120.8, 43.5, 36.0; ¹¹B NMR (128 MHz, Chloroform-*d*) δ – 26.72 (t, *J* = 90 Hz); ¹⁹F NMR (282 MHz, Chloroform-*d*) δ –61.80 (d, *J* = 13 Hz).

IX. Derivatization of Enzymatic Borylation Product (*R*)-18

(A) Conversion of (*R*)-18 to the corresponding pinacol boronate ester 19. The protocol was modified from that reported by Zhou *et al.*¹². To a 40 mL vial with screw cap were added 54 mg enzymatic product (**(R)-18**) (0.2 mmol) and a stir bar. The vial was evacuated and backfilled with argon three times. 4 mL acetonitrile solution of pinacol (33 mg, 0.28 mmol, 1.4 eq.) was added to the vial *via* syringe. The resulting solution was stirred for 5 min for (**(R)-18**) to dissolve, followed by the addition of 300 μL of 2 M HCl. The vial was stirred at 40 $^\circ\text{C}$. The reaction can be monitored by GC-MS (usual reaction time is 10 - 12 hours) or ^{19}F NMR (**19** has a chemical shift at $\delta -62.75$ ppm (d, $J = 12$ Hz)). After reaction completion, 10 μL fluorobenzene was added to the reaction mixture. An aliquot of reaction mixture was diluted with CDCl_3 to measure the yield via ^{19}F NMR. The formation of **19** was confirmed by GC-MS, and by conversion of **19** to alcohol **20** (see section **B** below).

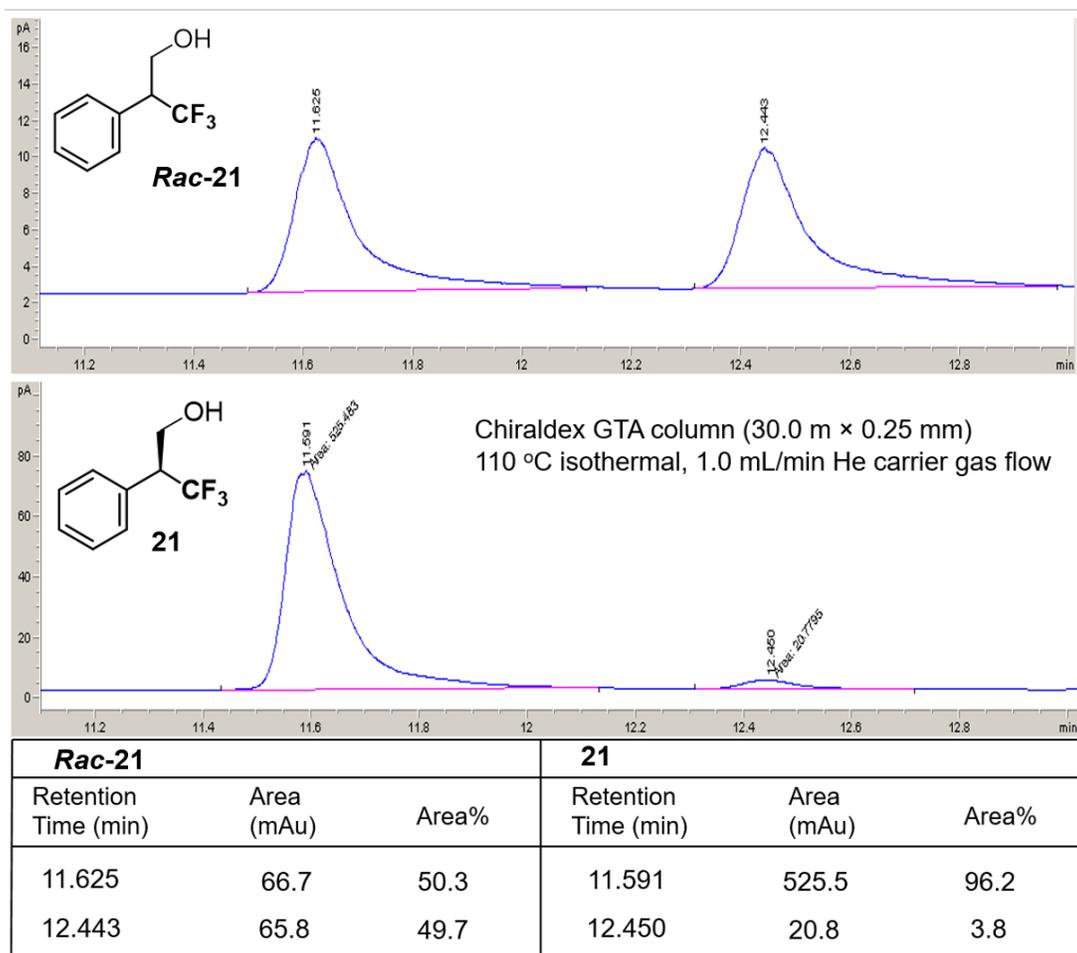


(B) Conversion of 19 to alcohol 20. To the reaction mixture obtained after ligand exchange with pinacol in step **A**, 5 mL of water was added, and the mixture was extracted with 15 mL of 1:1 hexanes:EtOAc three times. The solvent was removed under reduced pressure and the vial was backfilled with argon. The crude product **19** was dissolved in 15 mL of pentane and passed through a syringe filter to remove the insoluble materials. This process was repeated two additional times to ensure all soluble materials were extracted. The combined organic extracts were dried over anhydrous Na_2SO_4 and concentrated under reduced pressure. Under argon, THF (1 mL) was added to the vial to dissolve the crude product **19** followed by the addition of H_2O (1 mL) and 154 mg of $\text{NaBO}_3 \cdot 4\text{H}_2\text{O}$. The reaction mixture was stirred for 6 hours and extracted with EtOAc (15 mL) three times. The combined organic extracts were concentrated under reduced pressure and purified by flash column chromatography to yield alcohol **20** (0 - 30% hexanes/EtOAc). 27.4 mg alcohol **1** was obtained (78% yield). The e.r. was confirmed by chiral GC with FID detector using a Chiraldex GTA column (30.0 m \times 0.25 mm) (conditions: 120 $^\circ\text{C}$ isothermal at 1.0 mL/min He carrier gas flow). Retention time: 7.09 min for *R* enantiomer, 7.52 min for *S* enantiomer). This compound is known¹³. ^1H NMR (400 MHz, Chloroform-*d*) δ 7.56 – 7.34 (m, 5H), 5.03 (qd, $J = 6.7, 4.4$ Hz, 1H), 2.57 (dd, $J = 4.5, 1.5$ Hz, 1H) ^{13}C NMR (101 MHz, Chloroform-*d*) δ 134.3, 130.0, 129.0, 127.8, 124.6 (q, $J = 282.1$) Hz, 73.2 (q, $J = 32.0$ Hz). ^{19}F NMR (282 MHz, Chloroform-*d*) $\delta -78.40$ (d, $J = 7$ Hz).

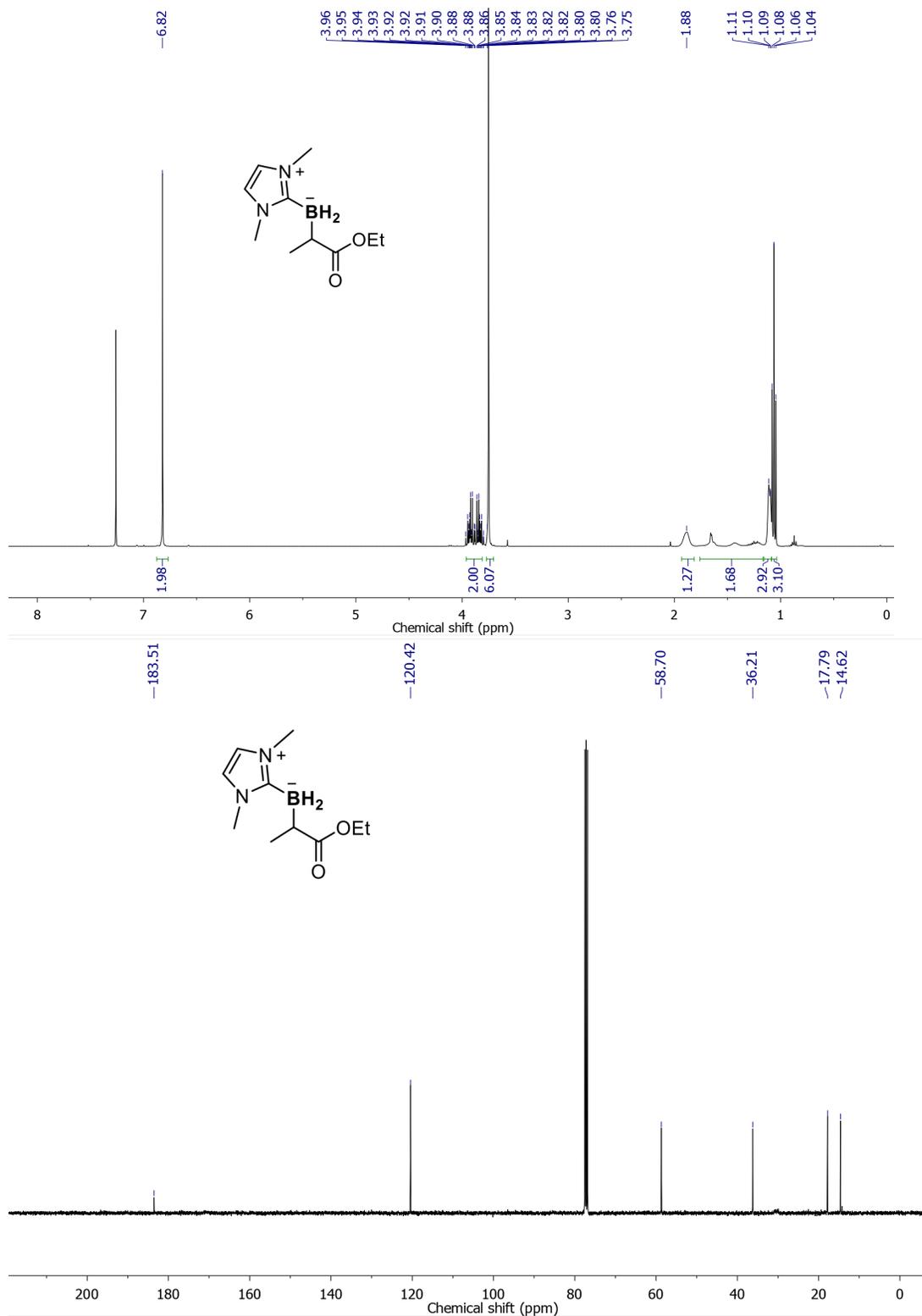


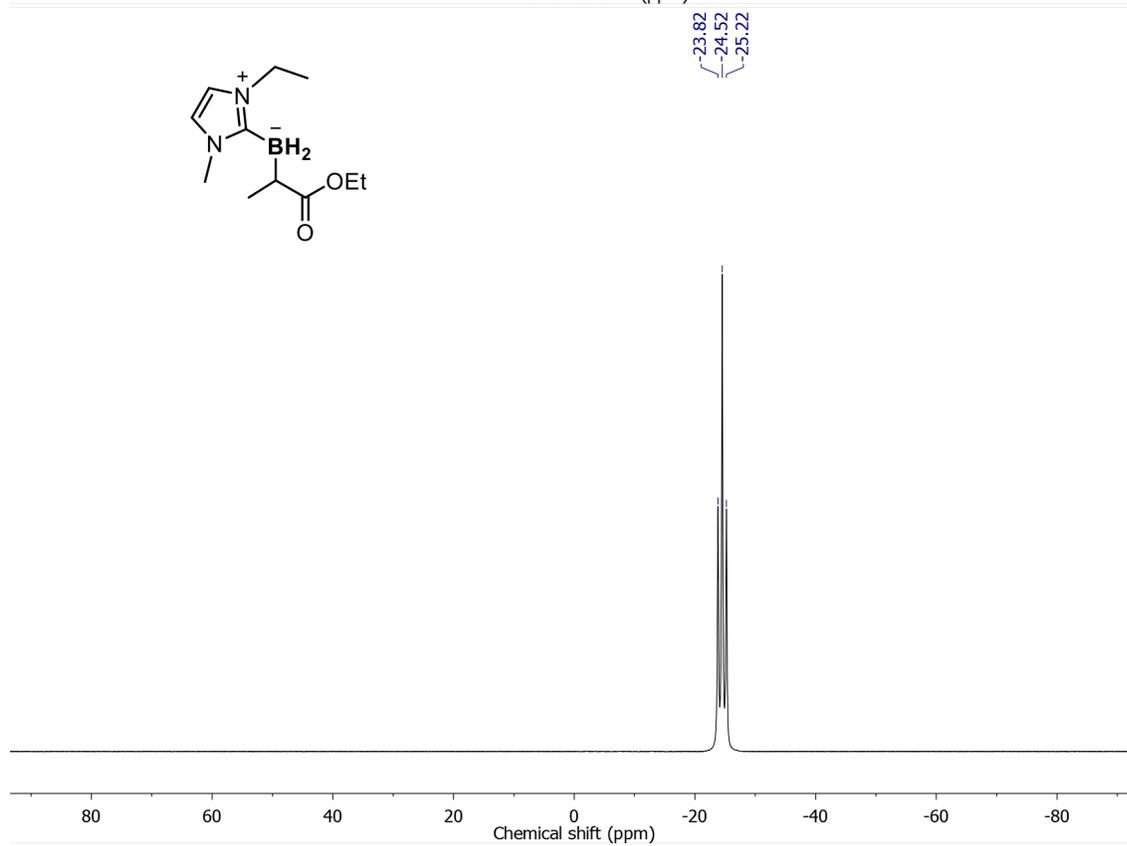
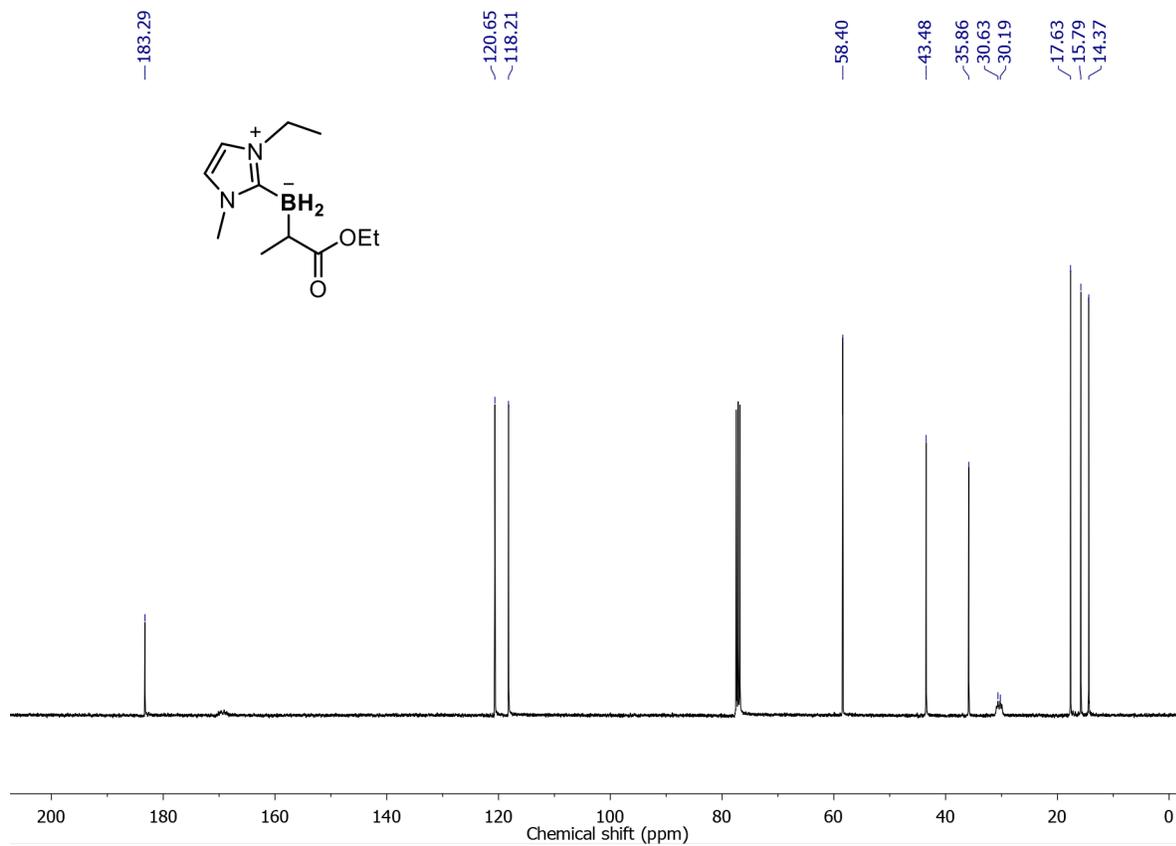
(C) Conversion of 19 to alcohol 21 via Matteson homologation and oxidation. To the reaction mixture obtained after ligand exchange with pinacol in step A, 5 mL of water was added and the mixture was extracted with 15 mL of 1:1 hexanes:EtOAc three times. The solvent was removed under reduced pressure and the vial was backfilled with argon. The crude product **19** was dissolved in 15 mL of pentane and passed through a syringe filter to remove the insoluble materials. This process was repeated two additional times to ensure all soluble materials were extracted. The combined organic extracts were concentrated under reduced pressure. Under argon, 2 mL of anhydrous THF and dibromomethane (35 μ L, 2.5 eq.) were added and the vial was cooled in a dry ice/acetone bath. *n*-BuLi (160 μ L, 2.5 M in hexanes, 2.0 eq.) was added dropwise over 30 min. The solution was allowed to warm to room temperature slowly. The reaction mixture was then diluted with 3 mL *sat.* NH_4Cl and extracted with EtOAc (15 mL) for three times. The combined organic extracts were dried over anhydrous Na_2SO_4 and concentrated under reduced pressure. The resulting crude mixture was dissolved in THF (1 mL). 1 mL of H_2O and $\text{NaBO}_3 \cdot 4\text{H}_2\text{O}$ (154 mg) were then added and the reaction mixture was stirred for 6 hours. The reaction was then extracted with EtOAc. The organic extracts were dried, concentrated under reduced pressure, and purified by flash column chromatography to yield alcohol **21** (0 - 30%

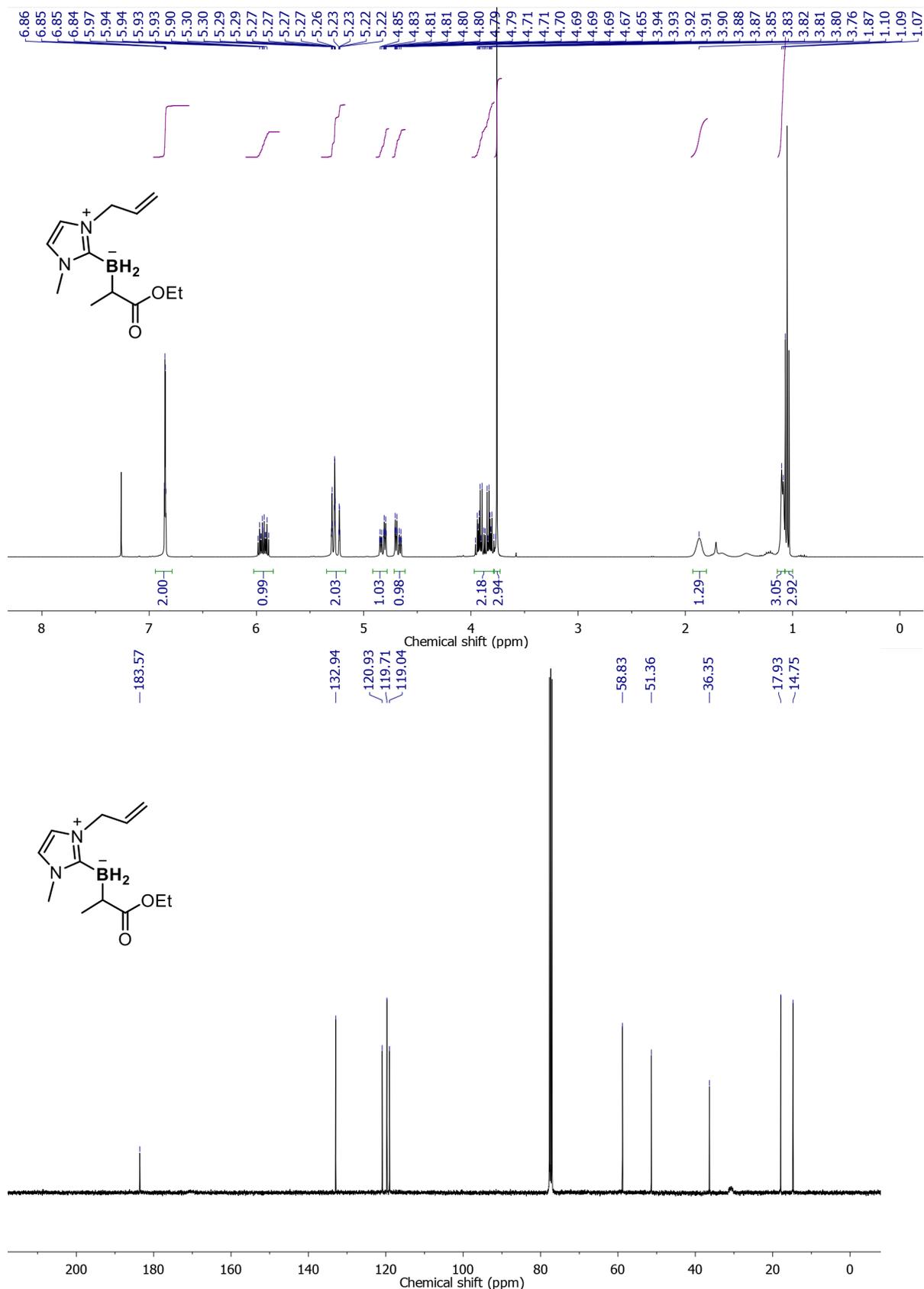
hexanes/EtOAc). 12.6 mg alcohol **21** was obtained (33% overall yield, 38% for the Matteson homologation and oxidation steps). The e.r. was confirmed by chiral GC with FID detector using a Chiraldex GTA column (30.0 m × 0.25 mm) (conditions: 110 °C isothermal at 1.0 mL/min He carrier gas flow). Retention time: 11.625 min for *S* enantiomer, 12.443 min for *R* enantiomer). This compound is known¹⁴. ¹H NMR (400 MHz, Chloroform-*d*) δ 7.44 – 7.31 (m, 5H), 4.20 (dd, *J* = 11.7, 5.7 Hz, 1H), 4.04 (dd, *J* = 11.4, 7.8 Hz, 1H), 3.56 (qdd, *J* = 9.4, 7.8, 5.8 Hz, 1H), 1.57 (s, 1H). ¹³C NMR (101 MHz, Chloroform-*d*) δ 132.8 (d, *J* = 2.2 Hz), 129.5, 129.4, 129.0, 126.4 (q, *J* = 280.5 Hz), 61.7 (q, *J* = 2.9 Hz), 52.9 (q, *J* = 25.5 Hz). ¹⁹F NMR (282 MHz, Chloroform-*d*) δ –67.47 (d, *J* = 9 Hz).

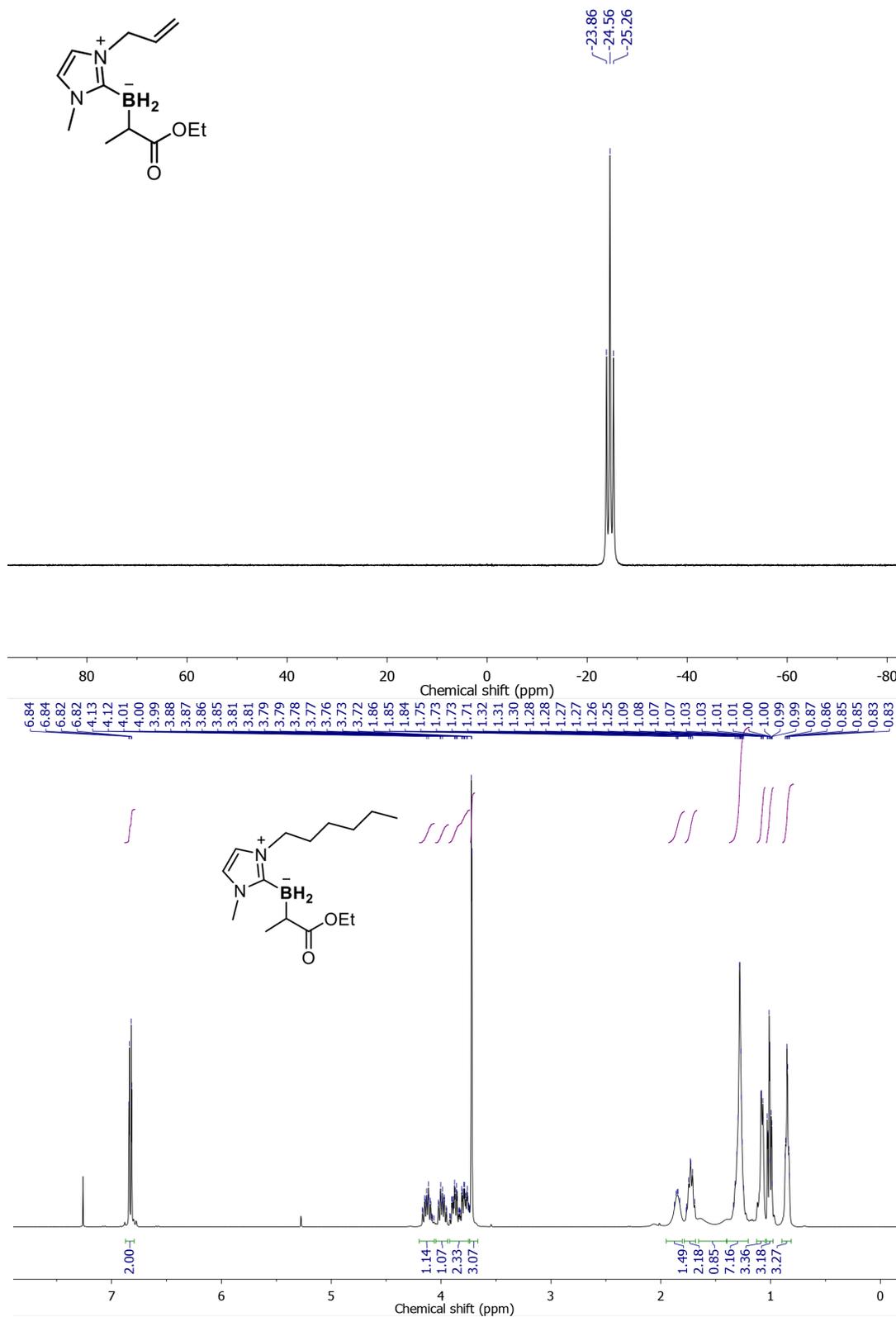


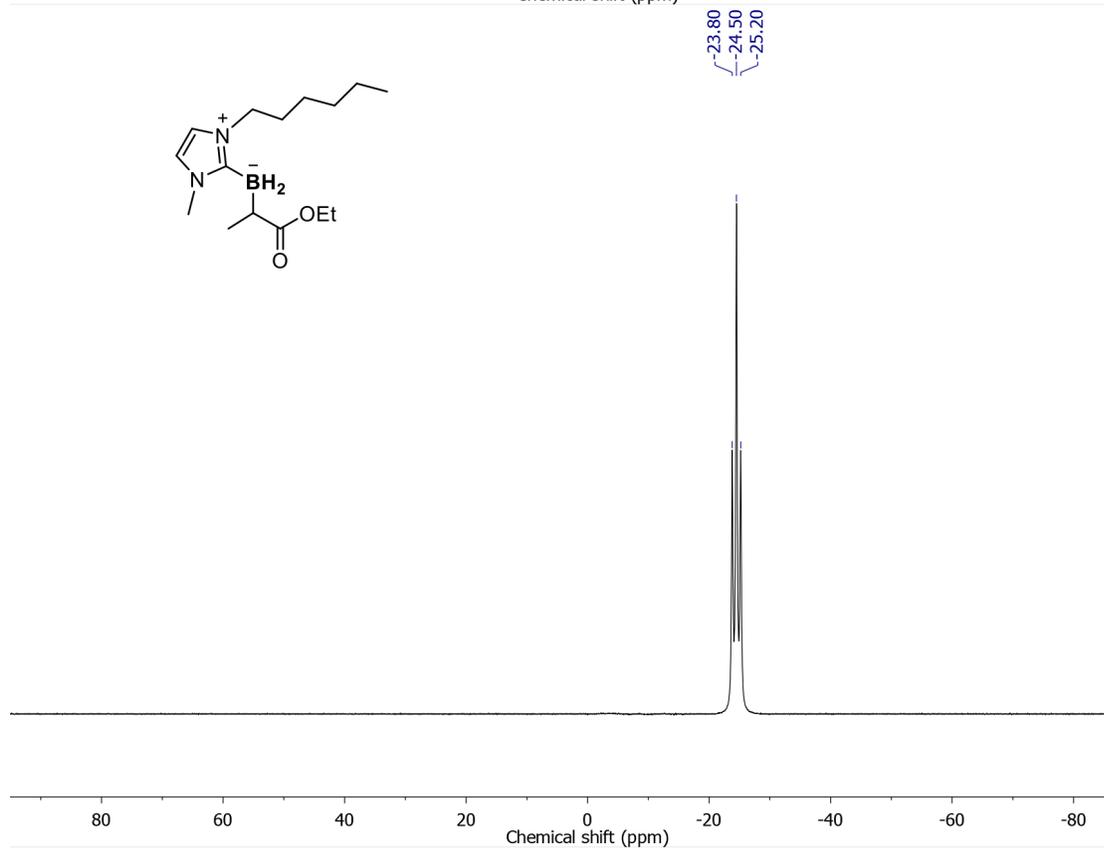
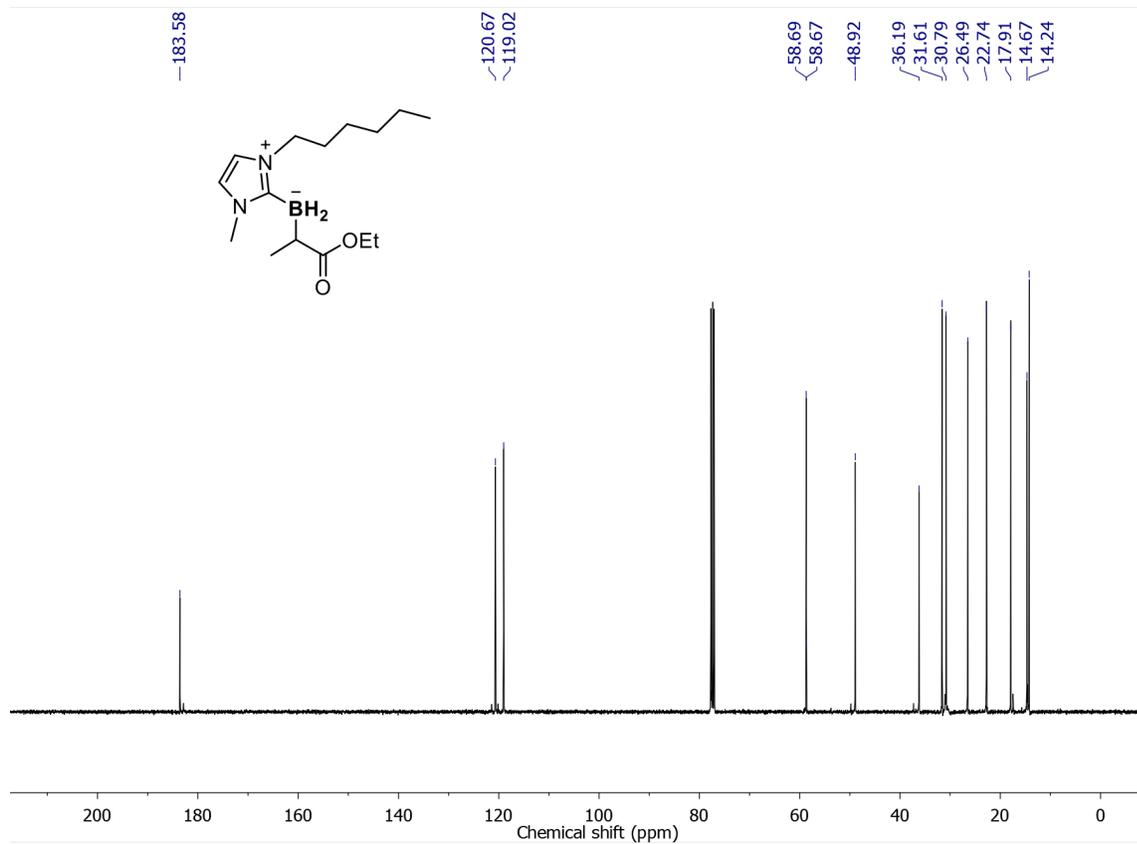
X. NMR Spectra

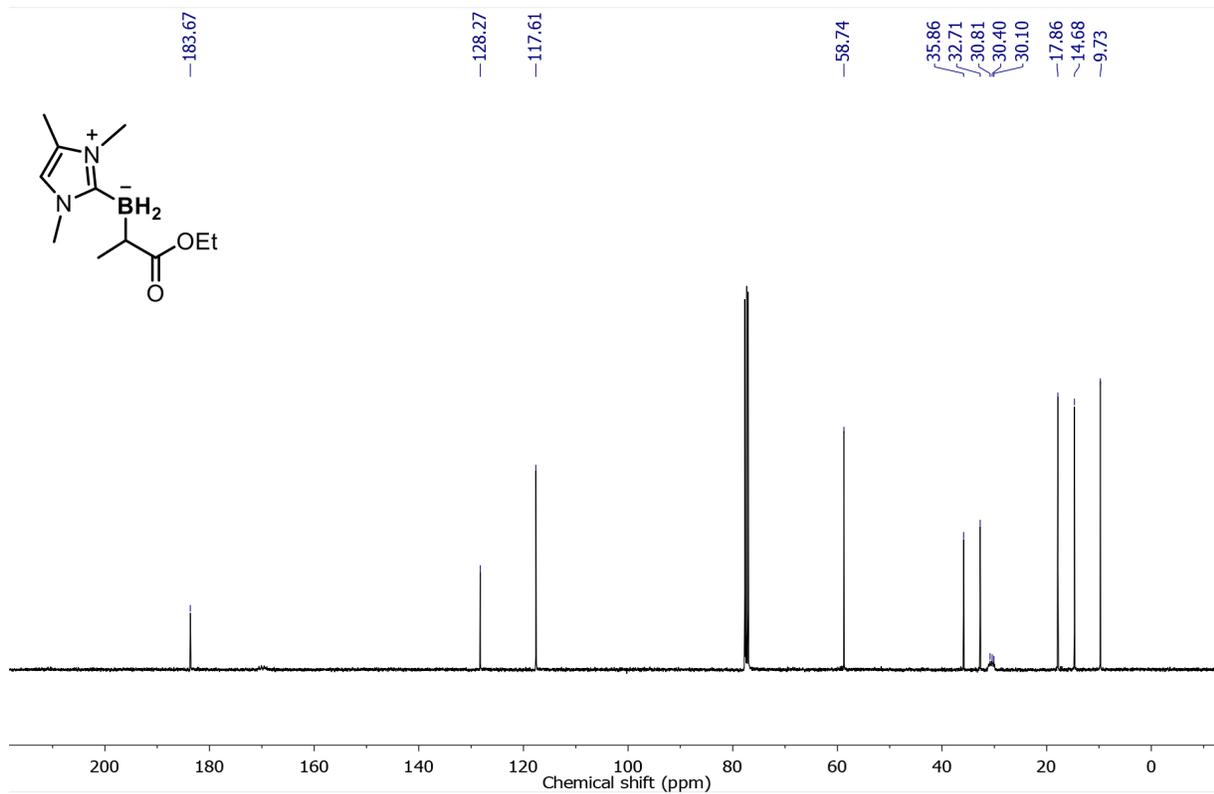
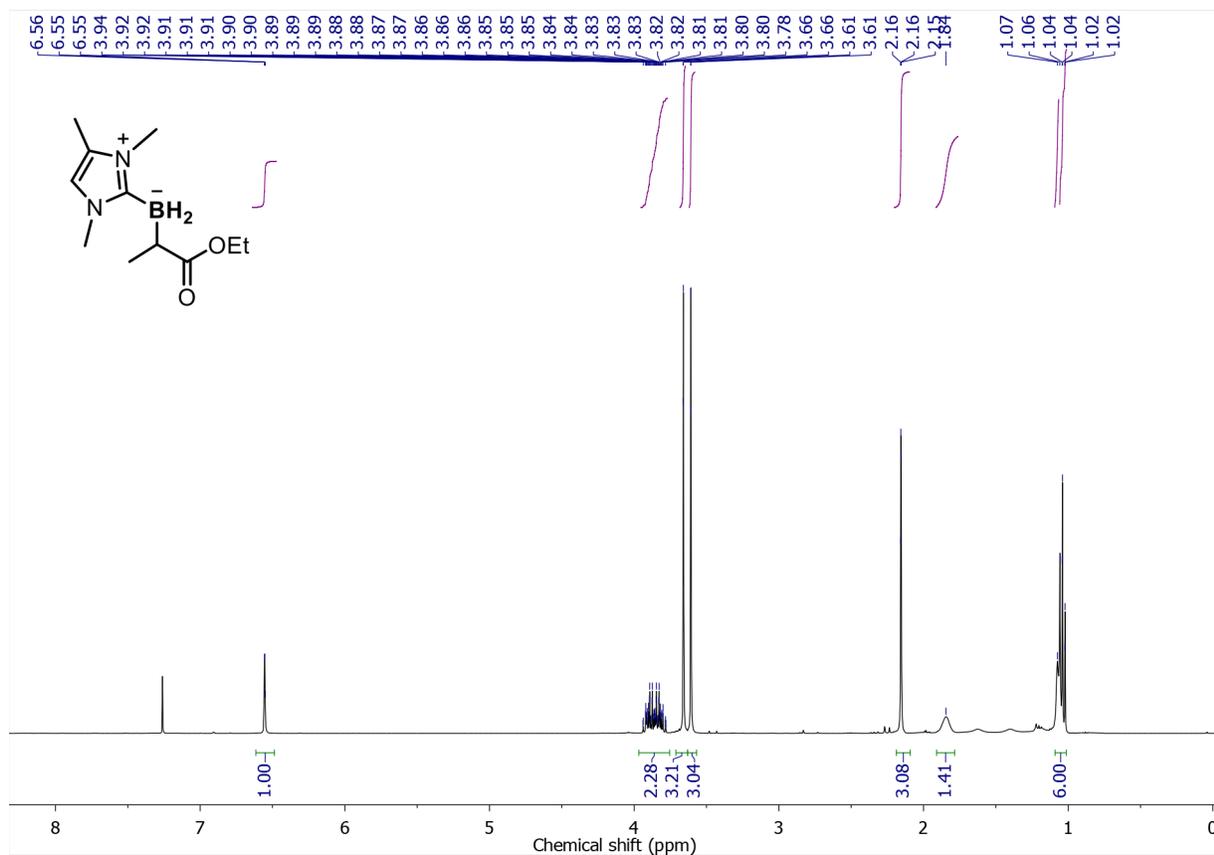


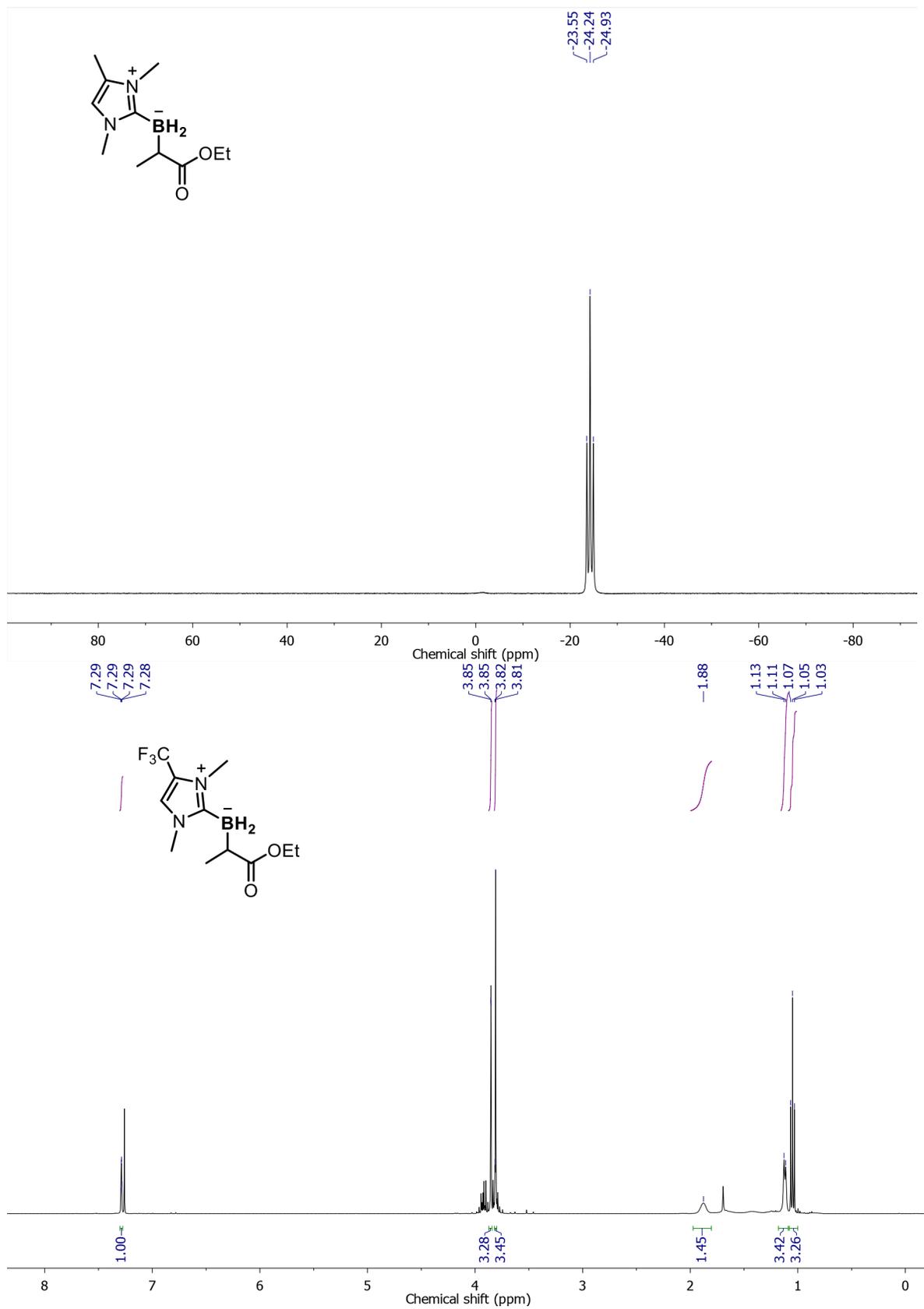


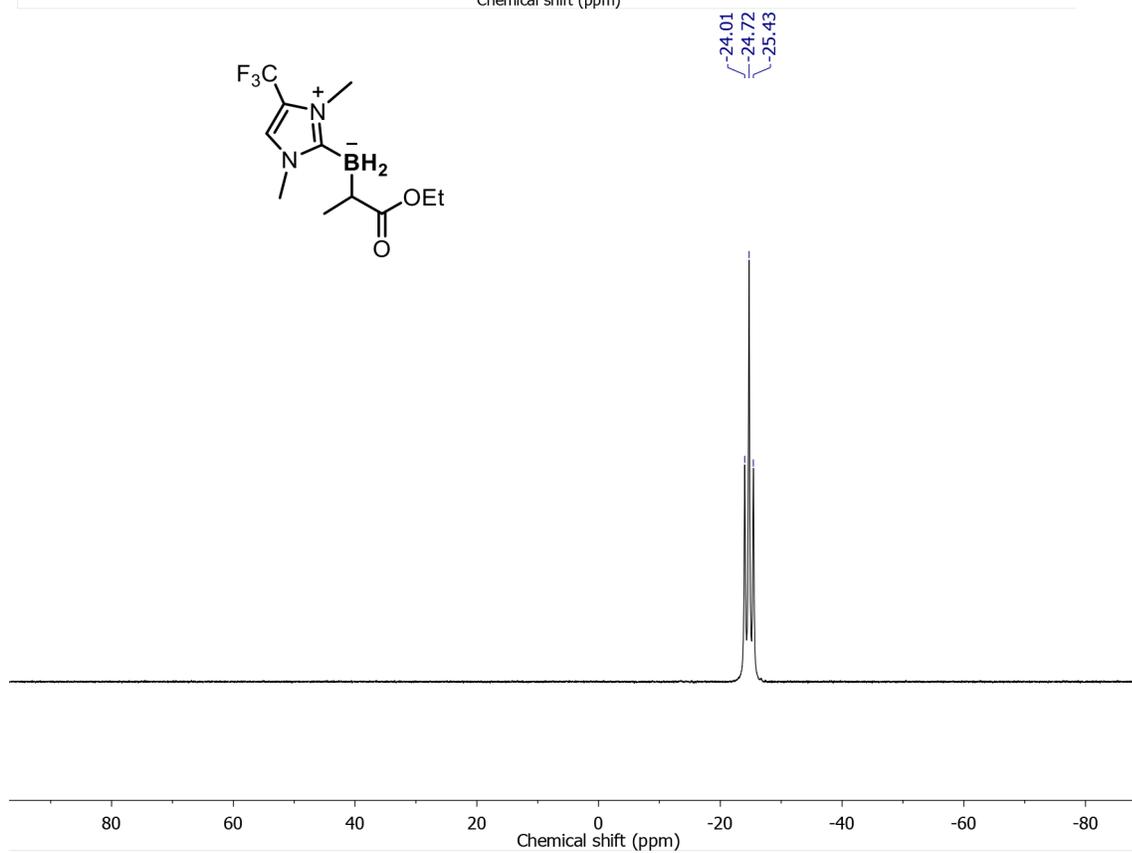
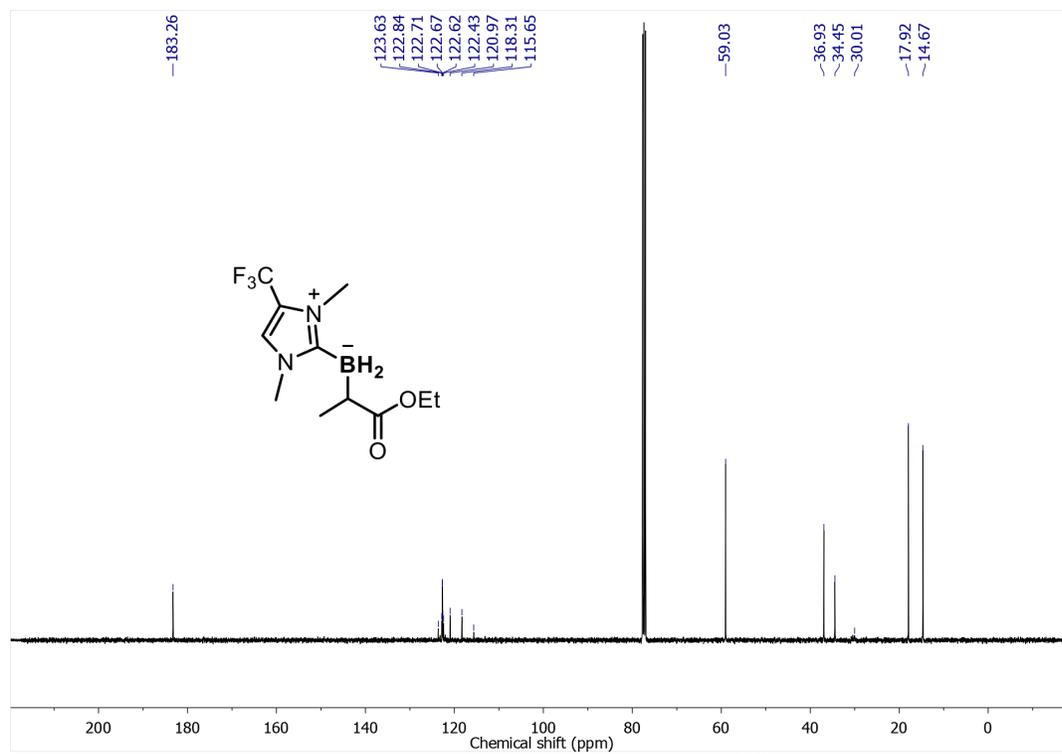


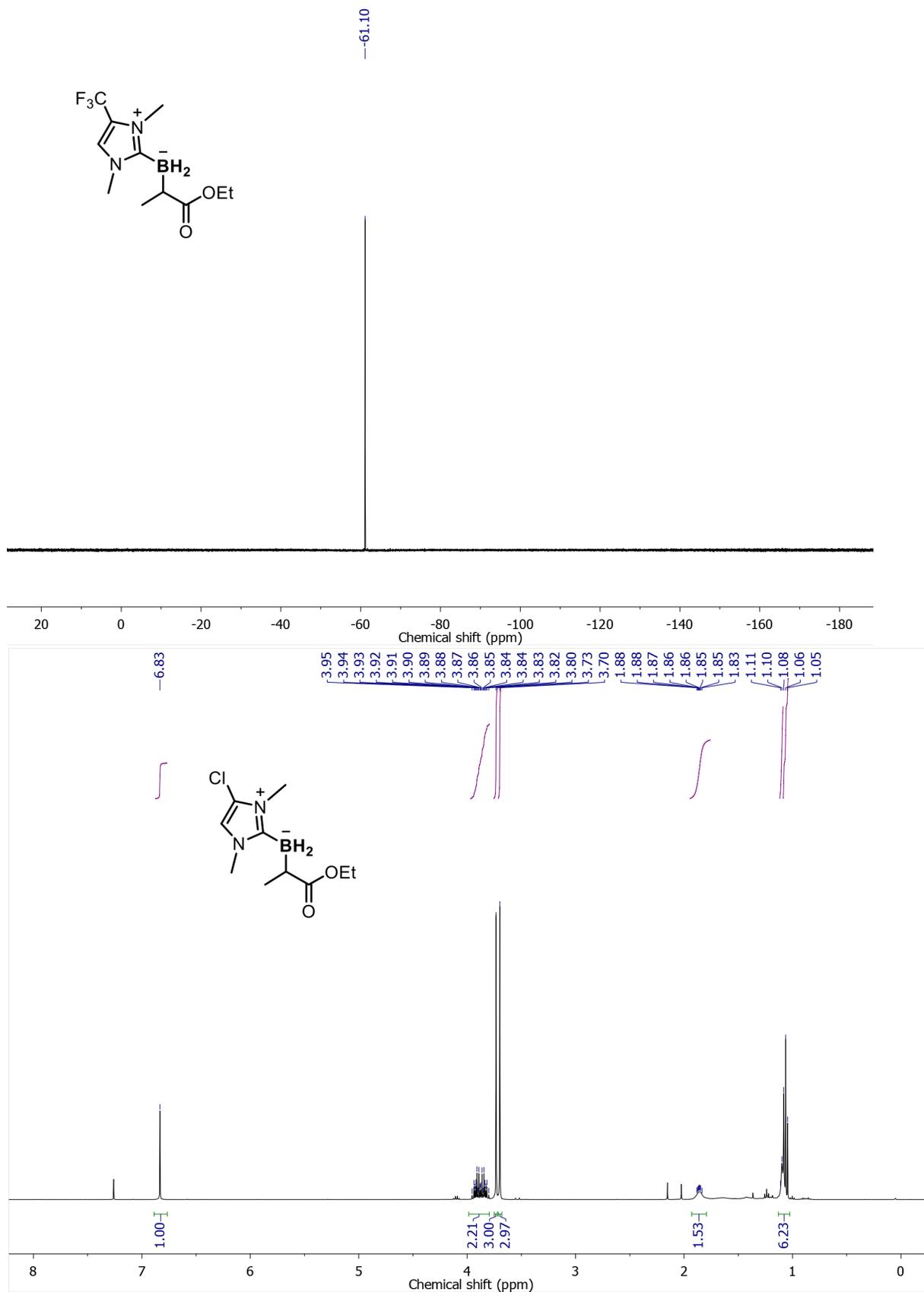


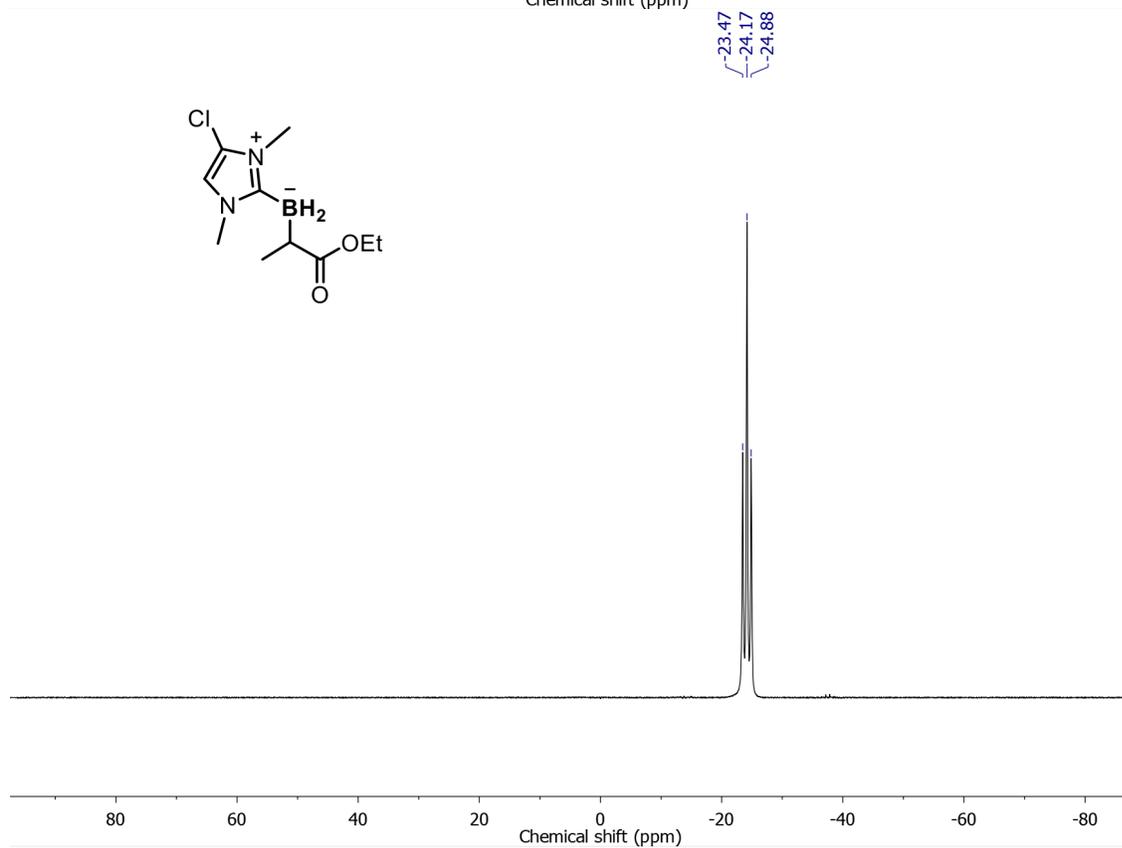
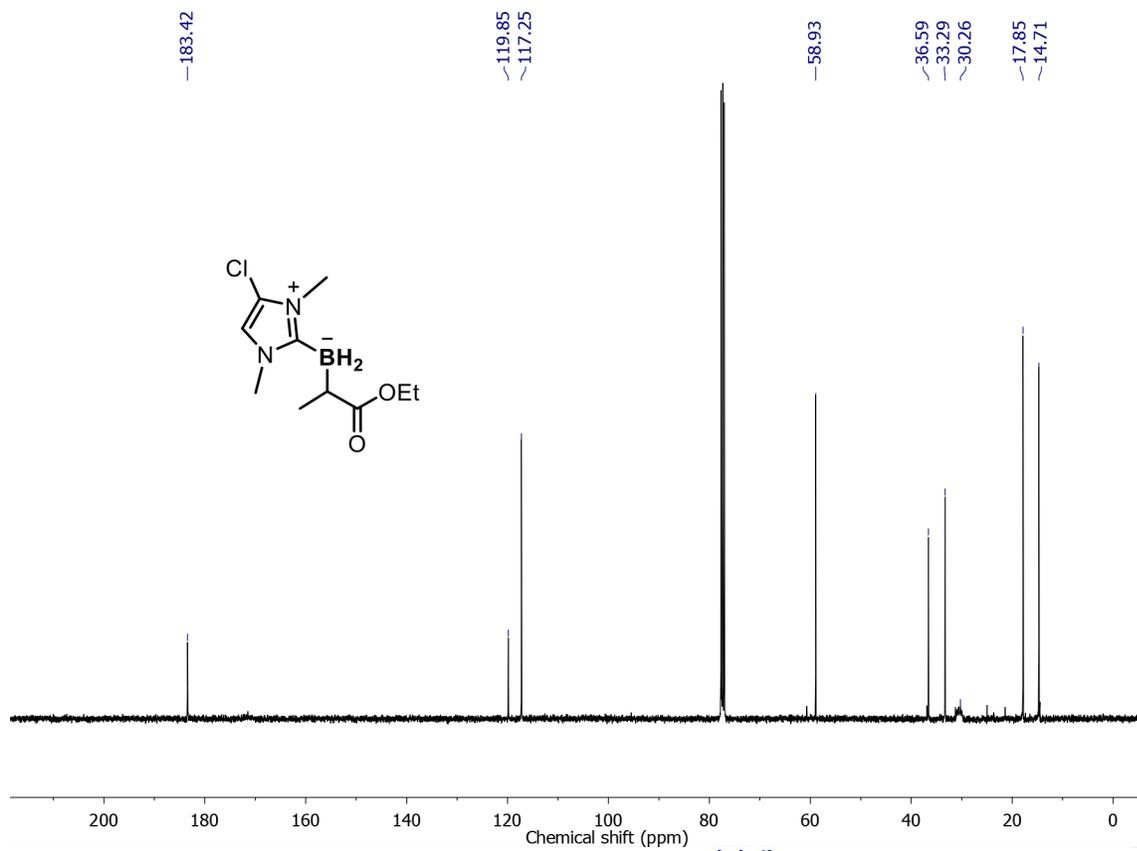


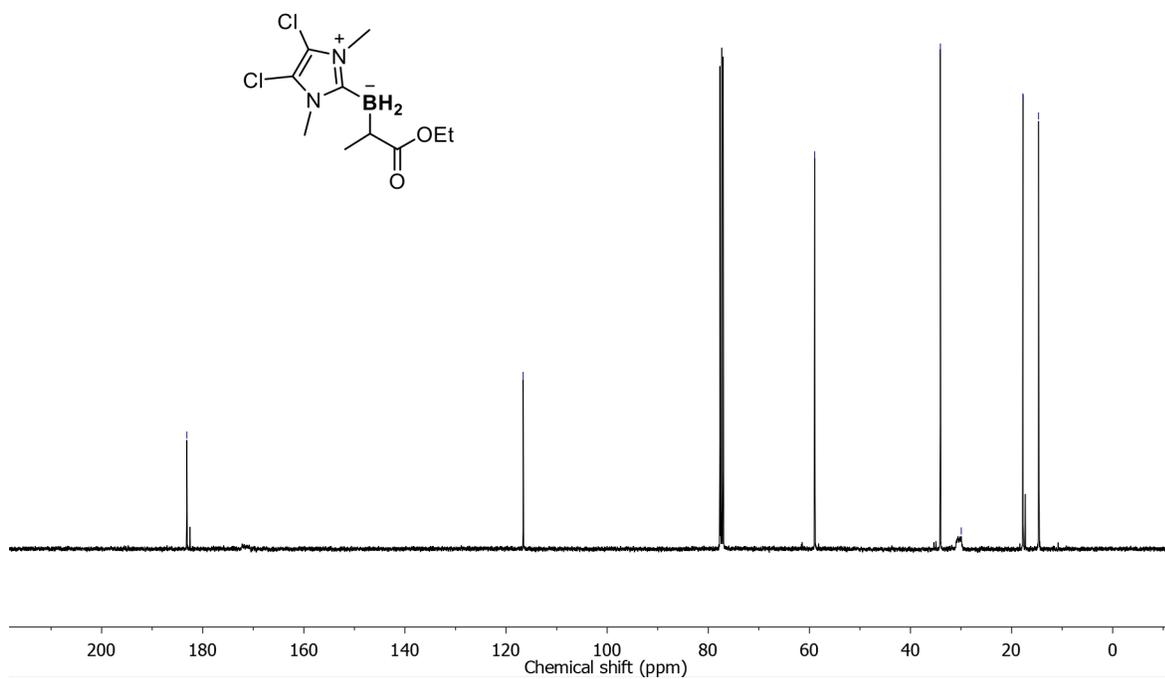
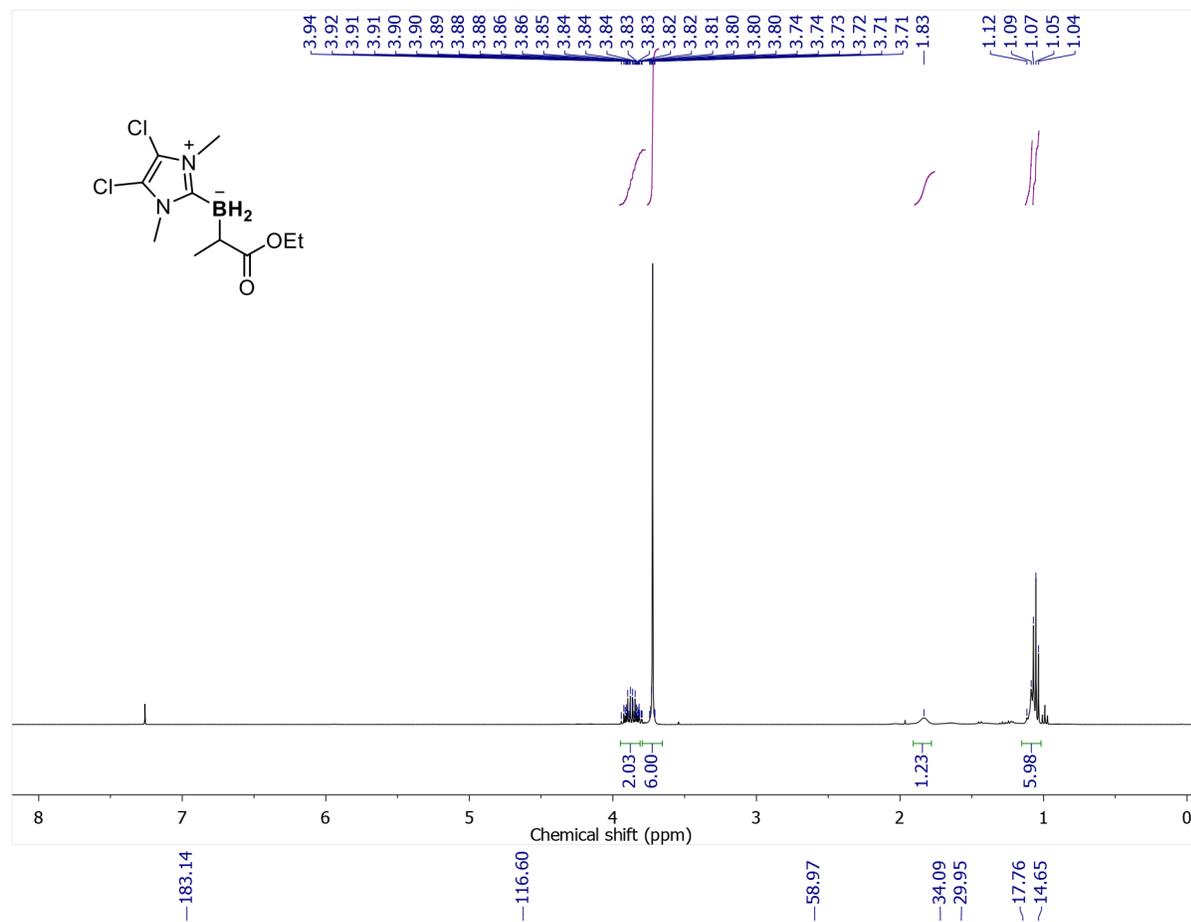


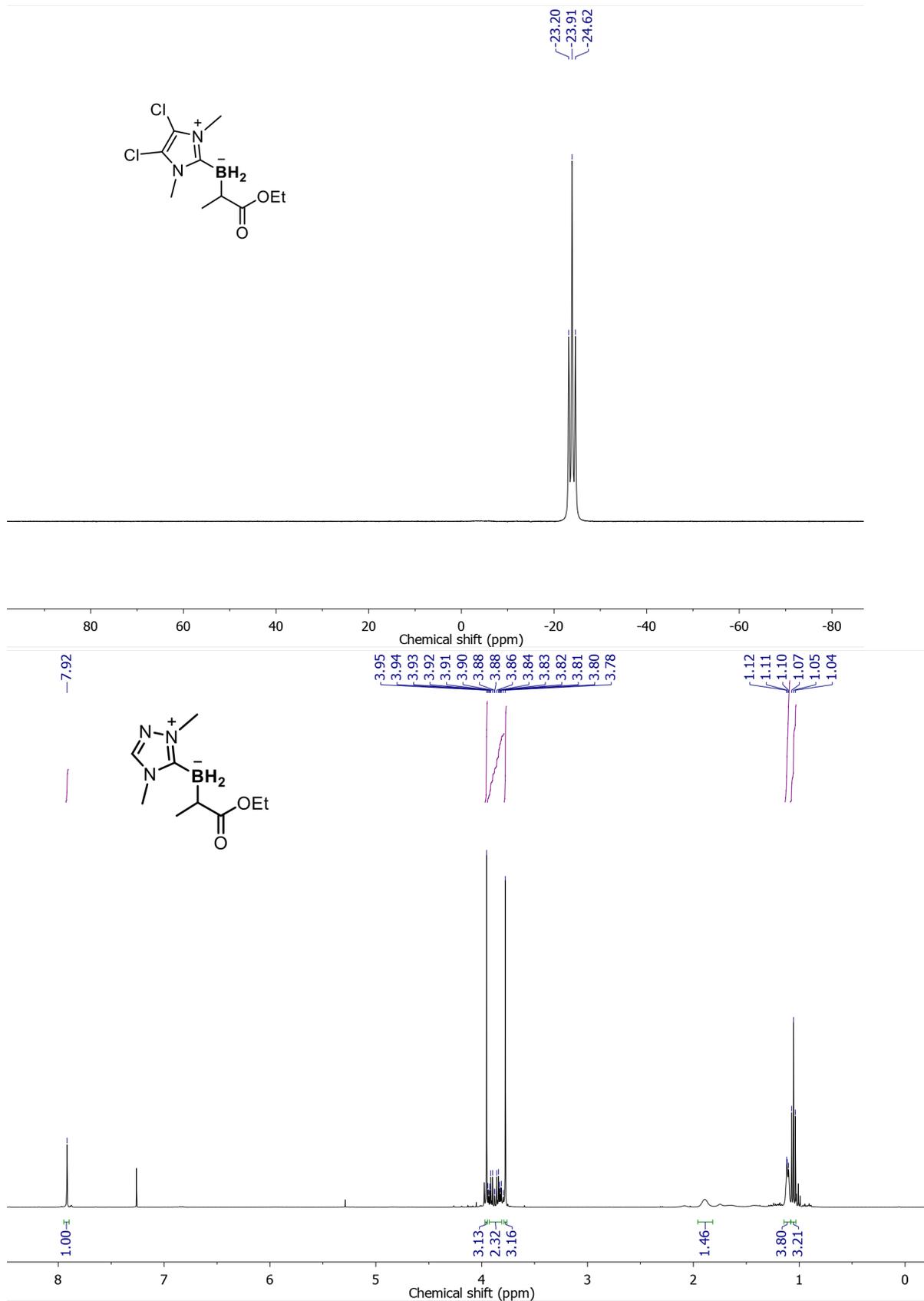


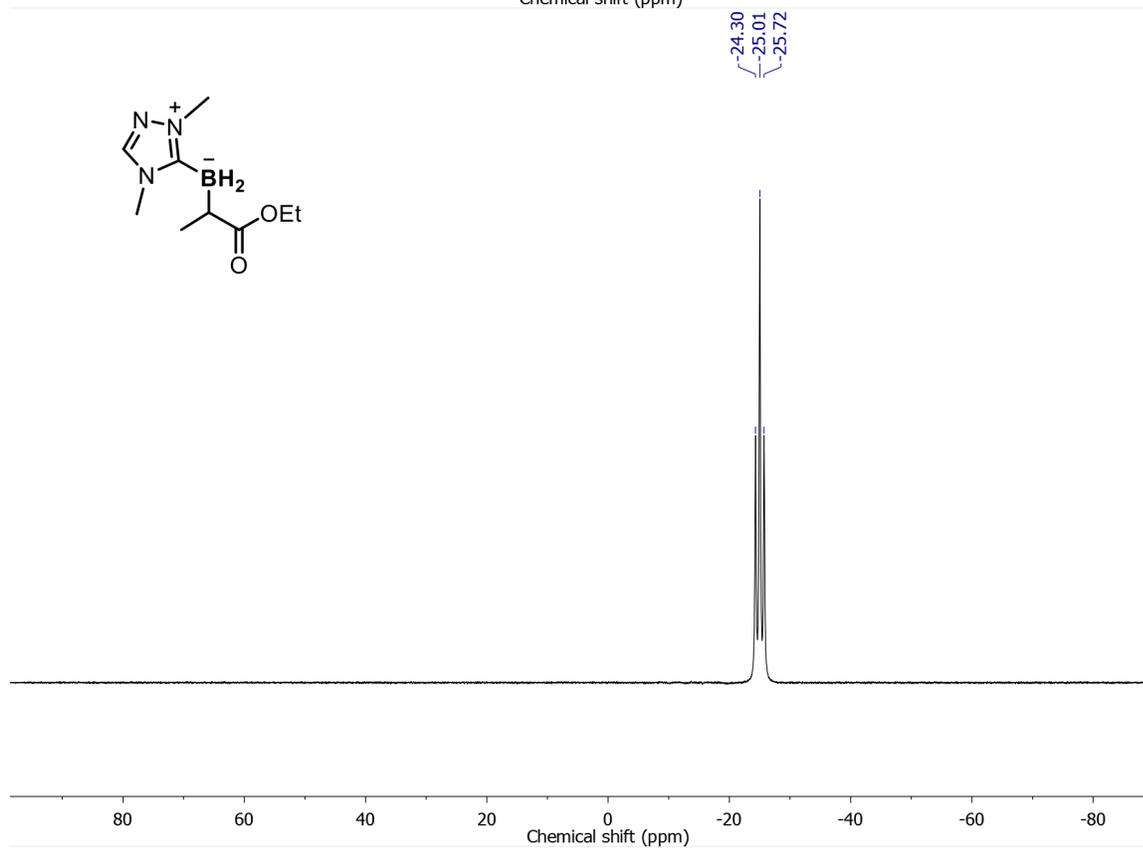
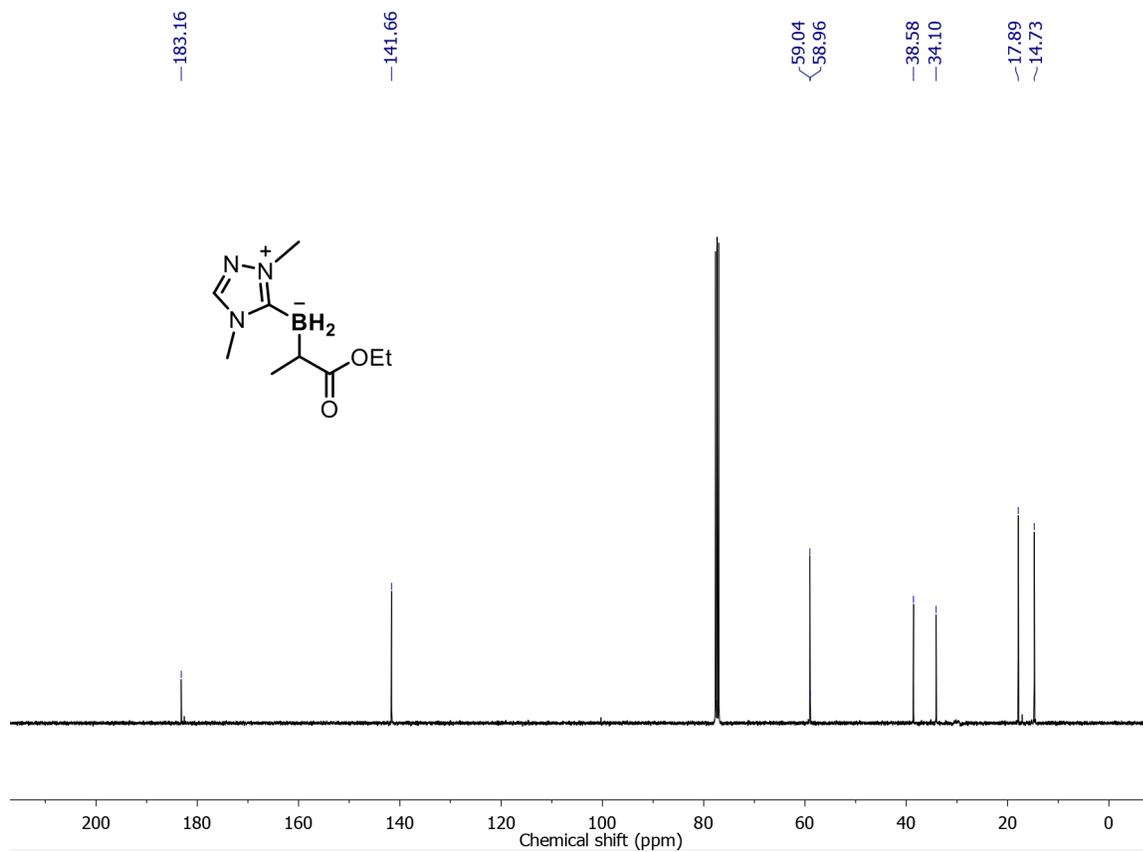


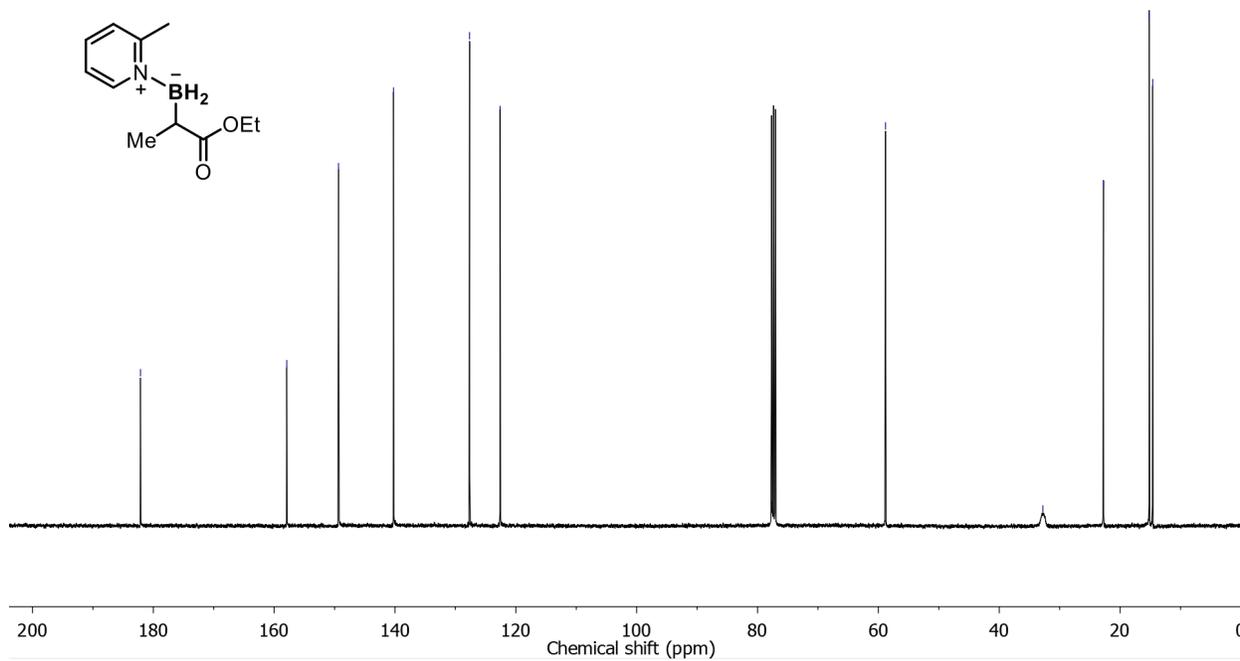
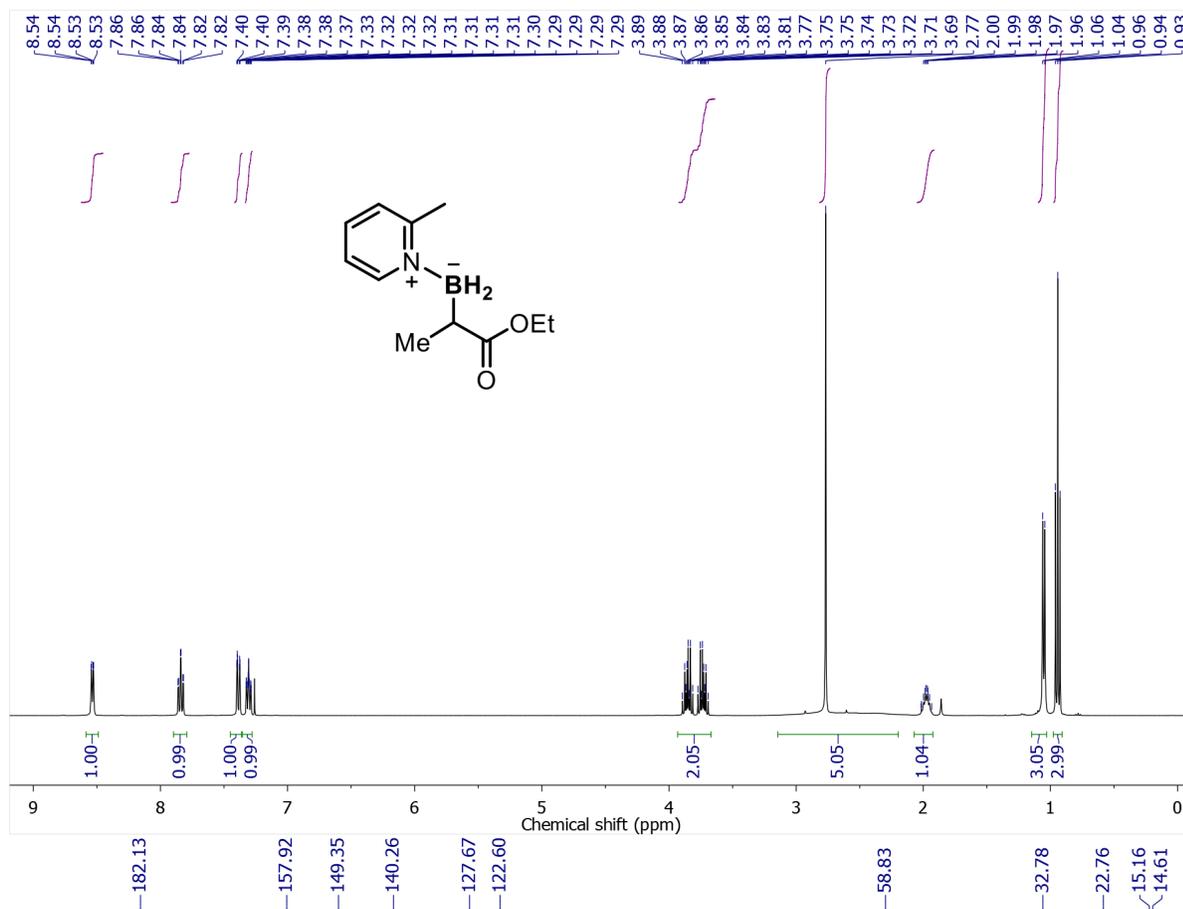


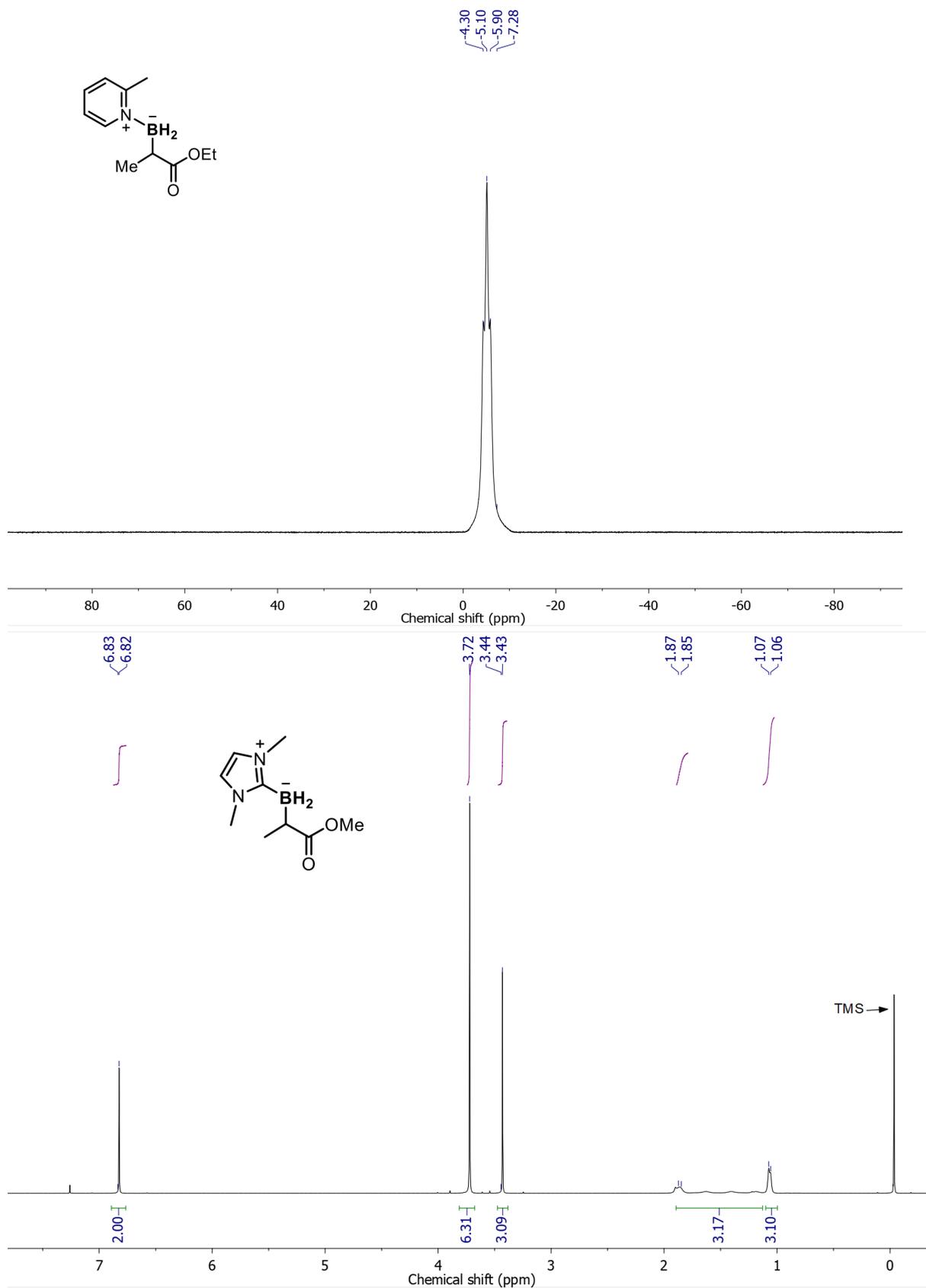


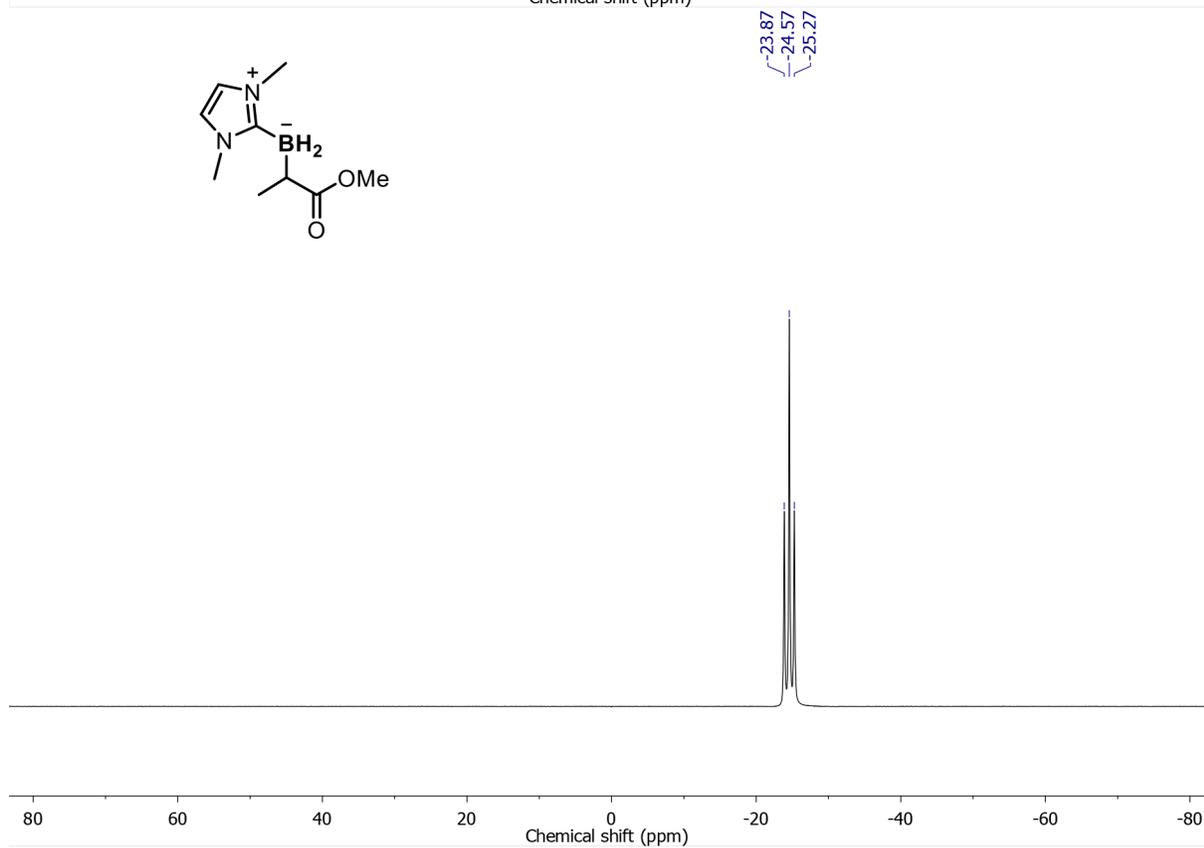
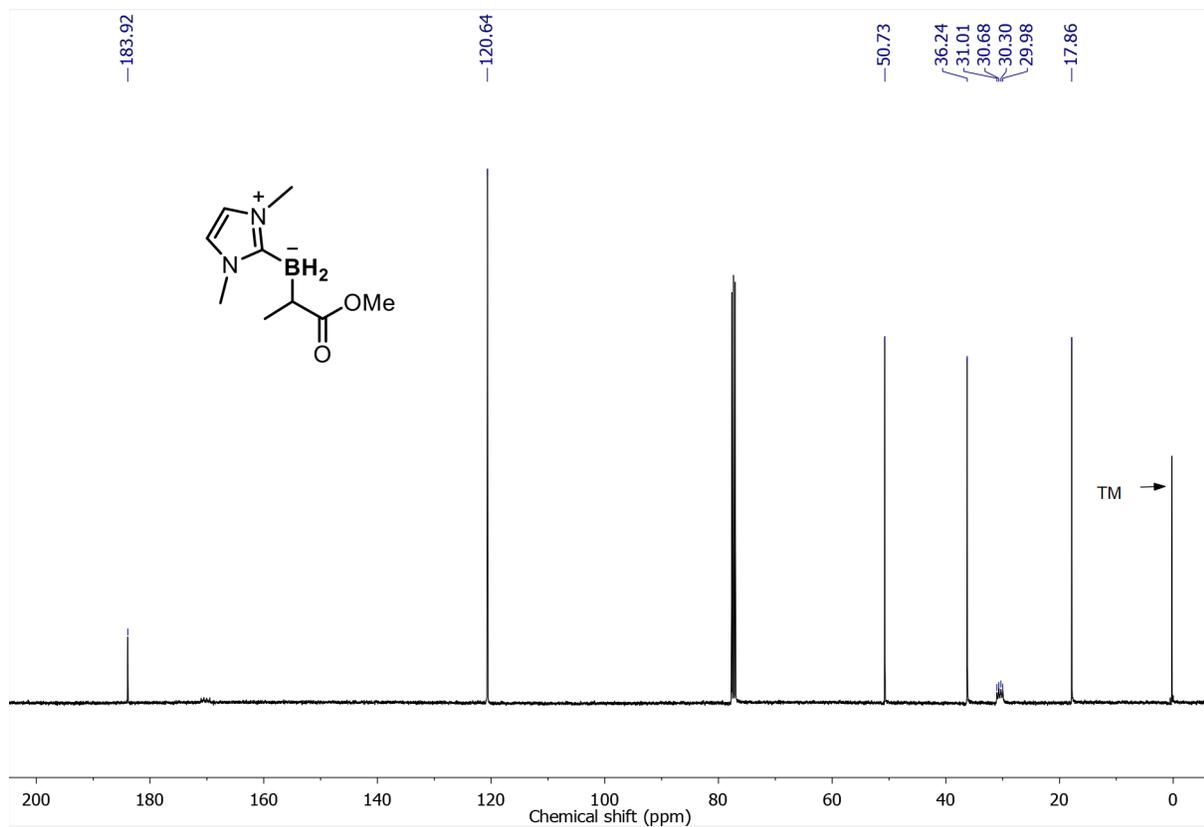


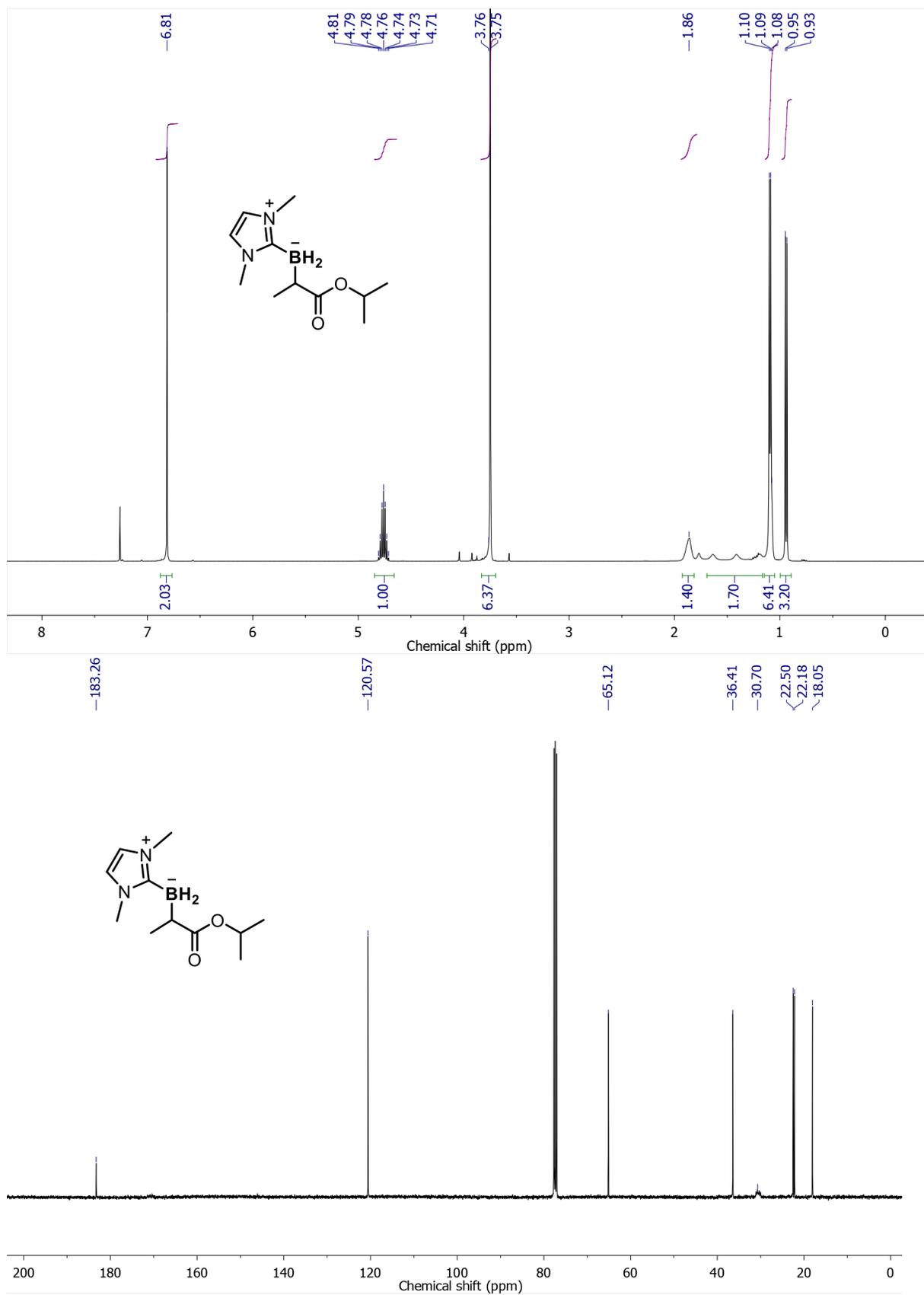


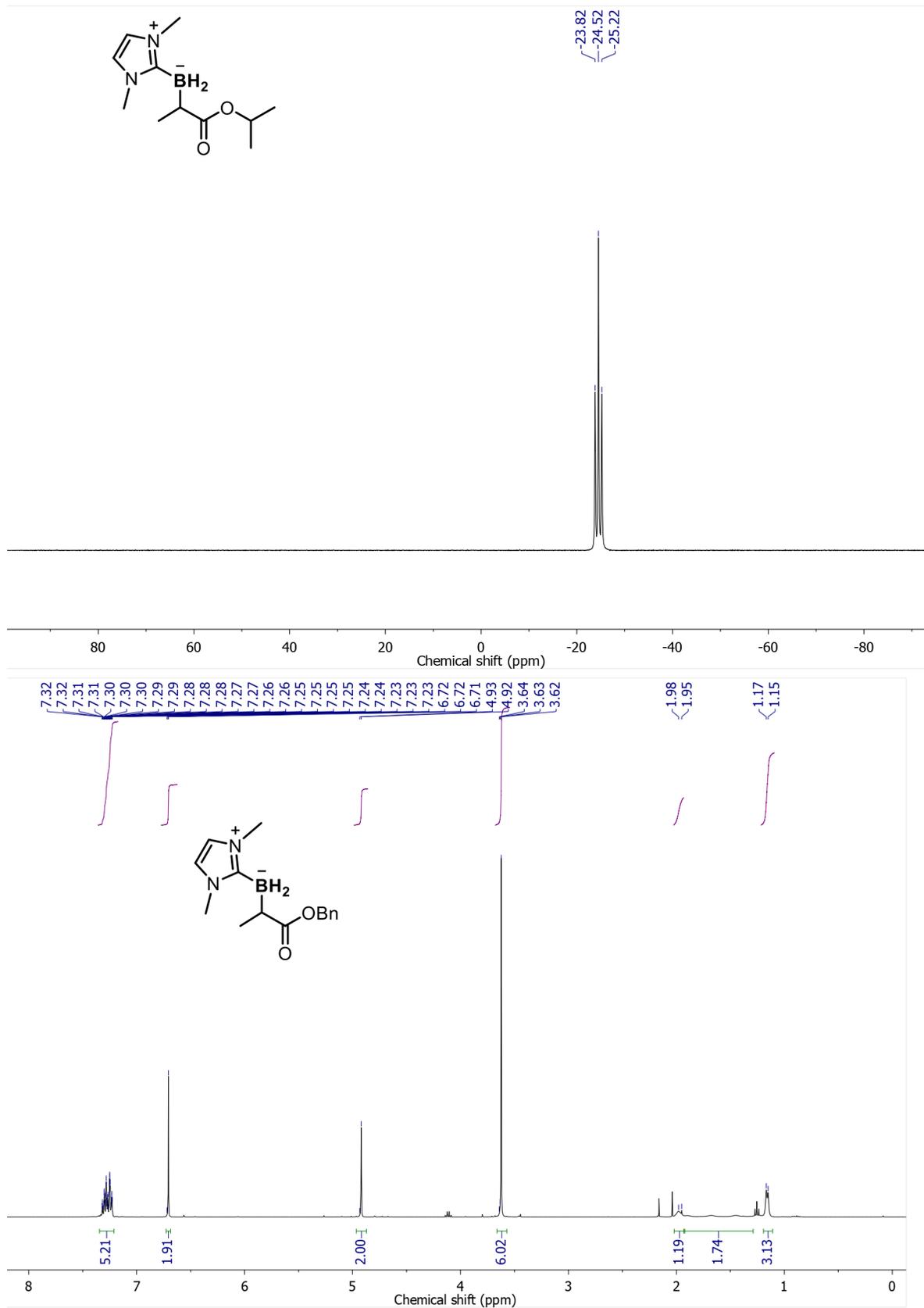


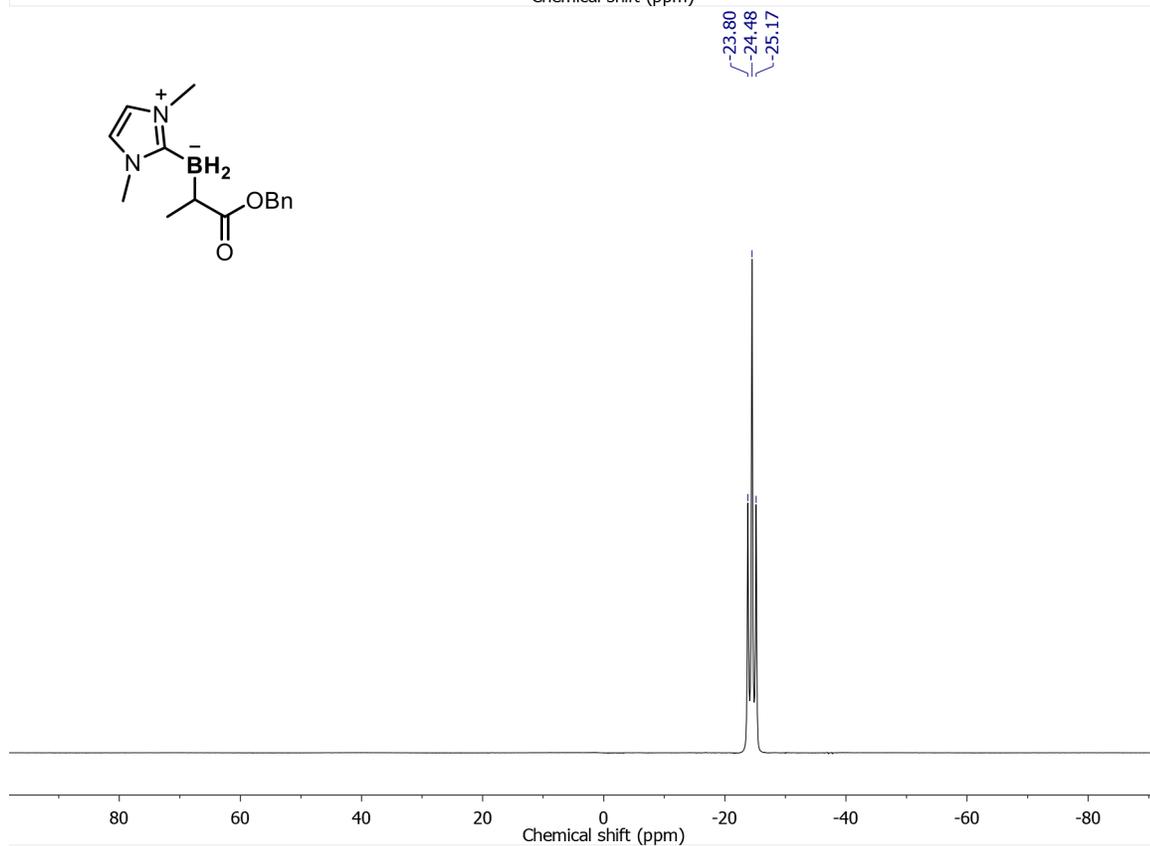
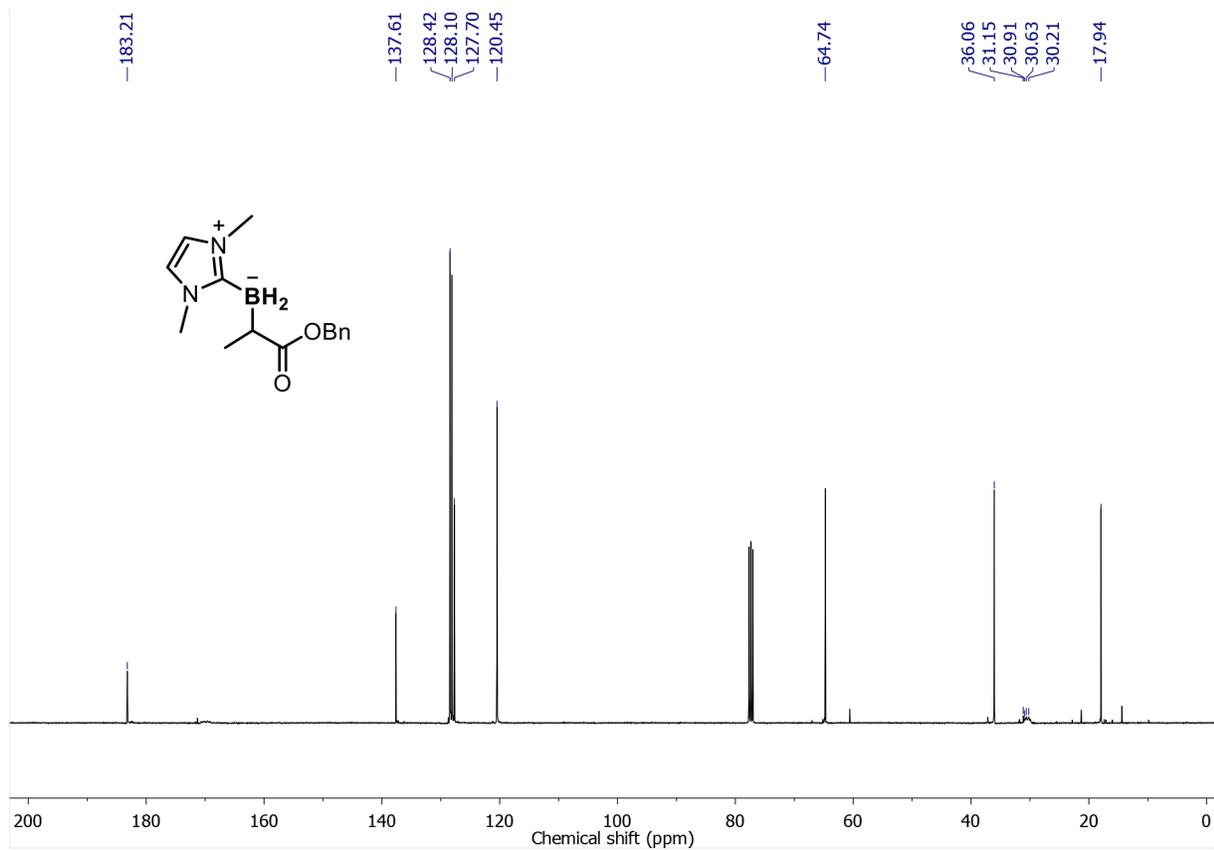


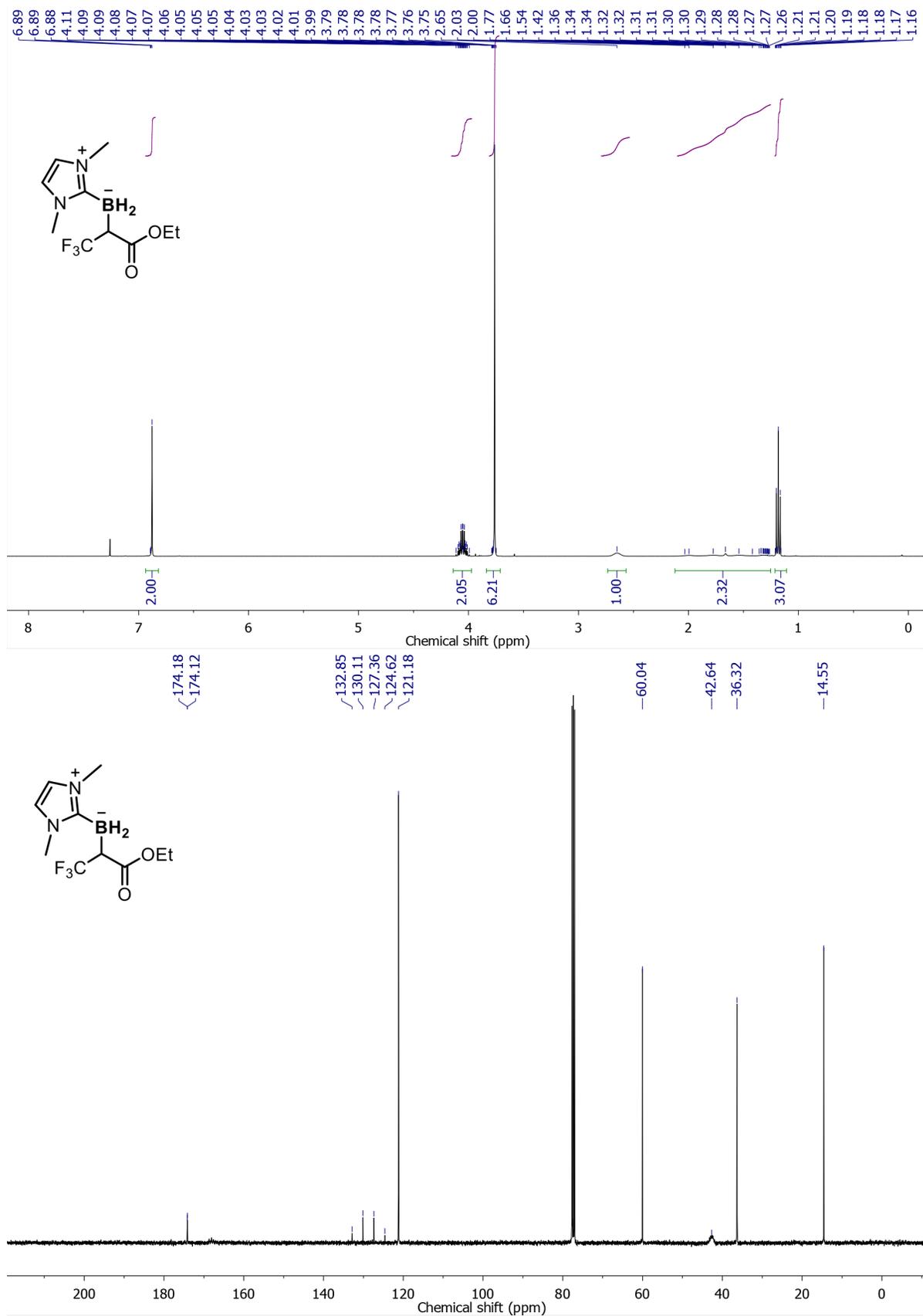


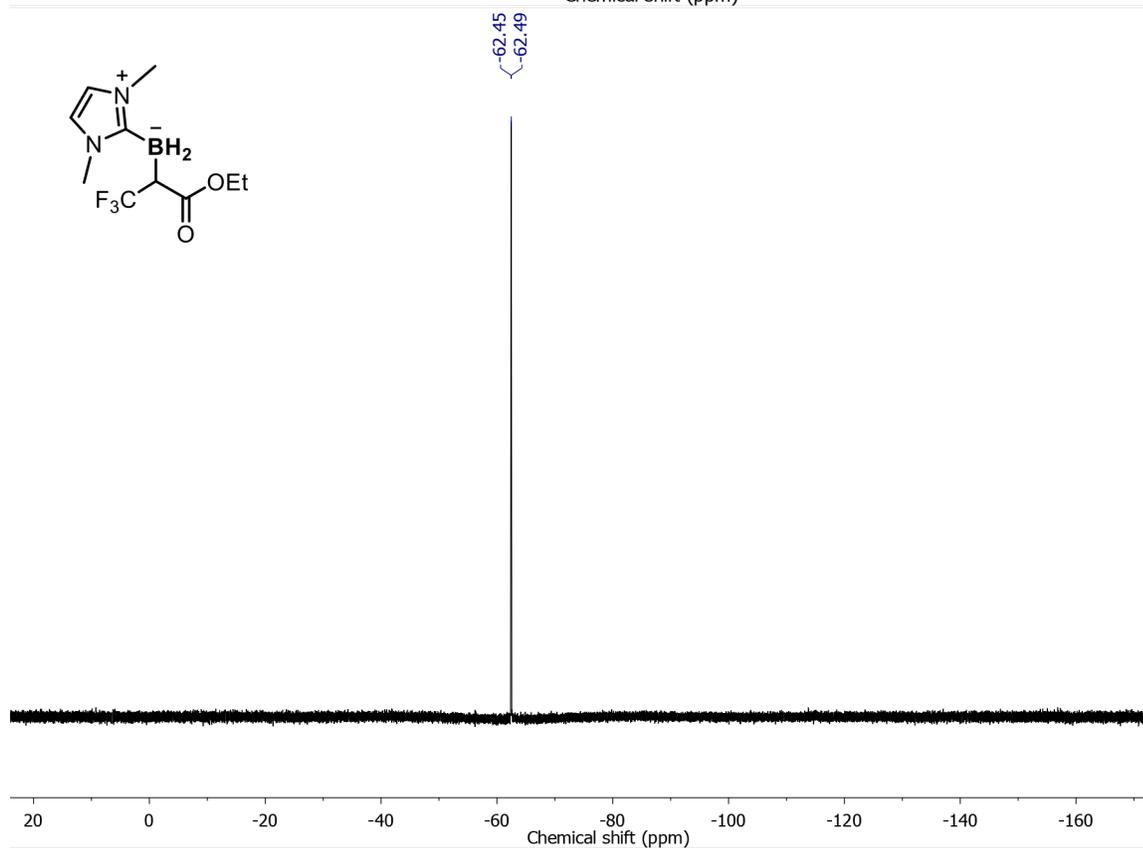
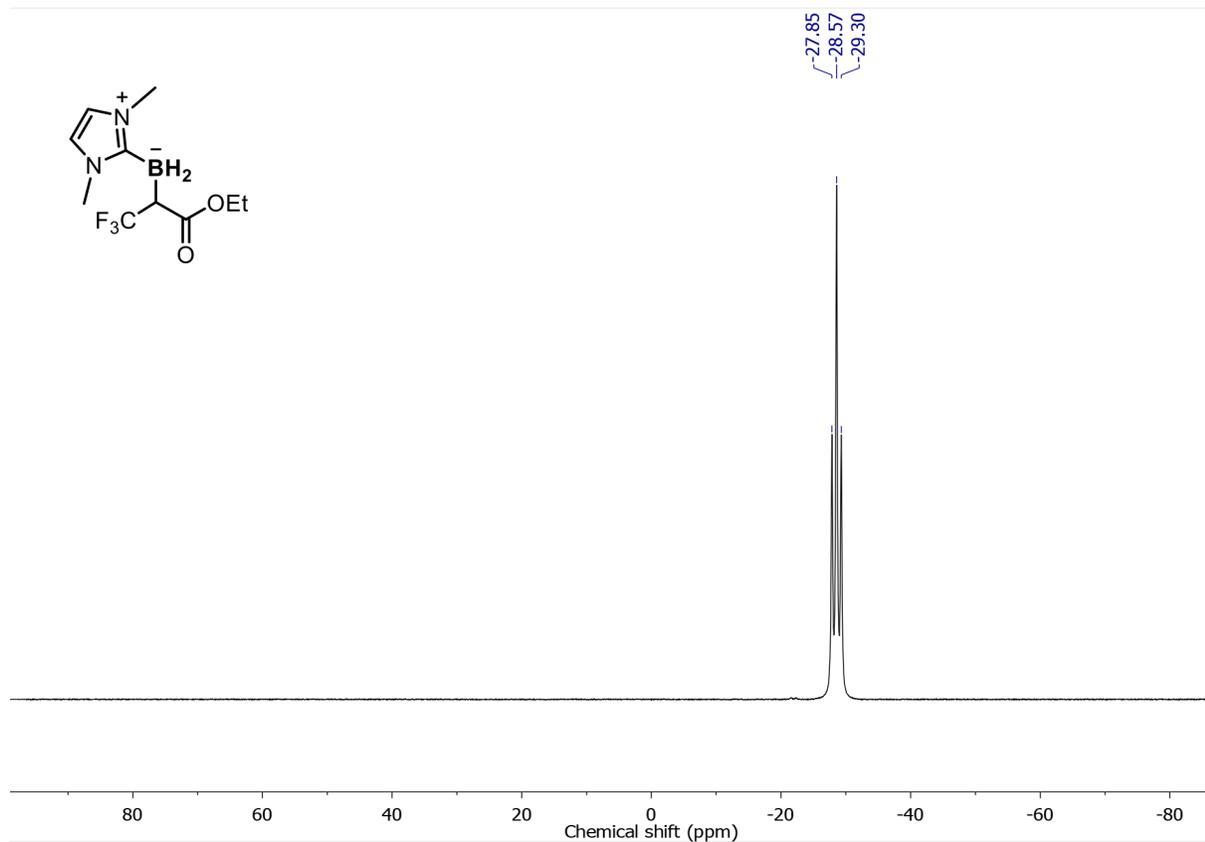


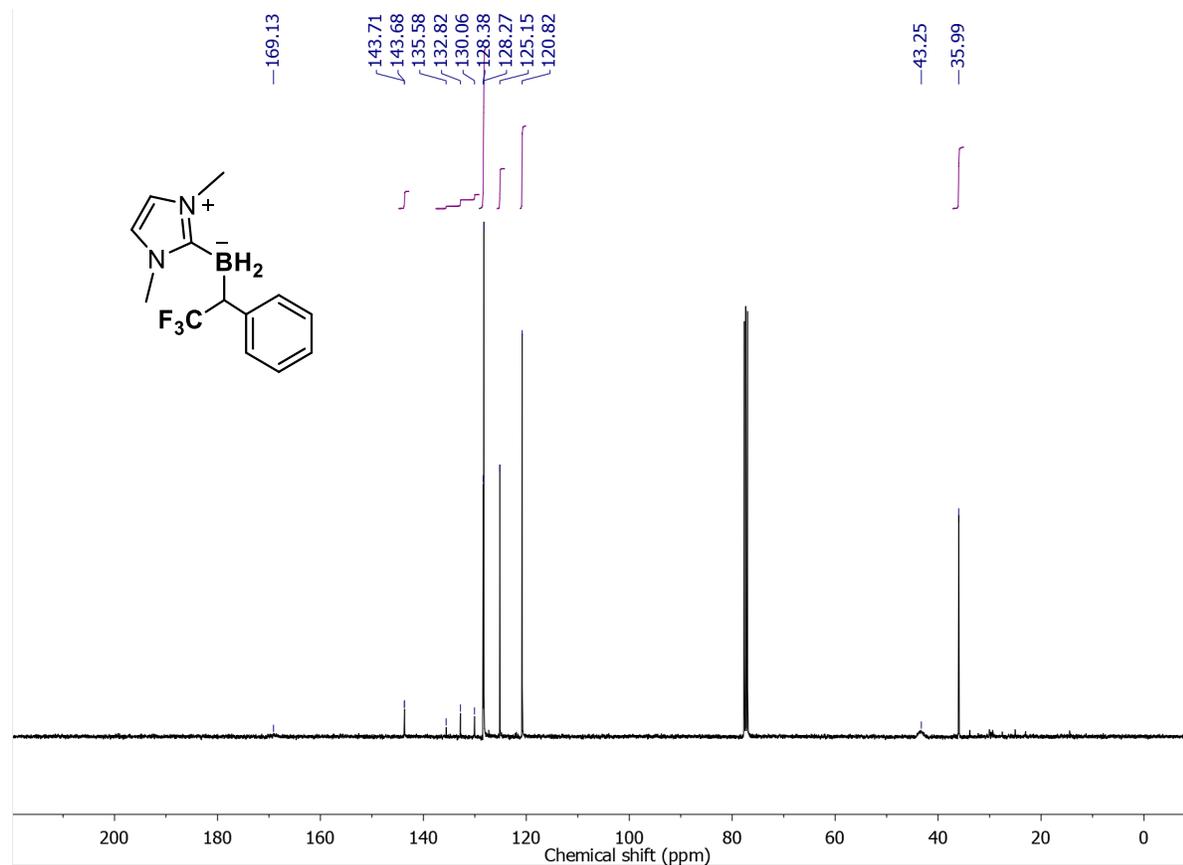
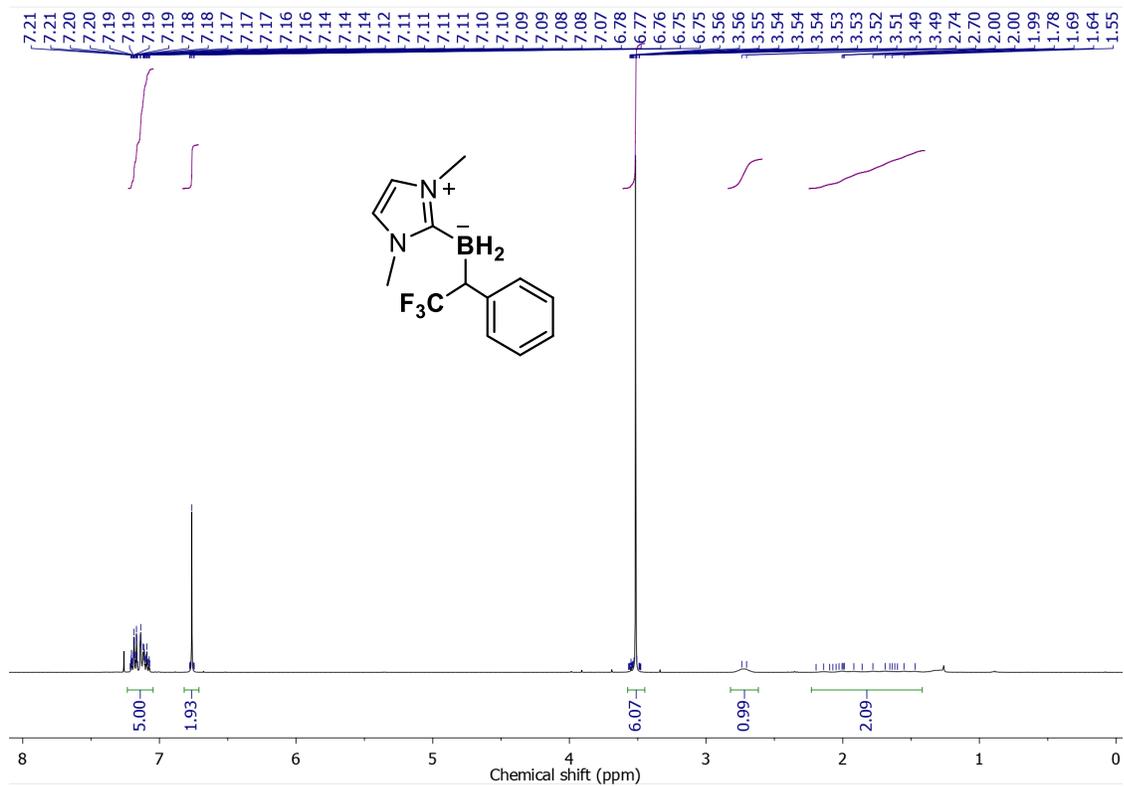


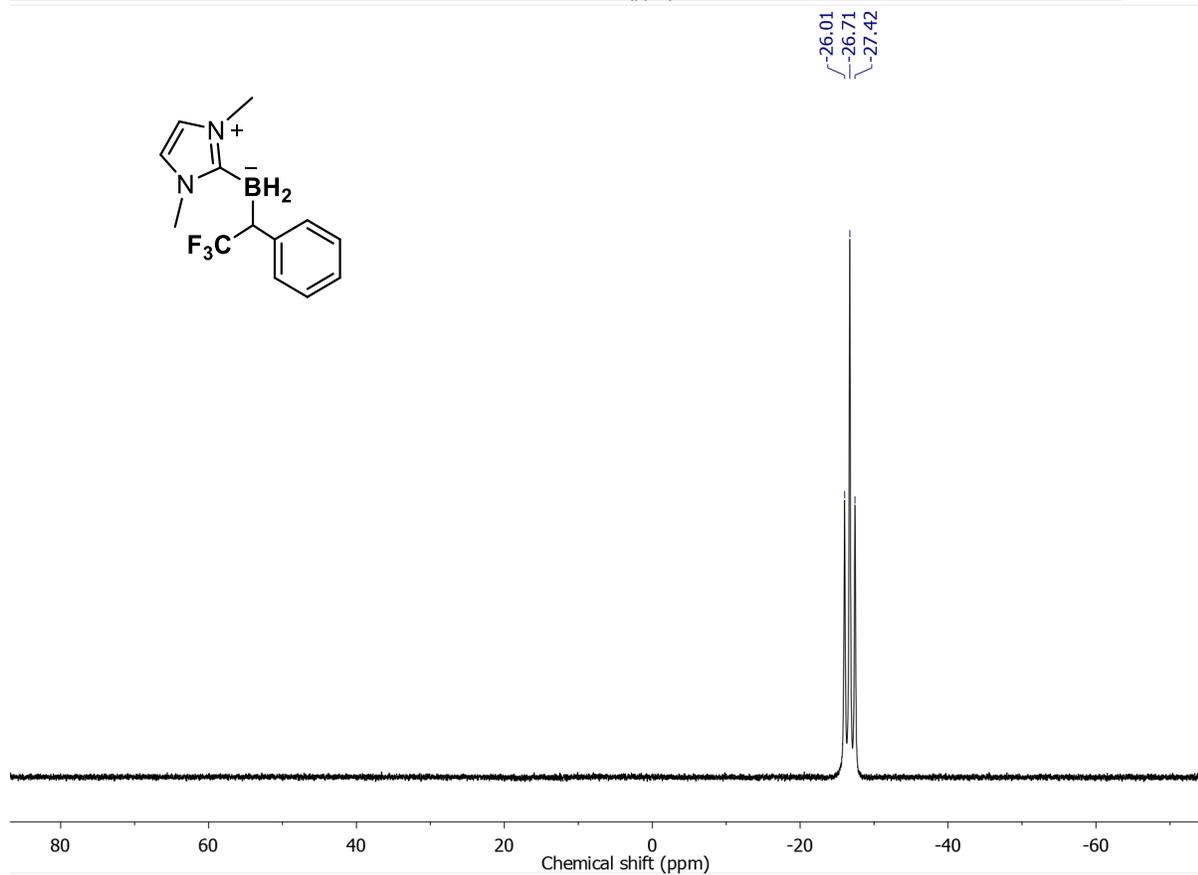
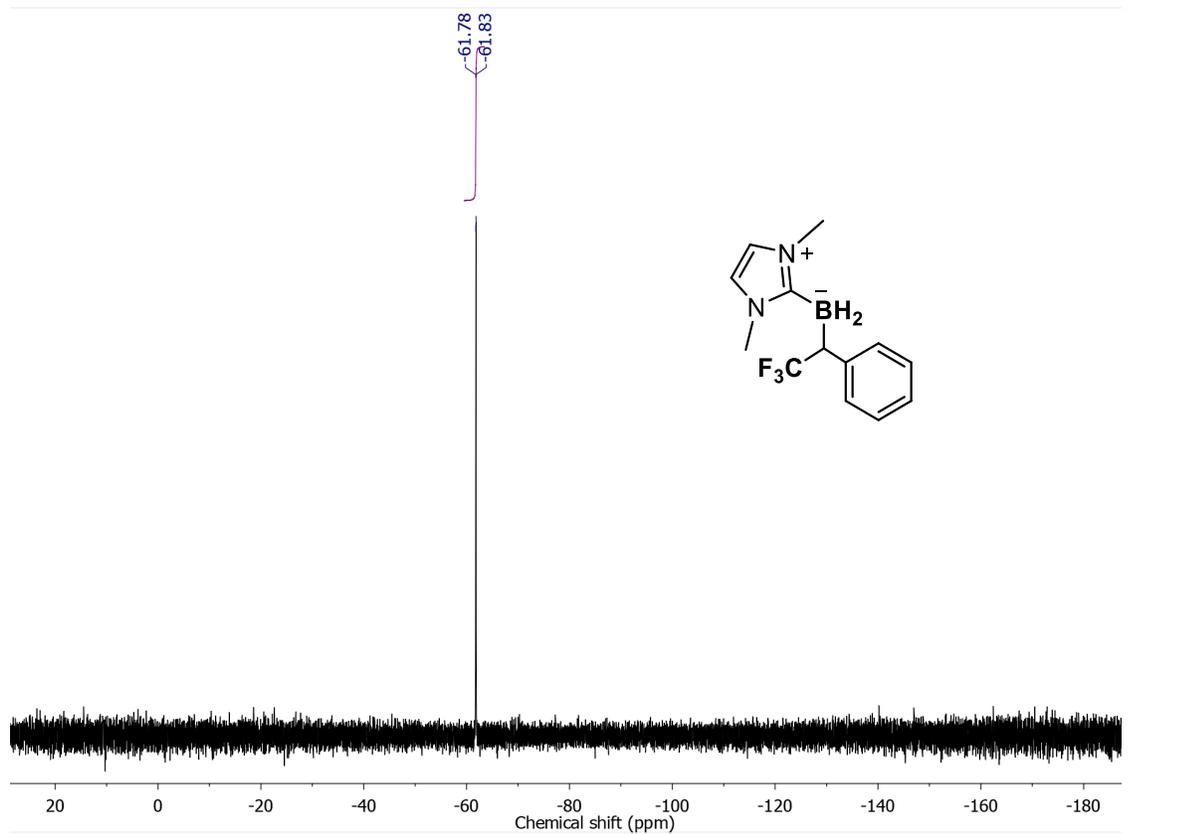


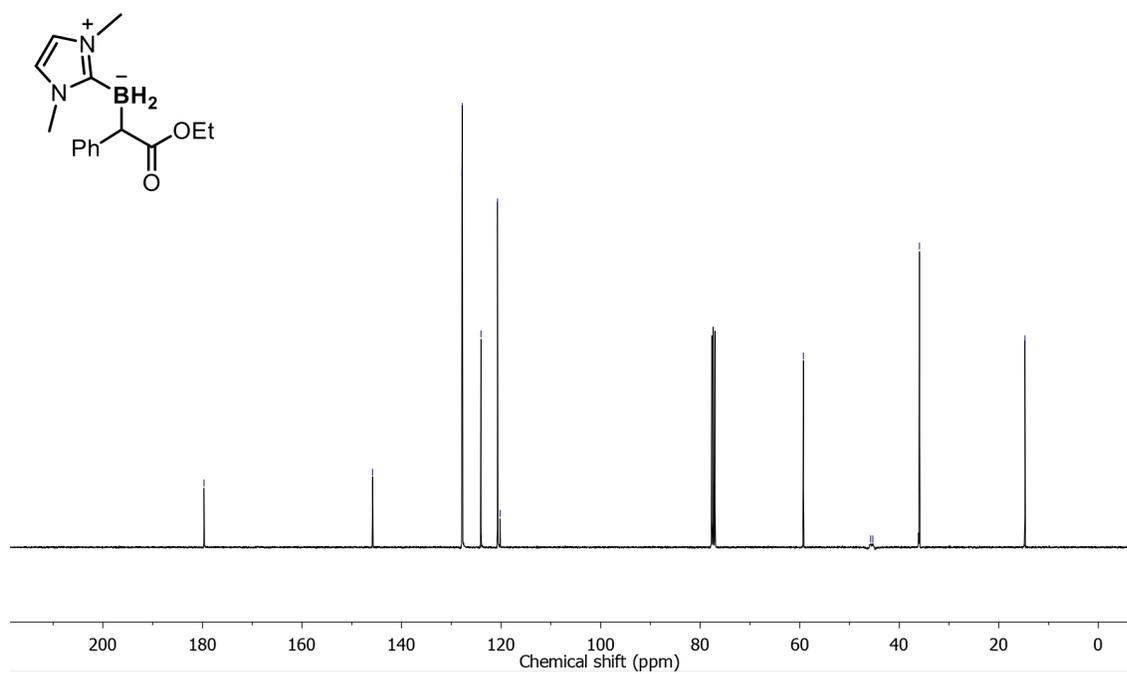
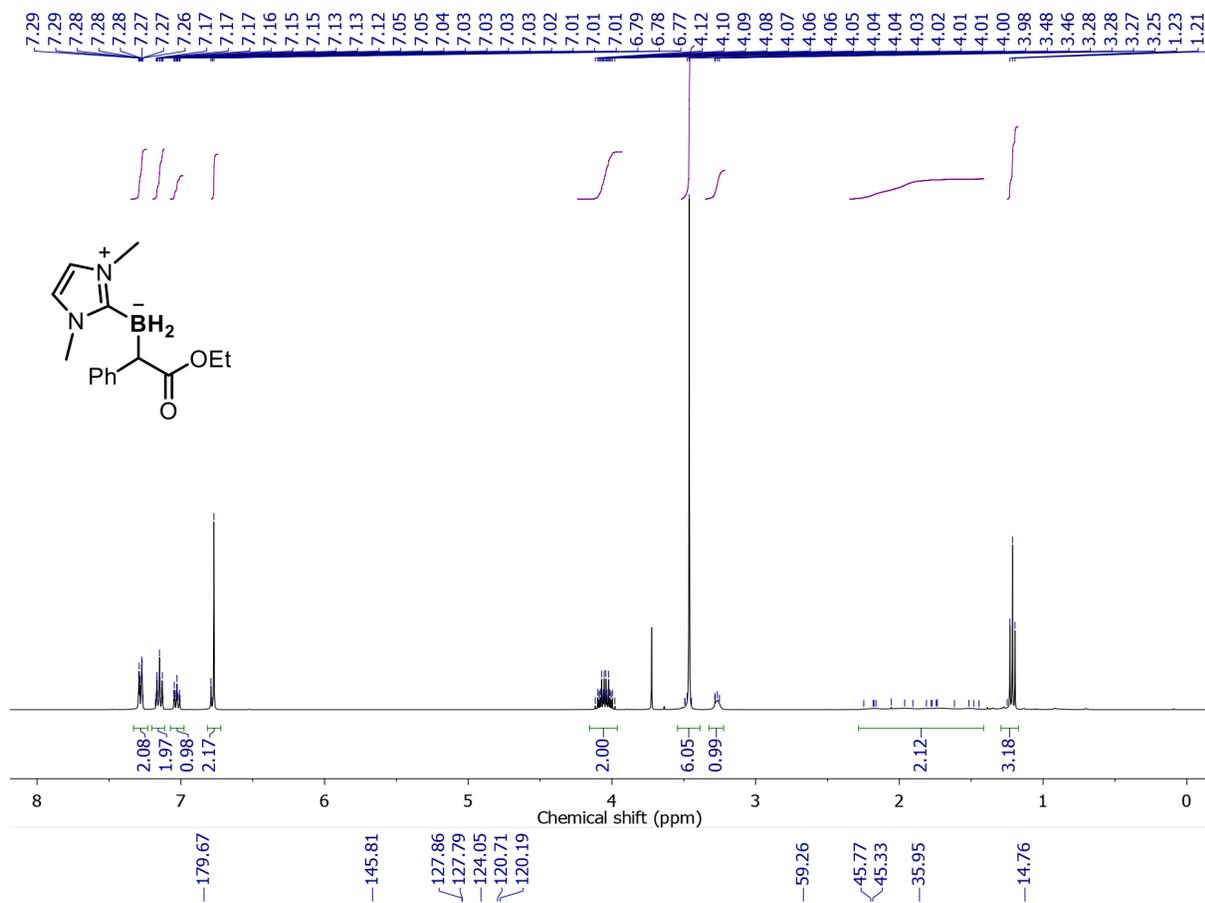


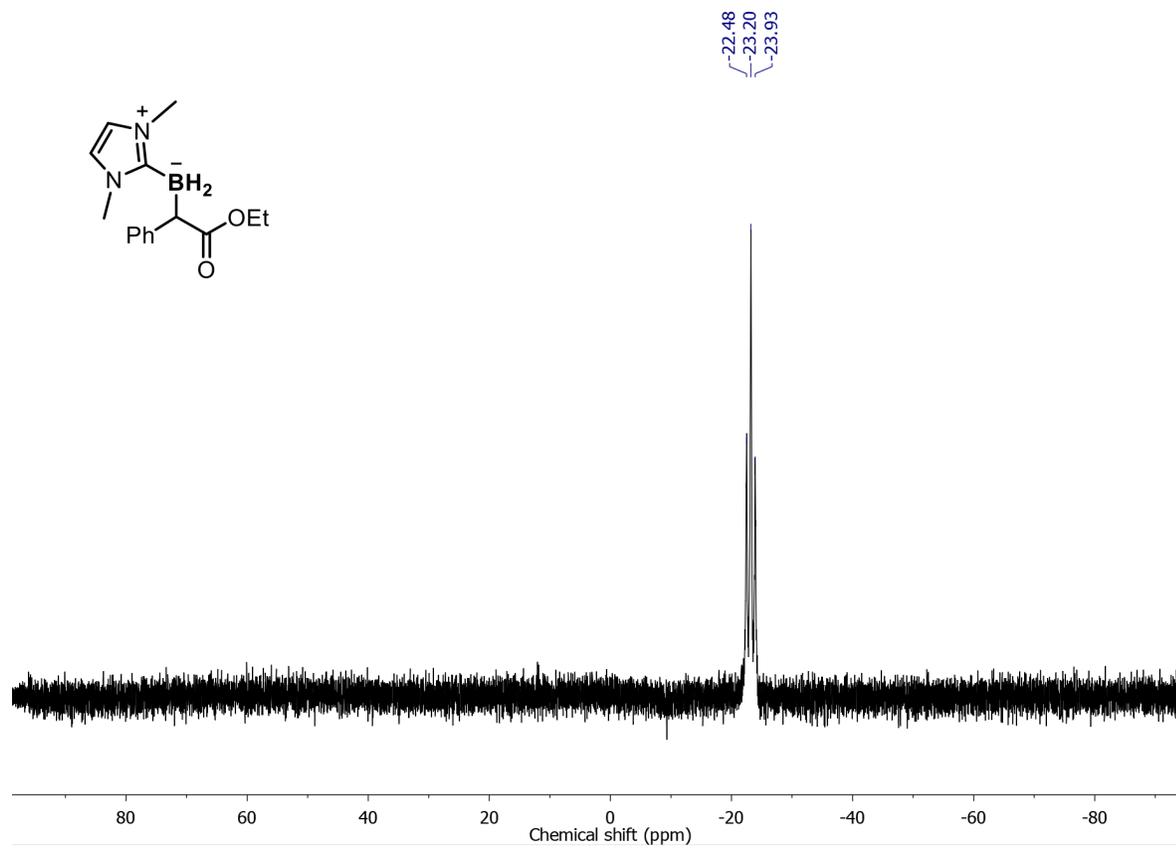












XI. X-ray Crystallography and the Assignments of Absolute Configuration

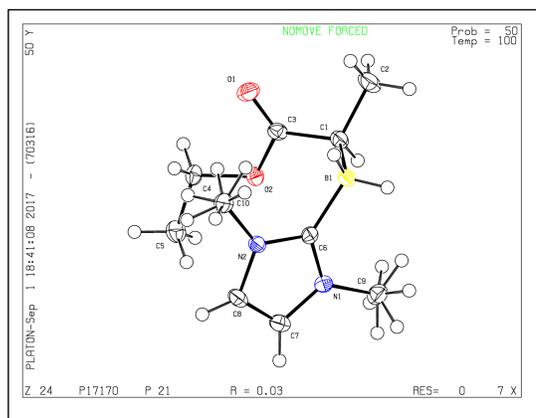
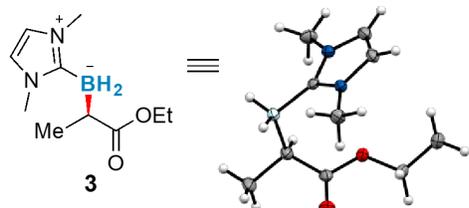
For products **3** and **18**, 10 mg of pure compound was dissolved in 0.5 mL ethylacetate and added to a 4 mL vial, which was then placed in a 20 mL vial containing 10 mL *n*-pentane. The 20 mL vial was capped, sealed with parafilm, and left undisturbed for three days at 4 °C. A suitable crystal was selected and mounted in a nylon loop in immersion oil. All measurements were made on a Bruker photon diffractometer with filtered Cu- $K\alpha$ radiation. Crystals of compound **12** were obtained via slow evaporation of an ethylacetate solution of **12** at room temperature.

Low-temperature diffraction data (ϕ - and ω -scans) were collected on a Bruker AXS D8 VENTURE KAPPA diffractometer coupled to a PHOTON 100 CMOS detector with Cu $K\alpha$ radiation ($\lambda = 1.54178 \text{ \AA}$) from an $I\mu S$ micro-source. The structure was solved by direct methods using SHELXS¹⁵ and refined against F^2 on all data by full-matrix least squares with SHELXL-2016¹⁶ using established refinement techniques¹⁷. All non-hydrogen atoms were refined anisotropically. Unless otherwise noted, all hydrogen atoms were included into the model at geometrically calculated positions and refined using a riding model. The isotropic displacement parameters of all hydrogen atoms were fixed to 1.2 times the U value of the atoms they are linked to (1.5 times for methyl groups). Compound **18** (sample No. P17253) crystallizes in the monoclinic space group $P2_1$ with two molecules in the asymmetric unit. The coordinates for the hydrogen atoms bound to B1 and B2 were located in the difference Fourier synthesis and refined semi-freely with the help of a restraint on the B–H distance (1.12(4) Å). The crystal was refined as a two-component twin.

The absolute configurations of compounds **3**, **12** and **18** were established by anomalous-dispersion effects using Cu $K\alpha$ radiation ($\lambda = 1.54178 \text{ \AA}$). For P17170 (compound **3**), the Flack x parameter of 0.02(7) was determined using 1062 quotients $[(I^+) - (I^-)] / [(I^+) + (I^-)]$. For P17207 (compound **18**), the Flack x parameter of 0.08(3) was determined using 2543 quotients $[(I^+) - (I^-)] / [(I^+) + (I^-)]$ and the Hooft y is 0.06(2). For P17253 (a two component twin, compound **12**), the Flack x parameter of 0.07(10) was determined using 2069 quotients $[(I^+) - (I^-)] / [(I^+) + (I^-)]$, the Hooft y is 0.16(9), and the PLATON P3 is 0.997. The Flack and van Hooft parameters are measures of the confidence of the absolute structure determination (zero (within several estimated standard deviation) for correct enantiomer, one for incorrect, intermediate for racemic twinning)^{18,19}.

PLATON version of 13/08/2017; check.def file version of 27/07/2017

Datablock P17170 - ellipsoid plot

Crystal data and structure refinement for organoborane **3**:

Identification code	P17170	
Empirical formula	C ₁₀ H ₁₉ B N ₂ O ₂	
Formula weight	210.08	
Temperature	100(2) K	
Wavelength	1.54178 Å	
Crystal system	Monoclinic	
Space group	P2 ₁	
Unit cell dimensions	a = 7.5635(4) Å	a = 90°.
	b = 9.4499(5) Å	b = 98.612(2)°.
	c = 8.3935(4) Å	g = 90°.
Volume	593.16(5) Å ³	
Z	2	
Density (calculated)	1.176 Mg/m ³	
Absorption coefficient	0.643 mm ⁻¹	
F(000)	228	
Crystal size	0.150 x 0.150 x 0.050 mm ³	
Theta range for data collection	5.330 to 74.490°.	
Index ranges	-9<=h<=9, -11<=k<=11, -10<=l<=9	
Reflections collected	11161	
Independent reflections	2410 [R(int) = 0.0311]	
Completeness to theta = 67.679°	100.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	1.000 and 0.9210	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	2410 / 1 / 138	
Goodness-of-fit on F ²	1.075	
Final R indices [I>2sigma(I)]	R1 = 0.0285, wR2 = 0.0703	
R indices (all data)	R1 = 0.0297, wR2 = 0.0711	
Absolute structure parameter	0.02(7)	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.150 and -0.204 e.Å ⁻³	

Datablock: P17170

Bond precision: C-C = 0.0020 Å Wavelength=1.54178
 Cell: a=7.5635(4) b=9.4499(5) c=8.3935(4)
 alpha=90 beta=98.612(2) gamma=90
 Temperature: 100 K

	Calculated	Reported
Volume	593.16(5)	593.16(5)
Space group	P 21	P 21
Hall group	P 2yb	P 2yb
Moiety formula	C10 H19 B N2 O2	?
Sum formula	C10 H19 B N2 O2	C10 H19 B N2 O2
Mr	210.08	210.08
Dx, g cm ⁻³	1.176	1.176
Z	2	2
Mu (mm ⁻¹)	0.643	0.643
F000	228.0	228.0
F000'	228.66	
h, k, lmax	9, 11, 10	9, 11, 10
Nref	2425[1290]	2410
Tmin, Tmax	0.908, 0.968	0.921, 1.000
Tmin'	0.908	

Correction method= # Reported T Limits: Tmin=0.921 Tmax=1.000
 AbsCorr = MULTI-SCAN
 Data completeness= 1.87/0.99 Theta(max)= 74.490
 R(reflections)= 0.0285(2341) wR2(reflections)= 0.0711(2410)
 S = 1.075 Npar= 138

The following ALERTS were generated. Each ALERT has the format

[test-name_ALERT_alert-type_alert-level](#).

Click on the hyperlinks for more details of the test.

Alert level C

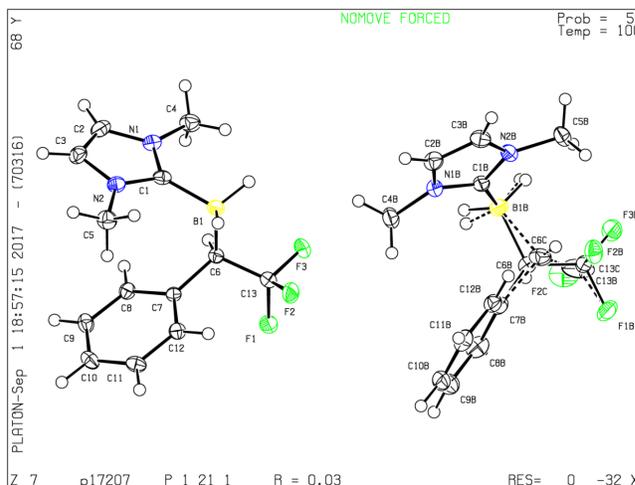
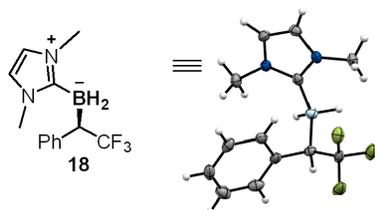
[PLAT413_ALERT_2_C](#) Short Inter XH3 .. XHn H2A .. H9F .. 2.09 Ang.
[PLAT978_ALERT_2_C](#) Number C-C Bonds with Positive Residual Density. 0 Info

Alert level G

[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H9A is Constrained at 0.5 Check
 And 11 other PLAT300 Alerts

[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H9B is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H9C is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H9D is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H9E is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H9F is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H10A is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H10B is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H10C is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H10D is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H10E is Constrained at 0.5 Check
[PLAT300_ALERT_4_G](#) Atom Site Occupancy of H10F is Constrained at 0.5 Check

[PLAT791_ALERT_4_G](#) The Model has Chirality at C1 (Chiral SPGR) R Verify
[PLAT912_ALERT_4_G](#) Missing # of FCF Reflections Above STh/L= 0.600 2 Note



Crystal data and structure refinement for organoborane **18**:

Identification code	p17207
Empirical formula	C ₁₃ H ₁₆ B F ₃ N ₂
Formula weight	268.09
Temperature	100 K
Wavelength	1.54178 Å
Crystal system	Monoclinic
Space group	P 1 21 1
Unit cell dimensions	a = 5.6677(7) Å a = 90° b = 15.3928(16) Å b = 98.957(6)° c = 15.4429(18) Å g = 90°
Volume	1330.8(3) Å ³
Z	4
Density (calculated)	1.338 Mg/m ³
Absorption coefficient	0.920 mm ⁻¹
F(000)	560
Crystal size	0.21 x 0.18 x 0.05 mm ³
Theta range for data collection	2.897 to 78.467°.
Index ranges	-7 <= h <= 7, -19 <= k <= 19, -19 <= l <= 19
Reflections collected	53323
Independent reflections	5657 [R(int) = 0.0403]
Completeness to theta = 67.000°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.0000 and 0.8917
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	5657 / 1 / 375
Goodness-of-fit on F ²	1.058
Final R indices [I > 2sigma(I)]	R1 = 0.0307, wR2 = 0.0757
R indices (all data)	R1 = 0.0318, wR2 = 0.0763
Absolute structure parameter	0.08(3)
Extinction coefficient	n/a
Largest diff. peak and hole	0.150 and -0.236 e.Å ⁻³

Datablock: p17207

Bond precision:	C-C = 0.0031 Å	Wavelength=1.54178
Cell:	a=5.6677(7) b=15.3928(16) c=15.4429(18)	
	alpha=90 beta=98.957(6) gamma=90	
Temperature:	100 K	
	Calculated	Reported
Volume	1330.8(3)	1330.8(3)
Space group	P 21	P 1 21 1
Hall group	P 2yb	P 2yb
Moiety formula	C13 H16 B F3 N2	C13 H16 B F3 N2
Sum formula	C13 H16 B F3 N2	C13 H16 B F3 N2
Mr	268.09	268.09
Dx, g cm ⁻³	1.338	1.338
Z	4	4
Mu (mm ⁻¹)	0.920	0.920
F000	560.0	560.0
F000'	562.00	
h,k,lmax	7,19,19	7,19,19
Nref	5722[2971]	5657
Tmin,Tmax	0.824,0.955	0.892,1.000
Tmin'	0.824	
Correction method=	# Reported T Limits: Tmin=0.892 Tmax=1.000	
AbsCorr =	MULTI-SCAN	
Data completeness=	1.90/0.99 Theta(max)= 78.467	
R(reflections)=	0.0307(5508) wR2(reflections)= 0.0763(5657)	
S =	1.058 Npar= 375	

The following ALERTS were generated. Each ALERT has the format
[test-name_ALERT_alert-type_alert-level](#).
 Click on the hyperlinks for more details of the test.

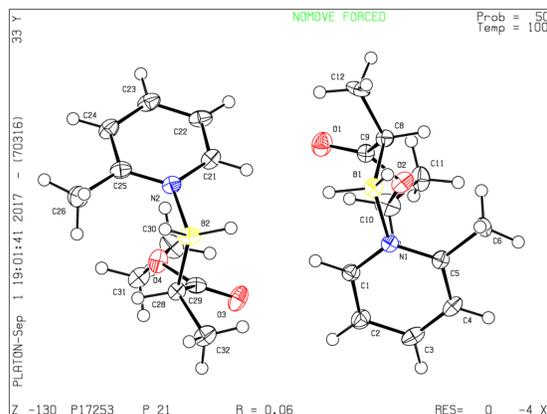
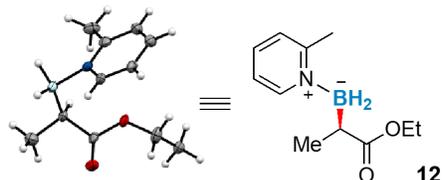
Alert level B

[PLAT410_ALERT_2_B](#) Short Intra H...H Contact H12B .. H6C . 1.84 Å.

Author Response: H6C is a 15% occupied disordered site. There is presumably some accomodation by the phenyl ring part of the time.

Alert level G

PLAT301_ALERT_3_G	Main Residue Disorder(Resd 1)..	16% Note
PLAT720_ALERT_4_G	Number of Unusual/Non-Standard Labels	10 Note
PLAT791_ALERT_4_G	The Model has Chirality at C6 (Chiral SPGR)	R Verify
PLAT791_ALERT_4_G	The Model has Chirality at C6B (Chiral SPGR)	R Verify
PLAT912_ALERT_4_G	Missing # of FCF Reflections Above STh/L= 0.600	5 Note
PLAT978_ALERT_2_G	Number C-C Bonds with Positive Residual Density.	2 Info



Crystal data and structure refinement for organoborane **12**:

Identification code	P17253
Empirical formula	C ₁₁ H ₁₈ B N O ₂
Formula weight	207.07
Temperature	100(2) K
Wavelength	1.54178 Å
Crystal system	Monoclinic
Space group	P2 ₁
Unit cell dimensions	a = 8.0292(3) Å a = 90° b = 19.3346(8) Å b = 106.631(2)° c = 8.0293(3) Å g = 90°
Volume	1194.34(8) Å ³
Z	4
Density (calculated)	1.152 Mg/m ³
Absorption coefficient	0.611 mm ⁻¹
F(000)	448
Crystal size	0.250 x 0.200 x 0.100 mm ³
Theta range for data collection	2.285 to 74.659°
Index ranges	-9<=h<=10, -24<=k<=24, -10<=l<=10
Reflections collected	15764
Independent reflections	4783 [R(int) = 0.0592]
Completeness to theta = 67.679°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7538 and 0.6598
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	4783 / 5 / 290
Goodness-of-fit on F ²	1.109
Final R indices [I>2sigma(I)]	R1 = 0.0592, wR2 = 0.1453
R indices (all data)	R1 = 0.0624, wR2 = 0.1496
Absolute structure parameter	0.07(10)
Extinction coefficient	n/a
Largest diff. peak and hole	0.624 and -0.287 e.Å ⁻³

Datablock: P17253

Bond precision:	C-C = 0.0083 Å	Wavelength=1.54178
Cell:	a=8.0292(3) b=19.3346(8) c=8.0293(3)	
	alpha=90 beta=106.631(2) gamma=90	
Temperature:	100 K	
	Calculated	Reported
Volume	1194.34(8)	1194.34(8)
Space group	P 21	P 21
Hall group	P 2yb	P 2yb
Moiety formula	C11 H18 B N O2	?
Sum formula	C11 H18 B N O2	C11 H18 B N O2
Mr	207.07	207.07
Dx, g cm ⁻³	1.152	1.152
Z	4	4
Mu (mm ⁻¹)	0.611	0.611
F000	448.0	448.0
F000'	449.28	
h,k,lmax	10,24,10	10,24,10
Nref	4856[2503]	4783
Tmin,Tmax	0.864,0.941	0.660,0.754
Tmin'	0.858	
Correction method=	# Reported T Limits: Tmin=0.660 Tmax=0.754	
AbsCorr =	MULTI-SCAN	
Data completeness=	1.91/0.98 Theta(max)= 74.659	
R(reflections)=	0.0592(4582) wR2(reflections)= 0.1496(4783)	
S =	1.109 Npar= 290	

The following ALERTS were generated. Each ALERT has the format

[test-name_ALERT_alert-type_alert-level](#).

Click on the hyperlinks for more details of the test.

● Alert level C

[DIFMX02 ALERT 1 C](#) The maximum difference density is > 0.1*ZMAX*0.75
The relevant atom site should be identified.

PLAT094 ALERT 2 C	Ratio of Maximum / Minimum Residual Density	2.17 Report
PLAT097 ALERT 2 C	Large Reported Max. (Positive) Residual Density	0.62 eA-3
PLAT340 ALERT 3 C	Low Bond Precision on C-C Bonds	0.00831 Ang.
PLAT480 ALERT 4 C	Long H...A H-Bond Reported H1 .. O3 ..	2.61 Ang.
PLAT480 ALERT 4 C	Long H...A H-Bond Reported H2 .. O3 ..	2.62 Ang.
PLAT911 ALERT 3 C	Missing # FCF Refl Between THmin & STh/L= 0.600	4 Report

● Alert level G

PLAT002 ALERT 2 G	Number of Distance or Angle Restraints on AtSite	6 Note
PLAT172 ALERT 4 G	The CIF-Embedded .res File Contains DFIX Records	1 Report
PLAT720 ALERT 4 G	Number of Unusual/Non-Standard Labels	4 Note
PLAT791 ALERT 4 G	The Model has Chirality at C8 (Chiral SPGR)	R Verify
PLAT791 ALERT 4 G	The Model has Chirality at C28 (Chiral SPGR)	R Verify
PLAT860 ALERT 3 G	Number of Least-Squares Restraints	5 Note
PLAT870 ALERT 4 G	ALERTS Related to Twinning Effects Suppressed ..	! Info
PLAT912 ALERT 4 G	Missing # of FCF Reflections Above STh/L= 0.600	9 Note
PLAT931 ALERT 5 G	Found Twin Law (1 0 -1)[] Estimated BASF	0.49 Check

XII. Supplemental References

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